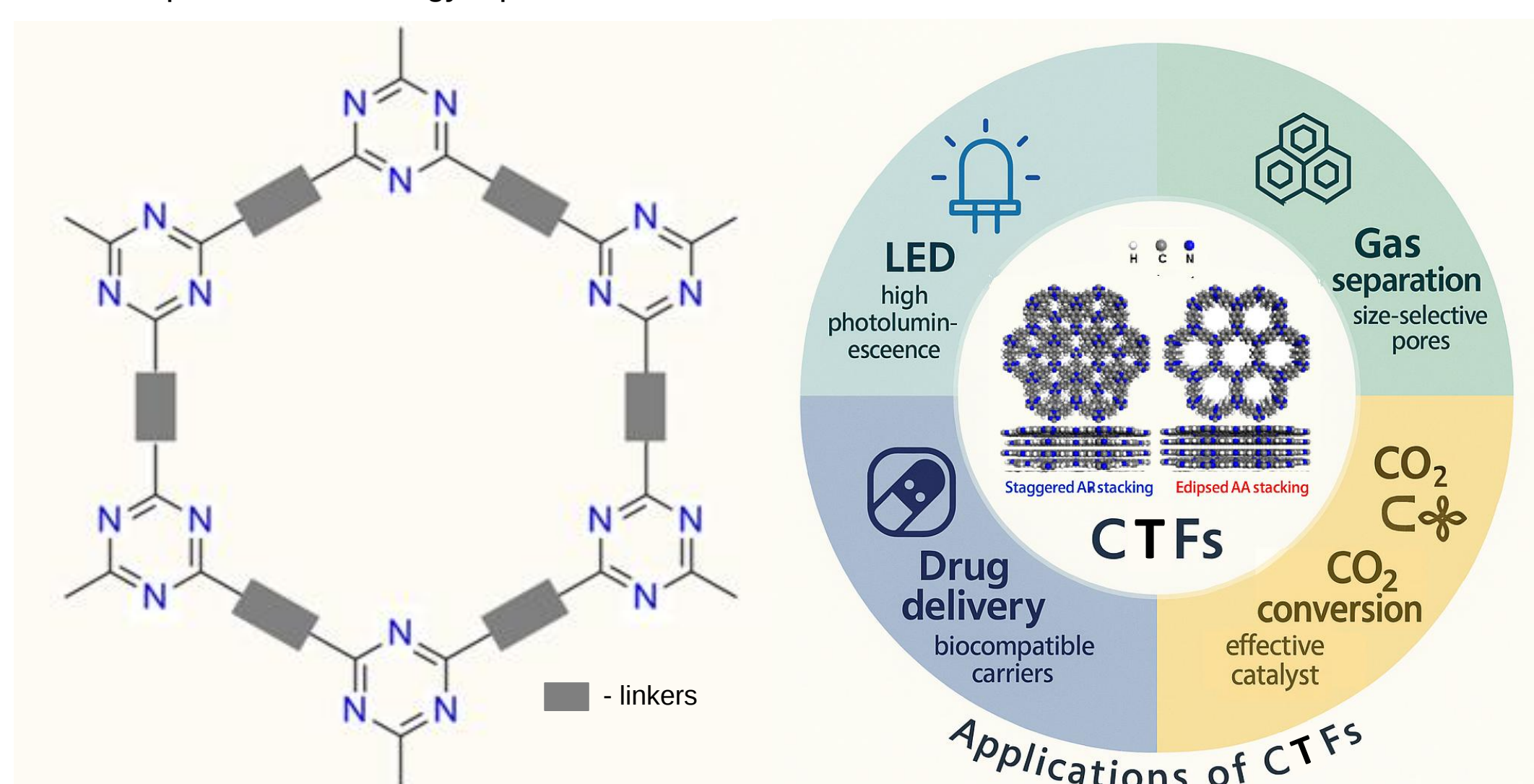


Facilitation of amine-bridged covalent triazine framework synthesis by microwave irradiation

