

Ab initio generation of binary alloy foams: the case of amorphous $\text{Cu}_{64}\text{Zr}_{36}$

Jonathan Galván

Ariel A. Valladares

Renela M. Valladares

Alexander Valladares

Materials Research Institute, UNAM

2nd International Electronic Conference on Materials, 2016



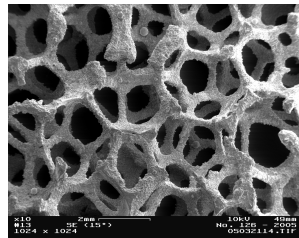
Outline

- 1 Background
 - Experimental and theoretical methods
- 2 Why amorphous porous metals?
 - Bulk metallic glasses and porous metals
- 3 Method
 - The Process
- 4 Results
 - Pair Distribution Functions
 - Bond-Angle Distribution (BAD)
- 5 Conclusions

Background

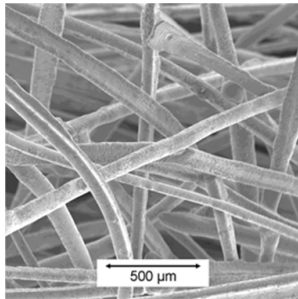
Experimental pore-making I

- Gas-generated foams
 - Bubbling gas into molten metals
 - Gas releasing from chemical-breaking at high temperatures when adding powders to molten metals
 - H can be dissolved in many liquid metals; upon solidification, H is rejected
- Template porous
 - Casting into a mold
 - Powder deposition
 - Decomposition
 - Electrodeposition



Experimental pore-making II

- Individual elements assembly
 - Spray forming
 - Sintering powders, wires or hollow spheres
 - Selective sintering
- Removable 2nd phase
 - Casting into space holders
 - Sintering with space holders
 - Replication
 - Dealloying (small scale pores ~ several nm)



Clyne, University of Cambridge (2010)

Early works

- The first successful method to create a bulk metallic glass foam was reported by Schroers *et al.* $\text{Pd}_{43}\text{Cu}_{27}\text{Ni}_{10}\text{P}_{20}$ with pores of 200-1000 μm
- Brothers and Dunand reported the foaming of a Zr-based alloy by rapid quenching in 2004: $\text{Zr}_{57}\text{Nb}_5\text{Cu}_{15.4}\text{Ni}_{12.6}\text{Al}_{10}$ with pore sizes of 25-50 μm
- Successive experimental works were focused on multicomponent alloys: $\text{Ni}_{59}\text{Zr}_{20}\text{Ti}_{16}\text{Si}_2\text{Sn}_3$, $\text{Fe}_{48}\text{Cr}_{15}\text{Mo}_{14}\text{Y}_2\text{C}_{15}\text{B}_6$, and $\text{Mg}_{60}\text{Cu}_{21}\text{Ag}_7\text{Gd}_{12}$

Schroers *et al.*, App. Phys. Lett. 82, 370 (2003)
Brothers, Dunand, App. Phys. Lett 84, 1108 (2004)
Lee, Sordélet, App. Phys. Lett. 89 (2006)
Demetriadou *et al.*, Scripta Mater. 57, 9 (2007)
Brothers *et al.*, J. App. Phys. 102, 023508 (2007)

Porous materials from simulations

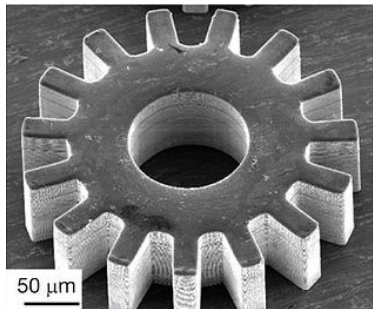
- Reconstructive approach:
 - The model structure is based on experimental reference data
 - Problem of uniqueness arises
 - Reverse Monte Carlo: energy determined by how well the structure reproduces empirical data, e.g., structure factor
- Mimetic approach
 - Mimics physical and chemical processes to generate the experimental sample
 - Monte Carlo or Molecular Dynamics are used to model the evolution of the structure from a starting state
 - Empirical or semi-empirical potentials have no transferability
- Pores are manually carved in the structure
- No *ab initio* approaches have been cited

Gelb, MRS Bull. 34, 592 (2009)

Why amorphous porous metals?

