



Microwave-assisted kinetic resolution of homochiral diols using lipase

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Aim of the project

Better understanding of the influence of the microwave interaction with enzyme in various conditions :

- ➔ Aqueous or Non-aqueous medium Solvent free system
- ➔ High or Low Temperatures
- ➔ Under Pressure or Not

To determine Microwave influence on the enzymatic properties :

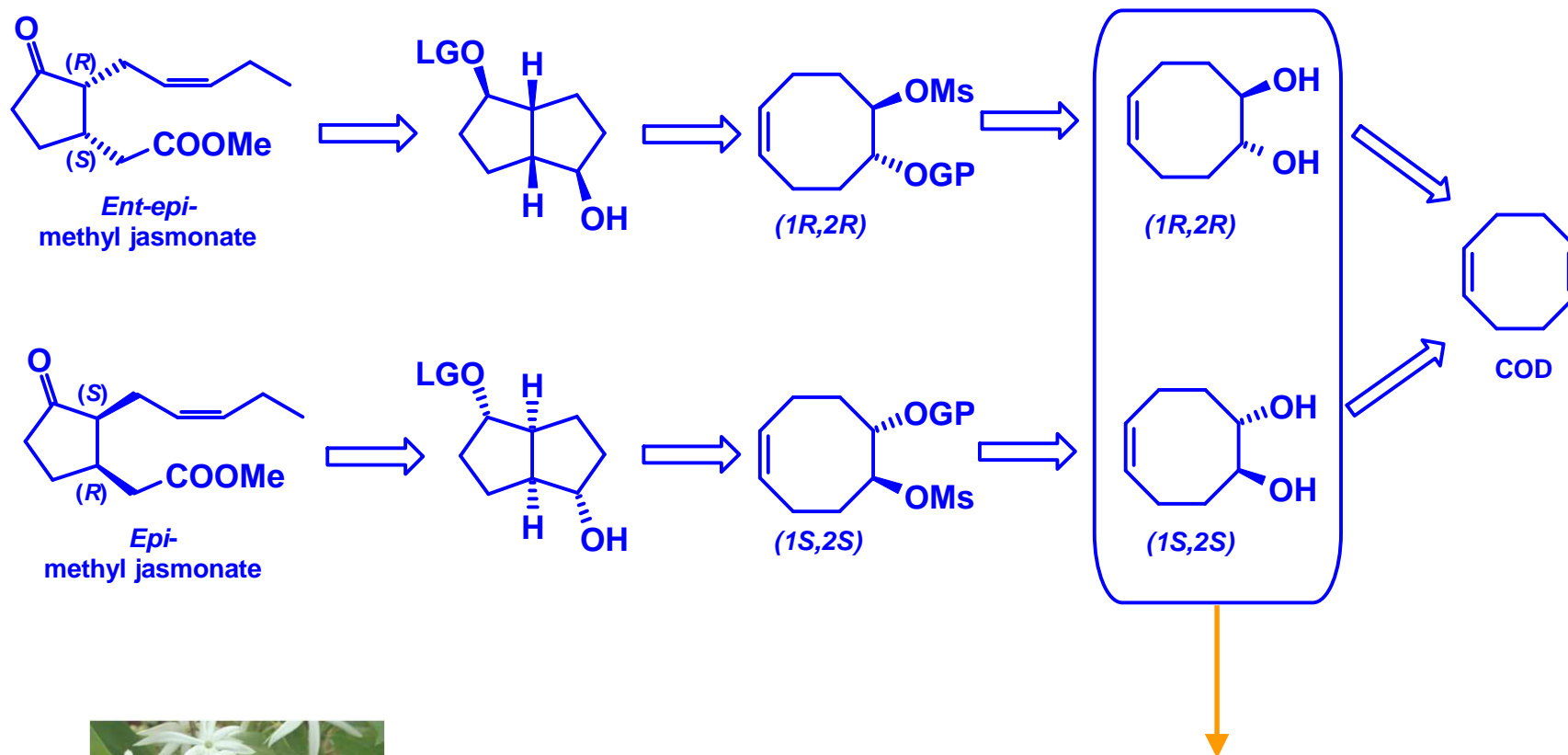
Activity
Stability

Selectivity
Reusability

Exploit the irradiation effects on the enzyme activation for the synthesis of interesting compounds