Dmitrii Gribanov¹, Elena Pankratova¹, Hassan Alqahtani², Khalid Alruwaili², Vera Solovyeva¹¹ Aramco Innovations LLC, Bld. 1, 9 Varshavskoye Highway, 117105 Moscow, Russia² EXPEC Advanced Research Center (EXPEC ARC), 31311 Dhahran, Saudi Arabia

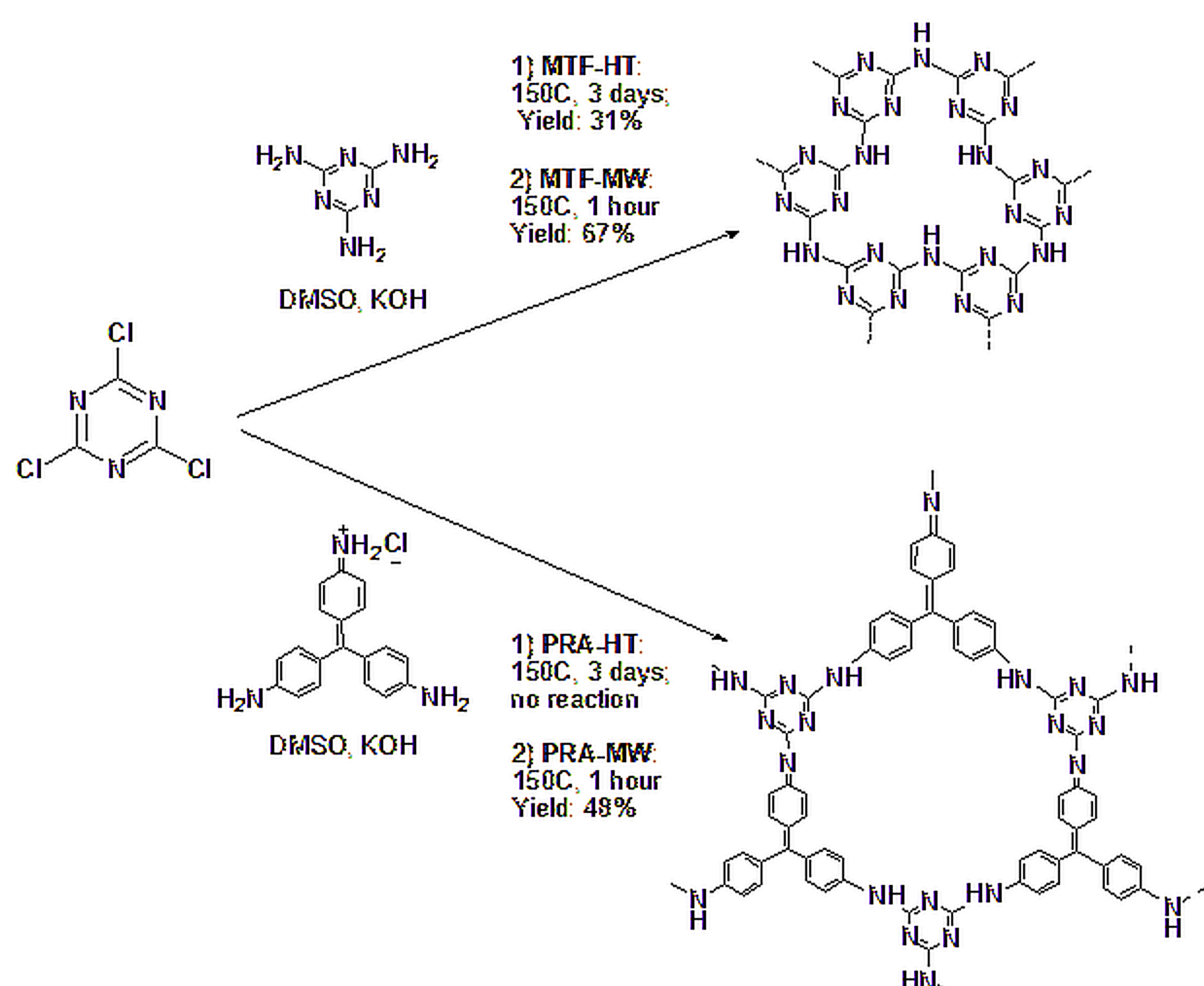
INTRODUCTION & AIM

Covalent triazine frameworks (CTFs) are N-rich porous polymers built from 1,3,5-triazine rings connected by covalent linkers. Their rigid, heteroatom-dense backbones typically show high thermal stability, appreciable surface areas, and sites well-suited for gas capture and catalysis^{1,2}. Here we