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Arylalkylamine Derivatives as Myeloperoxidase Inhibitors, Synthesis and Pharmacological Activity

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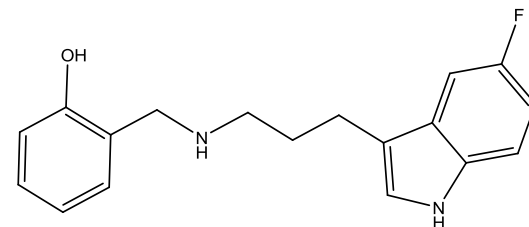
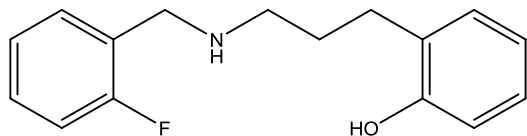
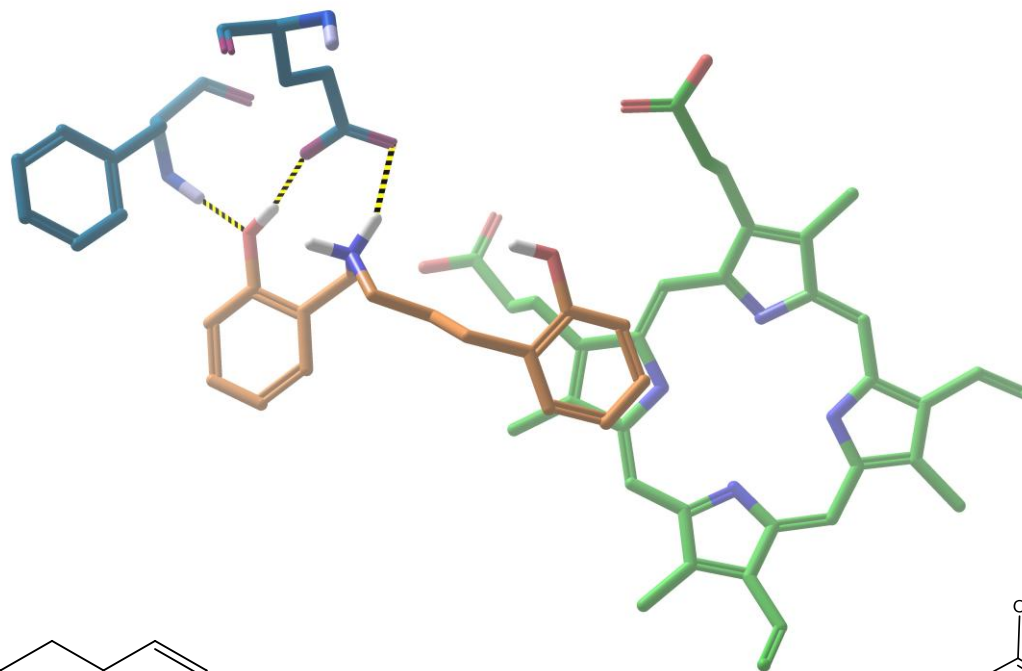


Feel inspired



Wallonia.be

Arylalkylamine Derivatives as Myeloperoxidase Inhibitors, Synthesis and Pharmacological Activity



Abstract: Myeloperoxidase (MPO) is an important target for drug design because of its contributing role in many inflammatory syndromes such as atherosclerosis, rheumatoid arthritis, end-stage renal disease or neurodegeneration. Rational drug design assisted by virtual screening is an interesting tool to design new chemical entities that could inhibit MPO. After a high throughput virtual screening of a database, bis-2,2'-[(dihydro-1,3(2H,4H)-pyrimidinediyl)bis(methylene)]phenol was chosen as a starting hit and we used different strategies of chemical synthesis to perform pharmacomodulation described by the three approaches. This led to 36 compounds that have been assessed in an in vitro inhibition MPO test. We found that the arylalkylamine compounds were active but to a lesser extent than the starting hit. Exception for propylamine derivatives with a phenyl cycle should be noticed. As indolic compounds have demonstrated interesting inhibiting properties, we combined indole ring with the phenolhydropyrimidine structure which led to compounds more active than the hit. Among them, propylamine derivatives were new MPO inhibitors with a nanomolar IC₅₀. Kinetics studies for the most potent inhibitors were conducted and reflected a fast reaction with compound I resulting in the accumulation of compound II Structure-activity.

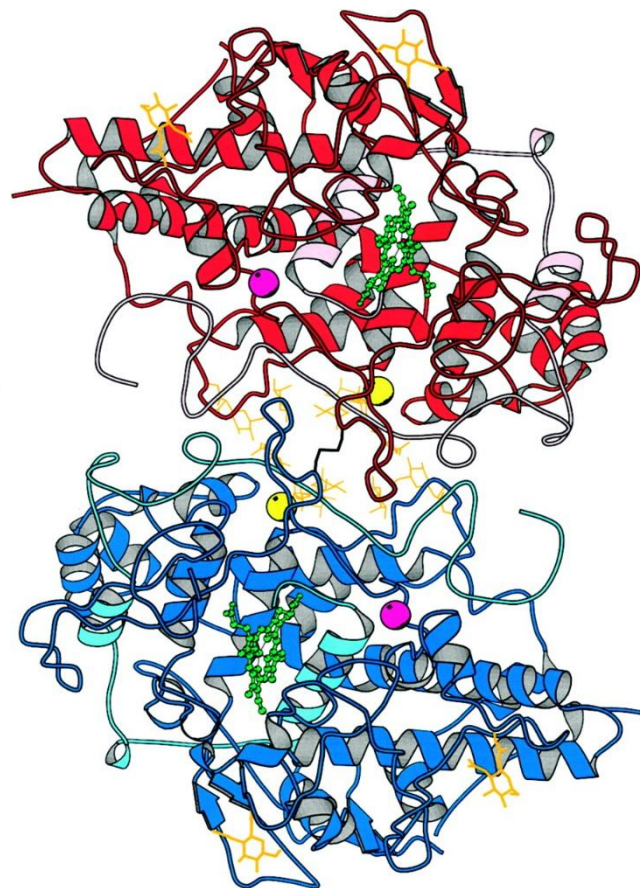
Keywords: Myeloperoxidase; Inhibitors; Arylalkylamine



