

Synthesis of New Nicotinamides Starting from Monothiomalonanilide [†]

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Abstract

It is known that thioamides containing hydrogen atoms in the α -position of the thiocarbamoyl group exhibit a CH-acidic character. The purpose of the work: synthesis of new nicotinamide derivatives based on monothiomalonanilide and investigation of their properties. According to the known method, we synthesized the initial monothiomalonanilide starting from phenyl isothiocyanate and acetylacetone. The reaction of monothiomalonanilide with arylidenemalonitriles, available by the Knoevenagel reaction, or with aromatic aldehydes and malonitrile in the presence of a base leads to the formation of cyclic Michael adducts—nicotinamide derivatives. New compounds have been studied spectrally (FTIR, NMR ¹H, ¹³C).

Keywords: thioamides; active methylene compounds; nicotinonitriles; nicotinamides

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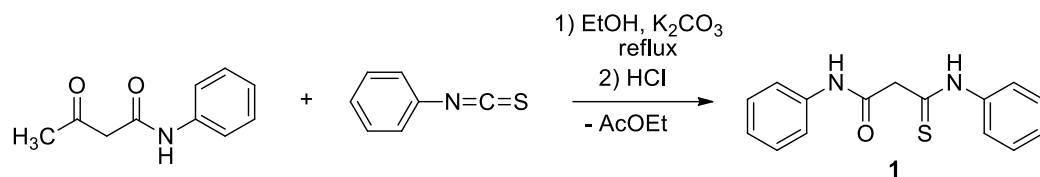
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1. Introduction

The chemistry of thioamides containing hydrogen atoms in the α -position to the thiocarbamoyl group represents a promising area of modern organic synthesis, owing to their considerable synthetic potential and the valuable practical properties of the obtained products [1]. Despite its multifunctional nature and obvious potential for diverse transformations, the reactions of monothiomalonanilide **1** remain insufficiently explored. Only a few papers were reported to date dealing with the chemistry of monothiomalonanilide **1** [2–7]. We therefore set out to investigate some new reactions.

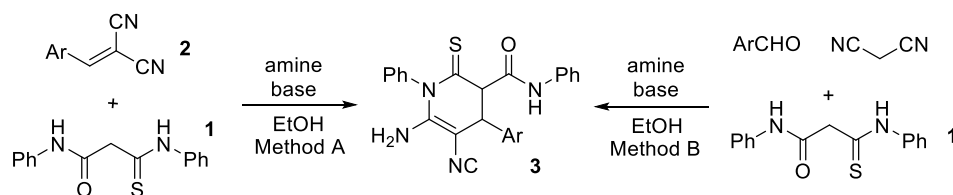
2. Results and Discussion

Using a reported procedure [8], the starting monothiomalonanilide **1** was synthesized by reacting commercially available acetoacetanilide with phenyl isothiocyanate (Scheme 1). The structure of compound **1** was confirmed by spectral data as well as X-ray diffraction analysis.



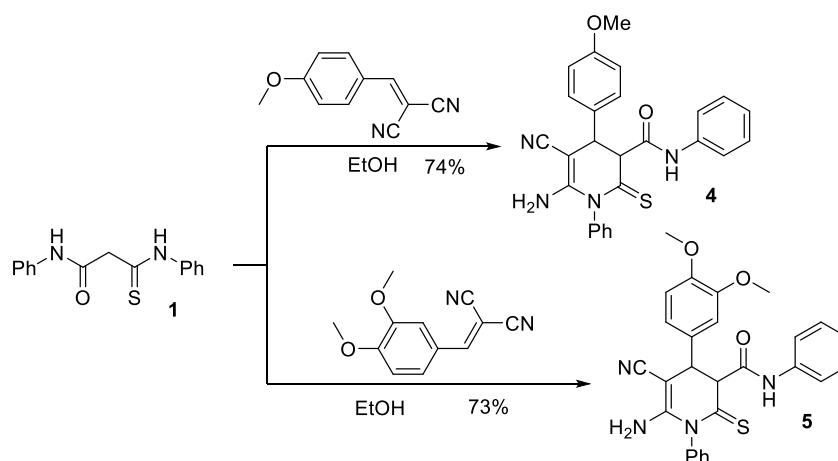
Scheme 1. XXX.

The reaction of thioamide **1** with arylidenemalononitriles **2**, either prepared via the Knoevenagel condensation (Method A), or in a multicomponent reaction involving aromatic aldehydes and malononitrile in the presence of a base (Method B), afforded cyclic Michael adducts, previously unreported nicotinamides **3** (Scheme 2).



Scheme 2. XXX.