Stereoselective synthesis of methyl jasmonates



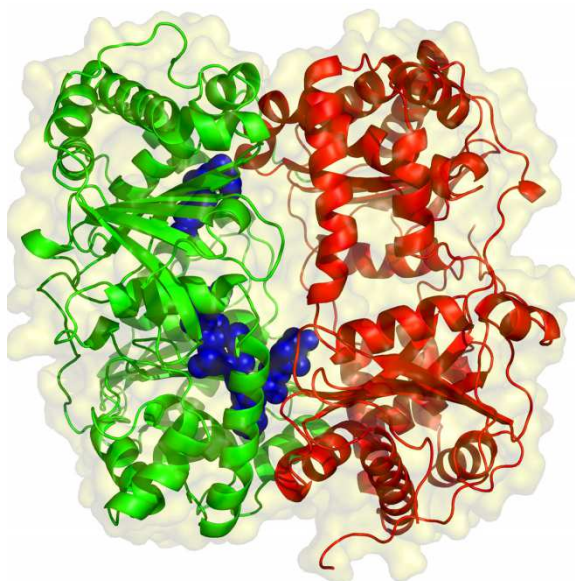
**Efficient resolution of homochiral diols
Used as precursor of methyl jasmonates**



Enzymes and green chemistry

Use of enzymes :

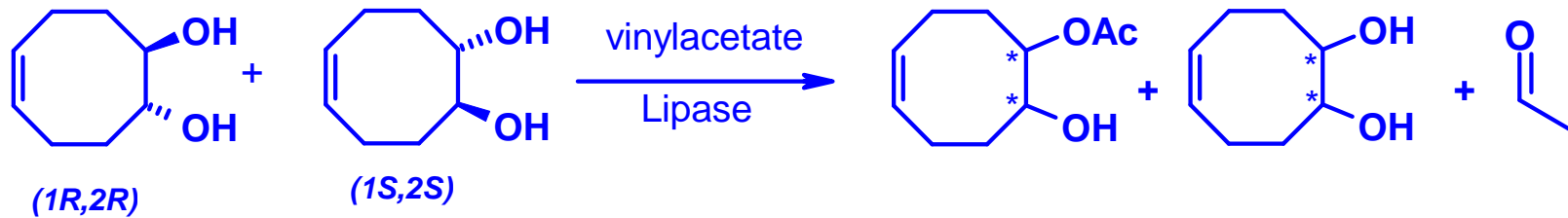
- Chimioselectivity
- Enantioselectivity
- Reusability



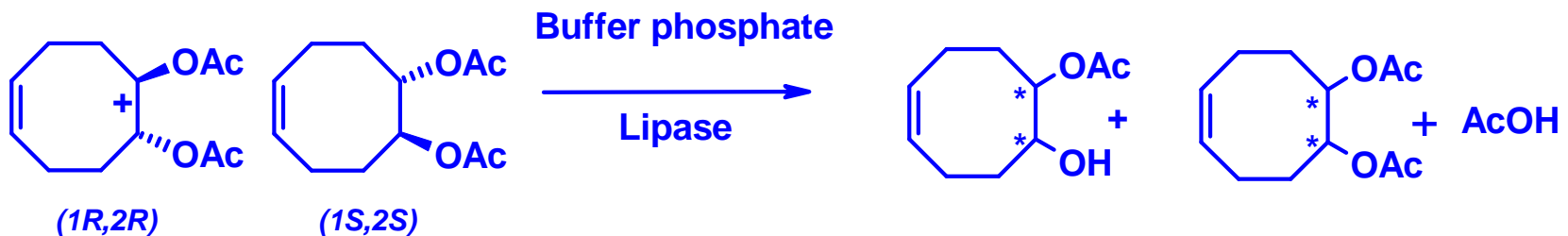
- Efficient catalyst for green chemistry
- The activity/selectivity depend on the reaction conditions