Introduction



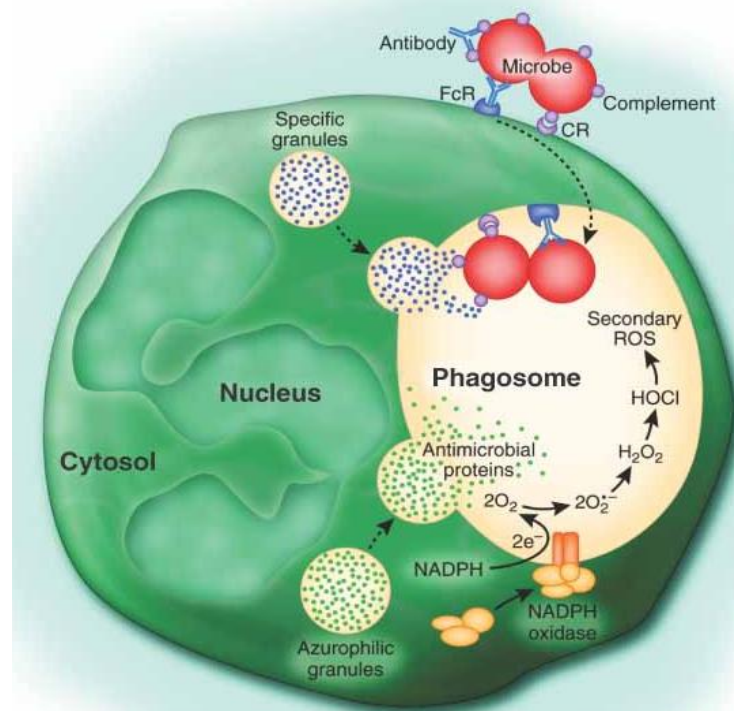
MPO



Introduction

- Neutrophils, monocytes, immune defense system
- Phagocytosis
- Kills microorganisms
- Produces HOCl

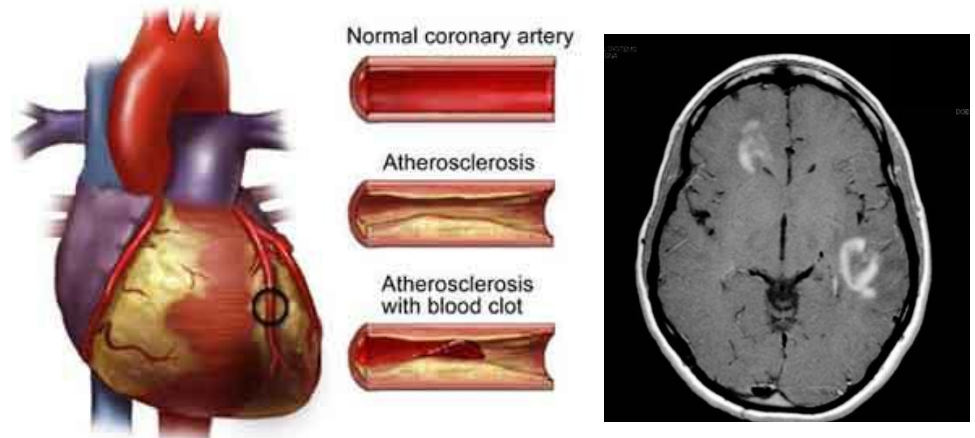
Myeloperoxidase MPO



Introduction

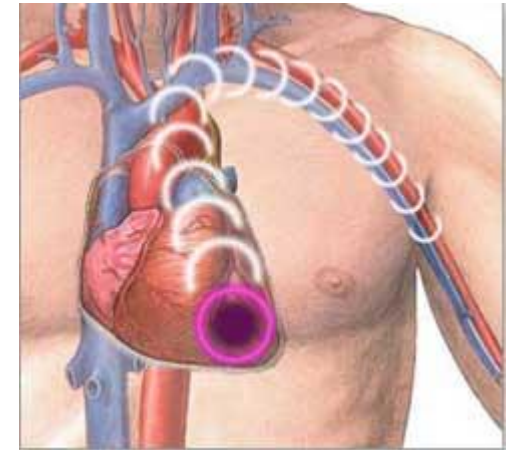
MPO is a contributing factor in many **inflammatory syndromes** such as:

- Atherogenic lesions
- Rheumatoid arthritis
- End-stage renal disease
- Neurodegeneration



