

Yttria partially stabilized zirconia crystals and co-doped with neodymium, erbium or ytterbium oxides

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Introduction

Zirconia based materials have a variety of unique physicochemical, electrical and mechanical properties including high strength, hardness, impact toughness, wear resistance, low coefficient of friction, high melting point, chemical inertness, low heat conductivity and biocompatibility. These properties account for the wide range of applications, from wear resistant bearings to medical and surgical instruments.

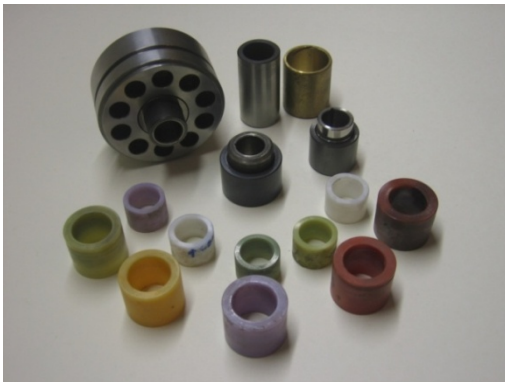
This work presents the results of studying the phase composition, structure, and mechanical properties of partially stabilized zirconia (PSZ) crystals with yttrium oxide and co-doped with cerium, neodymium, erbium, or ytterbium oxides with a total concentration of 3.0 mol.%.



Wire drawing dies



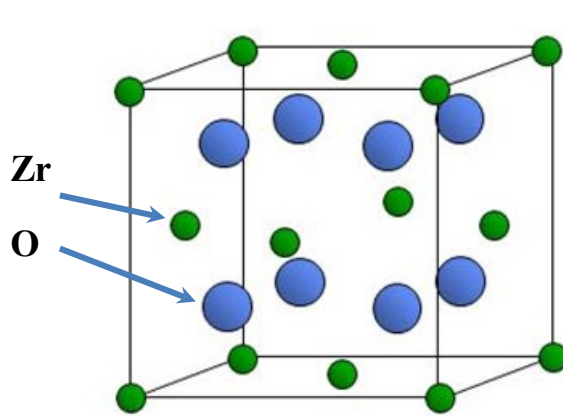
Scalpels and electrosurgical medical instrument



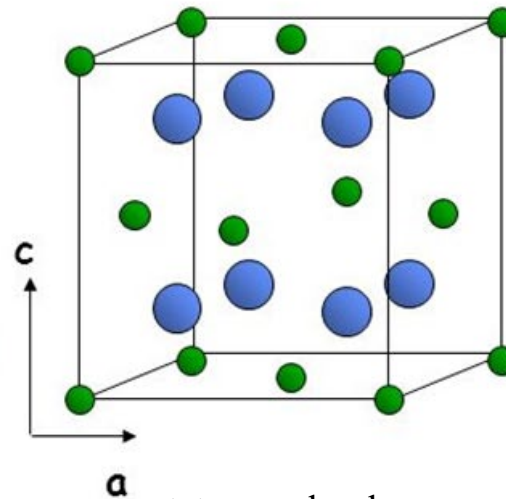
Sleeve bearings



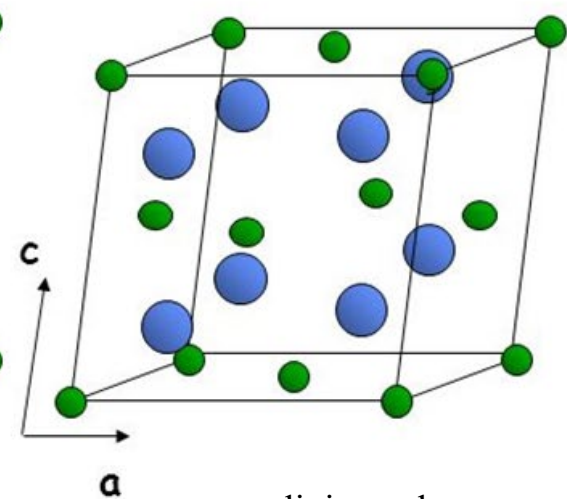
The three zirconia polymorphs



cubic *c*-phase ZrO_2
Fm $\bar{3}m$
2680 - 2370 °C
 $a=b=c$
 $\alpha=\beta=\gamma=90^\circ$



tetragonal *t*-phase
 $P4_2/nmc$
2370 - 1160 °C
 $a=b \neq c$
 $\alpha=\beta=\gamma=90^\circ$