Resolution of rac-diol and rac-diacetate



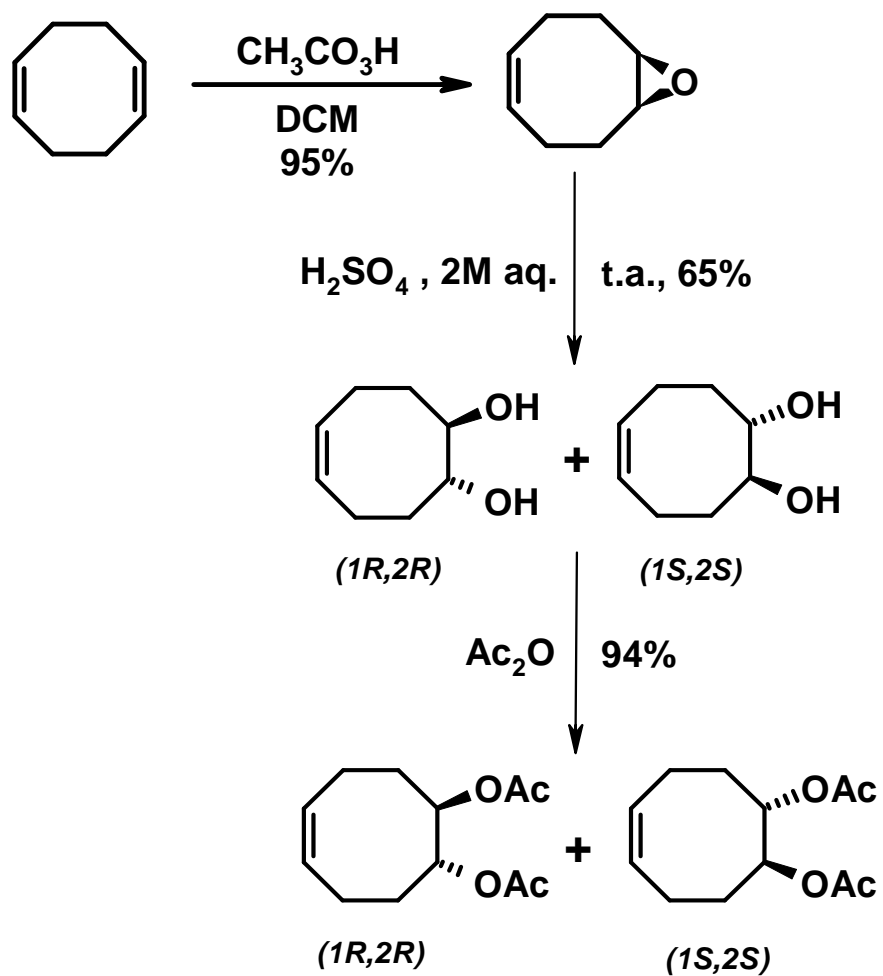
Resolution by acetylation

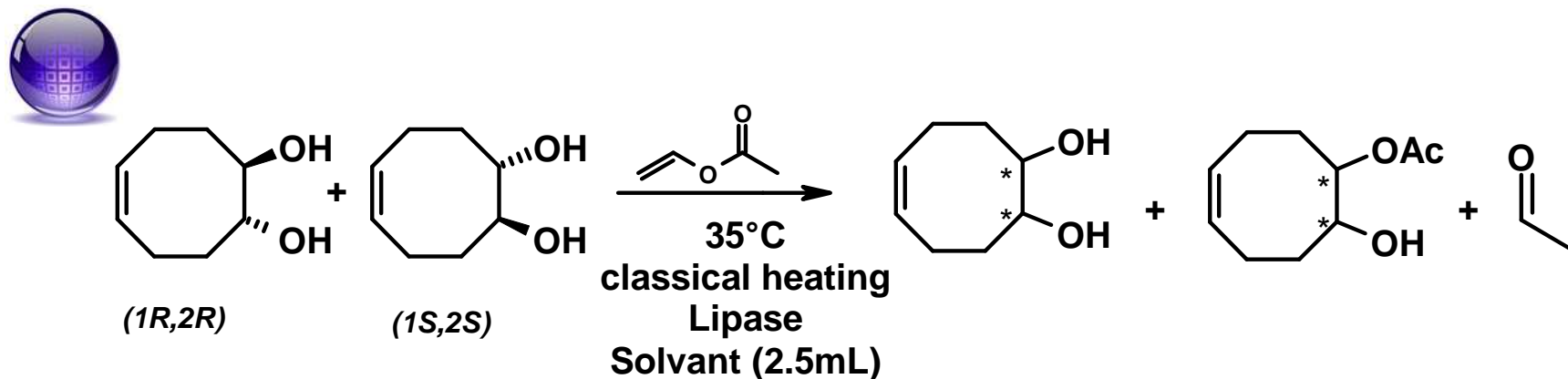


Resolution by hydrolysis



Synthesis of rac-diacetate and rac-diols

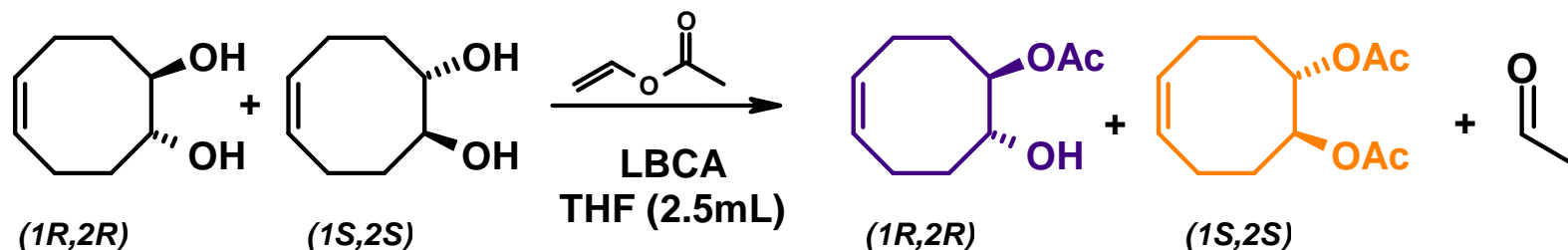




Lipase	c (%)	Diol ee (%)	Monoacetate ee (%)
<i>Aspergillus carneus</i>	3	2	50 (S,S)
<i>Rhizopus niveus</i>	4	5	38 (S,S)
<i>Rhizopus arrhizus</i>	-	-	-
<i>Muccor miehei</i>	-	-	-
<i>Candida cylindracea</i>	-	-	-
<i>Candida antarctica</i>	8	4	50 (R,R)
<i>Pseudomonas cepacia</i>	17	12	60 (S,S)
<i>Candida antarctica immobilisée (acrylique)</i>	16	9	55 (R,R)
<i>Pseudomonas cepacia immobilisée (diatomée)</i>	26	12	57 (S,S)



Lipase-catalyzed desymmetrization of rac-diol



Classical heating versus Microwave irradiation

Temperature (°C)	Time (day)	(1R,2R)		(1S,2S)	
		Yield (%)	ee (%)	Yield (%)	ee (%)
35	22	28	42	6	>99,9
50	7	30	50	20	>99,9