We studied the capabilities of approaches **A** and **B** and found that they produce comparable yields. The highest yields of the target products were obtained for arylidene malononitriles with donor substituents. Thus, the reaction with (3,4-dimethoxybenzylidene)malononitrile and (4-methoxybenzylidene)malononitrile afforded products **4** and **5**, respectively (Scheme 3). In the future, it is planned to expand the library of products, as well as study their properties.



Scheme 3. XXX.

The structures of new compounds were also confirmed by spectral studies. For instance, Figures 1 and 2 show the ¹H and ¹³C NMR spectra of compound **3** (Ar = 2-thienyl).

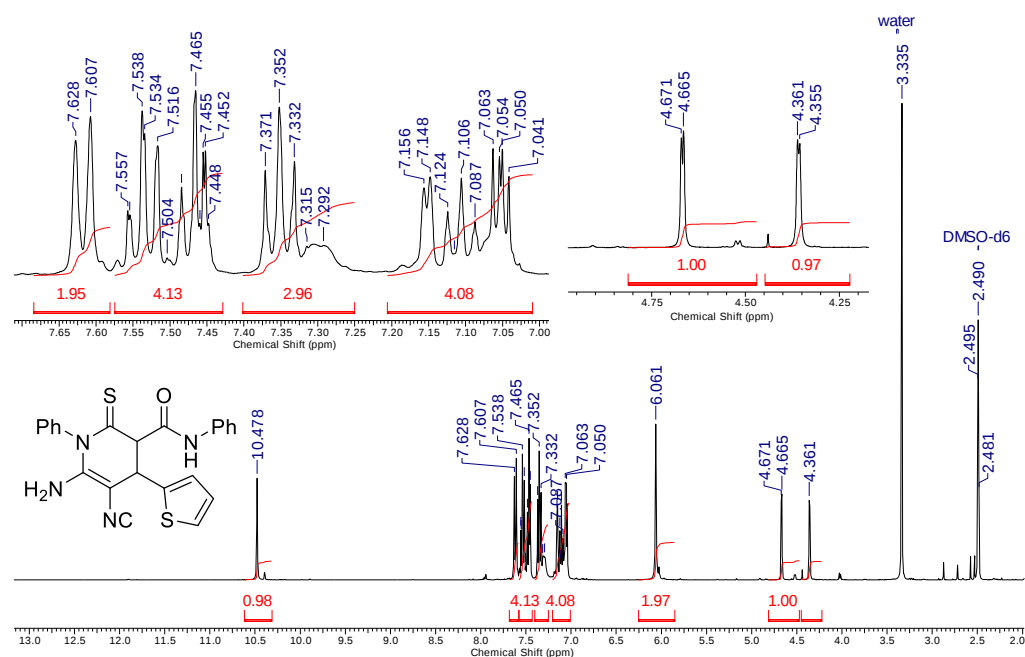


Figure 1. ^1H NMR spectrum of compound **3** (Ar = 2-thienyl).

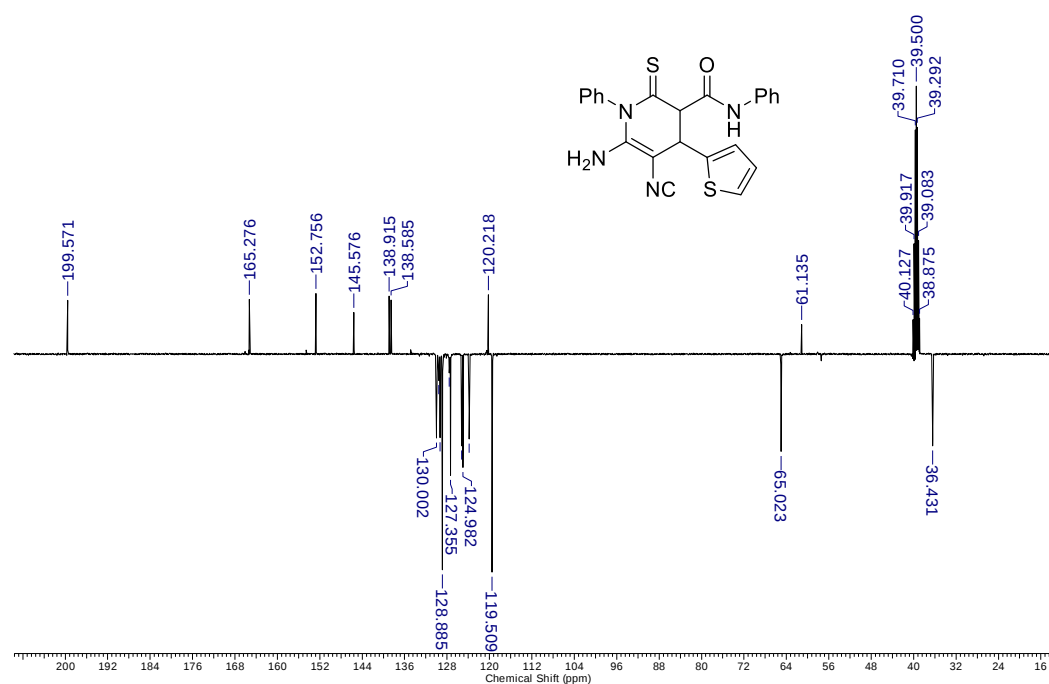


Figure 2. ^{13}C DEPTQ NMR spectrum of compound **3** (Ar = 2-thienyl).