Atherosclerosis

Multiple sclerosis



Parkinson

CVD



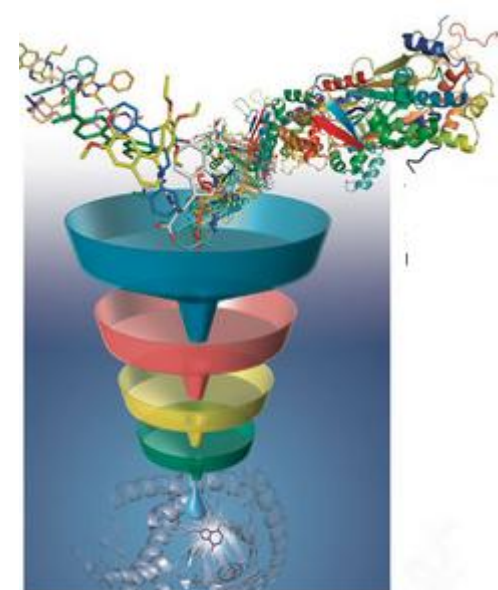
Finding New MPO inhibitors



MPO



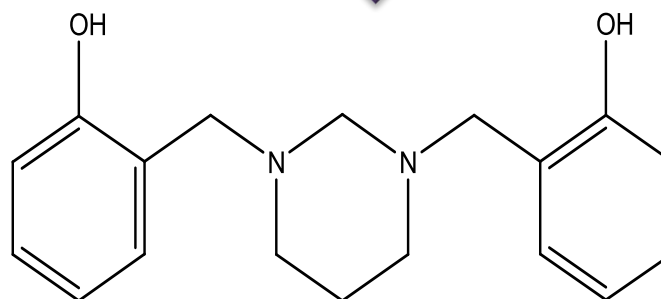
Pharmaceutical
Database



HTVS

8 HITS of MPO inhibitors

Selected Hit for
pharmacomodulation



A1
IC₅₀ = 0.5 μM



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Results and discussion

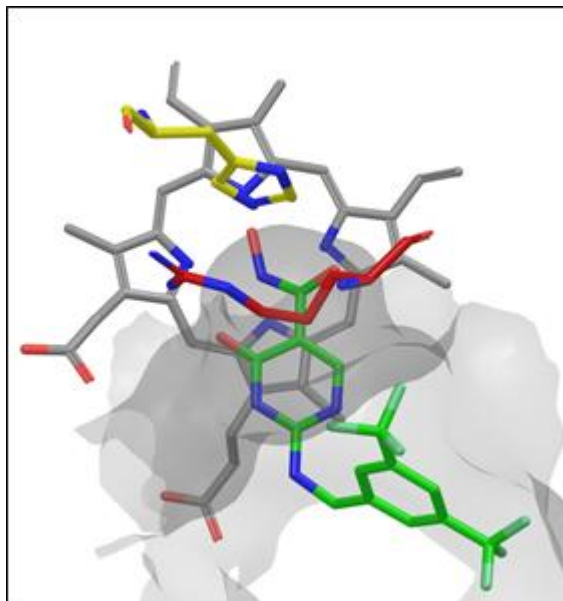
Pharmacomodulation and docking



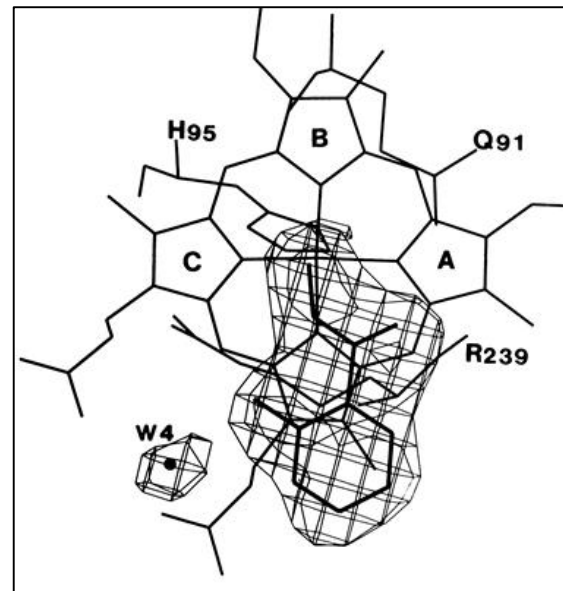
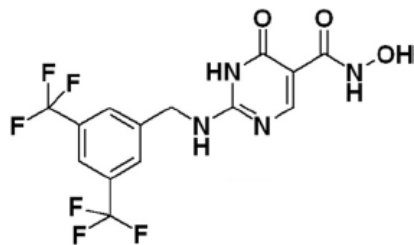
Results and discussion

Docking

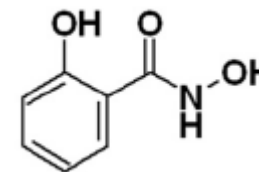
Validation
of docking
using poses
in HX1,
SHA
X-ray data



HX1 IN PDB 4C1M



SHA in PDB 1DNW



Pharmacomodulation and docking

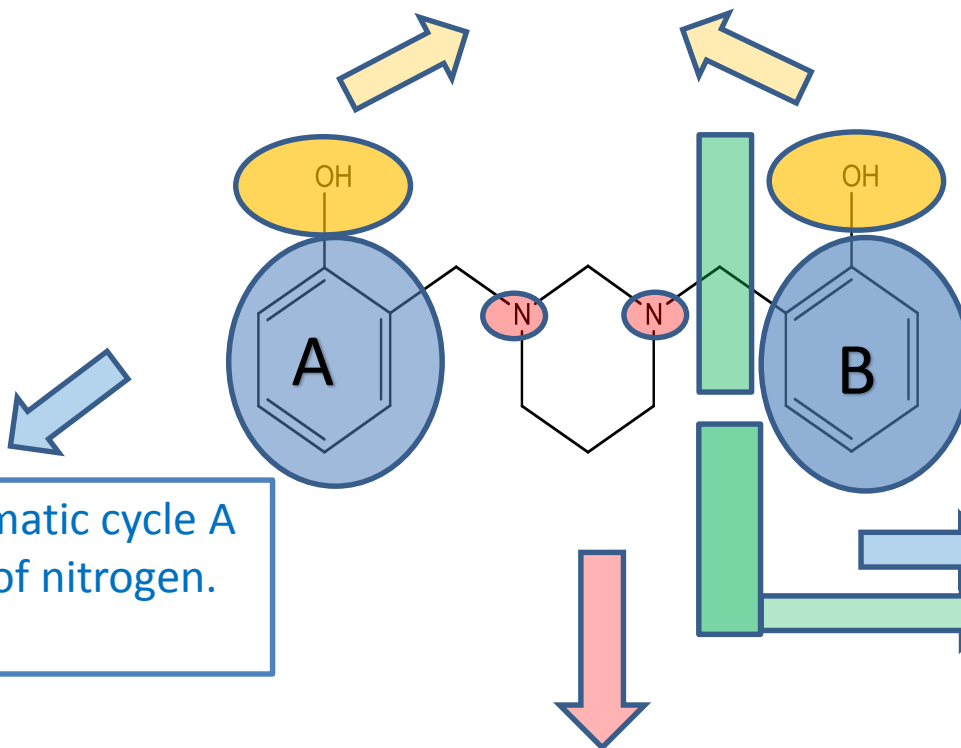
All compounds were designed and docked in
MPO receptors 1DNW -4C1M

Best poses of the docked compounds were
compared with X-Ray data of the known
inhibitors HX1 and SHA
And redocked in same receptors



Pharmacomodulation

The role of hydroxyl groups on both aromatic cycles A and B



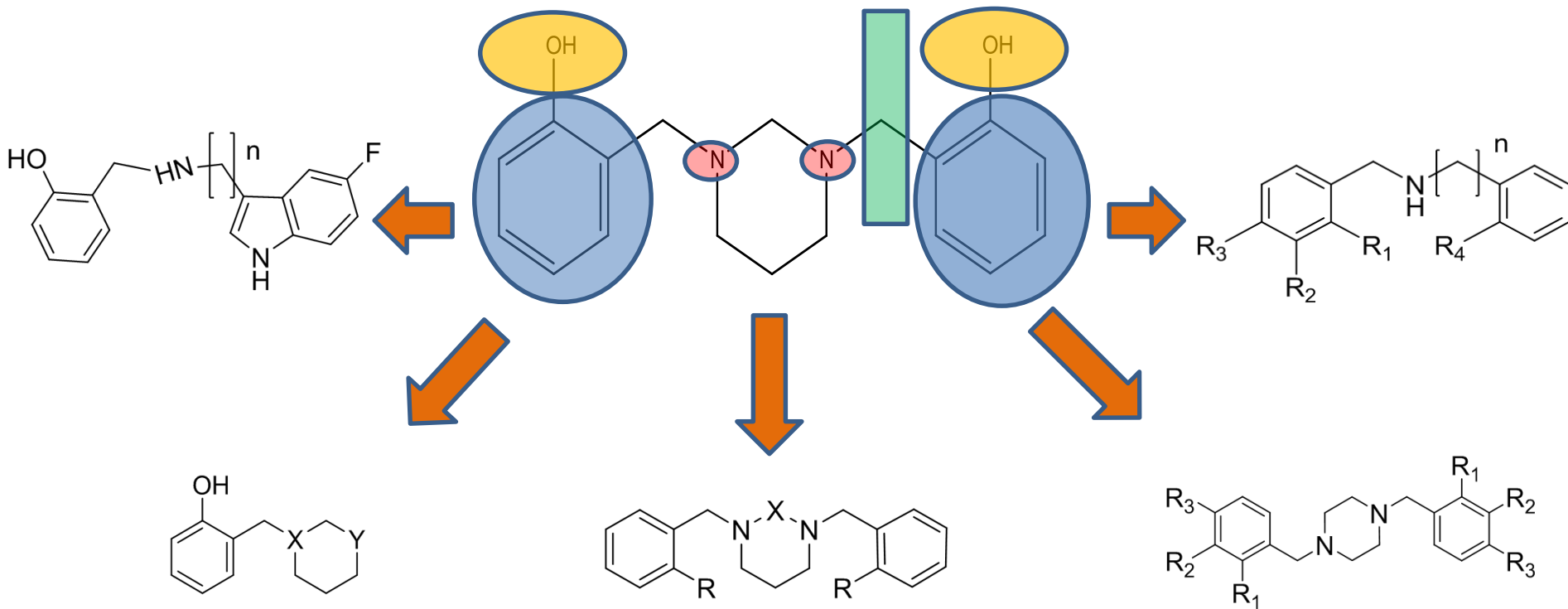
The role of aromatic cycle A and one atom of nitrogen.

