

Synthesis and New Reactions of 3,6-Diaminothieno[2,3-b]pyridine-5-carbonitriles [†]

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Abstract

6-Aminopyridine-3,5-dicarbonitriles and 3,6-diamino-5-cyanothieno[2,3-b]pyridines are well known as compounds with a broad spectrum of bioactivity. In particular, such thienopyridines are known as inhibitors of scrapie prion infection replication and accumulation, as well as selective inhibitors of malaria plasmodia kinase-3 with a pronounced antimalarial effect. To expand the range of such compounds, we studied the reaction of 3,6-diaminothieno[2,3-b]pyridine-5-carbonitriles with chloroacetyl chloride. The analysis of the Fukui indices showed that in these compounds the amino group at the C(3) atom is the most reactive. In fact, the reaction with $\text{ClCH}_2\text{C}(\text{O})\text{Cl}$ leads to the predicted product. Other reactions of thienopyridines as well as data on the biological activity of the products are discussed.

Keywords: thienopyridines; acylation; biological activity

1. Introduction

The chemistry of thieno[2,3-b]pyridine derivatives still remains a rapidly developing area of research, which has repeatedly become the subject of detailed consideration in a number of papers and reviews and dissertations [1–6]. In recent years, significant progress has been made in this area of heterocyclic chemistry, which is reflected in a significant number of new publications concerning methods for the preparation, modification, and especially issues of biological activity of thieno[2,3-b]pyridine derivatives. It is important to note that thieno[2,3-b]pyridine derivatives are used in various fields, which confirms their importance and prospects for further research. In this regard, further study of these compounds plays an important role in the development of scientific knowledge and practical application in the modern world.

3,6-Diamino-5-cyanothieno[2,3-b]pyridines are of great importance due to their diverse biological activities. For example, based on 3,6-diamino-5-cyanothieno[2,3-b]pyridines, drugs have been created for the treatment of Huntington's disease and dengue fever, and many compounds also have high antimicrobial and antibacterial activity [7,8]. Their synthetic accessibility also makes them valuable scaffolds for further

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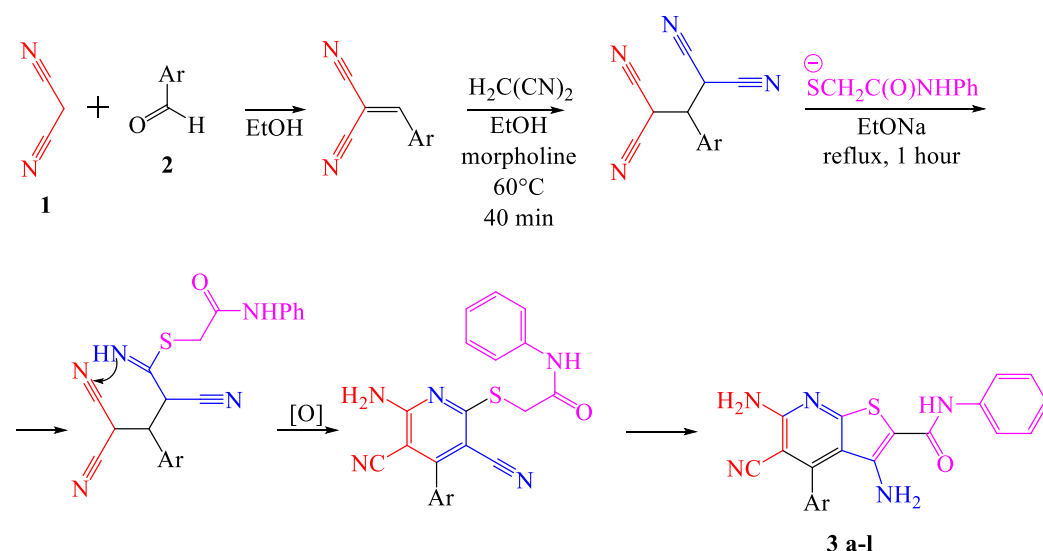
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transformations. In this work, we report the synthesis of starting compounds, their selective acylation, and intramolecular cyclization.

