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Synthesis and Investigation of Tungsten–Copper Oxide Composites for Enhanced Photocatalytic Applications

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INTRODUCTION & AIM $CB \qquad e^{-} \qquad H_2O \qquad reduction \qquad 2H^+ + 2e^- \rightarrow H_2$ $CB \qquad WO_{2.9} \qquad H^+/H_2$ $CU_2O \qquad WO_{2.9} \qquad CU_2O \qquad WO_{2.9} \qquad CU_2O \qquad CU_2$

- Most renewable energy sources have fluctuating power outputs and are commonly difficult to predict.
- Currently, most industrial H₂ production is extracted from fossil fuels.
- Technological innovations like photochemical water splitting offer a green alternative for H₂ production
- Hydrogen can be stored using existing technologies, making it a viable energy carrier..

Hot Wire Chemical Vapor Deposition

Pyrometer

Power

Supply

- Developing visible-light-driven hydrogen production is key to creating efficient, low-cost photocatalysts
- Tungsten oxide is an efficient at oxidizing H₂O, but ineffective at reducing H₂O.
- Copper oxide is an efficient at reducing H₂O, but ineffective at oxidizing H₂O.
- A Z-scheme geometry between $WO_{2.9}$ and Cu_2O will effectively split H_2O .

Growth Conditions

50

1200

500

15

 P_{0_2} (mTorr)

T_{filament} (°C)

T_{substrate} (°C)

Δt (min)

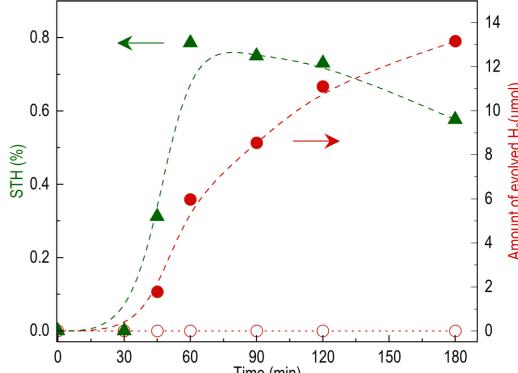
Under solar illumination, the synthesized samples exhibit substantial enhancement, confirming its effectiveness as a proficient photocatalyst

0.1 -0.3

-0.2 -0.1

 $Z'(k\Omega)$

During EIS measurements, we observe a twofold reduction in impedance regardless of whether solar light is present



X-ray photoemission spectroscopy

- Presence of multivalent W (W⁵⁺ and W⁶⁺)
 - Average oxidation state of W^{5.8+}
 - Corresponds to WO_{2.9}
- Presence of single valence Cu (Cu¹⁺)
 - Corresponds to Cu₂O
- Indicates presence of H₂ gas after solar illumination
- Maximum STH efficiency ~1%
- Suppression of STH with time due experiment at fixed volume

RESULTS & DISCUSSION

substrates (Cu₂O//Cu).

tungsten oxide.

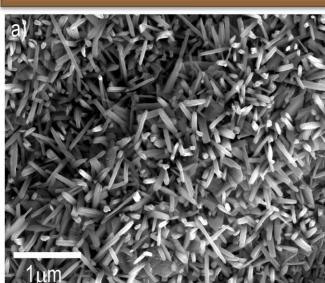
METHOD

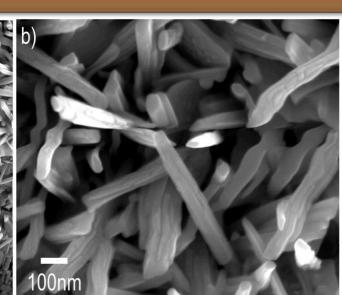
Tungsten filaments are heated

above the vapor temperature of

nanostructures

synthesized on oxidized copper





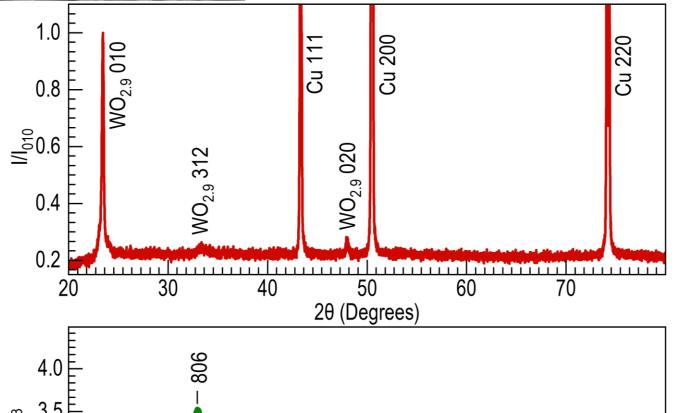
SEM characterization revealed that the copper tungsten oxide thin films exhibit a nanorod-shaped nanostructure with a diameter of ~50 nm

X-ray diffraction

- WO_{2.9} is dominant phase
- WO_{2.9} has preferential growth along b-axis
- Cu₂O signal is likely screened by WO_{2,9} nanostructures

Raman

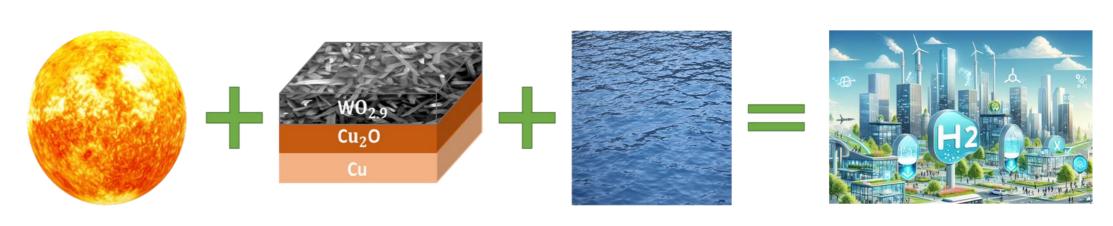
- Sensitive to M-O-M bond stretching and bending
- Features *above* 240 cm⁻¹ attributed to WO_{2.9}
- Features below 240 cm⁻¹ attributed to Cu₂O



20 30 40 50 60 70 20 (Degrees) 4.0 (SOLUTION 100) (SOL

CONCLUSION/ FUTURE WORK

- We successfully fabricated $WO_{2.9}$ nanostructures on Cu_2O thin films on Cu substrates utilizing the HWCVD technique.
- The samples were photo-catalytically active with an STH efficiency of about 1% without any external bias voltage in water splitting testing in deionized water under illumination of simulated solar light.
- The simple synthesis and alternative material for hydrogen production offer a cost-effective approach, showcasing potential for advanced photocatalysts in next-generation fuel production.



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