The role of bridge length between one nitrogen atom and cycle B after removing the hexadrodperimidine cycle and different substitution on both cycles A and B.

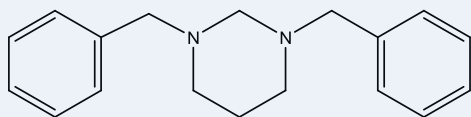
The role of the position of the two nitrogen atoms



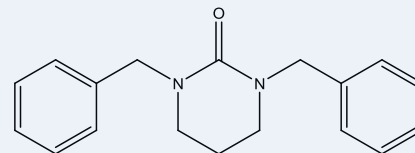
Pharmacomodulation



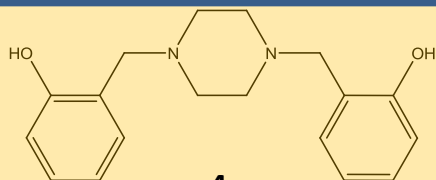
Designed compounds



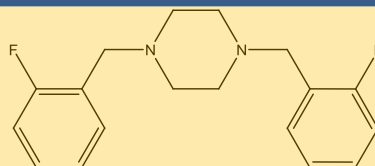
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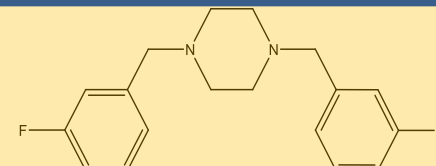
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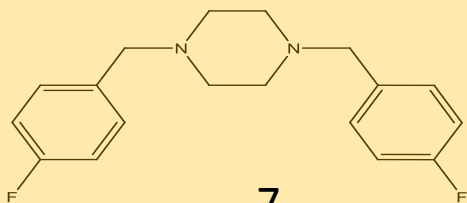
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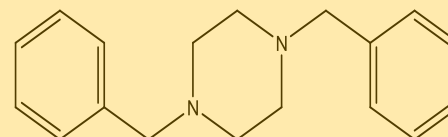
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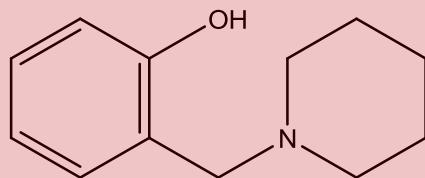
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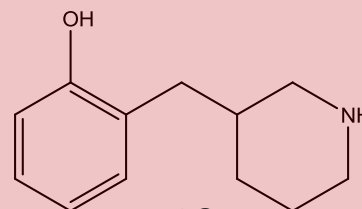
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8



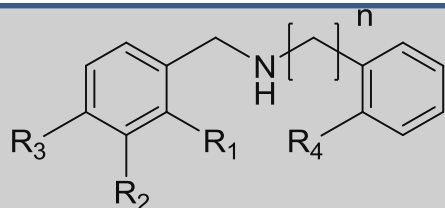
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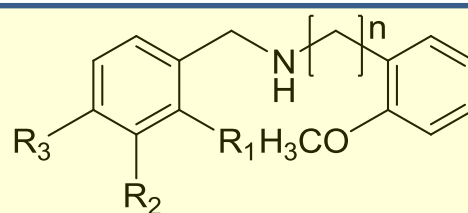
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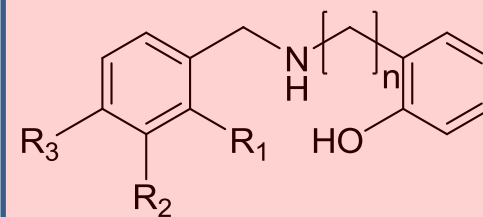
Designed compounds



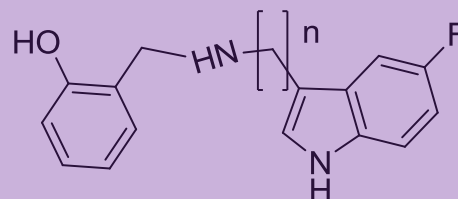
$R_1=OH, R_2=R_3=R_4=H$	$n=2$	11
$R_1=OH, R_2=R_3=R_4=H$	$n=3$	12
$R_1=R_2=R_3=R_4=H$	$n=2$	13
$R_1=R_2=R_3=R_4=H$	$n=3$	14
$R_1=OH, R_2=R_3=H, R_4=F$	$n=2$	15
$R_1=OH, R_2=R_3=H, R_4=F$	$n=3$	16



$R_1=OH, R_2=R_3=H,$	$n=2$	17
$R_1=OH, R_2=R_3=H,$	$n=3$	19
$R_1=R_2=R_3=H,$	$n=2$	21
$R_1=R_2=R_3=H,$	$n=3$	23
$R_1=F, R_2=R_3=H$	$n=2$	25
$R_1=F, R_2=R_3=H$	$n=3$	27
$R_1=R_2=H, R_3=F$	$n=2$	29
$R_1=R_2=H, R_3=F$	$n=3$	31
$R_1=H, R_2=F, R_3=H$	$n=2$	33
$R_1=H, R_2=F, R_3=H$	$n=3$	35



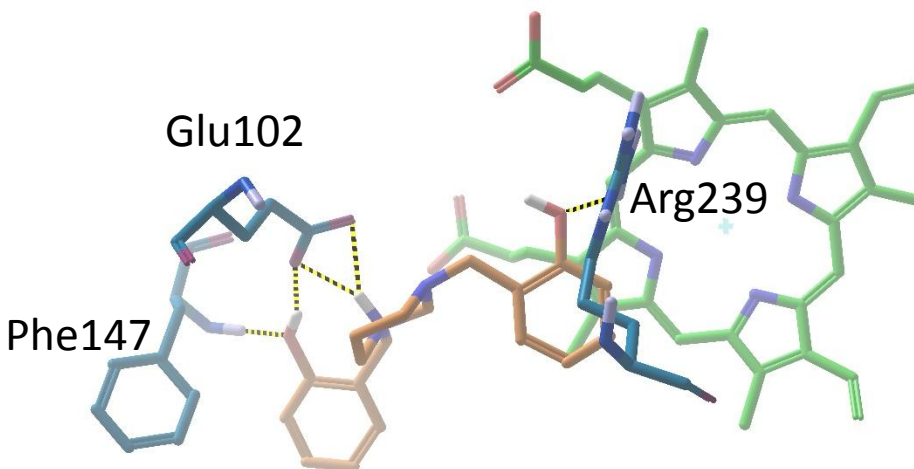
$R_1=OH, R_2=R_3=H,$	$n=2$	18
$R_1=OH, R_2=R_3=H,$	$n=3$	20
$R_1=R_2=R_3=H,$	$n=2$	22
$R_1=R_2=R_3=H,$	$n=3$	24
$R_1=F, R_2=R_3=H$	$n=2$	26
$R_1=F, R_2=R_3=H$	$n=3$	28
$R_1=R_2=H, R_3=F$	$n=2$	30
$R_1=R_2=H, R_3=F$	$n=3$	32
$R_1=H, R_2=F, R_3=H$	$n=2$	34
$R_1=H, R_2=F, R_3=H$	$n=3$	36