2. Results and Discussion

The starting 3,6-diaminothiopyridines were prepared in a one-pot, two-step process. In the first step, 2 equivalents of malononitrile react with an aromatic aldehyde to form Michael adducts. Mercaptoacetanilide was then added to the reaction, initiating the cascade process of assembling the thienopyridine system (Scheme 1) [9].



Scheme 1.

The ^1H NMR spectrum shows signals for the protons of the NH_2 group, located in the third position of the thienopyridine system: a singlet at δ 5.80–5.99 ppm, signals for the protons of the NH_2 group, located in the sixth position: a broadened singlet at δ 7.42–7.57 ppm and signals for the protons of the amide group: a singlet at δ 9.28–9.45 ppm.

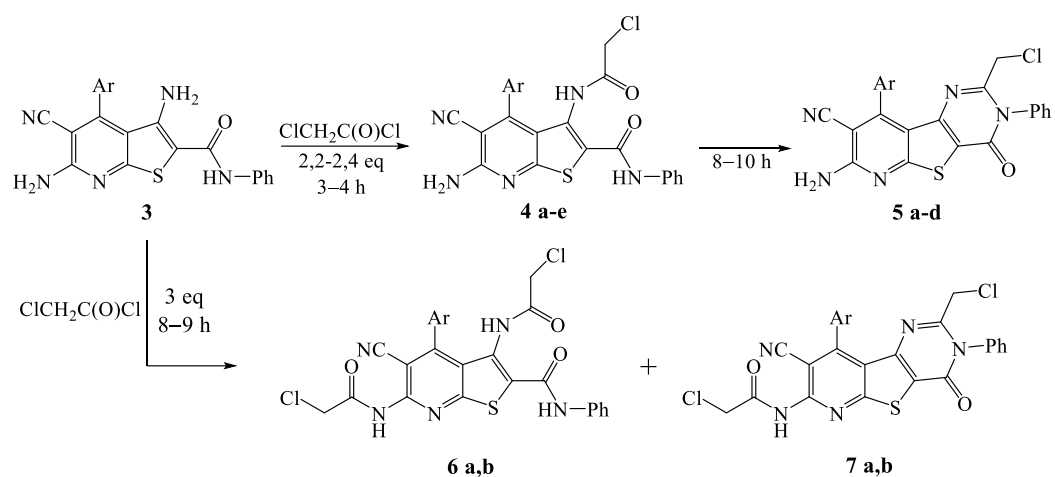
It is known that 3-aminothieno[2,3-b]pyridines are quite easily acylated [10]. 3,6-Diamino-5-cyanothieno[2,3-b]pyridines have two amino groups, so it seemed appropriate to study their behavior under this reaction conditions and to investigate the regional direction of the process.

An analysis of the Fukui indices showed that the amino group at the C(3) atom is the most reactive in compounds **3**. Therefore, it was expected that the reaction with ClCH_2COCl would proceed predominantly at C(3) NH_2 . Indeed, the reaction with 2.2–2.4 equivalents of chloroacetyl chloride for 3–4 h leads to the predicted product **4** (Scheme 2).

Further heating for 8–10 h leads to intramolecular cyclization and the formation of product **4**. It was also found that when using 3 equivalents of the ClCH_2COCl and refluxing for 8–9 h, the reaction can proceed at the amino group C(6) NH_2 , as evidenced by spectral data (IR, NMR ^1H and ^{13}C). This produces a mixture of four acylation products **4–7** in approximately equal proportions (Figure 1).

The ^1H NMR spectra are characterized by the presence of a signal of methylene protons in the form of a singlet for compounds **4** in the region of δ 3.35–3.54 ppm, and for compounds **5** in the region of δ 3.99–4.04 ppm.

The evidence of obtaining a mixture of four products in equal proportions using 3 equivalents ClCH_2COCl and refluxing for 8–9 h is provided by the set of data from the ^1H and ^{13}C NMR spectra, based on which it can be concluded that the mixture of products contains signals of four $-\text{OCH}_3$, six $-\text{NH}_2$ and six $-\text{CH}_2-$ groups (Figure 1).



