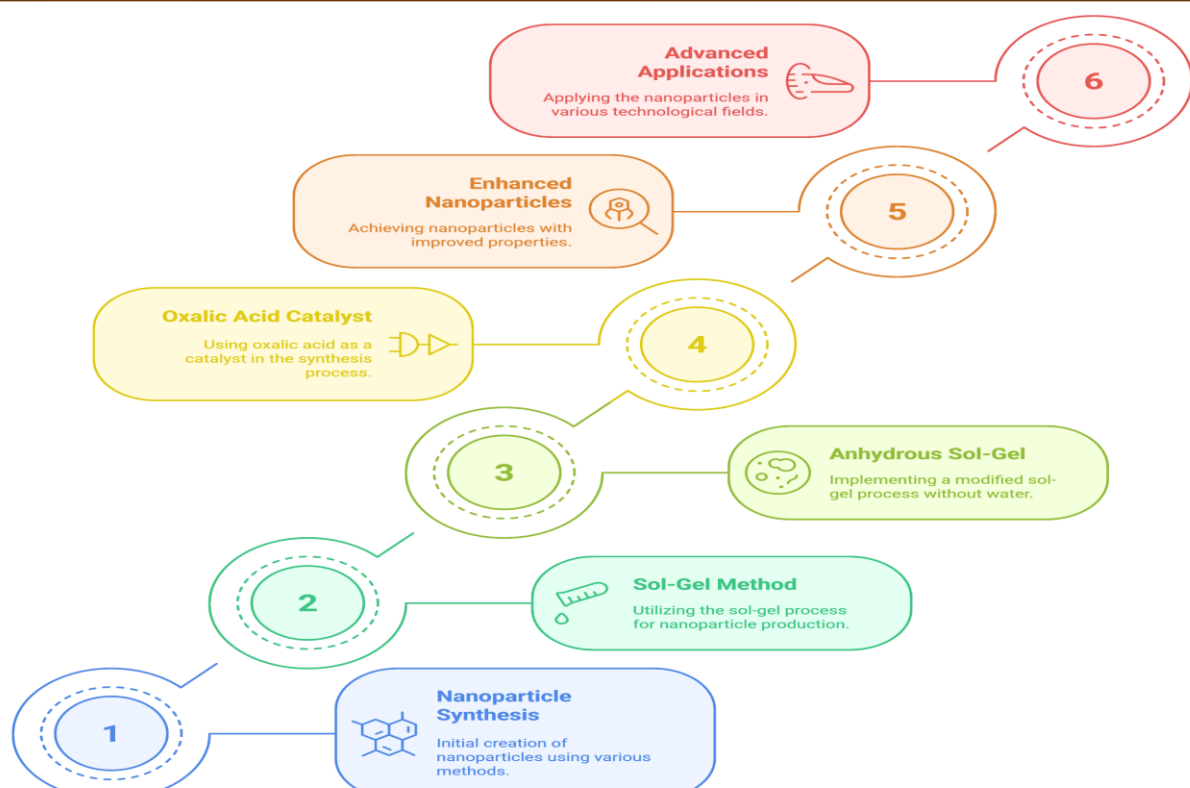
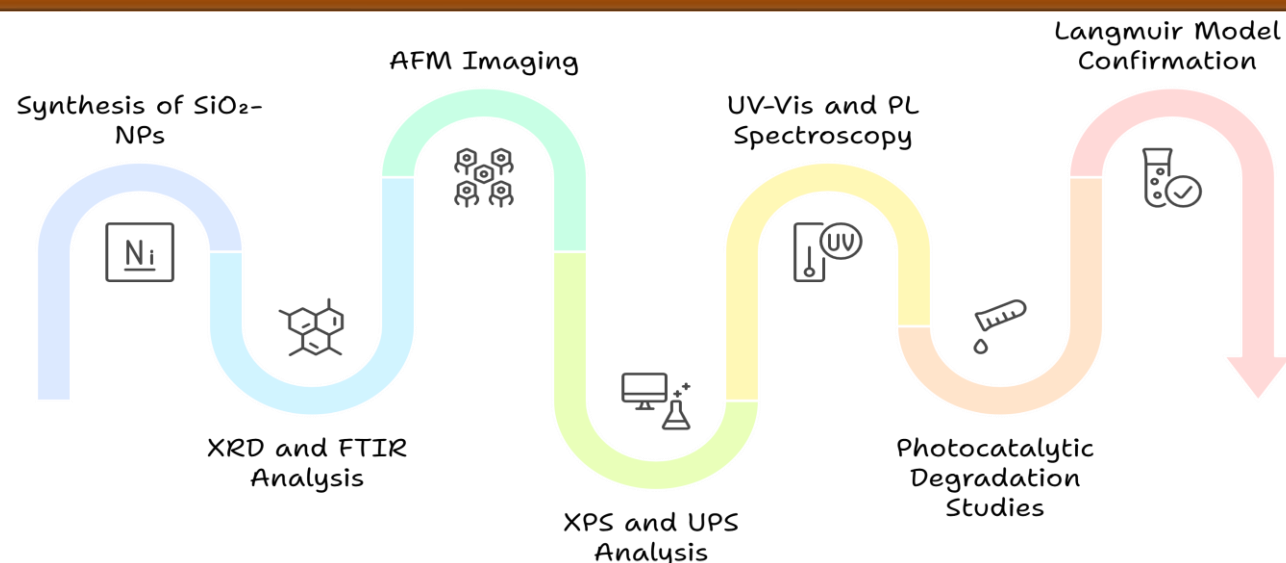


