

# On the roles of energy and entropy in thermodynamics

by

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$$q_i = -\kappa \frac{\partial T}{\partial x_i}$$



**J.B. Fourier**



**C.-L. Navier**

$$t_{ij} = 2\mu \frac{\partial v_{<i}}{\partial x_{j>}} + \nu \frac{\partial v_l}{\partial x_l} \delta_{ij}$$

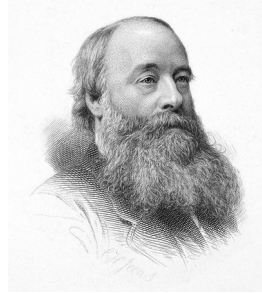


**G. G. Stokes**

**Derivations did not require the knowledge of the nature of heat, let alone the concepts of energy and entropy**



R. J. Mayer



J. P. Joule



H. v. Helmholtz

**First Law:**  $\frac{dE}{dt} = \dot{Q} + \dot{W}$



S. Carnot



R. Clausius

**Second Law:**  $\frac{dS}{dt} \geq \frac{\dot{Q}}{T_0}$

probabilistic interpretation

$$S = k \ln W$$



L. Boltzmann

$\implies \blacktriangleright \quad A \equiv E - T_0 S \rightarrow \text{Minimum in equilibrium}$

Minimal energy is conducive to equilibrium and so is maximal entropy. Temperature is control parameter.

Competition between **determinism** by which energy approaches a minimum

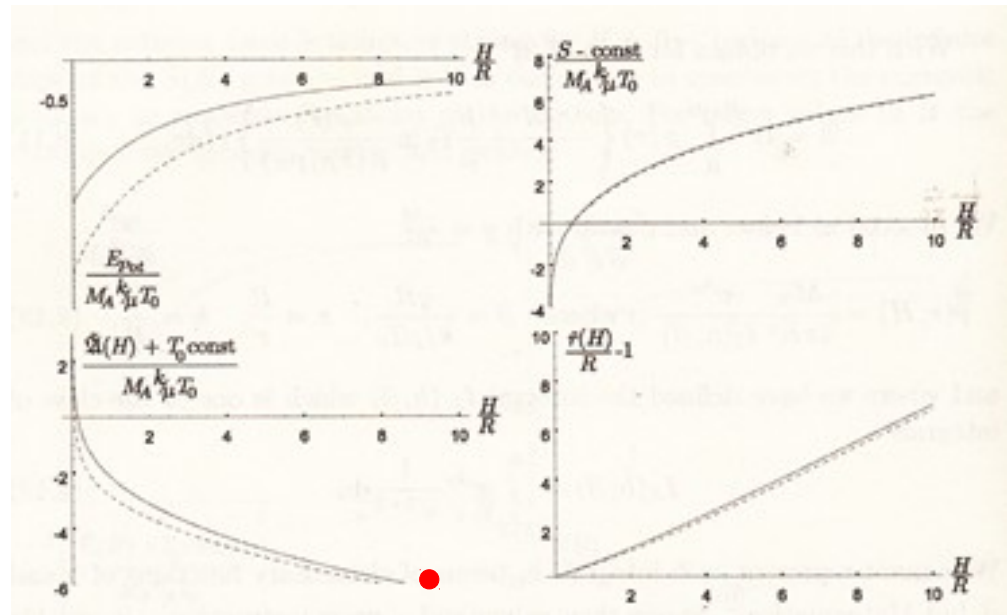
and **stochasticity** by which entropy approaches a maximum.

# Planetary Atmospheres

Energy of atmosphere is minimal when all air molecules lie on the solid surface.

Who wins?

Entropy is maximal when air molecules are evenly distributed throughout space..