Bulk metallic glasses - Properties

- Lack of long range order
- Glass Forming Ability depends on contents
- Enhanced mechanical properties than other crystalline materials
- High viscosity
- Micro- and nano-size forming molds
- Catalysts
- High wear and corrosion resistance



Kumar, Tang, Schroers, Nature 457, 868 (2009)

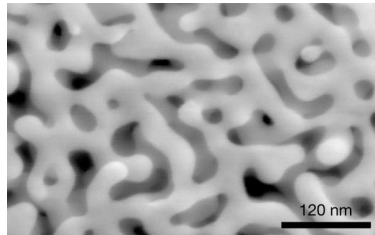
Bulk metallic glasses - Drawbacks

- Lack of ductility
- Size-growth limit
- High brittleness
- Generation techniques are under development
- Higher serial production costs than crystalline counterparts



Metallic foams and sponges - Properties

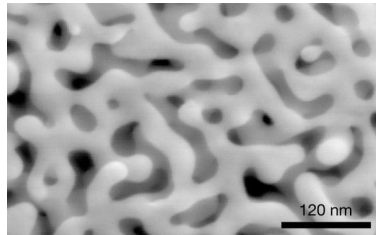
- Research dates back to early 1990s
- Improved mechanical properties under compression, bending and torsion
- Lightweight structures
- Impact/blast mitigation
- Fluid filters
- Gas storage



Erlebacher, Nature 410, 450 (2001)

Metallic foams and sponges - Drawbacks

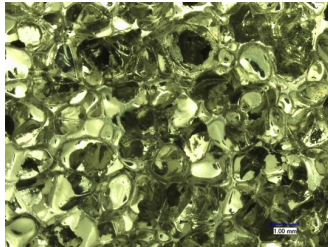
- Mechanical properties limited by the base alloys
- Difficult fabrication processes
- External phases while foaming due to blowing agents or solid placeholders
- Pore size control in the nano-regime



Erlebacher, Nature 410, 450 (2001)

Amorphous porous alloys

- Combine both properties of amorphous metals and metallic foams
- Experimental procedures design becomes complex
- Computer simulation modeling is useful to predict properties: topological, electronic, vibrational, mechanical, etc.
- Brand-new research field ~ 10 years



Method

The Expanding Lattice (EL) Method

The edge of the cubic cell is enlarged and the interatomic distances are proportionally increased. The volume increases and the density decreases as $l_2 = 2^{1/3}l_1$ for a 50 % of porosity.

- This approach has been applied to semiconductors and pure metals, such as: carbon, silicon, aluminum, copper and gold.
- Some properties as eDOS and vDOS have been calculated for amorphous nanoporous carbon which have agreed with other simulations and experimental data.
- Energetics of hydrogen adsorption have been calculated in amorphous nanoporous carbon and silicon.

Valladares *et al.*, Mater. Res. Soc. Symp. Proc. 971 (2007)

Valladares *et al.*, Mater. Res. Soc. Symp. Proc. 988 (2007)

Valladares *et al.*, Mater. Res. Soc. Symp. Proc. 1042 (2008)

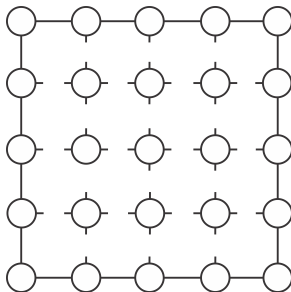
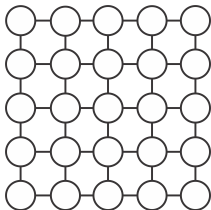
Romero *et al.*, Materials 3, 467 (2010)

Santiago-Cortés *et al.*, J. Non-Cryst. Solids 358, 596 (2012)

Romero *et al.*, J. Non-Cryst. Solids 362, 14 (2013)

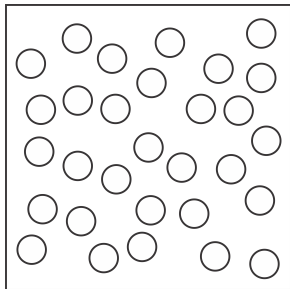
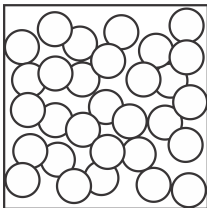
The EL method in 108-atom $\text{Cu}_{64}\text{Zr}_{36}$

