

Quaternary Chalcogenide Semiconductors $Tl_2M^{II}M^{IV}_3Se_8$ and $Tl_2M^{II}M^{IV}X_4$

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Abstract: New quaternary thallium-containing chalcogenides $\text{Tl}_2\text{M}^{\text{II}}\text{M}^{\text{IV}}\text{X}_8$ and $\text{Tl}_2\text{M}^{\text{II}}\text{M}^{\text{IV}}\text{X}_4$ were synthesized, and their crystal structure was determined by XRD. Three $\text{Tl}_2\text{M}^{\text{II}}\text{M}^{\text{IV}}\text{X}_8$ chalcogenides crystallize in orthorhombic symmetry (S.G. $P2_12_12_1$; $\text{Tl}_2\text{CdGe}_3\text{Se}_8$ lattice parameters $a=0.76023(9)$, $b=1.2071(2)$, $c=1.7474(2)$ nm), eight isostructural $\text{Tl}_2\text{B}^{\text{II}}\text{D}^{\text{IV}}\text{X}_4$ compounds crystallize in tetragonal symmetry, S.G. $I-42m$. These compounds form in the quasi-ternary systems $\text{Tl}_2\text{X}-\text{M}^{\text{II}}\text{X}-\text{M}^{\text{IV}}\text{X}_2$ ($\text{X} = \text{S}, \text{Se}, \text{Te}$) at the component ratio 1:1:1 and 1:1:3 at the sections $\text{Tl}_2\text{M}^{\text{IV}}\text{X}_3-\text{B}^{\text{II}}\text{X}$ and $\text{Tl}_2\text{M}^{\text{II}}\text{M}^{\text{IV}}\text{X}_4-\text{M}^{\text{IV}}\text{X}_2$, respectively. The composition of the $\text{Tl}_2\text{CdGe}_3\text{Se}_8$ compound was additionally confirmed by SEM and EDS.

Keywords: crystal structure, thallium-containing chalcogenides, phase equilibria, powder X-ray diffraction.

Results and Discussion

Phase equilibria

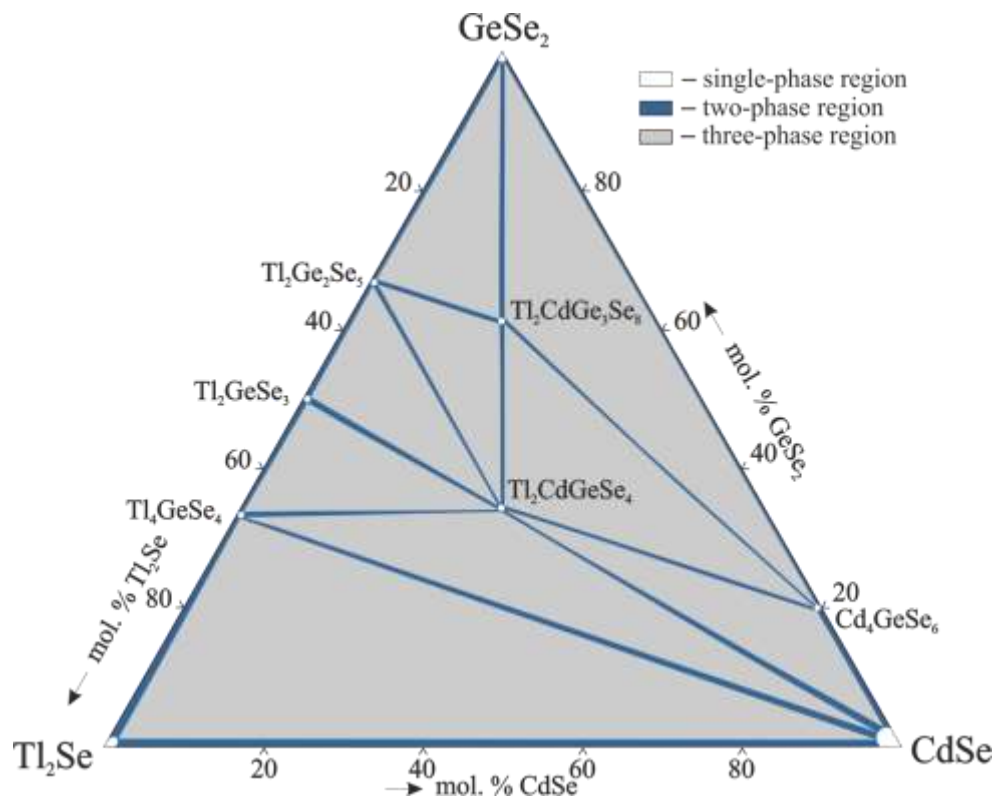


Fig. 1. Isothermal section of the quasi-ternary system Tl_2Se - CdSe - GeSe_2 at 570 K.