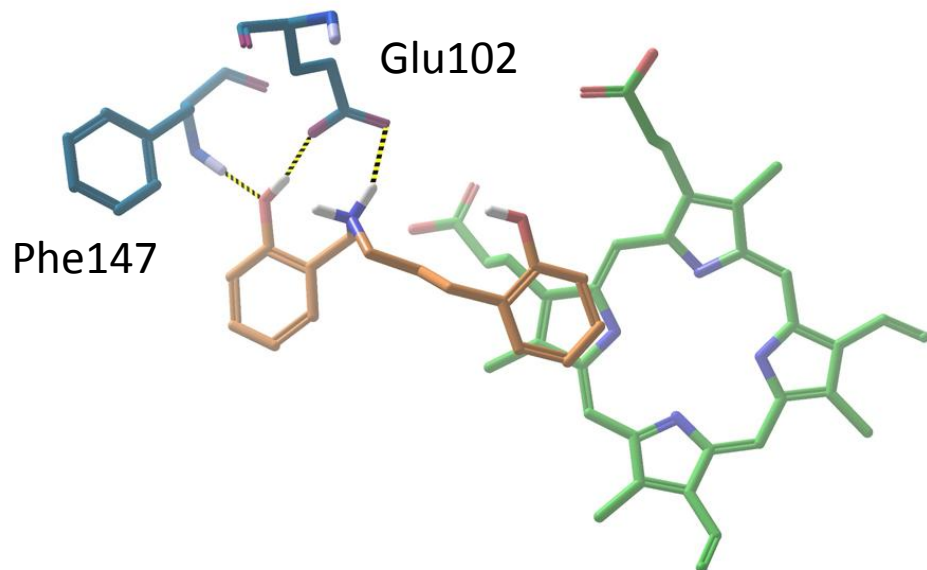
$n=2$	37
$n=3$	38



Some docked poses of the designed compounds in MPO Receptor



Compound 4 Shows hydrogen bonds with Glu102 and Phe147 Arg239 and salt bridge with Glu102



Compound 20 Shows hydrogen bonds with Glu102 and Phe147 and salt bridge with Glu102

Docking results gave some similar interactions as with HX1 and SHA A1 and different free Energy levels $-\Delta G$ or affinities with MPO receptors



Chemistry



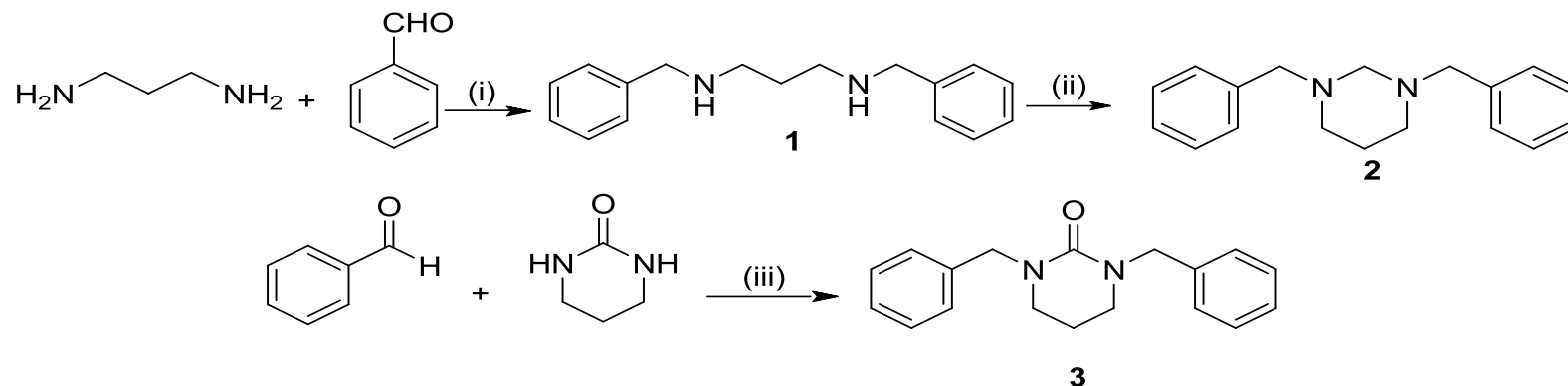
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Chemistry

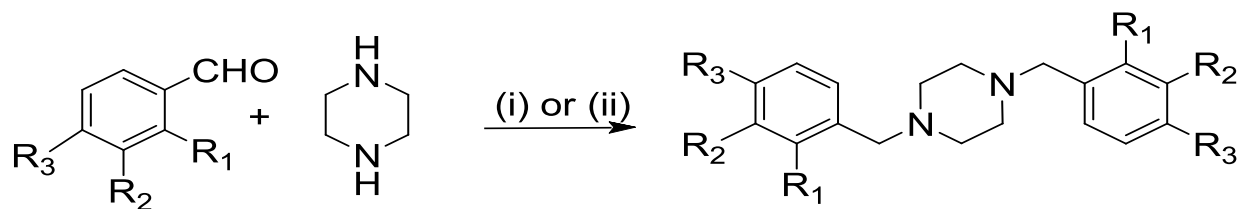


Reagents and conditions

(i) NaBH_4 , EtOH

(ii) formaldehyde (37 wt % aqueous solution), EtOH

(iii) BnCl , NaH , DMF, Reflux



Reagents and conditions:

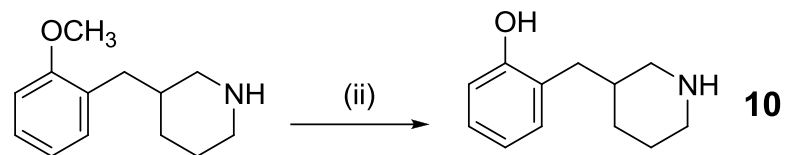
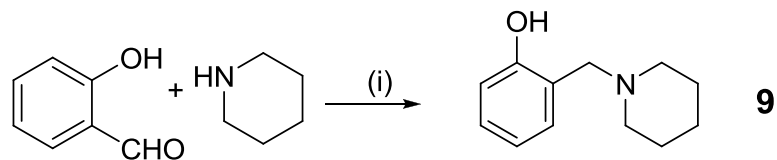
(i) NaBH_4 , acetic acid, EtOH for **4**

