

# AGGREGATION OF TWO DISAZO DIRECT DYES. EXPERIMENTAL AND THEORETICAL STUDY

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#### Introduction

Adsorption of dyes on solid adsorbents is influenced by the physicochemical properties of the dyes solutions.

Azo dyes are compounds whose molecules have a tendency to self-associate in the ground electronic state. That can lead to the formation of dimers, trimers, and tetramers.

The main factors influencing aggregation are: dye structure, dye concentration, electrolyte concentration, and temperature.



 Investigation of the molecular aggregation of two direct dyes in aqueous solution, as a function of dye concentration.

Mathematical analysis of the aggregation of the studied dyes.







DSA



#### DAN

Fig. 1. The structure of the studied disazo dyes

# **Results and Discussions**



## Fig. 2. Direct dyes DSA and DAN absorption spectra at different concentrations





# Figure 3. The trend in studied dye's absorption coefficient as concentration varies

# **Matematical analysis**

Using the experimental absorbance values, an n x m (n: values of absorption at different wavelength, m: concentrations) dimension matrix was build for a series of analytical concentrations.

The number of eigenvalues of this matrix different from zero (within accepted experimental errors) is equal to the number of chemical species (monomer, dimer, etc.) presented in the system.

### **Matriceal Analysis**

#### The eigenvalues value of the absorption matrix for DSA and DAN dyes

-3.084 x 10<sup>-15</sup> 3.985 x 10<sup>-14</sup> 1.399 x 10<sup>-13</sup> 1.566 x 10<sup>-4</sup> 1.001 x 10<sup>-3</sup> **0.087 3.289 163.642** 

DSA

DAN -3.983 x 10-15 2.371 x 10<sup>-5</sup> 1.938 x 10<sup>-4</sup> 3.783 x 10<sup>-4</sup> 2.584 x 10<sup>-3</sup> 0.025 8.339 116.417

#### Conclusion

- UV-Vis spectroscopy has been used to study the aggregation of two disazo direct dyes (DSA, DAN) in aqueous solution;
- The dyes aggregation depends on the concentration.

From the matriceal analysis, the number of chemical species presented in the system was determined, in domain of concentrations studied.

