

Photoinduced Nitric Oxide Release and Thermal Conversion Boosted by Electronic Modulation in N-Doped Carbon Nanodot Conjugates

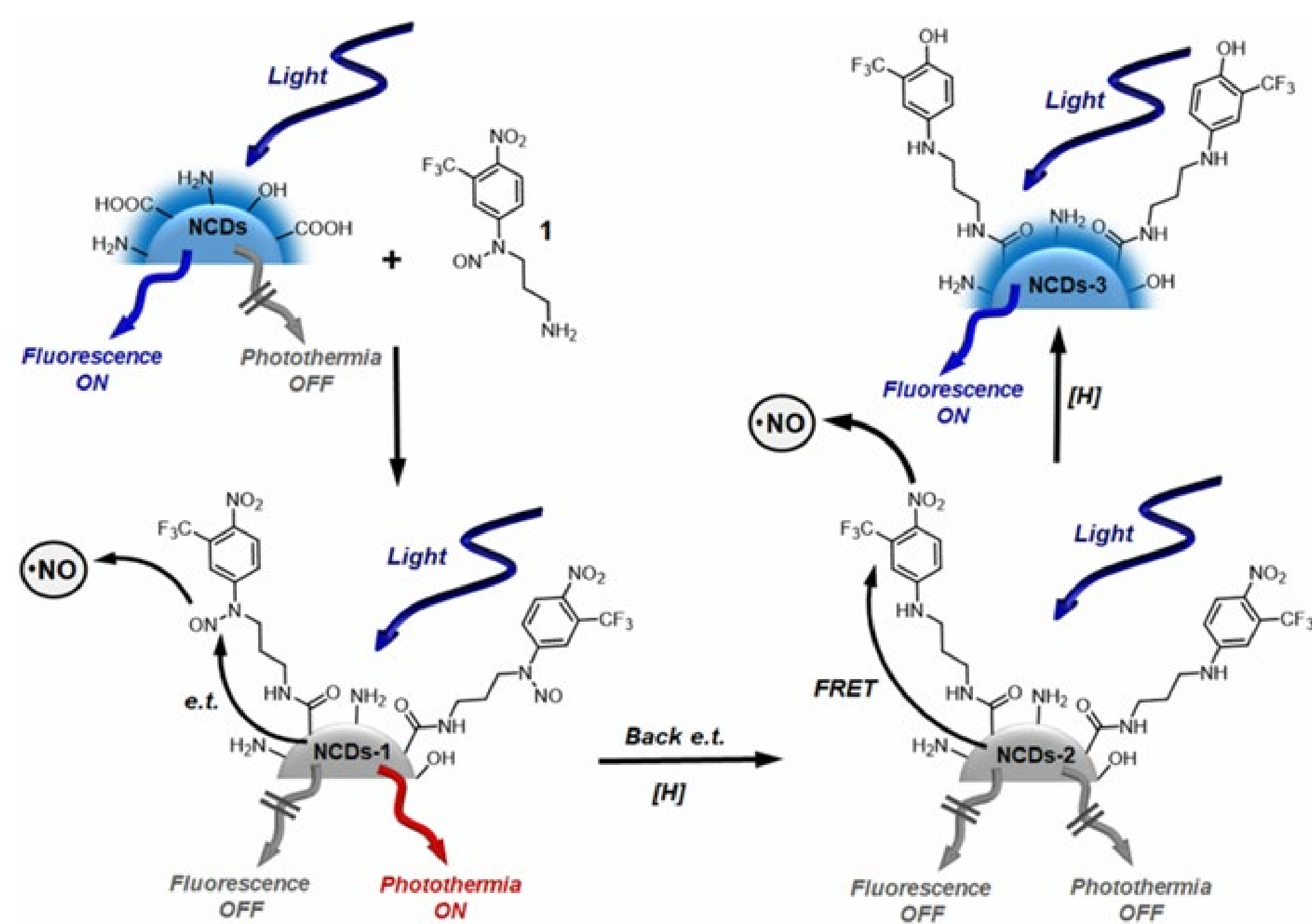
F. Laneri¹, C. Parisi¹, M. M. Natile², S. Sortino¹

¹PhotoChemLab, Department of Drug and Health Sciences, University of Catania, Catania, I-95125, Italy

²ICMATE-CNR, National Research Council, Department of Chemical Science, University of Padova, 35131 Padova, Italy

INTRODUCTION & AIM

Carbon Dots (CDs) are biocompatible carbon nanoparticles with tunable optical properties, applied in sensing, imaging, and phototherapy. Nitric oxide (NO), a key biological regulator with anticancer potential, requires controlled release, making NO photodonors (NOPDs) highly attractive. Combining CDs and NOPDs enables light-triggered release and fluorescent monitoring. In this work, the new nanoconjugate **NCDs-1**, obtained by linking N-doped CDs to NOPD1, shows enhanced NO photorelease, photothermal conversion, and fluorescence “off/on” reporting of NO generation [1].