(ii) NaBH_3CN , acetic acid, EtOH for **5-8**

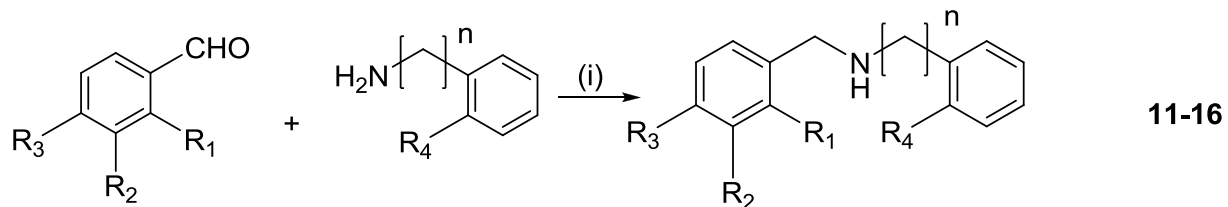
$\text{R}_1=\text{OH}; \text{R}_2=\text{R}_3=\text{H};$ **4**
 $\text{R}_1=\text{F}; \text{R}_2=\text{R}_3=\text{H}$ **5**
 $\text{R}_2=\text{F}; \text{R}_1=\text{R}_3=\text{H}$ **6**
 $\text{R}_3=\text{F}; \text{R}_1=\text{R}_3=\text{H}$ **7**
 $\text{R}_1=\text{R}_2=\text{R}_3=\text{H}$ **8**



Chemistry



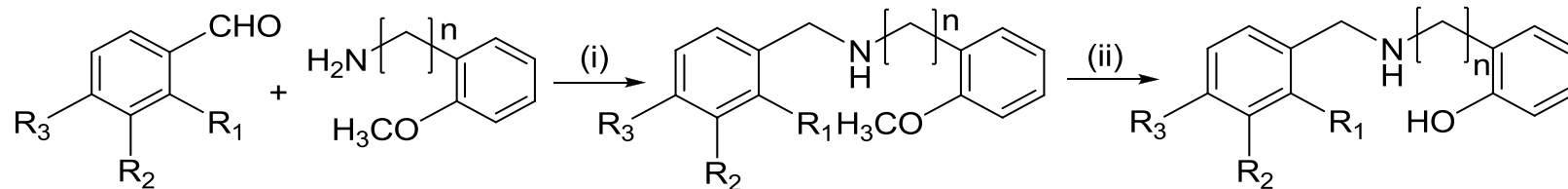
Reagents and conditions: (i) NaBH_3CN , acetic acid, EtOH (ii) BBr_3 , DCM



Reagents and conditions: (i) NaBH_4 , MeOH

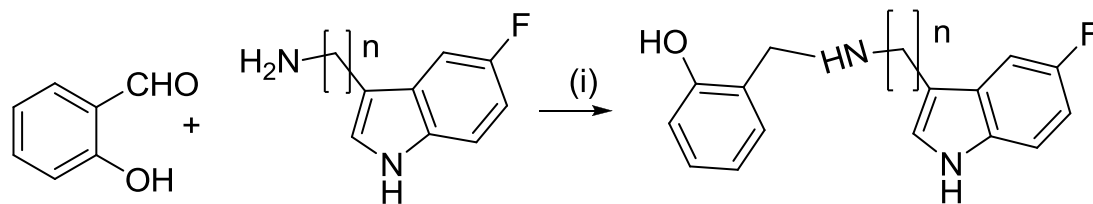


Chemistry



Reagents and conditions: (i) NaBH_4 , MeOH
(ii) BBr_3 , DCM

17-36



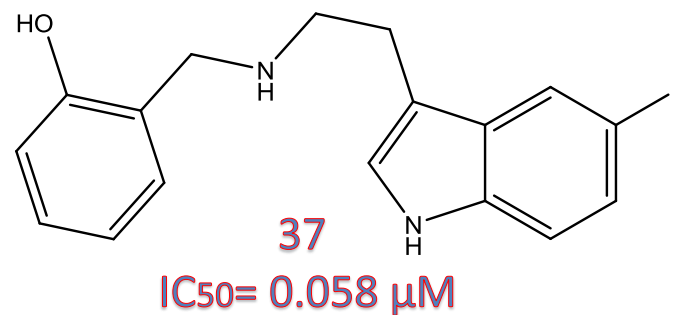
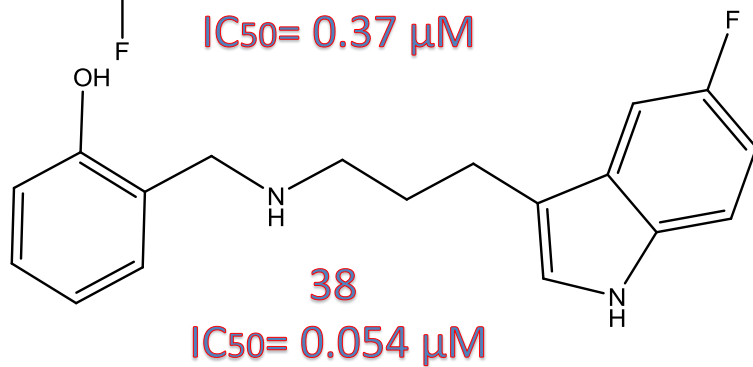
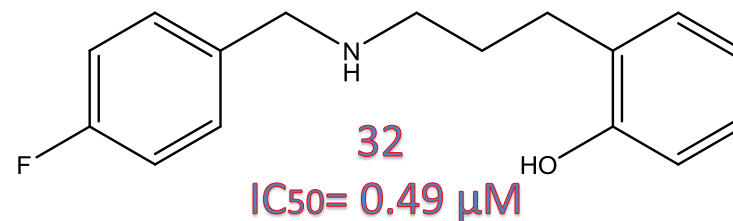
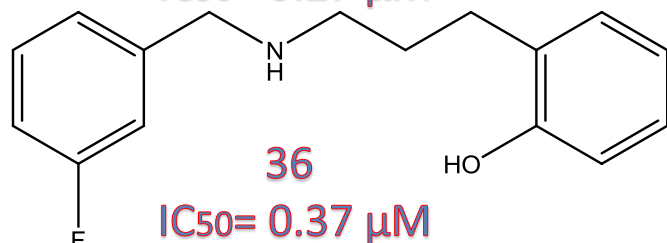
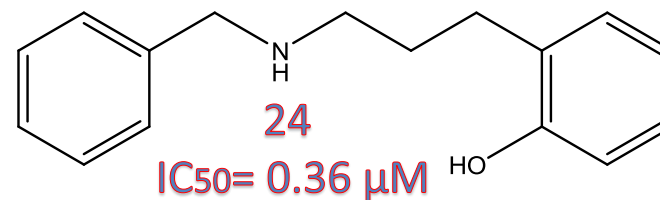
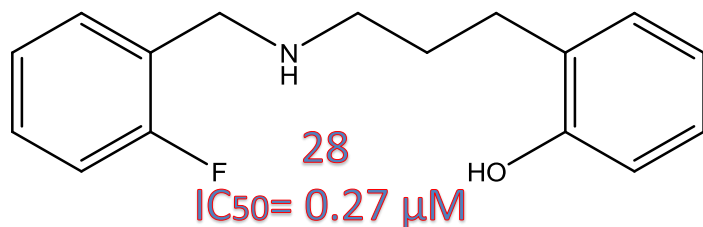
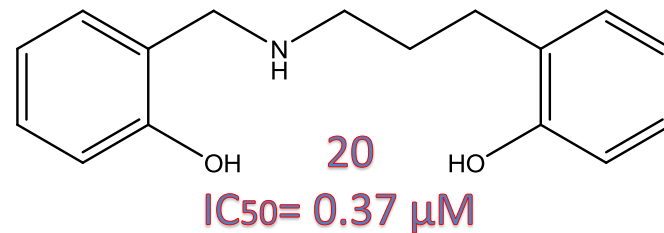
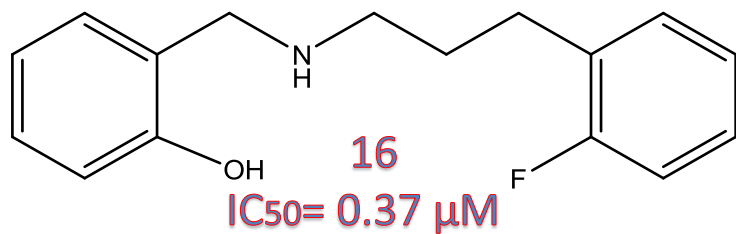
Reagents and conditions: (i) NaBH_4 , MeOH

