

Covalent functionalization of fullerene C₆₀ with polyethyleneimine

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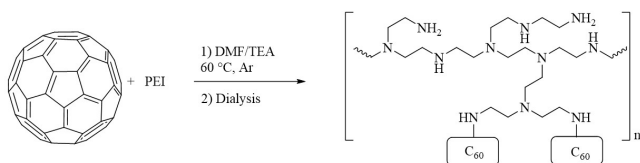
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INTRODUCTION & AIM

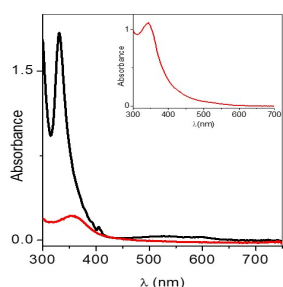
Several families of photosensitizers (PSs) have been evaluated for microbial inactivation [1,2]. These compounds are photostable, absorb visible light, and generate reactive oxygen species (ROS), but their poor solubility causes aggregation that limits photoactivity. Covalent functionalization, including polycationic fragments, enhances solubility, biological activity, and bacterial membrane interaction. The development of covalent functionalization strategies for C₆₀ fullerene has enabled the incorporation of chemical groups capable of improving its biological activity [2]. The inclusion of polycationic fragments in the PS structure increases its ability to interact with bacterial membranes, enhancing microbial inactivation. Polyethylenimine (PEI) is a polycationic polymer with protonable amino groups that confer positive charges, enabling interaction with microbial membranes [3]. Based on this property, a new C₆₀-PEI conjugate was synthesized as a potential photodynamic agent.

RESULTS & DISCUSSION

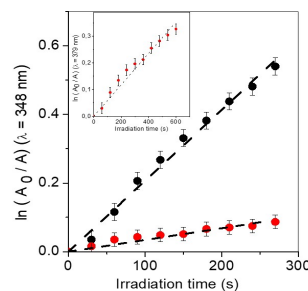
Fullerene derivative functionalized with polyethyleneimine (C₆₀-PEI) was synthesized from the reaction between C₆₀ fullerene and PEI using *N,N*-dimethylformamide (DMF) as solvent and in the presence of triethylamine (TEA) (Scheme 1). The reaction mixture was kept under stirring at 60 °C and subsequently purified by dialysis in aqueous solution, yielding the C₆₀-PEI conjugate as the final product [3].

Scheme 1. Synthesis of C₆₀-PEI

Spectroscopic studies of C₆₀-PEI were carried out and compared with reference C₆₀ fullerene using DMF and water as solvents (Figure 1). The conjugate showed weaker absorption bands in the 300–800 nm range, which are characteristic of substituted C₆₀ fullerenes (Figure 1) [1,3]. The absorption spectrum of C₆₀-PEI in water confirms the presence of PEI in the fullerene structure.

Figure 1. Absorption spectra of C₆₀-PEI (red solid line) and C₆₀ (black solid line) in DMF. Inset: absorption spectra of C₆₀-PEI in water.

The photodynamic properties of the fullerene derivative were evaluated in DMF and water using anthracene derivatives: 9,10-dimethylantracene (DMA) and 2'-(anthracene-9,10-diyl)bis(methylmalonate) tetrasodium salt (ABMM) [3]. From the observed rate constants (*k*_{obs}), singlet oxygen quantum yields (Φ_{Δ}) for C₆₀-PEI were determined, yielding values of 0.15 in DMF and 0.70 in water (Figure 2). The high Φ_{Δ} of the PS in water compared to the organic medium could indicate, in addition to considerable photodynamic activity, an association between the cationic PS and the anionic substrate ABMM. Nevertheless, these results demonstrate that the conjugate exhibits photodynamic activity in both organic and aqueous media, suggesting that the incorporation of the polymer improves the solubility of C₆₀ fullerene in polar systems.

Figure 2. First-order plots for the photooxidation of DMA photosensitized by C₆₀ (●) and C₆₀-PEI (●) in DMF. Inset: First-order plots for the photooxidation of ABMM photosensitized by C₆₀-PEI (●) in water. λ_{irr} = 450–700 nm.

CONCLUSION

A C₆₀ fullerene derivative modified with a polymer precursor of positive charges was synthesized to be used as a potential photosensitizer for the photodynamic inactivation of microorganisms. Absorption spectroscopic studies indicated that the conjugate exhibits a characteristic spectrum of substituted fullerenes and that the presence of PEI allows the solubilization of fullerene in water. Likewise, photodynamic studies confirmed that C₆₀-PEI generates oxygen singlet (O₂(¹ Δ_g)) in both organic and aqueous media. Taken together, the results suggest that C₆₀-PEI possesses suitable properties for application as a photodynamic agent in microbial inactivation.

FUTURE WORK / REFERENCES

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