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Hybrid nanostructures of transition metal oxides on vertical graphene for enhanced electrochemical performance

> Alina MATEI*, Cosmin ROMANIȚAN, Oana BRÎNCOVEANU, Marius STOIAN, **Octavian-Gabriel SIMIONESCU, Vasilica ȚUCUREANU***



National Institute for Research and Development in Microtechnologies IMT-Bucharest

*Correspondence: alina.matei@imt.ro; vasilica.tucureanu@imt.ro

INTRODUCTION & AIM

Transition metal oxide (TMO) nanostructures have attracted particular interest due to their multifunctionality, ranging from biomedical devices to electrochemical sensors for wastewater treatment in the textile industry, food processing and packaging, energy storage systems, catalysts, and solar cells. Among the different materials studied, In₂O₃ nanostructures have the advantages of remarkable physicochemical properties, high specific surface area, high surface-to-volume ratio, substantial chemical and environmental stability, and high electron mobility. Over the years, different substrates have been studied for the deposition of TMO thin films to meet the requirements of the targeted fields.

In the present work, In₂O₃ nanostructures were obtained by chemical synthesis, and the process conditions and thermal treatment parameters were controlled, with these factors being considered the determining factors with effects on particle size and morphology. Graphene hydrophilization was performed to ensure the oxide nanostructures were compatible with the substrate of interest. The next step consisted of dispersion of In_2O_3 powders in different media, drop-casting

> 1,4477 θ (right)

46,88

2 Bottom

θ (left) 43.60

Baseline Settings

🗘 BaseLine 🔶 Tilt

🗘 Тор

a suspension of oxide particles on the surface of the vertical graphene substrate, and evaporation of the solvent by heat treatment.



SEM micrograph of In₂O₃ - VG



The SEM images indicate a slight tendency of the particles to agglomerate at the surface and penetrate between the graphene sheets.



4000 Wavenumber (cm⁻¹)

The spectrum confirms the presence of functional groups assigned to the In–O vibration mode characteristic of cubic In_2O_3 .

The XRD diffractogram shows characteristic In_2O_3 planes at angles of 21°, 30.5°, 35.3°, 45°, 51°, and 60°, with an average crystallite size of 16 nm.

The oxidation and reduction peaks were observed during the anodic and cathodic sweeps, indicating the electrochemical performance of In_2O_3 .

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