Relevant parameter  $\beta = \frac{\gamma \frac{M}{R}}{\frac{k}{\mu} T}$

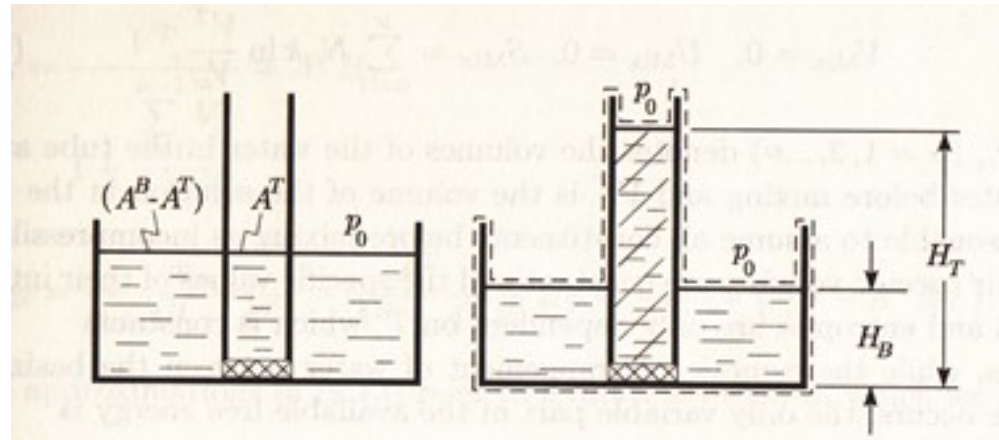
Mercury and Moon have already lost their atmospheres

Jupiter, Saturn and Uranus have kept even light gases

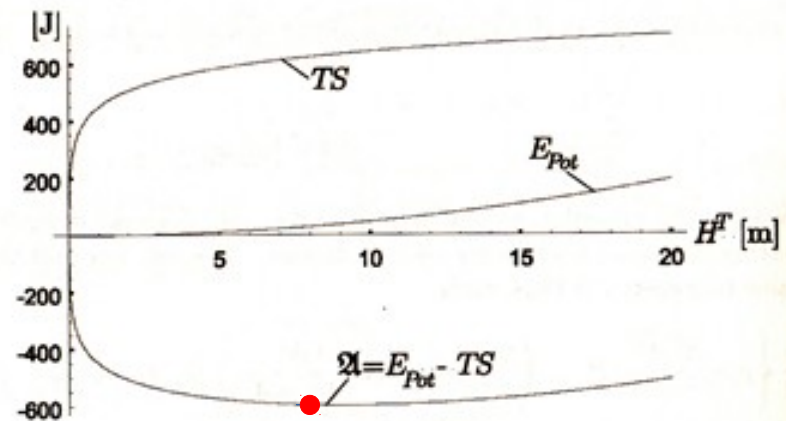
Earth hangs on to oxygen and nitrogen – **for he the time being !**