Modified Sol–Gel Synthesis of SiO₂ NanoparticlesOmar Mahmoud¹ omadr07@gmail.com, M'Hamed Guezzoul³, Abdelkader Nebatti Chergui⁴, Moumene Taqiyeddine², El Habib Belarbi², Hafida Miloudi¹, Djamila Bouazza¹.¹ Laboratory of Chemistry of Materials, University of Oran 1, Algeria² Laboratory of Synthesis and Catalysis, University of Tiaret, Algeria³ Laboratory of Materials (LABMAT), National Polytechnic School of Oran Maurice Audin (ENPO-MA), Algeria⁴ Laboratory of Materials Sciences and Applications (LSMA), University of Ain Temouchent Belhadj Bouchaib, Faculty of Sciences and Technology, Algeria

INTRODUCTION & AIM

Figure 1: Progressive Development of Carbon-Modified SiO₂ Nanoparticles via Anhydrous Sol-Gel Synthesis and Oxalic Acid Catalysis for Advanced Applications

METHOD

FIGURE 2: Synthesis and Characterization of Carbon-Modified SiO₂ Nanoparticles

RESULTS & DISCUSSION

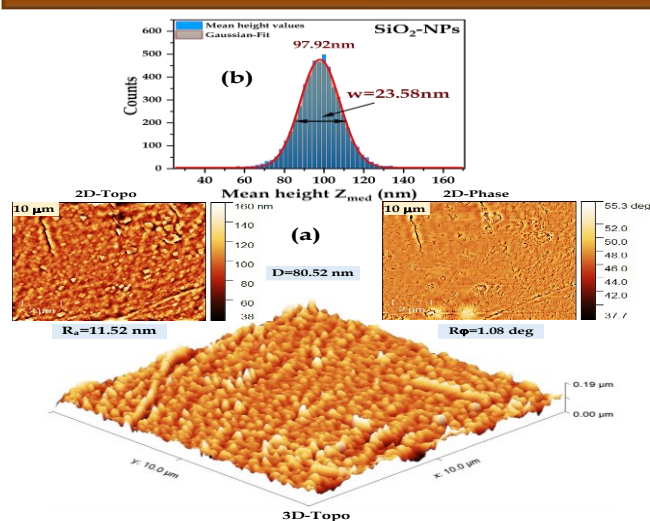
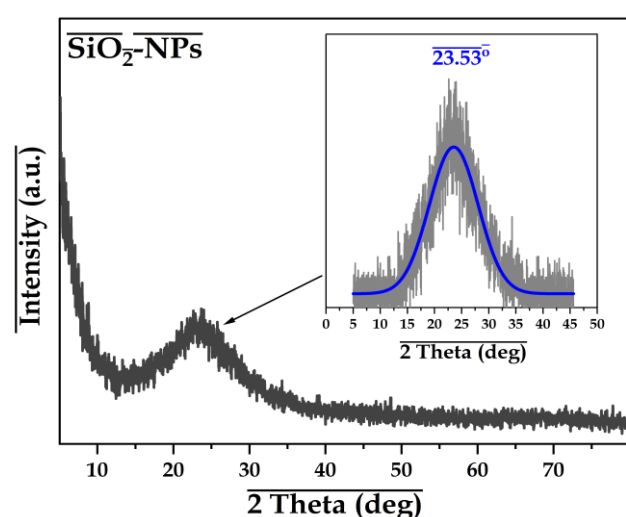
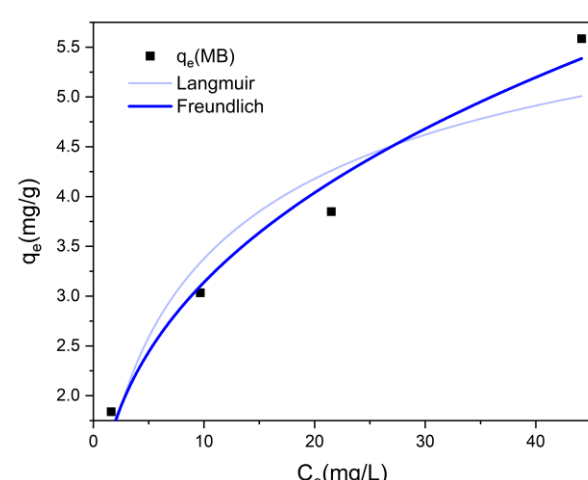
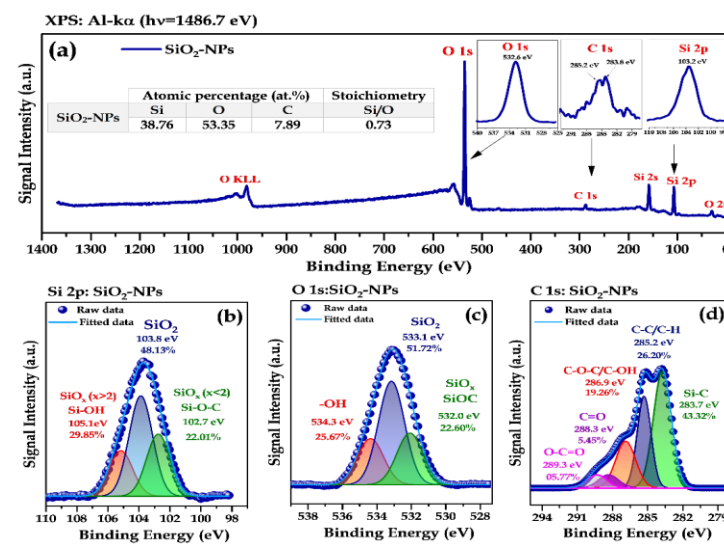
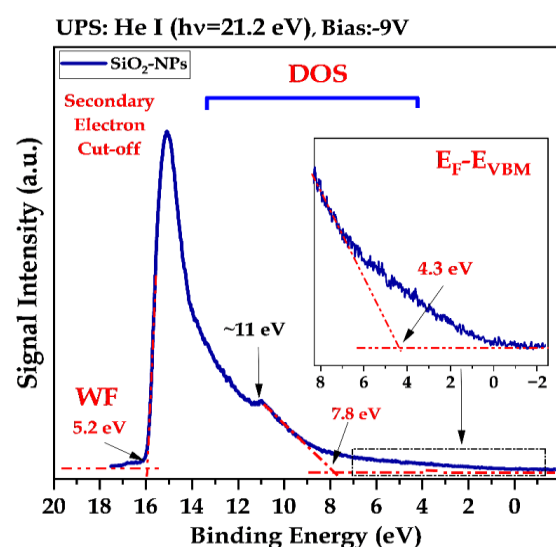
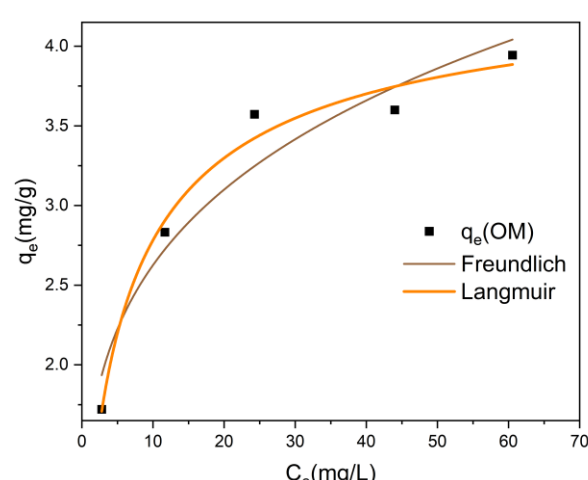
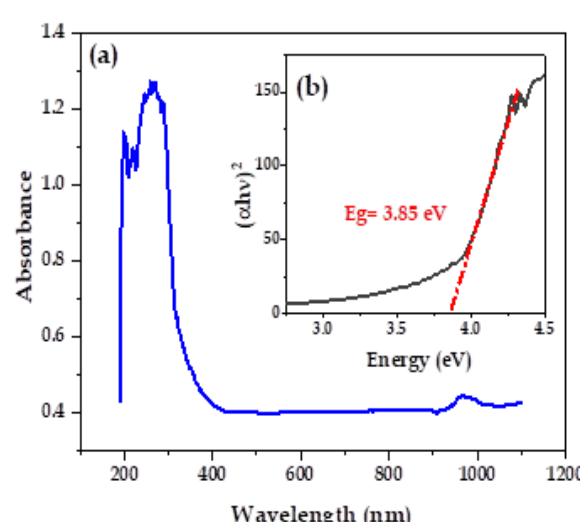
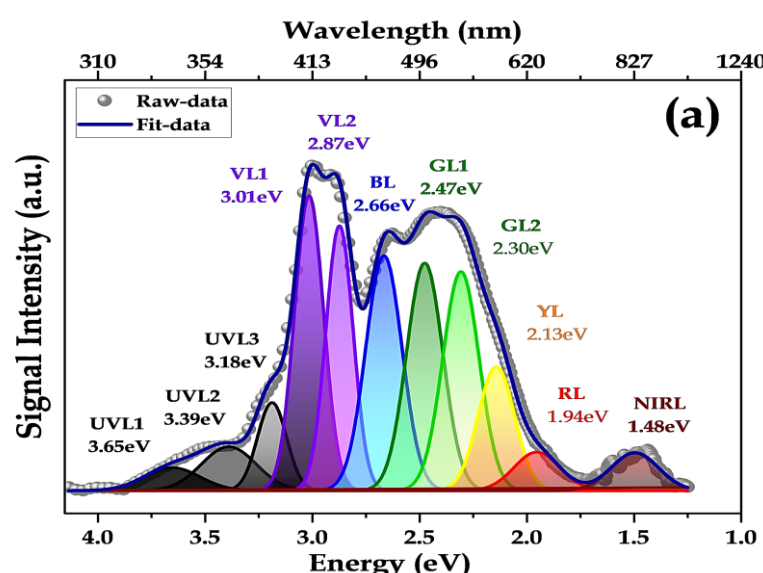
Fig. 3. (a) 2D/3D AFM topography and phase images of SiO₂ NPs (10×10 μm scan; Ra: roughness, φa: phase contrast, D: grain size). (b) Height histogramFig. 4: XRD pattern of SiO₂ nanoparticles.