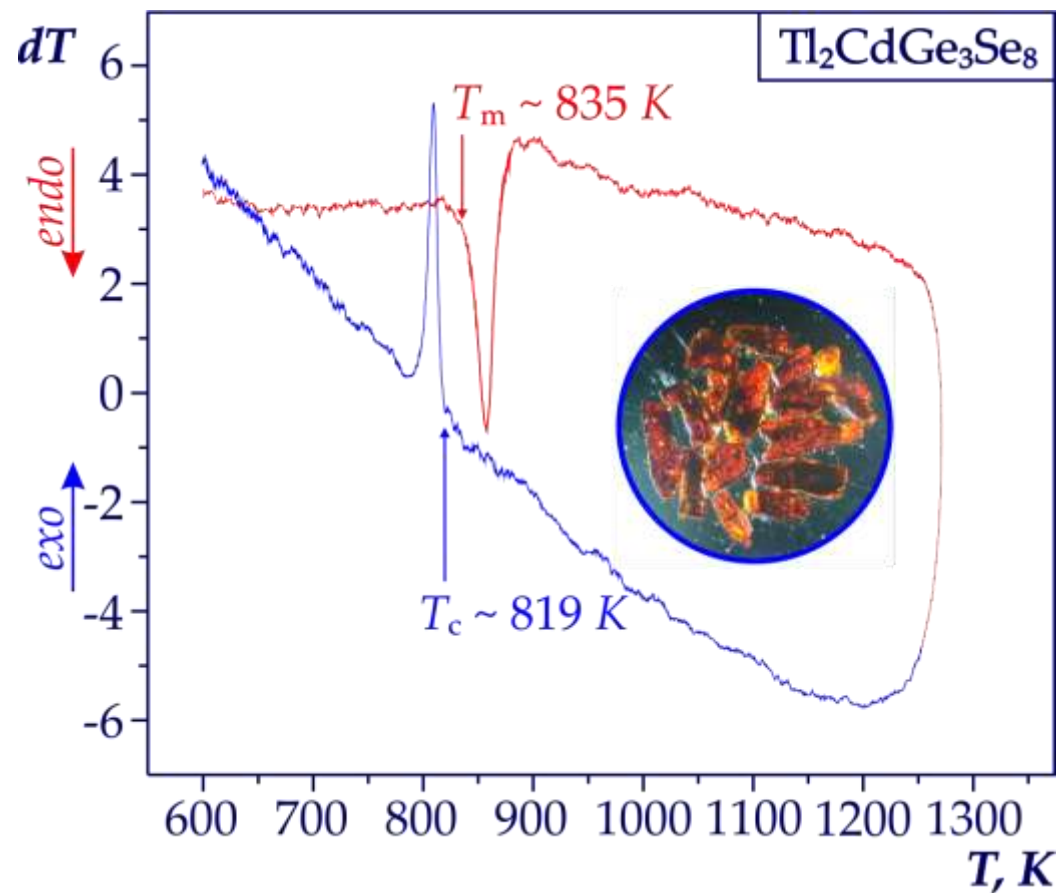


Fig. 2. DTA curve of the $\text{Tl}_2\text{CdGe}_3\text{Se}_8$ compound and photo of this compound

