The 5th International Online Conference on Nanomaterials



22-24 September 2025 | Online

Sn-doped NiFe LDH/Carbon Nanotube composite: An Efficient Bifunctional Electrocatalyst for Water Splitting

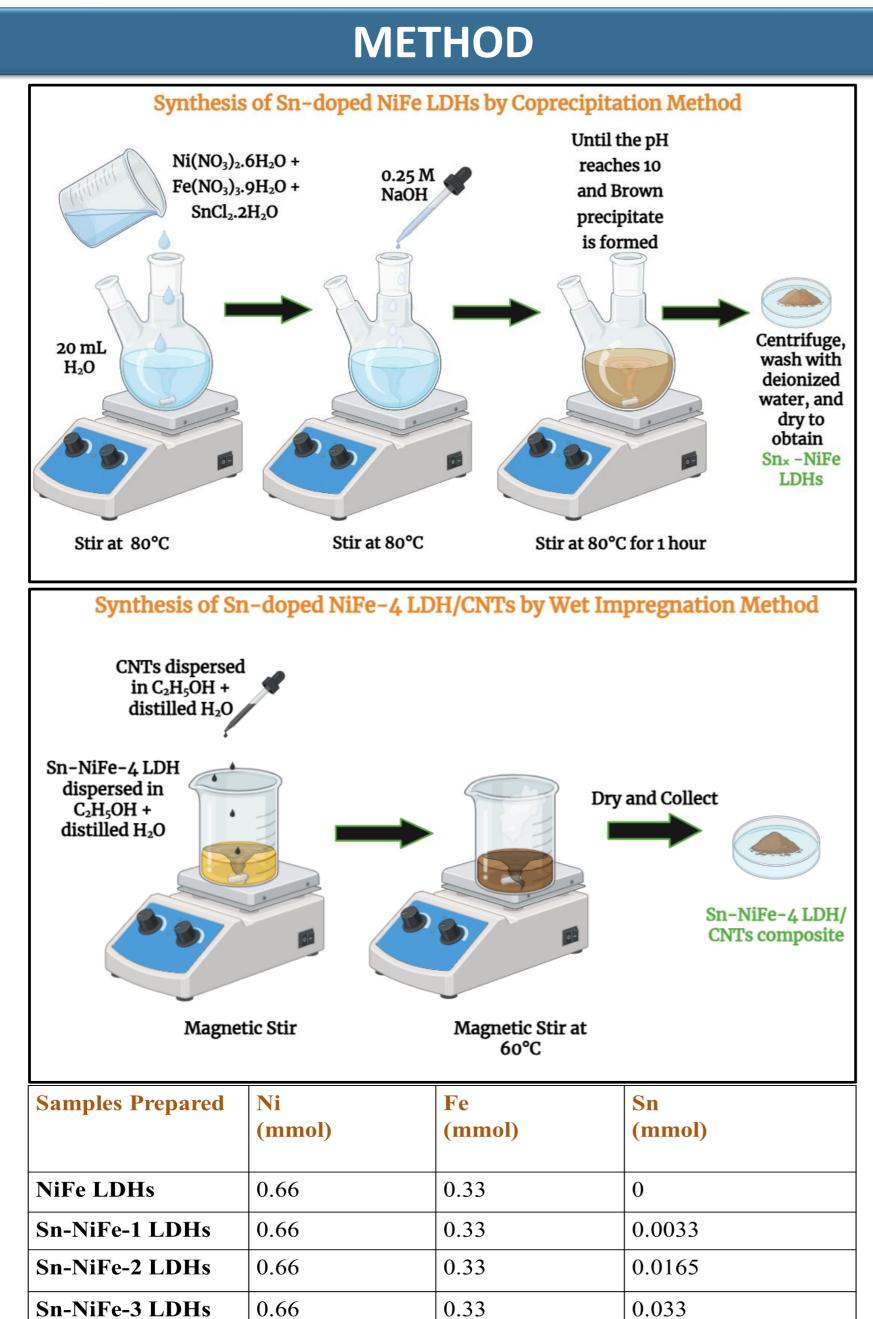
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INTRODUCTION & AIM Why hydrogen? **Sustainable Development Goals** (SDG) A versatile, zero-emission, efficient energy carrier LDH's general formula: High energy density $[M(II)_{1-x}M(III)_x(OH)_2](A^{n-})_{x/n}.yH_2O$ NEGATIVE Water **Electrolysis** Basal spacing (d) **ANODE** CATHODE CATION **OER** HER

AIM: Synthesis and Optimization of Sn-doped NiFe LDH/Carbon Nanotube composite as an Electrocatalyst for Water Splitting.