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According to the results of molecular docking using the GalaxyWeb Sagittarius protocol, one of the products has affinity ($\Delta G = -22.586$ kcal/mmol) to the protein complex H-Ras:SOS (PDB ID 6pf6, UniprotIDP04818), which indicates the potential prospects of studying the antitumor properties of the compound **3** (Ar = 4-MeOC₆H₄) (Figure 3).

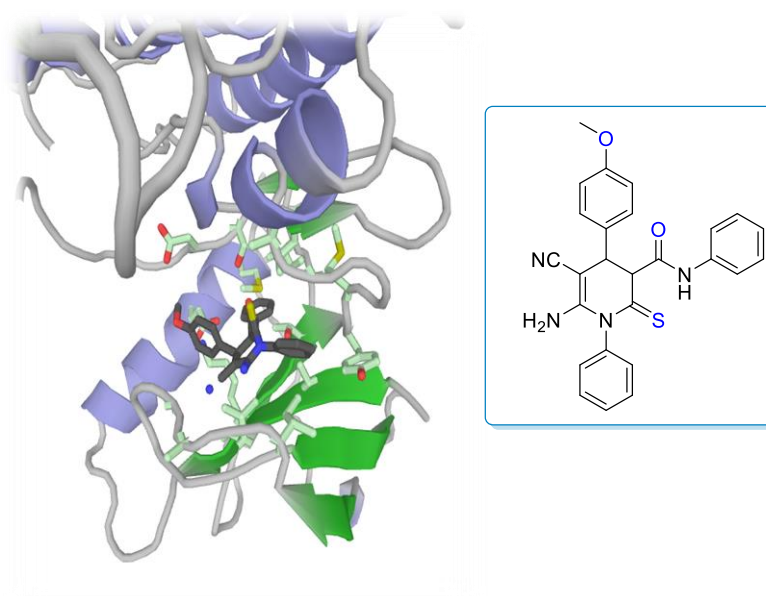
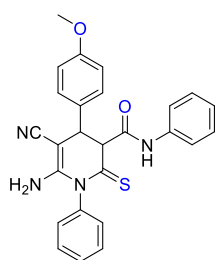


Figure 3. The predicted pose for the protein-ligand complex of H-Ras:SOS (PDB ID 6pf6, Uniprot-IDP04818) with compound **3** (Ar = 4-MeOC₆H₄).

3. Experimental

Synthesis of compounds **3 (Approach A).** Arylidene malononitrile (0.03 mol) was dissolved in 12 mL of EtOH with gentle heating. Then monothiomalonanilide **1** (0.03 mol) was added followed by treatment with 0.03 mol of organic base (diethylamine, morpholine etc.) The mixture was stirred until starting reagent were consumed (TLC control), and left to stand overnight for crystallization. A yellow, fine crystalline powder was filtered off and washed with EtOH to give **3**. For analytical purposes, the product can be recrystallized from suitable solvent (acetone, MeCN, dioxane).

6-Amino-5-cyano-4-(4-methoxyphenyl)-N,1-diphenyl-2-thioxo-1,2,3,4-tetrahydropyridine-3-carboxamide (3**)** Ar = 4-MeOC₆H₄



NMR ¹H spectrum (400 MHz, DMSO-*d*₆), δ_H, ppm: 3.75 s (3H, MeO), 4.03 d (1H, CH Ar), 4.55 d (1H, C³H), 5.94 br.s (2H, NH₂), 6.99-7.62 m (14H, 3 Ar), 10.40 br.s (1H, CONH).

NMR ¹³C DEPTQ (DMSO-*d*₆), δ_C, ppm: 39.6* (CH); 55.1* (OCH₃); 60.6 (C⁵); 64.9* (CH); 114.1* (2C, C³H C⁵H 4-MeOC₆H₄); 119.5* (2C, CH Ar); 120.5 (CN); 123.8* (CH Ph); 127.4* (CH Ph); 128.3* (2CH Ph); 128.8* (2CH Ph); 129.6* (2CH Ph); 130.0* (2CH Ph); 132.5 (C¹ Ar); 138.7 (C¹ Ar); 139.0 (C¹ Ar); 152.6 (C⁶); 158.5 (C-OMe); 165.9 (CONH); 199.9 (C=S).

*Signals in antiphase.

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