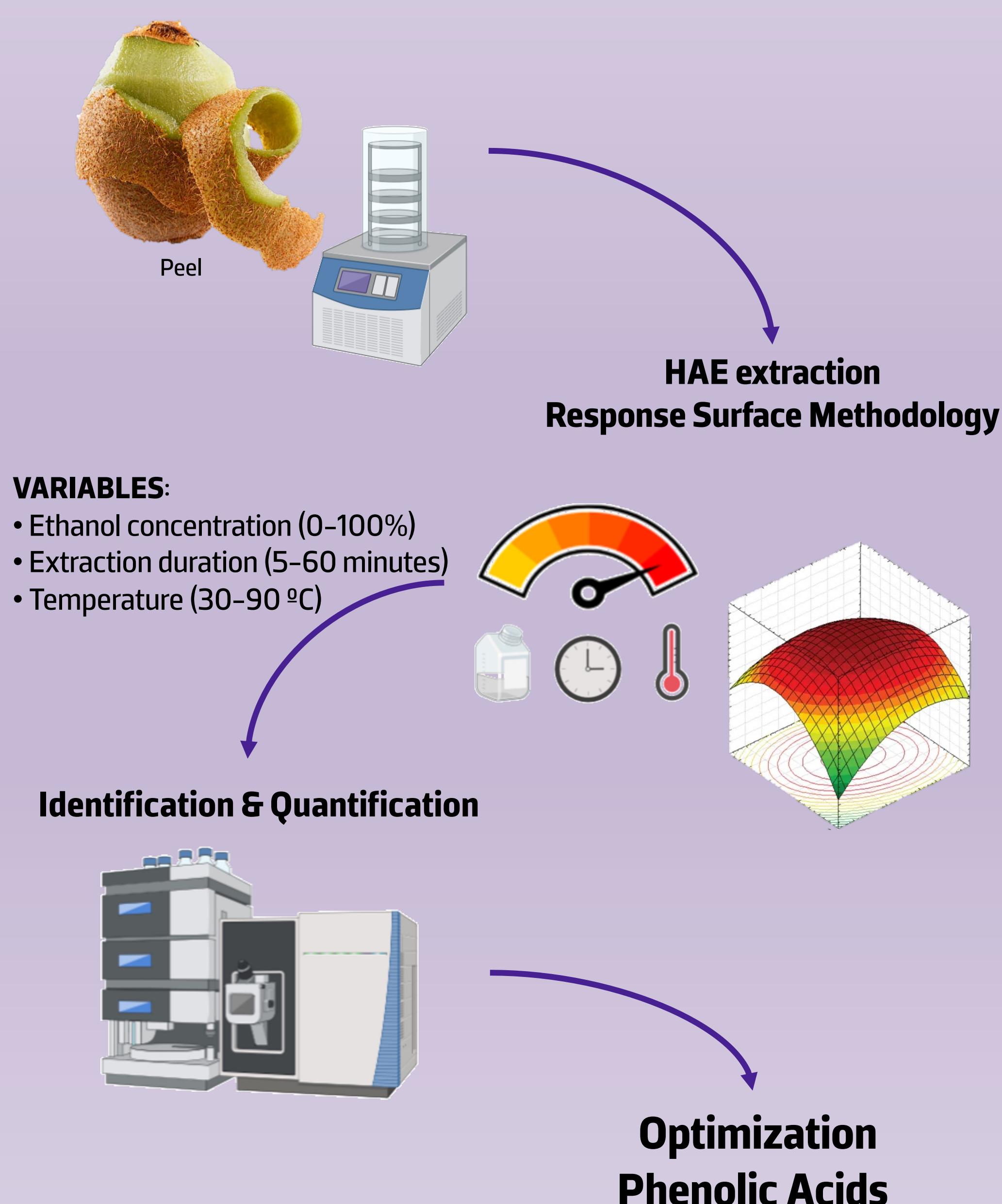


Harnessing Antioxidant Potential: Optimized Extraction of Bioactive Compounds from Yellow Kiwi Peels (*Actinidia chinensis*)Rafael Nogueira-Marques<sup>1</sup>, F. Chamorro<sup>1</sup>, A. Perez-Vazquez<sup>1</sup>, A.O.S. Jorge<sup>1,2</sup>, Aurora Silva<sup>1,3</sup> and M.A. Prieto<sup>1,\*</sup><sup>1</sup> Universidade de Vigo, Nutrition and Bromatology Group, Department of Analytical Chemistry and Food Science, Instituto de Agroecoloxía e Alimentación (IAA) – CITEXVI, 36310 Vigo, Spain.<sup>2</sup> REQUIMTE/LAQV, Department of Chemical Sciences, Faculty of Pharmacy, University of Porto, R. Jorge Viterbo Ferreira 228, 4050-313 Porto, Portugal<sup>3</sup> REQUIMTE/LAQV, Instituto Superior de Engenharia do Porto, Instituto Politécnico do Porto, Rua Dr. António Bernardino de Almeida 431, 4249-015 Porto, Portugal.