0.33

0.33

0.066

0.078

Sn-NiFe-4 LDHs

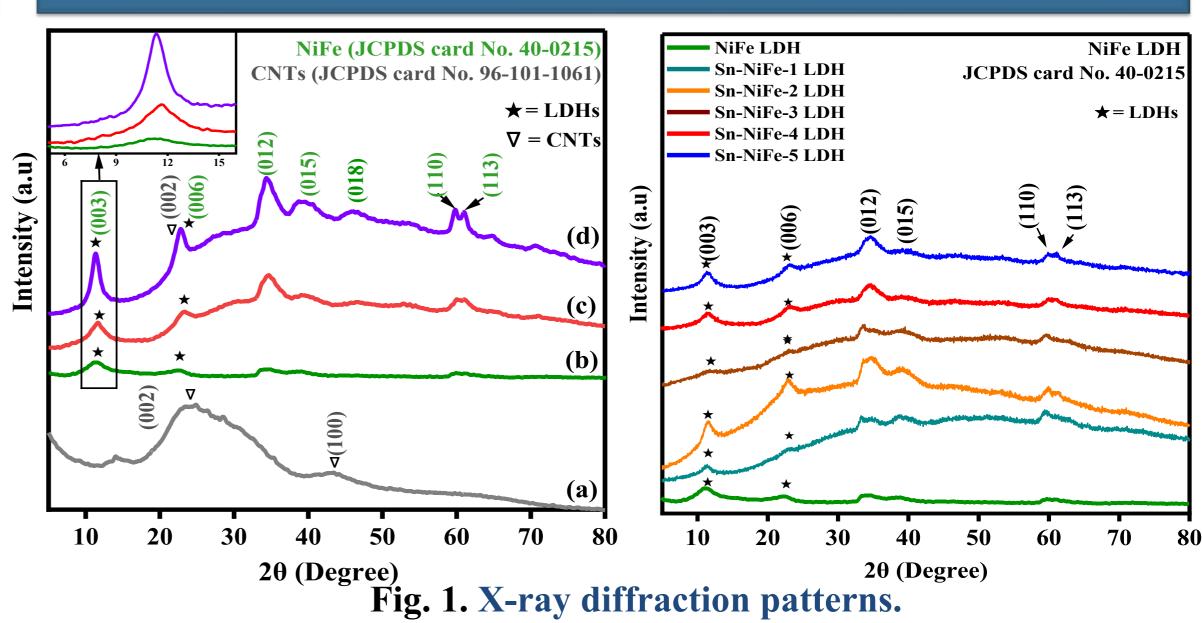
Sn-NiFe-5 LDHs

0.66

0.66

Sn-NiFe-4/CNTs (1.2wt% of CNT optimized)

RESULTS & DISCUSSION



(a) (b) (c) (d) CNTs

Sn-NiFe-4
LDHs

Fig. 2. FESEM images of (a) NiFe LDH, (b) Sn-NiFe-4 LDH, (c) CNTs, (d) Sn-NiFe-4/CNTs.

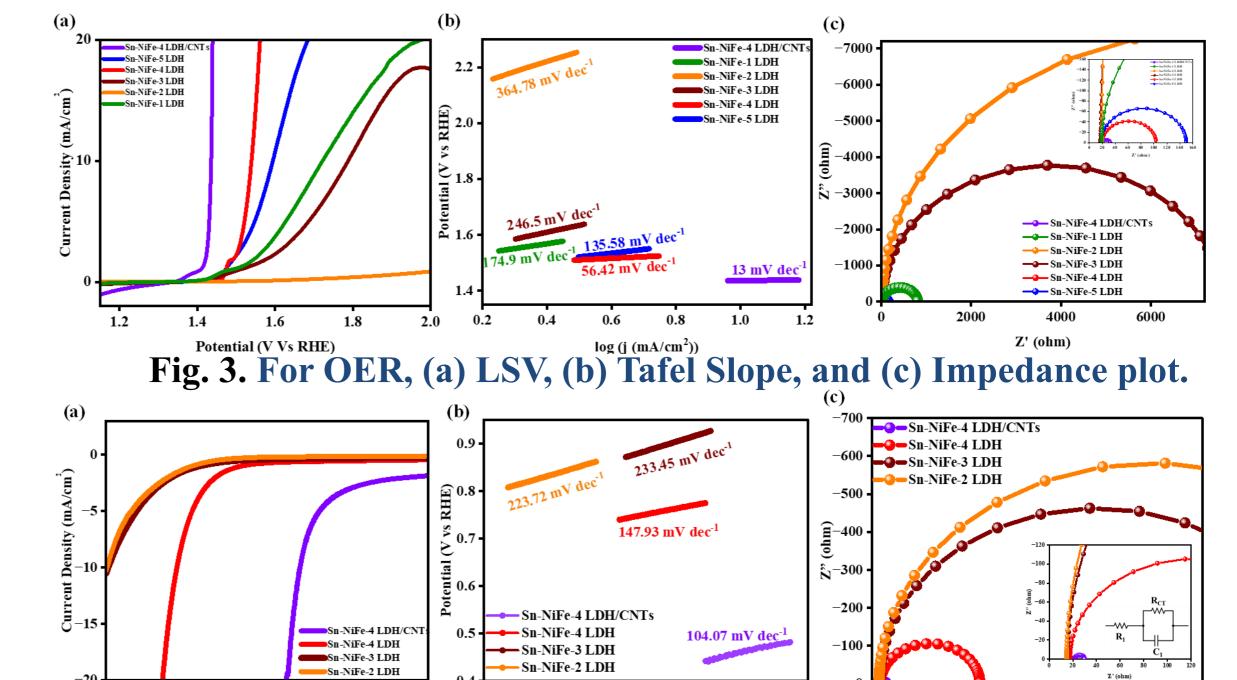


Fig. 4. For HER, (a) LSV, (b) Tafel Slope, and (c) Impedance plot.

CONCLUSION

 $log (j (mA/cm^2))$

- XRD revealed that the incorporation of CNTs enhances both the peak intensity and decreases the peak broadness, suggesting enhanced crystallinity or better stacking.
- Sn-NiFe-4 LDH/CNTs have the lowest HER (488) and OER (203) overpotential.
- Sn-NiFe-4 LDH/CNTs have the lowest R_{CT} .

Potential (V Vs RHE)

Thus, Sn-NiFe-4 LDH/CNTs outperformed other prepared electrocatalysts for both OER/HER activity.

FUTURE WORK / REFERENCES

- **XPS:** Analyze surface composition & electronic structure after Sn doping.
- HRTEM: Visualize lattice fringes & confirm Sn incorporation.
- Stability Tests: Long-term chronoamperometry at high current density.

Ali, A.; Shen, P. K. J. E. E. R., Recent progress in graphene-based nanostructured electrocatalysts for overall water splitting. 2020, 3, 370-394.

Z' (ohm)