Fig. 5: Freundlich isotherm for Methylene Blue (MB); Langmuir isotherm for Methyl Orange (MO).

Fig. 7: XPS of SiO₂-NPs: (a) wide scan with atomic %, (b–d) Si 2p, O 1s, C 1s with Gaussian fits.Fig. 7. (a) UV-Vis absorption of SiO₂-NPs (max at 266 nm); (b) Tauc plot for bandgap energy.Fig. 8. (a) PL spectrum of SiO₂-NPs with Gaussian fitTable 1. PL emission peaks and corresponding defect states in SiO₂-NPs.

Emission Peak (eV)	Color	Origin
3.65	Ultraviolet (UV)	Oxygen-deficient centers (ODC I), Singlet-Singlet transition
3.39		Oxygen vacancy related states in SiO ₂
3.18		Oxygen deficient centers ODC (II) (≡Si:Si=)
3.01	Violet	Oxygen deficient centers ODC (II) (≡Si:Si=)
2.87		Oxygen deficient centers ODC (II) (≡Si:Si=)
2.66		Self-Trapped Excitons (STEs)
2.47	Green	Dioxasilirane group (DOSG),
2.30		SiC-related centers
2.13		Silanone groups (Si=O double-bond states)
1.94	Red	The nonbridging oxygen hole centers (NBOHC) ≡Si-O·
1.48		Hydrogen-related defects (the OH group)
1.48		The quantum confinement effect
1.48	Near-infrared	Defect luminescence from Silanol groups (Si-OH)
1.48		Si nanocluster

CONCLUSION

Table 2. Summary of Key Structural, Optical, and Functional Properties of SiO₂-NPs

Parameter	Summary
Crystallite Size	~7.7 nm (short-range order in amorphous matrix)
Synthesis	Sol-gel with oxalic acid, 400°C, low energy
Defects	Oxygen vacancies, Si–O–C bonds, NBOHCs
Surface	Hydroxylated, ~80 nm grains, roughness Sa ~ 11.5 nm
Optical	Bandgap 3.85 eV; PL emissions 1.48–3.65 eV (defects)
Adsorption	MB: Freundlich (R ² = 0.975); MO: Langmuir (R ² = 0.983)
Photocatalysis	Defect-enhanced charge separation, visible-light active
Applications	Wastewater treatment, optoelectronics, sensing

Conflict of Interest: The authors declare no conflicts of interest.

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