Scheme 1. Sketch of the nanoconjugate **NCDs-1** and of its stepwise photoconversion to **NCDs-2** and then to **NCDs-3** under visible light.

RESULTS & DISCUSSION

Luminescent, water-soluble, nitrogen-doped CDs (**NCDs**) were synthesized via a hydrothermal route from citric acid and urea [2]. The **NCDs-1** nanoconjugate exhibited a distinct absorption band in the blue region (Fig. 1A), absent in the simple physical mixture of the reagents, evidencing strong electronic interactions between **NCDs** and **1**, which persist even in the excited states (Fig. 1B). Under blue light irradiation, **NCDs-1** showed photolysis behavior similar to free **1** (Fig. 1C) but with markedly enhanced photoreactivity (Fig. 1D). Upon prolonged irradiation, a second NO molecule was released (Fig. 2A) accompanied by a fluorescence decrease (Fig. 2B). Moreover, **NCDs-1** displayed strongly enhanced photothermal properties compared to bare **NCDs** (Fig. 2C).

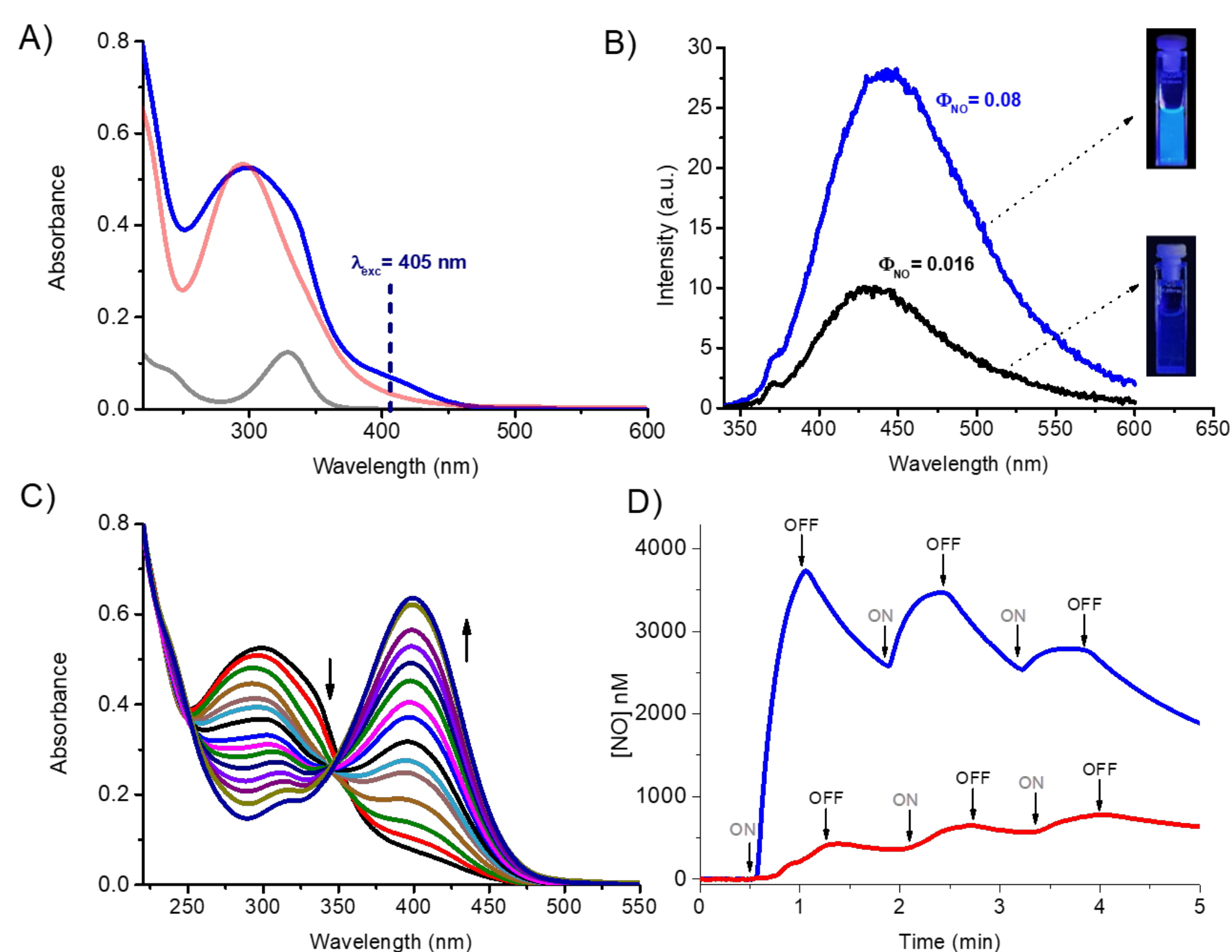


Fig. 1 (A) Absorption and (B) emission spectra ($\lambda_{\text{exc}} = 330 \text{ nm}$) of **NCDs-1** (blue), **NCDs** (black), and **1** (red) in aqueous solution (1% MeOH); inset: sample images under UV light. (C) Absorption changes of **NCDs-1** upon irradiation at 405 nm (0–39 min). (D) NO release profiles of **NCDs-1** (blue) and **1** (red) under alternating 405 nm light cycles.