37-38

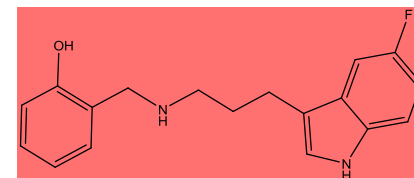
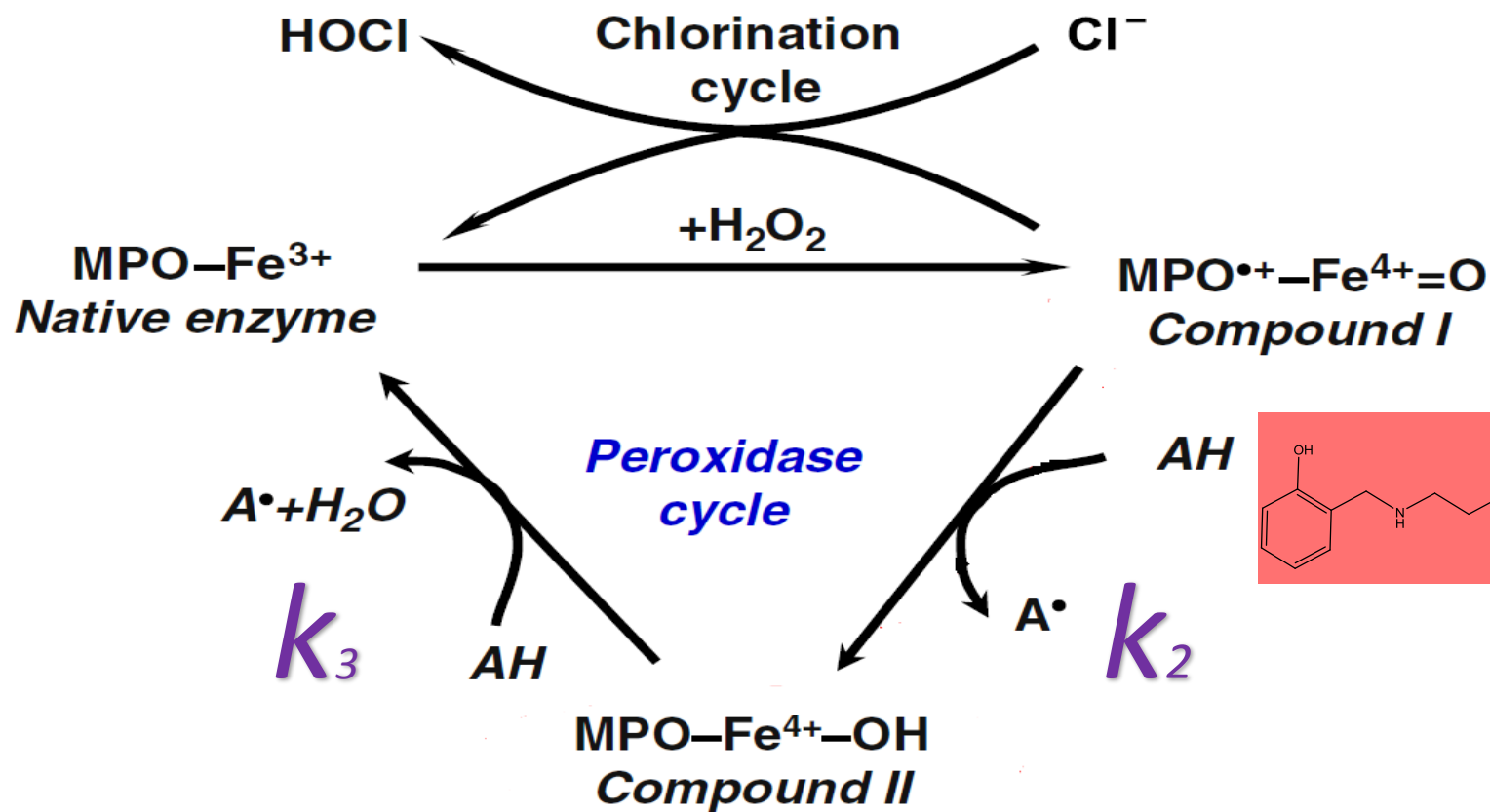


MPO inhibition assay

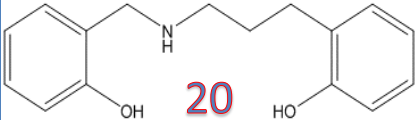
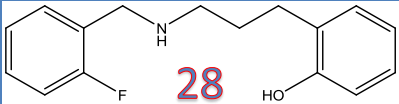
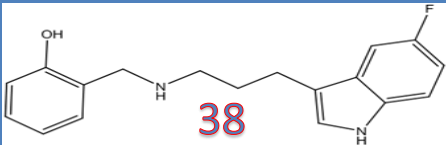
Best synthesized compounds with its IC50