# Osmosis

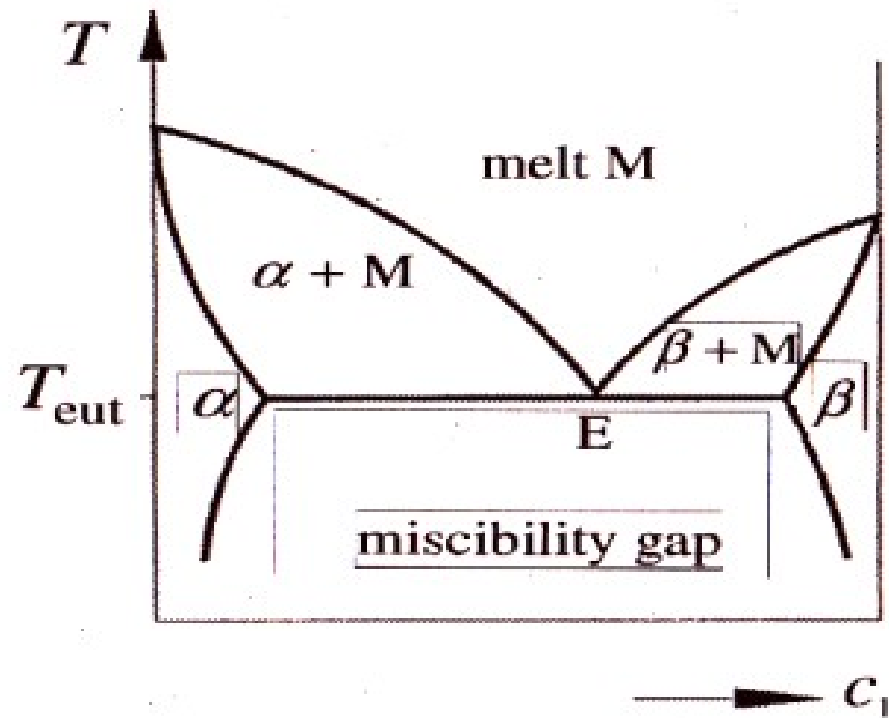
## Pfeffer tube



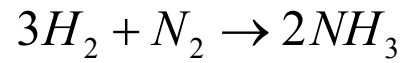
W. Pfeffer



# Phase Diagrams (for alloys and solutions)



# Ammonia Synthesis (Haber-Bosch)

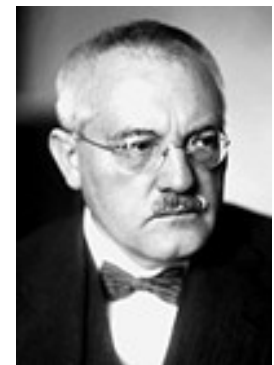
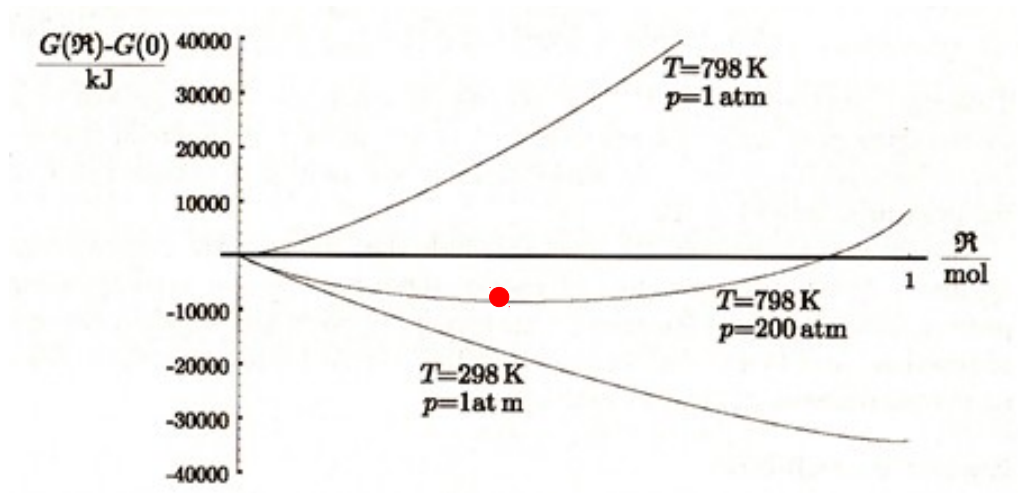


$$\Delta h = -3h^{H_2} - h^{N_2} + 2h^{NH_3} = -92.4 \frac{kJ}{mol}$$

$$\Delta s = -3s^{H_2} - s^{N_2} + s^{NH_3} = -178.6 \frac{J}{mol K}$$



F. Haber



K. Bosch

## Doctrine of Forces and Fluxes (TIP)

**Gibbs equation**

$$\rho \frac{ds}{dt} = \frac{1}{T} \left( \frac{du}{dt} - \frac{p}{\rho^2} \frac{d\rho}{dt} - g_\alpha \frac{dc_\alpha}{dt} \right)$$

$\uparrow\uparrow$                        $\uparrow\uparrow$                        $\uparrow\uparrow$   
 energy                      mass                      partial mass  
 balance                      balance                      balance

**Entropy Inequality**

$$\rho \frac{ds}{dt} + \frac{\partial}{\partial x_i} \left( \frac{q_i - \sum_{\alpha=1}^v g_\alpha J_i^\alpha}{T} \right) =$$

$$= -\frac{1}{T} \sum_{a=1}^n \left( \sum_{\alpha=1}^v g_\alpha \gamma_\alpha^a \mu_\alpha \right) \lambda^a + q_i \frac{\partial}{\partial x_i} \frac{1}{T} - J_i^\alpha \frac{\partial}{\partial x_i} \frac{g_\alpha}{T} + \frac{1}{T} t_{\langle ij \rangle} \frac{\partial v_{\langle i}}{\partial x_{j \rangle}} + \frac{1}{T} \left( \frac{1}{3} t_{ii} + p \right) \frac{\partial v_l}{\partial x_l} \geq 0$$

Thus follows a semi systematic derivation for the laws of Fourier, Navier-Stokes and Fick by linear relations between forces and fluxes.