From crystal

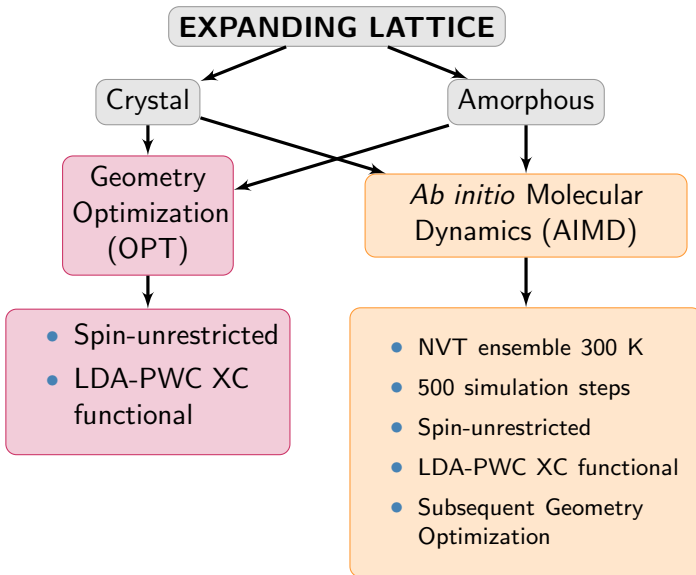


The EL method in 108-atom $\text{Cu}_{64}\text{Zr}_{36}$

From amorphous



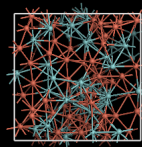
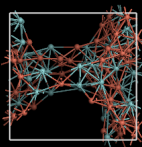
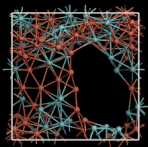
Valladares *et al.*, *Materials* 4 (2011)
Galván-Colín, *et al.*, *Mater. Res. Soc. Symp. Proc.* 1517 (2013)
Galván-Colín, *et al.*, *Physica B* 475 (2015)



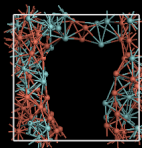
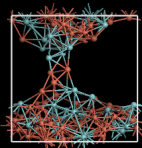
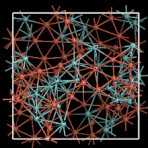
Results

OPT-generated ap-Cu₆₄Zr₃₆ structures

From crystal

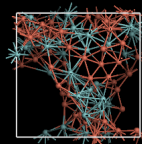
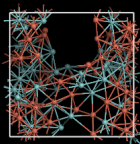
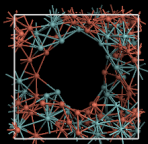


From amorphous

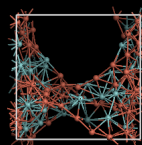
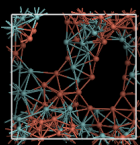
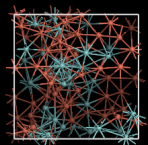