Scheme 2.

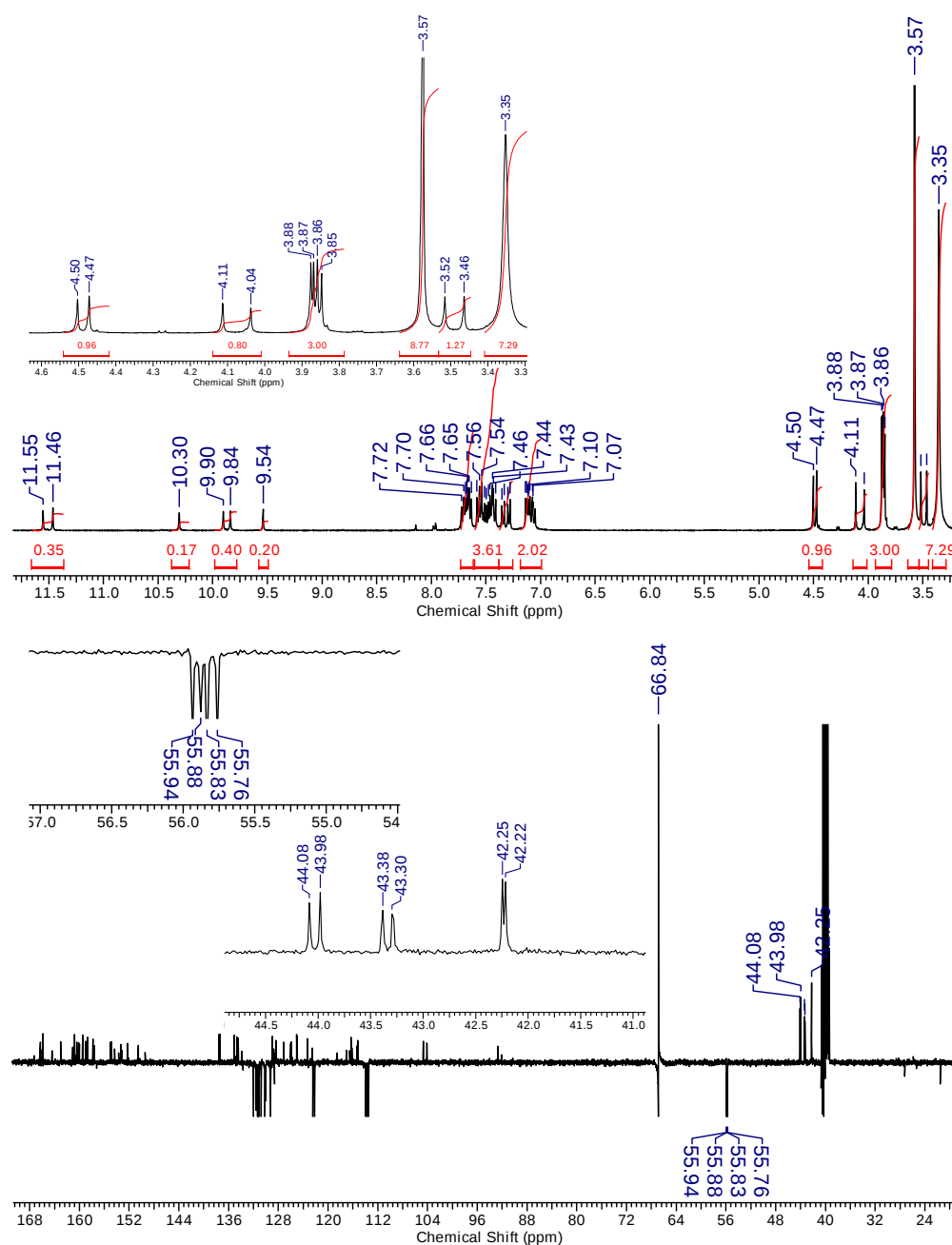


Figure 1. NMR data set ^1H and ^{13}C for a mixture of compounds 4–7.