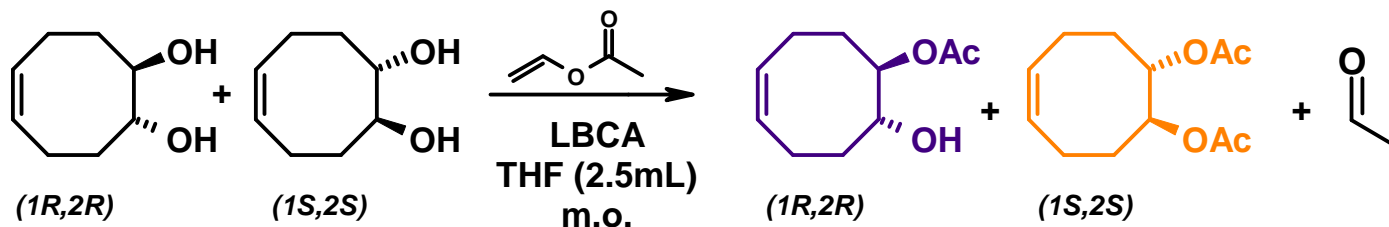
MW open vessel (10W), 14h

(1R,2R)		(1S,2S)	
Yield (%)	ee (%)	Yield (%)	ee (%)
58	55	37	>99,9

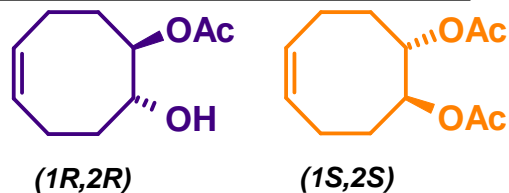
Optimisation of reaction



Lipase-catalyzed desymmetrization of rac-diol



Influence of temperature



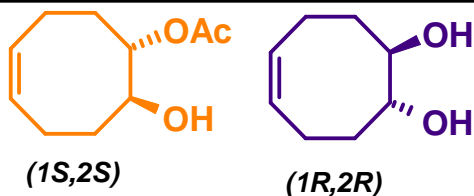
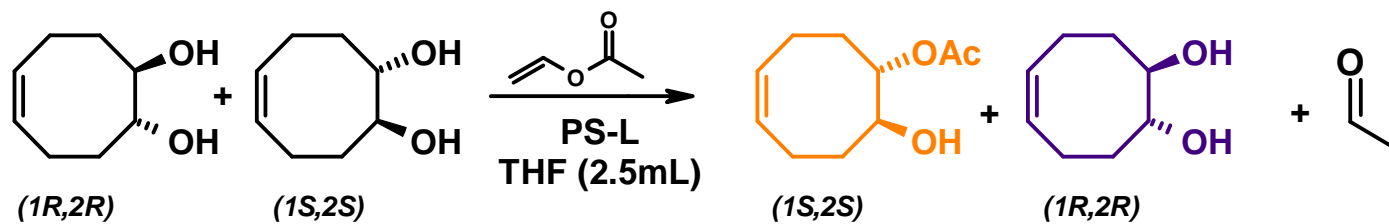
Temperature (°C)	Temps (h)	Yield (%)	ee (%)	Yield (%)	ee (%)
50 (10W)	14	58	55	37	99
80 (45W)	7	55	57	30	94
100 (syst. fermé, 40W)	7	-	-	-	-
35 (300W)	7	42	67	2	99