target for obtaining of NH-linked CTFs from amines that react sluggishly under standard conditions by contrasting efficiency of conventional synthesis with microwave assisted, which delivers rapid, uniform energy input to accelerate substitution at milder conditions³.



METHODS

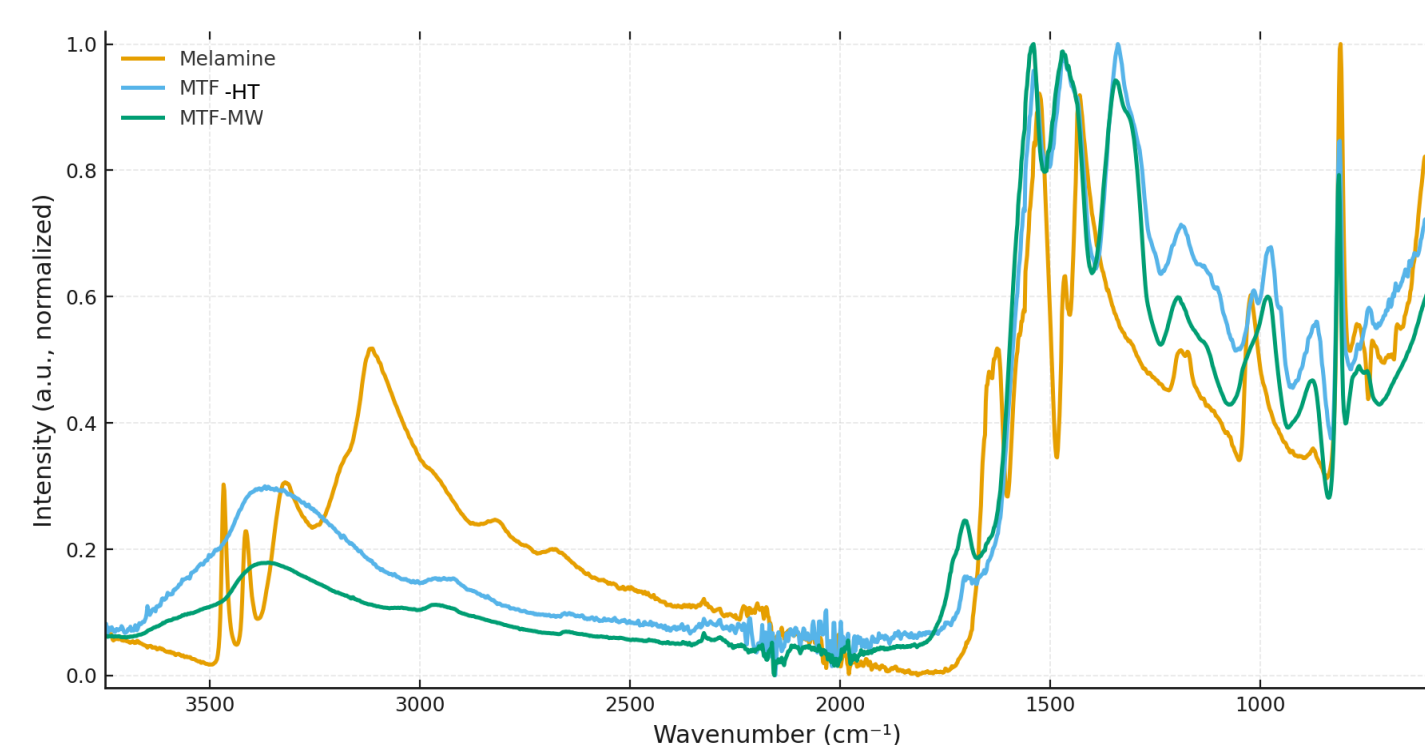
Reactions were performed as nucleophilic substitutions of cyanuric chloride with melamine or pararosaniline hydrochloride. The reactions were conducted under conventional heating during 3 days at 150°C or under microwave irradiation (50–200 W) during 1 hour at the same temperature.