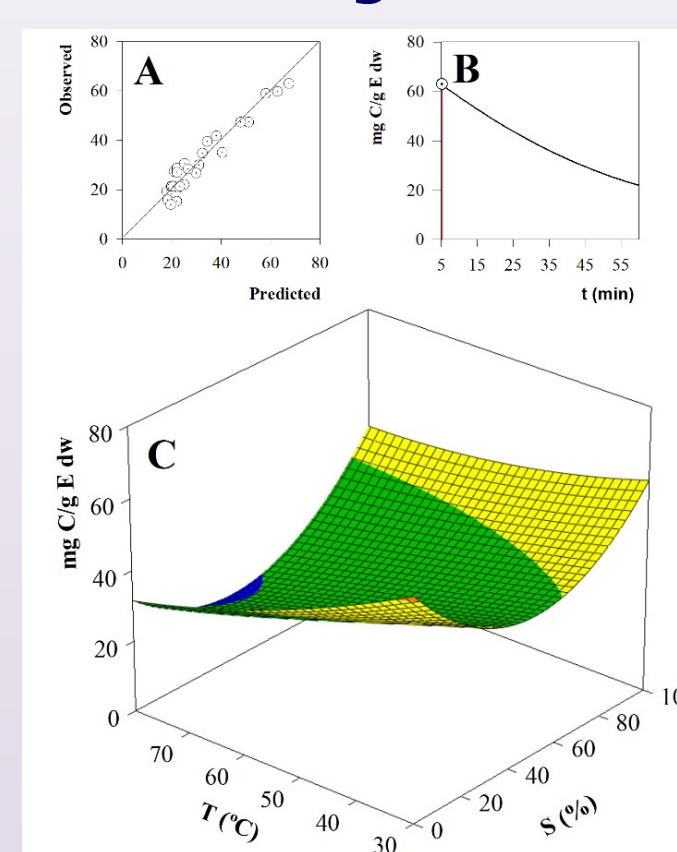
## INTRODUCTION &amp; AIM

With the increasing problem of food waste, research on agro-industrial **by-products** has proven to be an asset in the search for **bioactive compounds with biological activities**, such as antioxidant properties. The **kiwi** industry generates many by-products such as the peel, bringing the possibility of using these residues effectively, recovering target molecules and developing new products, including nutritional supplements, medicines and food additives. **Optimization** plays a crucial role in identifying the "best" conditions and potential interactions among the parameters involved in the extraction process to maximize desired outcomes. This process not only boosts the **circular economy** but also promotes consumer health. The main objective is to evaluate the phenolic acids present in the peel of the species *Actinidia chinensis* (yellow kiwifruit) as a potential source of antioxidants, through the optimization of the **heat-assisted extraction (HAE)** technique through the response surface model.