The starting 3,6-diamino-5-cyanothieno[2,3-b]pyridines were tested for activity against the new coronavirus infection. The compounds were found to be inactive to SARS-CoV-2 infection model. Almost all compounds are poorly soluble in common organic solvents. Modifications to increase solubility and bioavailability are required.

Potential protein targets for the obtained compounds were predicted using the new protein-ligand docking protocol GalaxySagittarius [11] based on the GalaxyWeb web server. All synthesized 3,6-diaminothieno[2,3-b]pyridine-5-carbonitriles are inhibitors of Hsp90 (PDB ID 3wha), a chaperone protein that stabilizes oncoproteins, making them potential antitumor agents (Figure 2).

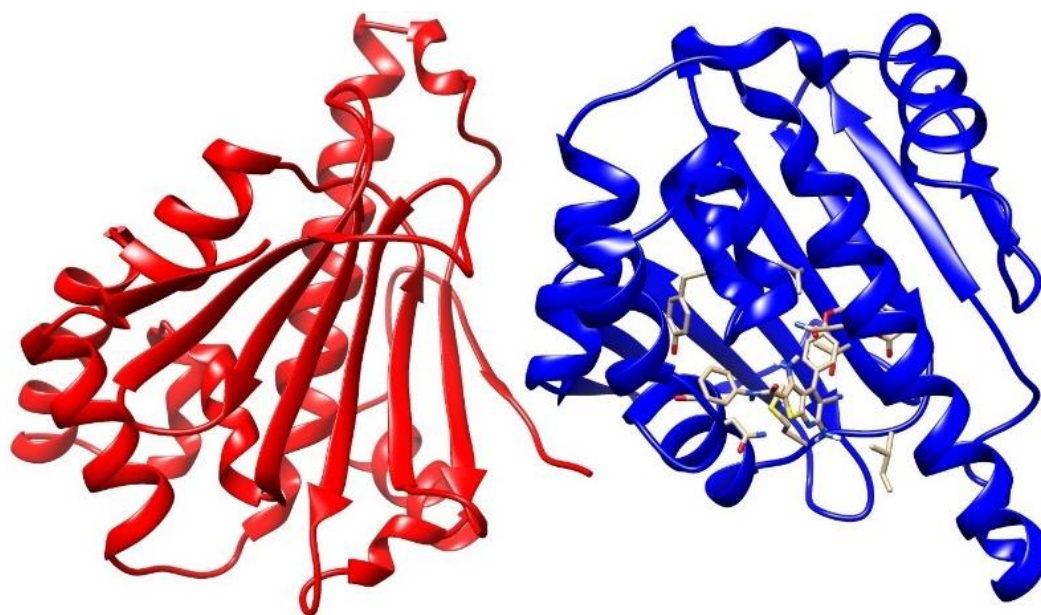


Figure 2. Predicted structure of the protein-ligand complex for compound 4g and the chaperone protein Hsp90 (PDB ID 3wha).

According to molecular docking data, acylated thienopyridines 4 bind to the Ephrin type-A receptor 2 (EPHA2) (PDB ID 5nkh). This receptor is responsible for the resistance of BRAF-mutant melanomas (where the mutation is caused by a thymine to adenine transversion at nucleotide 1799 of the gene) to BRAF inhibitors and for the entry of Kaposi's sarcoma-associated herpesvirus (KSHV) into cells. Furthermore, these compounds inhibit the Replication Protein A (RPA) (PDB ID 4lwc), disrupting DNA repair in cancer cells and increasing their sensitivity to damage.

3. Experimental

Synthesis of 3,6-diamino-4-(2-chlorophenyl)-5-cyano-N-phenylthieno [2,3-b]pyridine-2-carboxamide (3a): a solution of 2 g (0.014 mol) of 2-chlorobenzaldehyde and 1.88 g (0.028 mol) of malononitrile in 15 mL of ethyl alcohol is placed in a flat-bottomed flask. Four drops of morpholine are added. The reaction mixture is heated to 60 °C with stirring. After some time, a precipitate of arylidene malononitrile forms. To ensure complete formation of the reaction product, the reaction mixture is stirred with heating for 40 min. For the second stage of the reaction, a solution of sodium ethoxide is prepared (the calculated amount of metallic sodium is dissolved in anhydrous ethyl alcohol). Then, 2.37 g (0.014 mol) of mercaptoacetanilide is added to the sodium ethoxide. The resulting mixture is then added to the reaction mixture containing the Michael adduct, 2-aryl-1,1,3,3,-