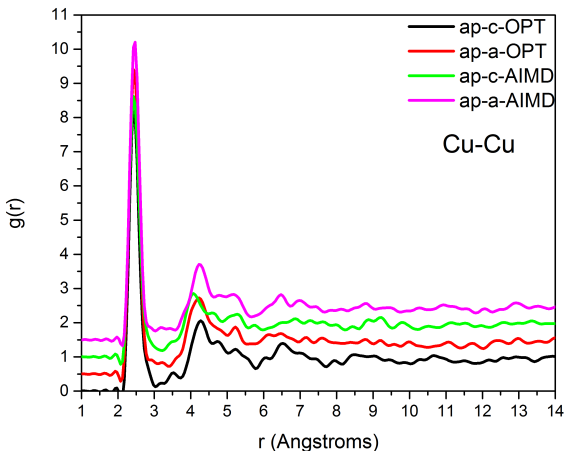
AIMD-generated ap-Cu₆₄Zr₃₆ structures

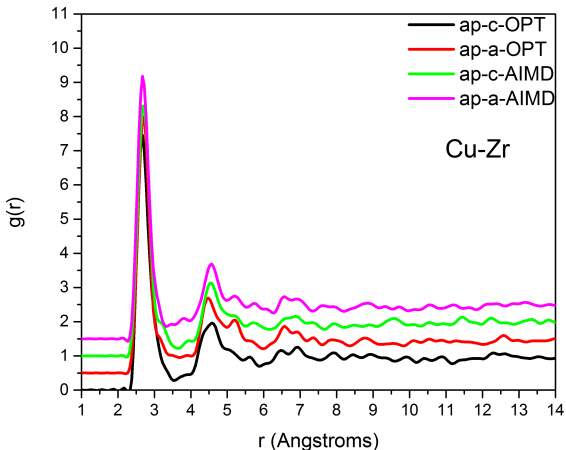
From crystal

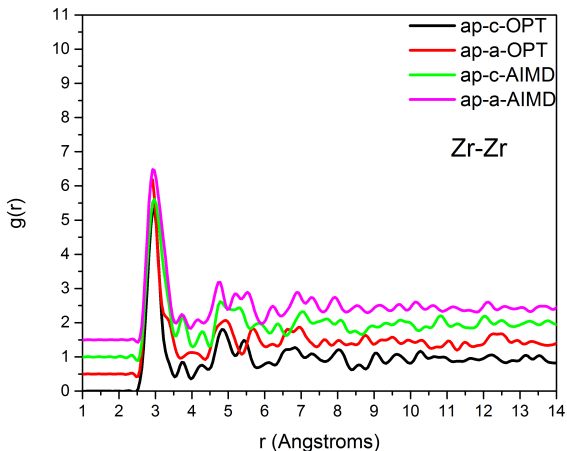


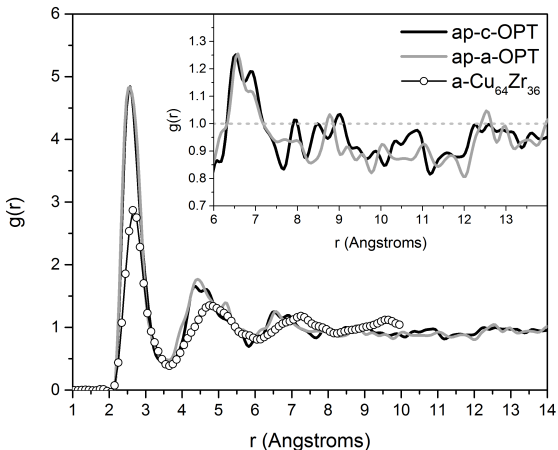
From amorphous

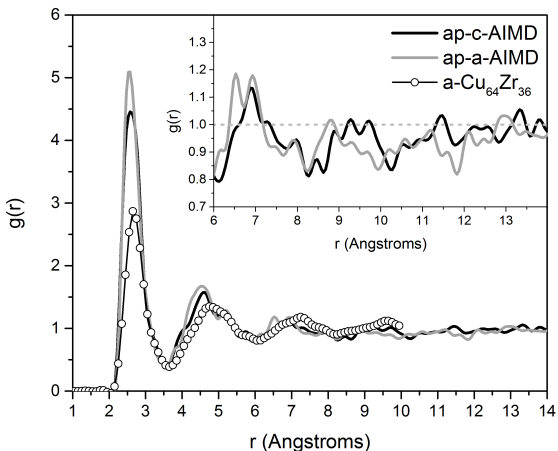


ap-Cu₆₄Zr₃₆ Cu-Cu partials

ap-Cu₆₄Zr₃₆ Cu-Zr partials

ap-Cu₆₄Zr₃₆ Zr-Zr partials

ap-Cu₆₄Zr₃₆ OPT comparison of total PDF

ap-Cu₆₄Zr₃₆ AIMD comparison of total PDF

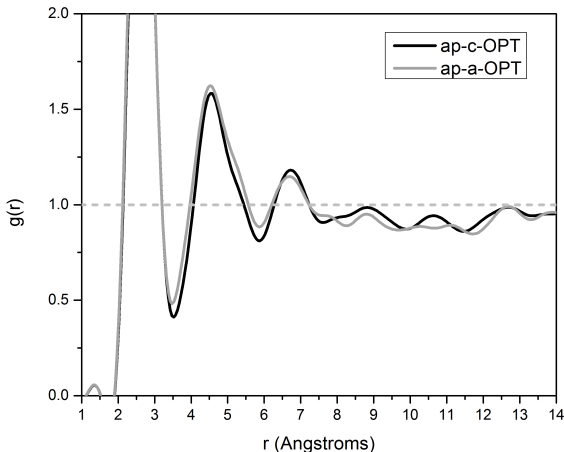
1st neighbor positions and coordination numbers