Fully **satisfactory** for liquids and dense gases.

But **deficient** when rates of change are rapid and gradients are steep **as may easily happen in rarefied gases**



# Extended Thermodynamics

Fields  $u_\alpha \quad (\alpha=1,2, \dots, N)$

Field equation  $\frac{\partial u_\alpha}{\partial t} + \frac{\partial F_\alpha^i(u_\beta)}{\partial x_i} = \Pi(u_\beta)$

Solutions: Thermodynamic processes

Entropy Principle  $\frac{\partial h}{\partial t} + \frac{\partial h^i(u_\beta)}{\partial x_i} = \Sigma(u_\beta) \geq 0$  for all thermodynamic processes **and**  $h(u_\alpha)$  concave

$$\Rightarrow \frac{\partial h}{\partial t} + \frac{\partial h^i(u_\beta)}{\partial x_i} - \Lambda_\alpha \left( \frac{\partial u_\alpha}{\partial t} + \frac{\partial F_\alpha^i}{\partial x_i} - \Pi \right) \geq 0 \quad \text{for all fields } u_\alpha(x_i, t)$$

Change of fields  $u_\alpha \Leftrightarrow \Lambda_\alpha$       Field equations  $\frac{\partial^2 h'}{\partial \Lambda_\alpha \partial \Lambda_\beta} \frac{\partial \Lambda_\alpha}{\partial t} + \frac{\partial^2 h^i}{\partial \Lambda_\alpha \partial \Lambda_\beta} \frac{\partial \Lambda_\alpha}{\partial x_i} = \Pi_\alpha$

↑↑  
symmetric hyperbolic !!

**Conclusion:** Entropy Principle guarantees that the field equations are symmetric hyperbolic.

Initial value problems well-posed:

- **existence and uniqueness of solutions**
- **continuous dependence of solutions on initial data**
- **finite characteristic speeds**

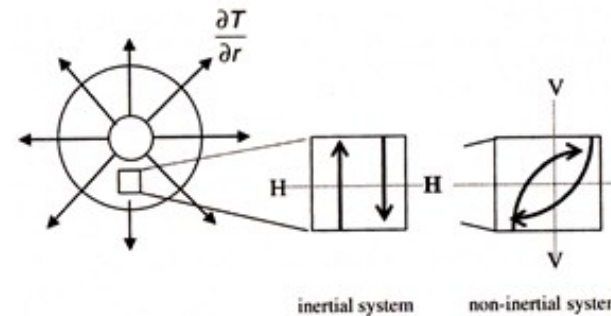


# Heat conduction in the gap between two coaxial cylinders

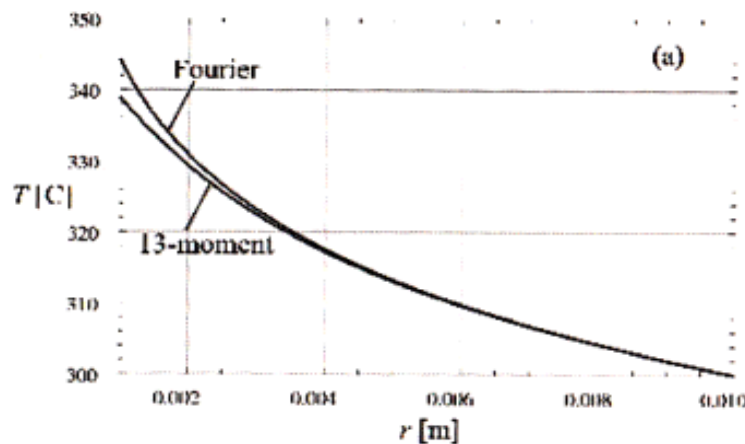
## Comparison between Fourier's law and Grad's 13-moment theory