Thermal denaturation of enzyme

Diol 51% ee: 50%



Use of free PS

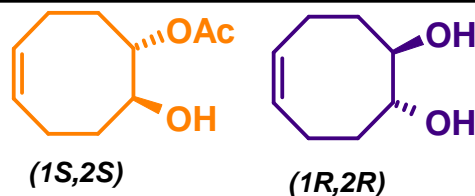
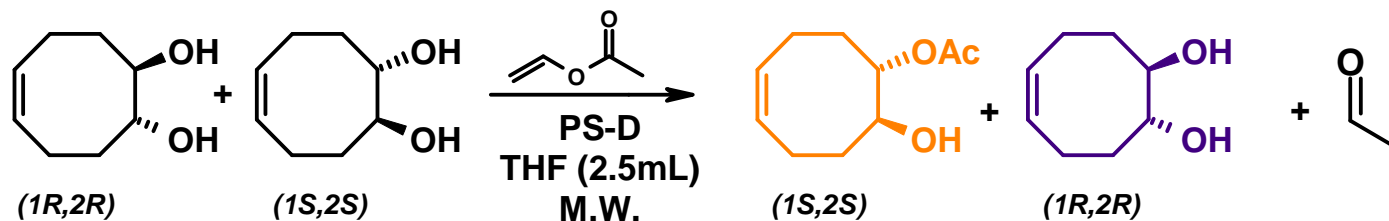


Temperature (°C)	Time (j)	(1S,2S)		(1R,2R)	
		Yield (%)	ee (%)	Yield (%)	ee (%)
35°C	21	55	0	45	0
55°C	7	47	0	51	0

Good conversion
No selectivity at all !



Use of immobilized PS



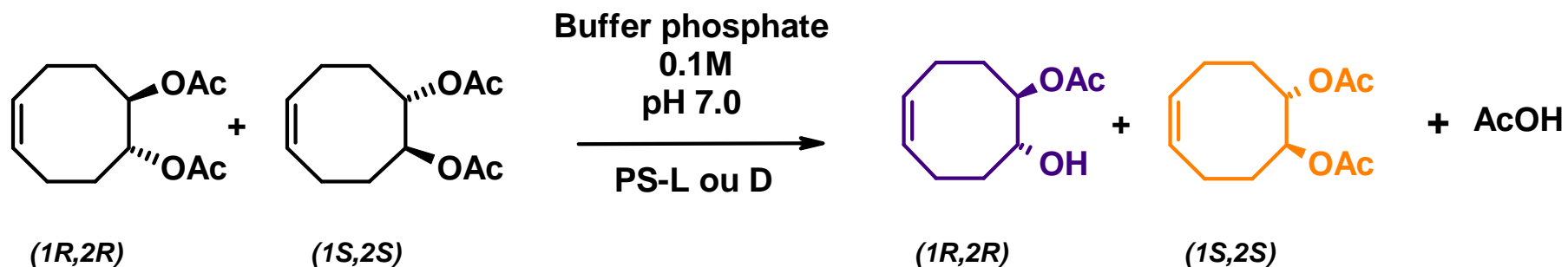
Temperature (°C)	Time (h)	(1S,2S)		(1R,2R)	
		Yield (%)	ee (%)	Yield (%)	ee (%)
50 (15W)	14	41	57	58	35
80 (35W)	7	12	35	66	5
100 (closed vessel)	7	-	-	-	-

Modulation of selectivity

High and fast thermal denaturation



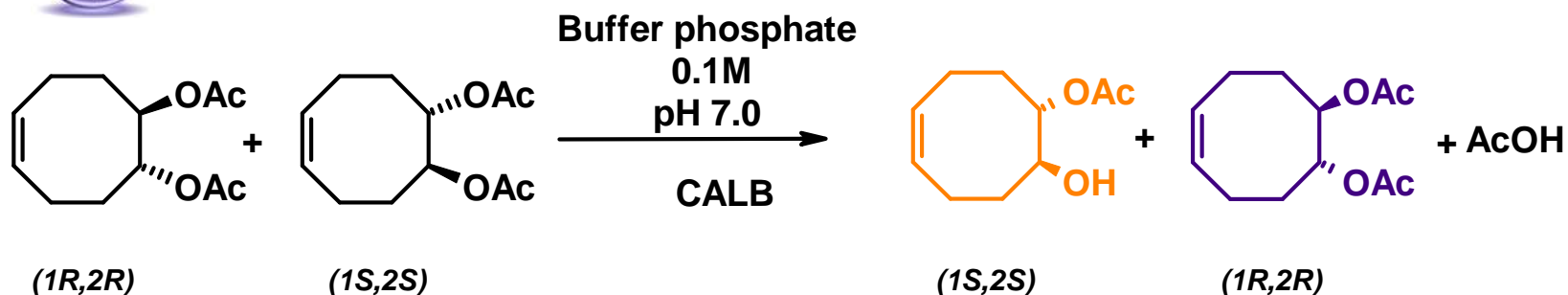
Resolution by hydrolysis under MW irradiation