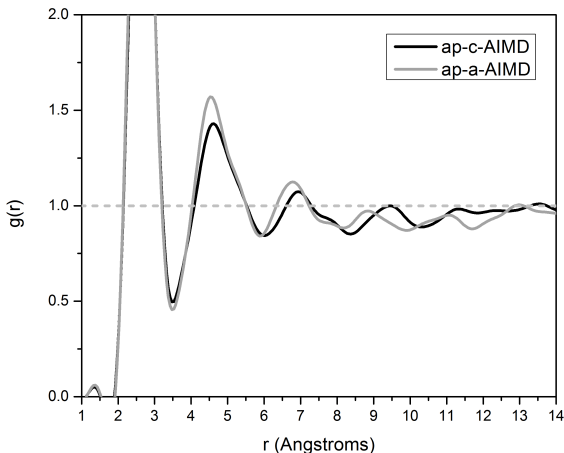
	$r_{\text{Cu-Cu}}$	$r_{\text{Cu-Zr}}$	$r_{\text{Zr-Zr}}$	r_{Total}
ap-c-OPT	2.43	2.68	2.98	2.58
ap-a-OPT	2.43	2.73	2.93	2.58
ap-c-AIMD	2.43	2.73	2.98	2.58
ap-a-AIMD	2.48	2.68	2.93	2.58
a-Cu ₆₄ Zr ₃₆	2.45	2.75	3.15	2.70

1st neighbor positions and coordination numbers

	$r_{\text{Cu-Cu}}$	$r_{\text{Cu-Zr}}$	$r_{\text{Zr-Zr}}$	r_{Total}
ap-c-OPT	2.43	2.68	2.98	2.58
ap-a-OPT	2.43	2.73	2.93	2.58
ap-c-AIMD	2.43	2.73	2.98	2.58
ap-a-AIMD	2.48	2.68	2.93	2.58
a-Cu ₆₄ Zr ₃₆	2.45	2.75	3.15	2.70

	$N_{\text{Cu-Cu}}$	$N_{\text{Cu-Zr}}$	$N_{\text{Zr-Cu}}$	$N_{\text{Zr-Zr}}$	N_{Cu}	N_{Zr}	N_{Total}
ap-c-OPT	4.5	3.8	6.7	3.7	8.3	10.4	9.0
ap-a-OPT	4.8	3.7	6.6	3.7	8.5	10.3	9.1
ap-c-AIMD	4.4	3.8	6.7	4.2	8.2	10.9	9.1
ap-a-AIMD	4.6	3.5	6.2	4.1	8.1	10.3	8.9
a-Cu ₆₄ Zr ₃₆	6.7	5.3	9.4	6.3	12	15.7	13.2