H. Grad

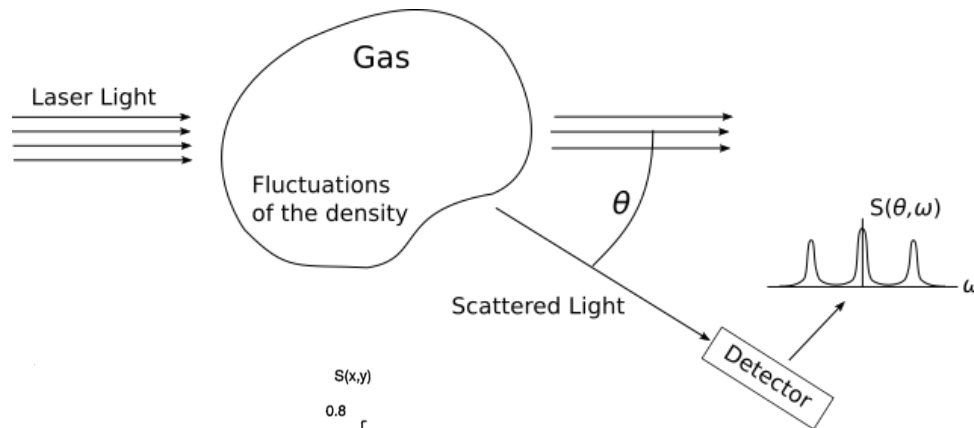


- A gas cannot rotate rigidly between the cylinders, if there is heat.
- A gas between the cylinders cannot be at rest on a turn table, if there is heat flow.

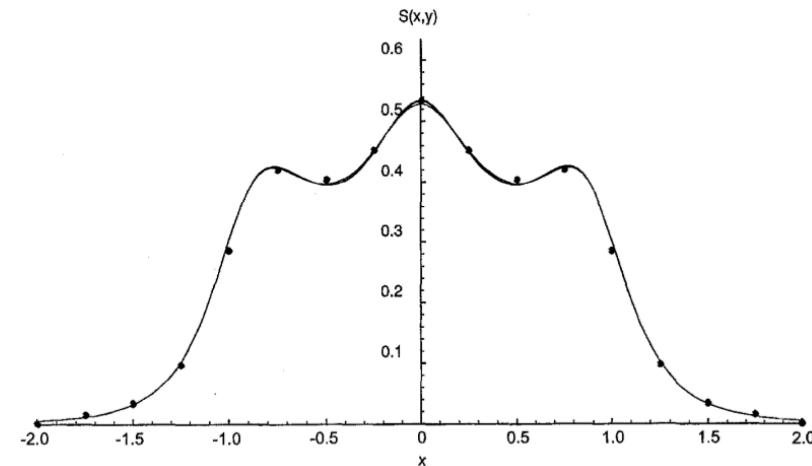
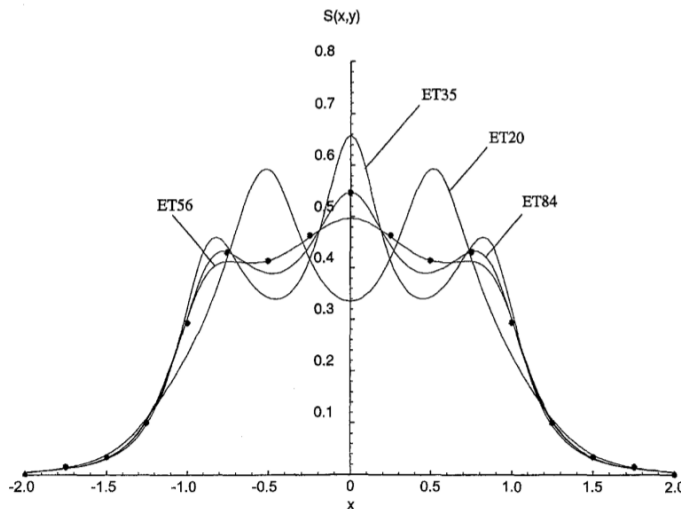


$$T = c_2 - \frac{c_1}{5^{k/\mu} p \tau} \ln \left( r^2 + \frac{56\tau}{75p} c_1 \right)$$

# Light Scattering



L. Onsager



ET 120, 165, 220, 286

ET is a theory of many theories with only **one** parameter: The number of fields.

For light scattering the theory provides results which are

- satisfactory (because continuum theory works)
- surprising (because theory provides its own limit of applicability)
- disappointing (because so many moments are needed)

## **Literature**

**Müller,I., Weiss,W. Entropy and Energy, a universal competition. Springer Verlag, Heidelberg (2005)**

**Müller,I., Ruggeri,T. Rational Extended Thermodynamics. Springer Verlag, New York (1998)**

**Müller,I., Weiss,W. Thermodynamics of irreversible processes -- past and present.  
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