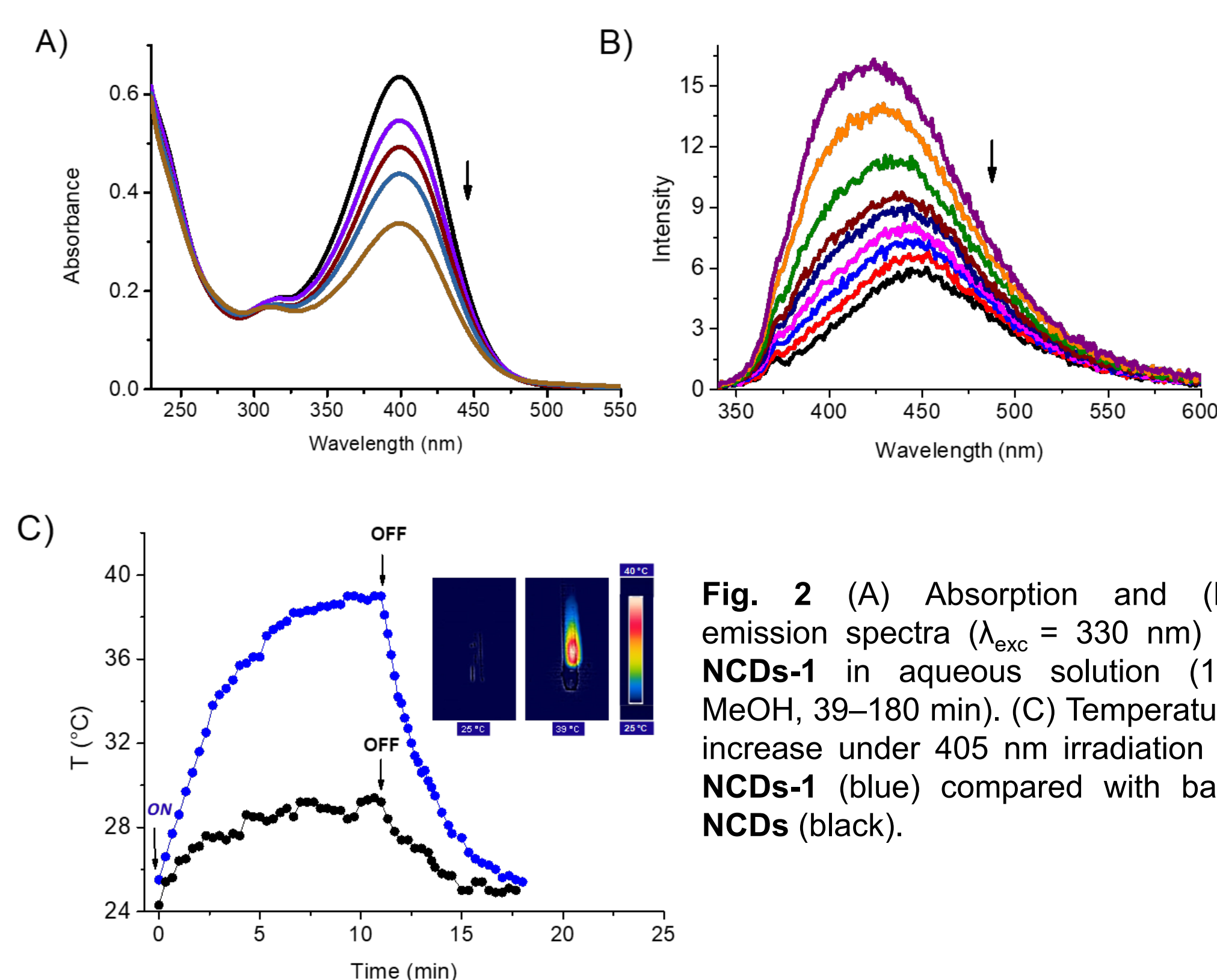


Fig. 2 (A) Absorption and (B) emission spectra ($\lambda_{\text{exc}} = 330 \text{ nm}$) of **NCDs-1** in aqueous solution (1% MeOH, 39–180 min). (C) Temperature increase under 405 nm irradiation of **NCDs-1** (blue) compared with bare **NCDs** (black).

REFERENCES

- [1] Laneri, F., Parisi, C., Natile, M. M., & Sortino, S. (2024). *Journal of Materials Chemistry B*, 12(45), 11817–11825.
- [2] Qi, H., Qiu, L., Zhang, X., Yi, T., Jing, J., Sami, R., Alanazi, S.F., Alqahtani, Z., Aljabri, M.D., Rahman, M.M. (2023). *RSC advances*, 13(4), 2663–2671.