PS-L	35°C	15j	34% ee : 99%	ee 99%
PS-L	35°C (MW)	14h	-	-
PS-D	35°C (MW)	14h	-	-
PS-D	50°C (15W)	14h	10% ee 81%	90% ee: 28%
			$[\alpha]_D = - 5.1^\circ$ (c=1.0 CHCl ₃)	$[\alpha]_D = - 21^\circ$ (c=1.0 CHCl ₃)
			1rst Enantio-enrichment:	35% ee: 70%
				$[\alpha]_D = - 52^\circ$ (c=1.0 CHCl ₃)



Resolution by hydrolysis under MW irradiation



35°	4j	17%	ee : 67%	diol: 3%
35°	14h	-		
50° (15W)	14h	20%	ee : 97%	80% ee 34%
80°	7-14h	-		
Open vessel (25W) closed vessel (35W)				

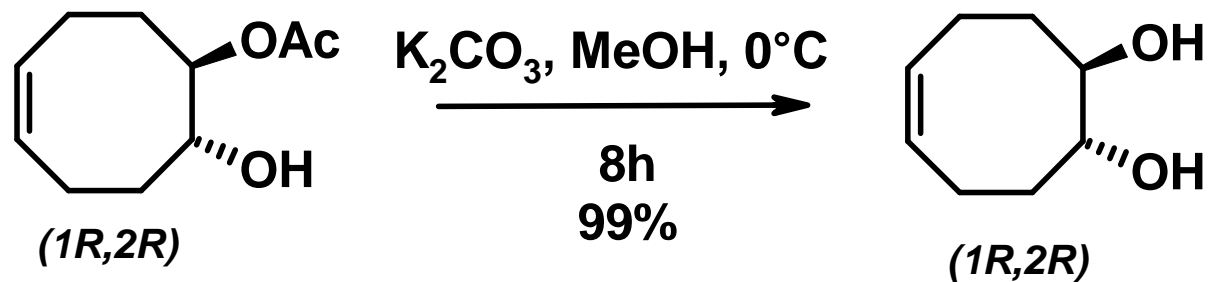
20 %	80%
ee : 97%	ee : 34%
$[\alpha]_D = + 5,6^\circ$	$[\alpha]_D = +26^\circ$
(c=1.0 CHCl₃)	(c=1.0 CHCl₃)

Enantio-enrichment 50°C (15W): 47% ee: 99%

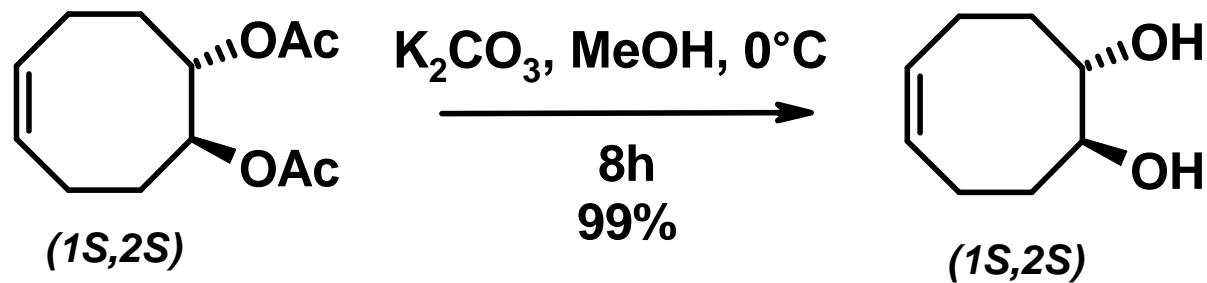
Control of the reaction selectivity by the choose of enzyme