Characterization included FT-IR (linkage formation, triazine retention), TGA (to 600 °C with low-temperature desorption), N₂ sorption at 77 K with BJH pore-size analysis, and CO₂-uptake estimates from ≤150 °C TGA mass loss of preliminary CO₂-saturated CTF samples.

RESULTS & DISCUSSION

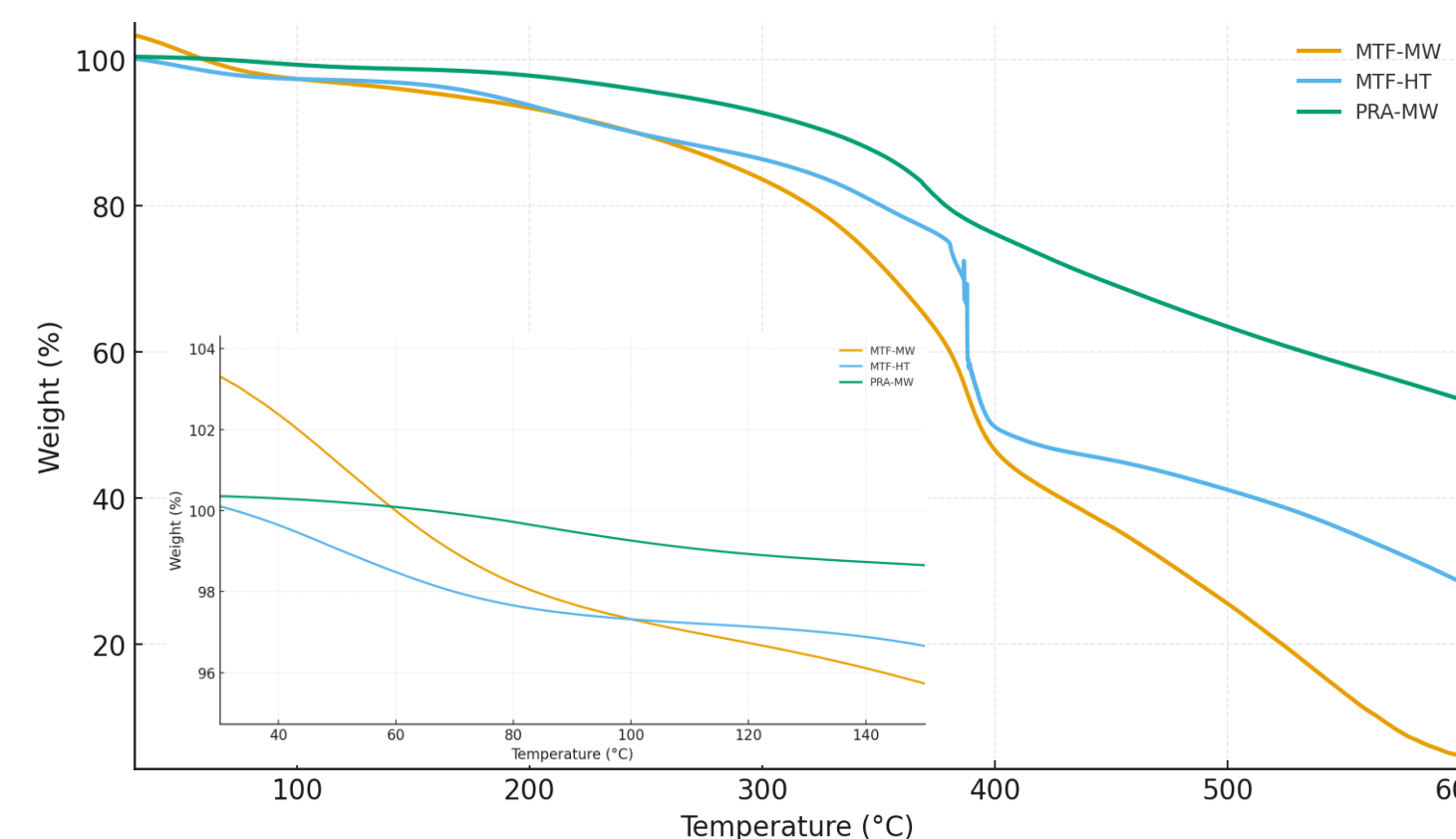
Conventional heating gave the –NH–linked framework **MTF-HT** with 31% yield, whereas microwave irradiation raised the yield of **MTF-MW** more than twice (to 67%). Moreover, the reaction of cyanuric chloride with pararosaniline failed under conventional conditions, but gave the **PRA-MW** in 48% yield under microwave assistance.