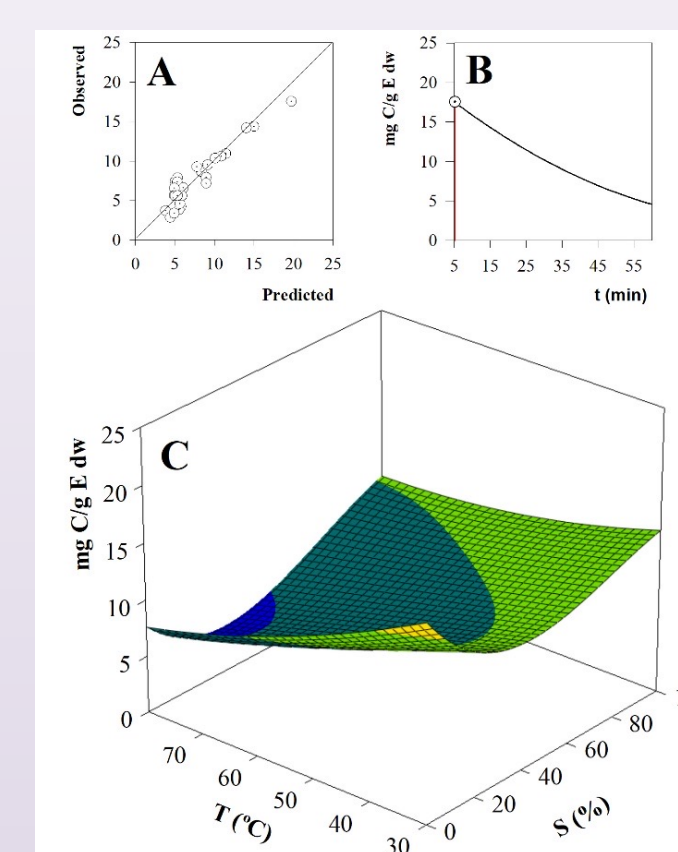
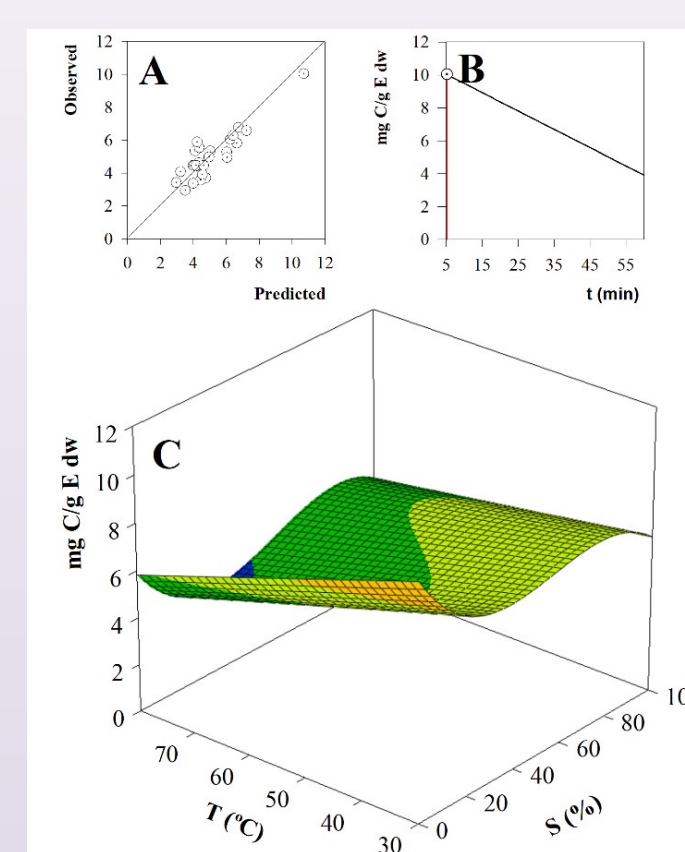
## METHOD



## RESULTS &amp; DISCUSSION

Dihydroferulic acid  
4-O-glucur

## Quinic acid

5-5-Dehydrodiferulic  
acid

The peel of *A. chinensis* was shown to be rich in phenolic acids:

- **Dihydroferulic acid 4-O-glucuronide** (67.32 mg/g E) > **quinic acid** (23.15 mg/g E) > **5-5-dehydrodiferulic acid** (4.08 mg/g E);
- All of them reported for their **antioxidant power**.
- The conditions that maximized HAE extraction of phenolic acids were **5 min**, **30 °C** and **100% water**. This suggests that the compounds **degrade** with **time** and **high temperature** and have a **strong affinity for water**.
- All significant parameters were **highly consistent** ( $p < 0.01$ ) and the high  $R^2$  values also confirmed this hypothesis by indicating the percentage of variability calculated by the model.

## CONCLUSION

The high levels of these compounds in yellow kiwifruit waste underscore their **potential for nutraceutical and industrial applications**. Additionally, the **antioxidant capabilities** of these residues are directly linked to their phenolic acids content, suggesting that they could be effectively used in **health-promoting products**.

## REFERENCES

- [1] – Patel, A., Hrůzová, K., Rova, U., Christakopoulos, P., & Matsakas, L. (2019). Sustainable biorefinery concept for biofuel production through holistic valorization of food waste. *Bioresource Technology*, **294**, 122247. <https://doi.org/10.1016/j.biortech.2019.122247>
- [2] – Gorinstein, S., Haruenkit, R., Poovarodom, S., Park, Y.-S., Vearasilp, S., Suhaj, M., Ham, K.-S., Heo, B.-G., Cho, J.-Y., & Jang, H. G. (2009). The comparative characteristics of snake and kiwi fruits. *Food and Chemical Toxicology*, **47**(8), 1884–1891. <https://doi.org/10.1016/j.fct.2009.04.047>
- [3] – Box, G. E. P., & Hunter, J. S. (1957). Multi-factor experimental designs for exploring response surfaces. *The Annals of Mathematical Statistics*, **28**(1), 195–241.

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