Enantiopure homochiral diols



$[\alpha]_D = -21^\circ$
($c = 1.0 \text{ CHCl}_3$)

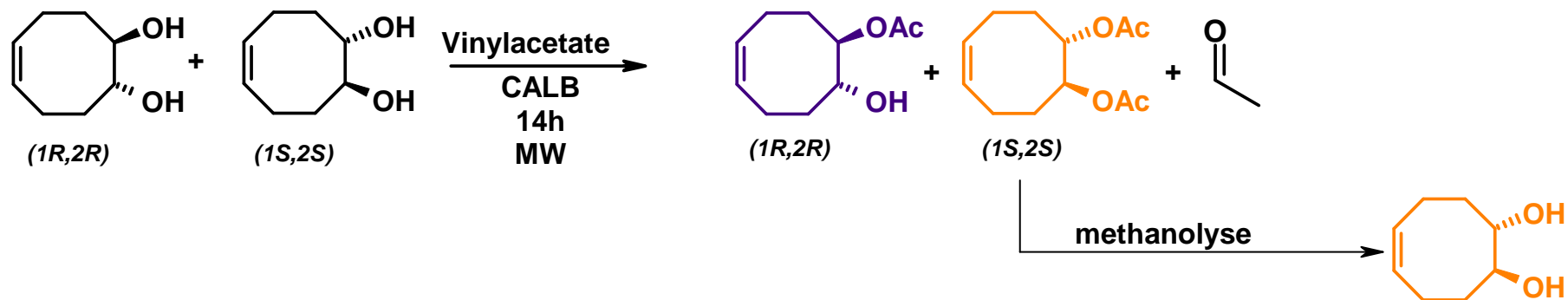


$[\alpha]_D = +20.8^\circ$
($c = 1.0 \text{ CHCl}_3$)

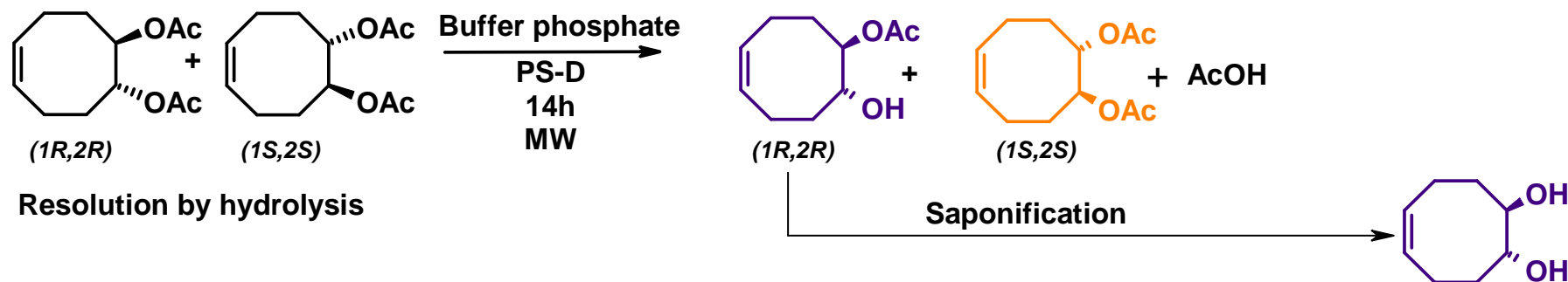


Conclusion

Optimisation d'une résolution enzymatique. Par choix judicieux des conditions :



Diol (1S,2S) Yield. 37 % under MW



Resolution by hydrolysis

Diol (1R,2R) Yield 20% under MW.



Thank you