EDS analysis

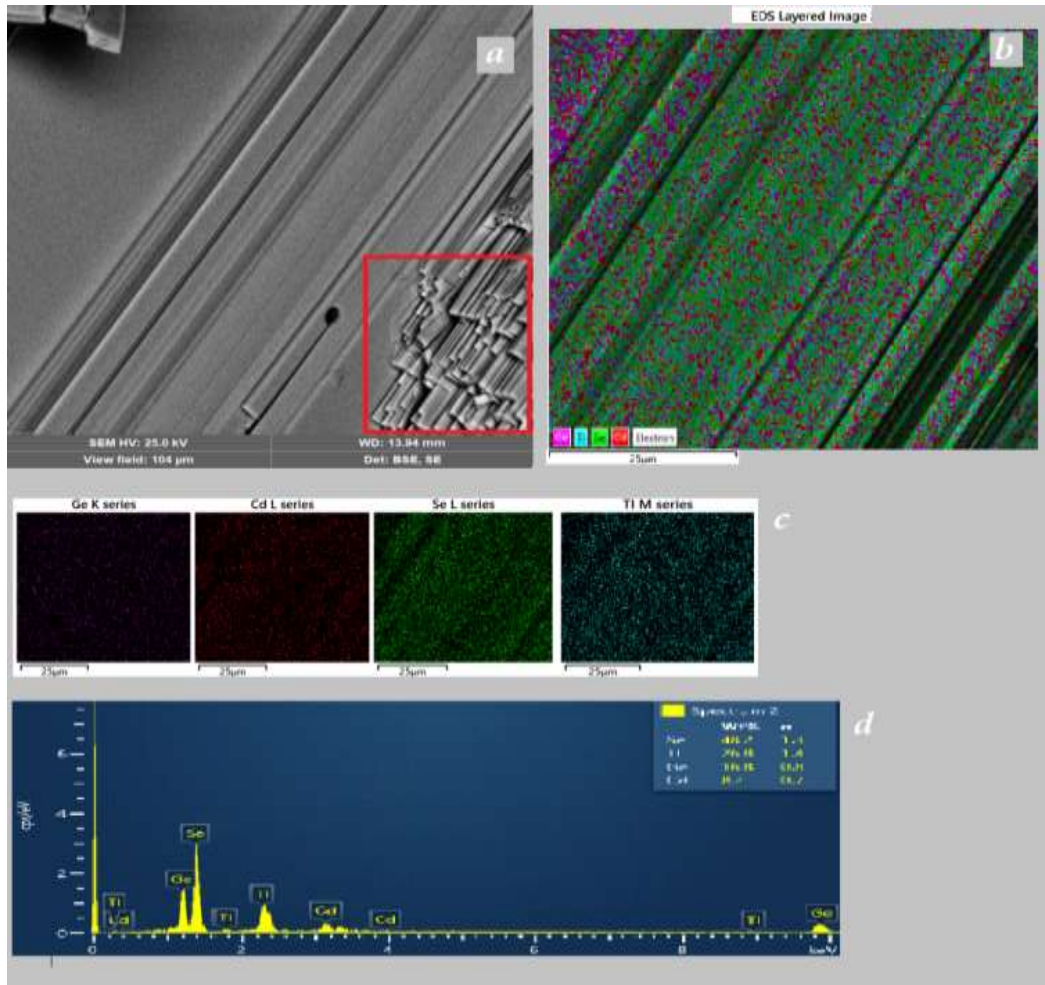


Figure 3. SEM/EDS results of the $\text{Tl}_2\text{CdGe}_3\text{Se}_8$ sample: microphotograph of the sample chip (*a*), EDS results with general mapping, element mapping, elemental composition (*b, c, d*)

Averaged formula of the investigation of 6 probes is $\text{Tl}_{1.79}\text{Cd}_{1.00}\text{Ge}_{2.99}\text{Se}_{7.83}$, which is fairly close to $\text{Tl}_2\text{CdGe}_3\text{Se}_8$.

Red square in Figure 3a shows the region where the formation of the layered structure is observed.