FT-IR analysis:

The formation of –NH– bridged CTFs was confirmed by FT-IR for both amines. More specifically, the collapse of the –NH₂ stretching multiplet of both amines (~3500–3250 cm⁻¹) into a broad peak at 3700–3100 cm⁻¹, signaling extensive H-bonding of bridging N–H groups. The given spectrum shows the case of the CTFs obtained from melamine.

BET and BJH measurements:

The results of BET and BJH measurements showed that all obtained frameworks have mesoporous structure. **MTF-HT** (BET surface area ~122.6 m²/g) and **PRA-MW** (BET surface area ~25.4 m²/g) have mesopores with wide size distribution and «bottle-like» forms, however, the latter has low total pore volume (0.1133 cm³/g). **MTF-MW** (BET surface area ~14.1 m²/g) has ordered, «net-like» structure and uniformly distributed mesopores.

TGA analysis:

We investigated thermal stability of obtained frameworks by TGA. All obtained frameworks are stable till 300°C. Thermal stability of synthesized frameworks adheres to the series MTF-MW ≥ MTF-HT > PRA-MW. Following the interest in CO₂-capture properties, we preliminarily saturated the powders of these materials with CO₂ and then explored the substitution of CO₂ with N₂. The values of weight loss of CTFs preliminarily saturated with CO₂, due to nitrogen displacement in 30–150°C temperature range, resulted for **MTF-MW** – 7.6 wt% (corresponds to CO₂ uptake ~4.7 mmol/g), **MTF-HT** – 3.5 wt% (~2.6 mmol/g), **PRA-MW** – 1.7 wt% (~1.1 mmol/g). The maximal weight loss corresponding for **MTF-MW** demonstrates the highest affinity of this framework to CO₂.

CONCLUSION

As a result of current investigation, it was demonstrated that synthesis of CTFs via microwave irradiation:

- Sharply decreases in CTF formation time and boosts yields;
- Facilitates the formation of ordered, «net-like» structures of CTFs;
- Leads to the formation of materials possessing uniformly distributed mesopores;
- Promotes increased CO₂ uptake by the framework.

FUTURE WORK / REFERENCES

FUTURE WORK:

The current work demonstrated the developed synthetic approach could be promising for further directions:

- Optimization and application of microwave synthesis conditions for upscaling of CTFs;
- Expansion of the scope of starting materials for CTF synthesis;
- Tests of functional properties of obtained CTFs for gas separation.

REFERENCES:

1. Liao L. et al., ACS Omega 2023, 8, 5, pp. 4527–4542
2. Silpa E. P. et al., Materials Advances 2024, 5, 23, pp. 9175–9209
3. Chaudhary M. and Mohanty P., New Journal of Chemistry 2018, 42, 15, pp. 12924–12928