ap-Cu₆₄Zr₃₆ OPT depletion in total PDF

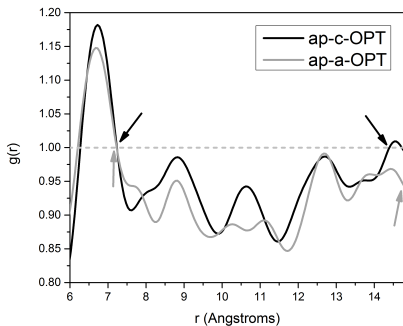
ap-Cu₆₄Zr₃₆ AIMD depletion in total PDF

Pore size estimation

	Pore size (Å)
ap-c-OPT	7.1
ap-a-OPT	7.5
ap-c-AIMD	2.1/3.9
ap-a-AIMD	5.6

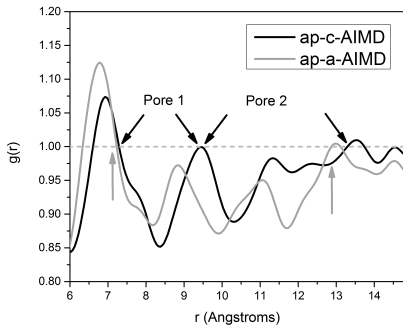
Pore size estimation

	Pore size (Å)
ap-c-OPT	7.1
ap-a-OPT	7.5
ap-c-AIMD	2.1/3.9
ap-a-AIMD	5.6

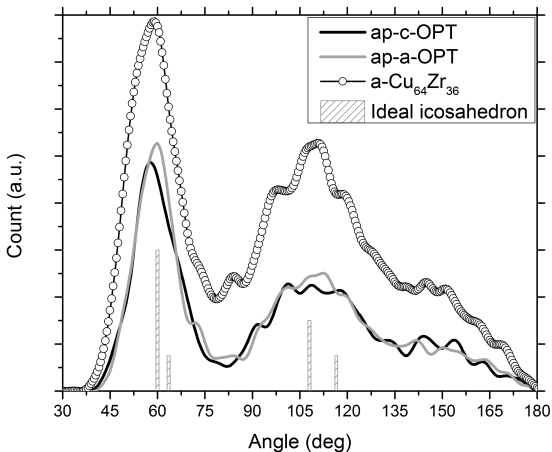


Pore size estimation

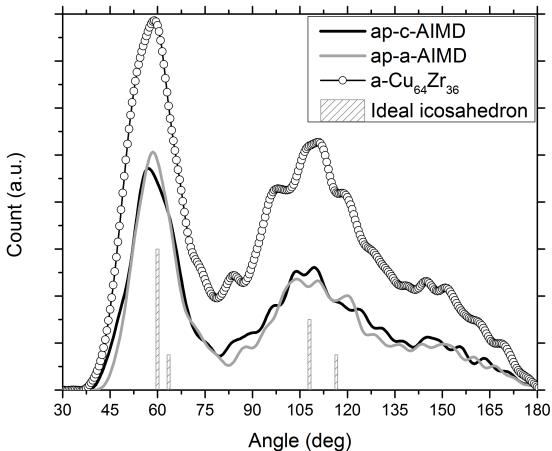
	Pore size (Å)
ap-c-OPT	7.1
ap-a-OPT	7.5
ap-c-AIMD	2.1/3.9
ap-a-AIMD	5.6



Bond-Angle Distribution (BAD)

BAD ap-Cu₆₄Zr₃₆ OPT

Bond-Angle Distribution (BAD)

BAD ap-Cu₆₄Zr₃₆ AIMD

Conclusions and future development

- We applied the expanding lattice method to obtain nano porous amorphous $\text{Cu}_{64}\text{Zr}_{36}$ at room temperature by means of *ab initio* molecular dynamics and geometry optimization only.
- The initial amorphous structures became porous whereas the initial crystal cells become both porous and amorphous.
- Larger pores that appeared in simulations with an initial amorphous configuration may be related to a higher energetic cost of disordering initial crystalline structures after EL.
- Low coordination is a consequence of a competition between surface and bulk contributions.
- BAD sharp peak around 60° suggests the existence of distorted Frank-Kasper polyhedra.
- Larger simulation cells would provide insight to verify whether pore topology is cell-size dependent (through and dendritic pores).

What's next

- Use a larger number of atoms to study pore size and geometry.
- Study other binary and ternary alloys.
- Try different concentrations of Cu-Zr alloy.
- Generate amorphous nanoporous structures with different porosity.
- AIMD at higher temperatures to see its influence in the amorphization process of initial crystalline models.

Our workgroup

- Prof. Ariel A. Valladares
- Prof. Renela Valladares
- Prof. Alexander Valladares

Students

- Martín Mejía
- Juan Carlos Noyola
- Cristina Romero
- Zaahel Mata
- Sebastián Tamariz
- Jozra Garrido
- Isaías Rodríguez

Former students

- Ph.D. Fernando Álvarez
- Ph.D. Rubén Estrada
- Ph.D. Emilye Rosas
- Ph.D. Ángel Reyes
- Ph.D. Andrés Díaz
- Ph.D. Ulises Santiago

Thank you for your attention!