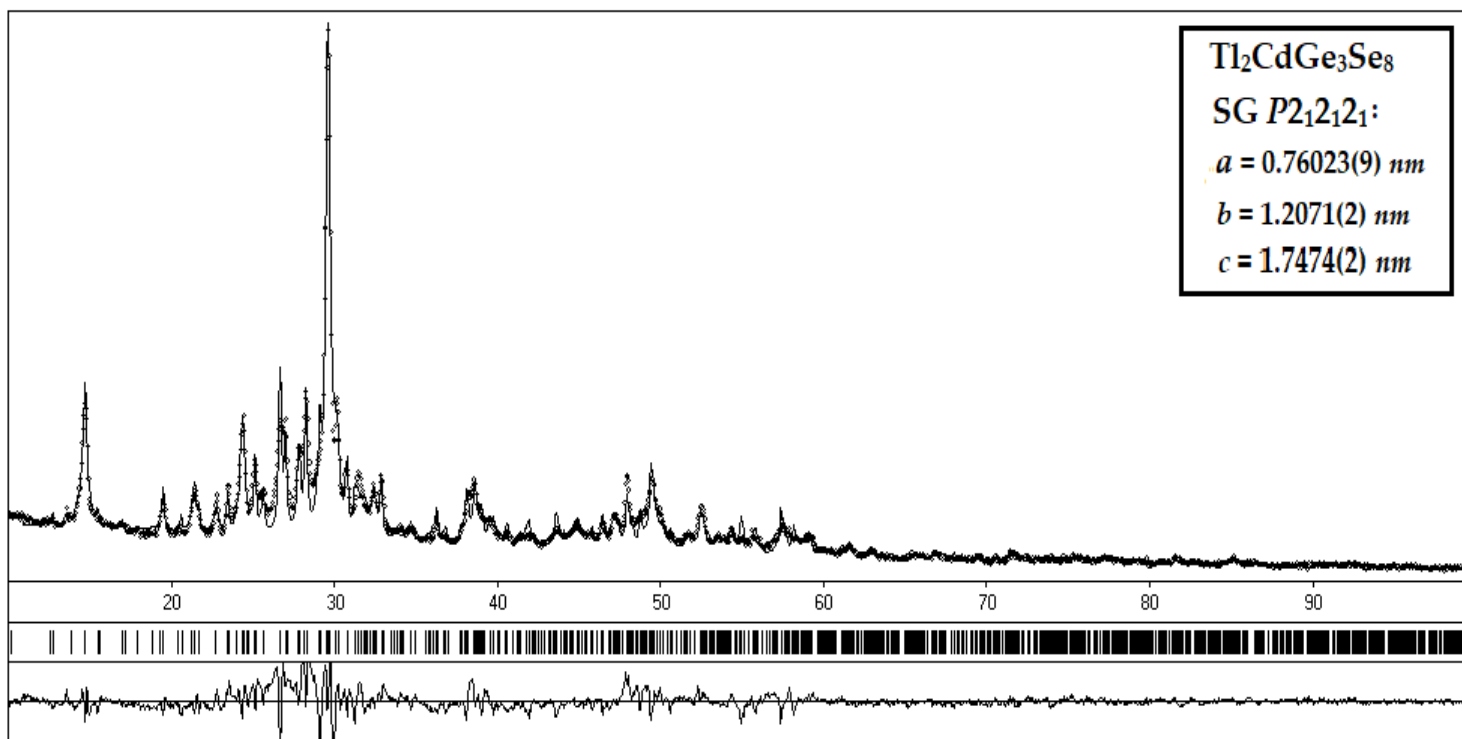
Crystal structure

Empirical formula	$\text{Tl}_2\text{CdGe}_3\text{Se}_8$
Space group	$P2_12_12_1$
Unit cell dimensions:	
a (nm)	0.76023(9)
b (nm)	1.2071(2)
c (nm)	1.7474(2)
V (nm ³)	1.6036(6)
F(000) (electrons)	2312.0
Number of atoms in cell	56.0
Calculated density (g/cm ³)	5.676(2)
Absorption coefficient (1/cm)	794.03
Radiation and wavelength	CuK_α 1.54185
Diffractometer	Powder
Mode of refinement	Full profile
Number of atom sites	14
Number of free parameters	58
2θ and $\sin\theta/l$ (max)	100.05 0.497
$h(\text{min}), k(\text{min}), l(\text{min})$	0 0 0
$h(\text{max}), k(\text{max}), l(\text{max})$	7 11 17
Number of atom sites	14
R_I	0.1204
R_p	0.2783
Scale factor	1.98(6)
Texture axis and parameter	58
Number of free parameters	[0 1 0] 0.134(5)

Atoms coordinates and isotropic temperature displacement parameters in the $\text{Tl}_2\text{CdGe}_3\text{Se}_8$ structure

Atom	x/a	y/b	z/c	$B(\text{is/eq}) \times 10^2, \text{nm}^2$	N
Ge1	0.512(4)	0.9622(12)	0.162(2)	1.37(8)	4
Ge2	0.762(5)	0.9173(11)	0.4877(12)	1.41(8)	4
Ge3	0.777(4)	0.9496(12)	0.8254(13)	1.57(8)	4
Se1	0.484(4)	0.8903(14)	0.7767(12)	1.75(8)	4
Se2	0.995(4)	0.4034(12)	0.7412(12)	1.87(8)	4
Se3	0.262(4)	0.6263(10)	0.8461(12)	1.20(8)	4
Se4	0.764(3)	0.6314(10)	0.8456(13)	1.43(8)	4
Se5	0.973(3)	0.6168(12)	0.4835(12)	0.88(8)	4
Se6	0.009(3)	0.3833(11)	0.9433(13)	1.52(8)	4
Se7	0.792(3)	0.8585(11)	0.9563(12)	1.54(8)	4
Se8	0.758(4)	0.8535(10)	0.3510(14)	1.81(8)	4
Cd	0.476(3)	0.9874(9)	0.6677(10)	1.45(8)	4
Tl1	0.284(2)	0.8871(6)	0.9490(6)	3.30(8)	4
Tl2	0.7094(14)	0.6949(6)	0.6650(7)	2.39(7)	4

Experimental and calculated diffraction patterns and their difference for $\text{Tl}_2\text{CdGe}_3\text{Se}_8$



Crystal structure of the $\text{Tl}_2\text{M}^{\text{II}}\text{M}^{\text{IV}}\text{X}_4$ compounds

Obtained quaternary equimolar 2-1-1-4 compounds ($\text{Tl}_2\text{CdGe}(\text{Sn})\text{Se}_4$, $\text{Tl}_2\text{HgSnS}_4$, $\text{Tl}_2\text{HgSi}(\text{Ge}, \text{Sn})\text{Se}_4$ and $\text{Tl}_2\text{Cd}(\text{Hg})\text{SiTe}_4$) are isostructural and crystallize in S.G. $I-42m$.