tetracyanopropane. The precipitate dissolves immediately, and after some time, crystals of the product begin to form. The synthesis is carried out for one hour at 100 °C. The precipitate was filtered off and washed with cold alcohol. Yield was 57%. M.p. 254–255 °C. FT-IR, ν , cm^{-1} : 3456, 3408 (NH_2); 3302 (NH); 2210 (CN), 1633 ($\text{C}=\text{O}$).

NMR ^1H spectrum (400 MHz, $\text{DMSO}-d_6$), δ_{H} , ppm: 5.68 s (2H, NH_2); 7.05 dd (1H, $J = 7.33$, $J = 1.4$, H_{ph}); 7.29 dd (2H, $J = 8.32$, $J = 1.4$, H_{ph}); 7.57 s (2H, NH_2); 7.58–7.71 m (5H, H_{ph}); 7.77 d (1H, $J = 7.33$, H_{ph}); 9.36 s (1H, NH). NMR ^{13}C DEPTQ ($\text{DMSO}-d_6$), δ_{C} , ppm: 90.69 (C5), 93.64 (C2), 114.33 (C3a), 115.64 (CN), 121.62 (C2'', 6''), 123.83 (C4''), 128.68 (C4'), 128.85 (C3', 5'), 130.60 (C3'', 5''), 131.81 (C1''), 132.58 (C6'), 132.73 (C3), 139.36 (C2'), 147.57 (C1'), 149.69 (C4), 159.1 (C7a), 164.08 (C6), 164.27 ($\text{C}=\text{O}$).

3,6-Diamino-4-aryl-5-cyano-N-phenylthieno[2,3-b]pyridine-2-carboxamides **3b–3l** were prepared using a similar procedure.

6-amino-3-(2-chloroacetamido)-4-(2-chlorophenyl)-5-cyano-N-phenylthieno[2,3-b]pyridine-2-carboxamide (4a): A round-bottomed flask was charged with 1 g (0.00238 mol) of 3,6-diamino-4-(2-chlorophenyl)-5-cyano-N-phenylthieno[2,3-b]pyridine-2-carboxamide (**3a**) and 30 mL of anhydrous dioxane and heated until completely dissolved. Then 0.42 mL (0.00524 mol) of chloroacetyl chloride was added. The reaction mixture is refluxed for 3–4 h. The progress of the synthesis is monitored by TLC (acetone: petroleum ether 2:1). The precipitate was filtered off and purified by recrystallization from acetone. Light yellow crystals were obtained in 51% yield.

6-Amino-3-chloroacetylaminothieno[2,3-b]pyridine-2-carboxamides (**4b–4e**) were obtained by a similar procedure. Purification of some compounds (**4**) was carried out by recrystallization from dioxane. M.p. 233–234 °C. FT-IR, ν , cm^{-1} : 3455 (NH_2); 3328, 3214 (NH); 2216 (CN), 1670, 1618 ($\text{C}=\text{O}$). NMR ^1H spectrum (400 MHz, $\text{DMSO}-d_6$), δ_{H} , ppm: 3.35 d (1H, $J = 14.18$, CH_2); 3.44 d (1H, $J = 14.30$, CH_2); 7.10 dd (1H, $J = 7.34$, $J = 1.9$, H_{ap}); 7.30–7.37 m (3H, H_{ph}); 7.47–7.55 m (2H, H_{ph}); 7.56 s (2H, NH_2); 7.62 d (1H, $J = 7.83$, H_{ph}); 7.65 (1H, Δ, Δ , $J = 0.85$, $J = 7.95$, H_{ap}); 7.95 dd (1H, $J = 1.23$, $J = 8.44$, H_{ap}); 9.635 s (1H, NH); 9.835 s (1H, NH).

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