Transient-State Kinetics



Mechanism of action Transient-State Kinetics

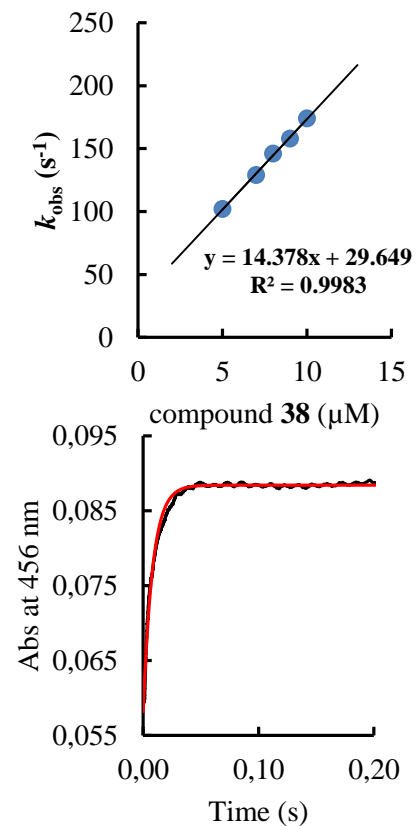
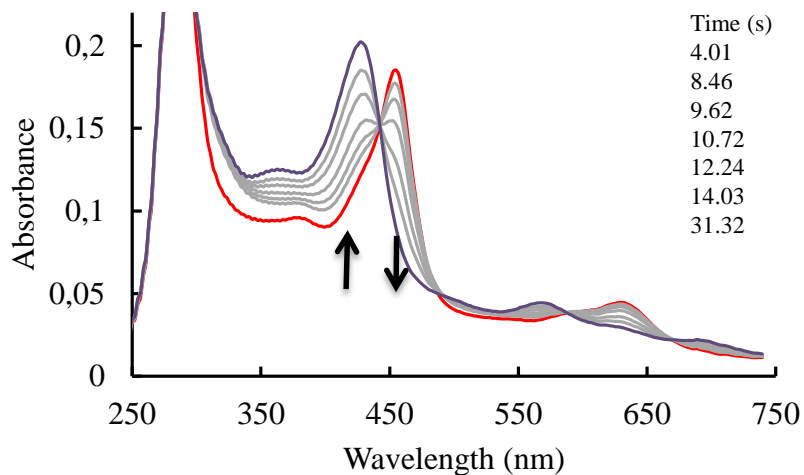
Compound	Compound I reduction rate constant (M ⁻¹ s ⁻¹)	Compound II reduction rate constant (M ⁻¹ s ⁻¹)	Ratio of compound I rate to compound II rate
 <p style="text-align: center;">20</p>	1.5×10^6	4.8×10^3	313
 <p style="text-align: center;">28</p>	5.7×10^6	1.4×10^3	4071
 <p style="text-align: center;">38</p>	1.4×10^7	3.5×10^3	4000



Mechanism of action

Transient-State Kinetics

The kinetic rate constant of reaction of the three best compounds with MPO/compound I /compound II have been measured.



The reaction from compound I to compound II is too fast but the reaction with compound II is slow leading to the accumulation of compound II



Conclusion



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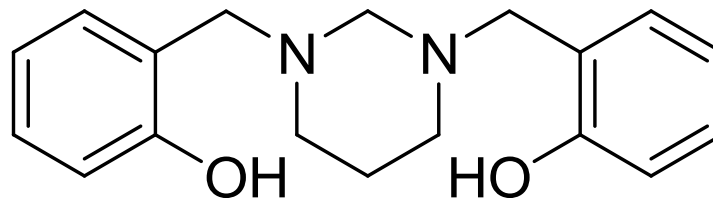
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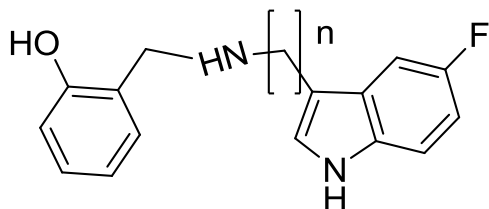
Conclusion

Pharmacomodulation



A1-IC₅₀ 0.5 μM

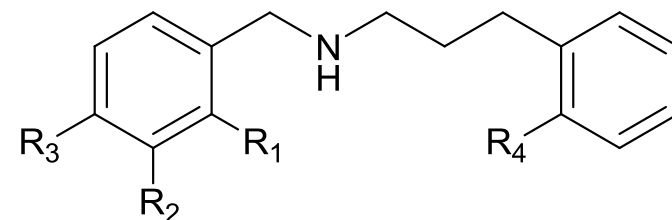
MPO
IC₅₀



n=2 **37**

n=3 **38**

with IC₅₀ 0.3-0.05 μM .
More active to 2-10 times
than the A1



R₁=OH, R₂=R₃=H, R₄=F **16**

R₁=OH, R₂=R₃=H, R₄=OH **20**

R₁=R₂=R₃=H, R₄=OH **24**

R₁=F, R₂=R₃=H, R₄=OH **28**

R₁=R₂=H, R₃=F, R₄=OH **32**

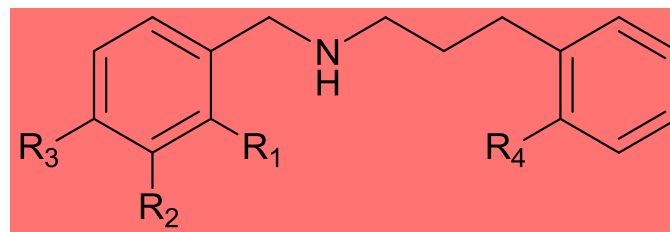
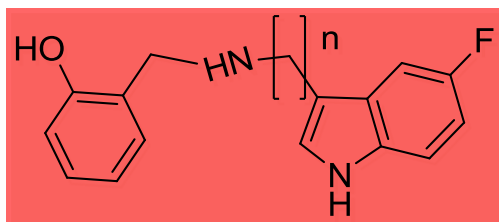
R₁=H, R₂=F, R₃=H, R₄=OH **36**



Conclusion

***best compounds have shown high reduction rate constants of compound I and II and their ratio can explain the accumulation of compound II, illustrating a reversible mechanism of inhibition..**

***Arylpropylamine derivatives and adding the indole structure to the original scaffold A1 have given us new effective MPO inhibitors**



$IC_{50} = 0.3-0.05 \mu M$



Acknowledgments to our team in the Organic Pharmaceutical Chemistry LAB -CPO [Therapeutic Chemistry] – Faculty of Pharmacy-ULB

Dr Jalal Soubhye , Dr. Gilles BERGER
Ana CERNE , Mélissa CORTESE
Dr. Cédric DELPORTE , Damien DUFOUR
Caroline NOYON , Florence REYE
Prof. François DUFRASNE
Prof. Michel GELBCKE
Prof. Pierre VAN ANTWERPEN
Prof. Jean NEVE



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