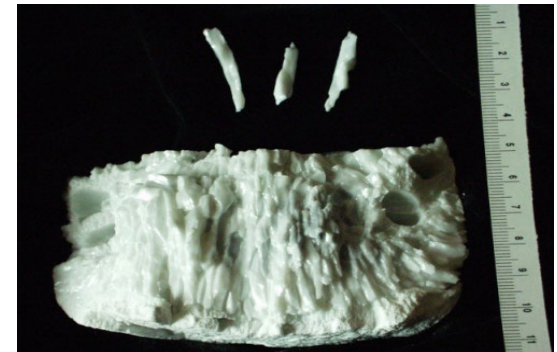


monoclinic *m*-phase
 $P2_1/C$
< 1160 °C
 $a \neq b \neq c$
 $\alpha=\gamma=90^\circ$ $\beta > 90^\circ$



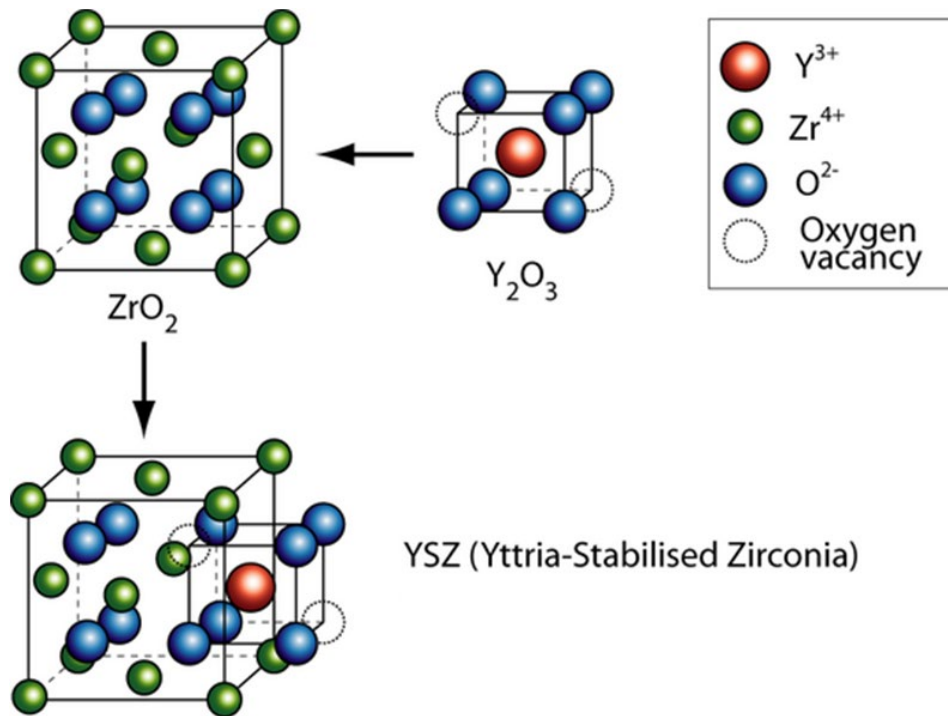
Crystals of the monoclinic phase of zirconia are needle-shaped and very small, so they do not find practical application.

The tetragonal and cubic phases are of interest!

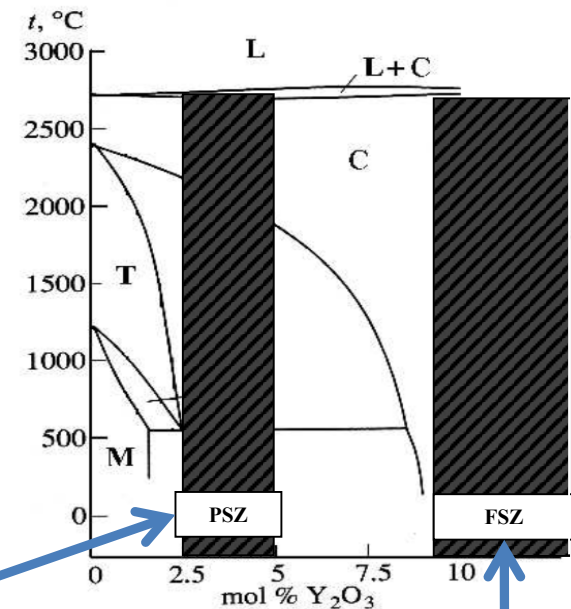


Stabilization of high-temperature zirconia phases

The cubic and tetragonal phases can be stabilized by doping with Y_2O_3 , Sc_2O_3 , CeO_2 ...



Phase diagram of the zirconia rich portion of $ZrO_2 - Y_2O_3$ system [1]



ZrO_2 - (2.5-5) mol.% Y_2O_3
 Partially Stabilized Zirconia (PSZ)
tetragonal phase

ZrO_2 - (9-40) mol.% Y_2O_3
 Fully Stabilized Zirconia (FSZ)
cubic phase

Crystal synthesis technology

The crystals were grown by directional crystallization technique with direct high-frequency heating in a cold container (skull melting)

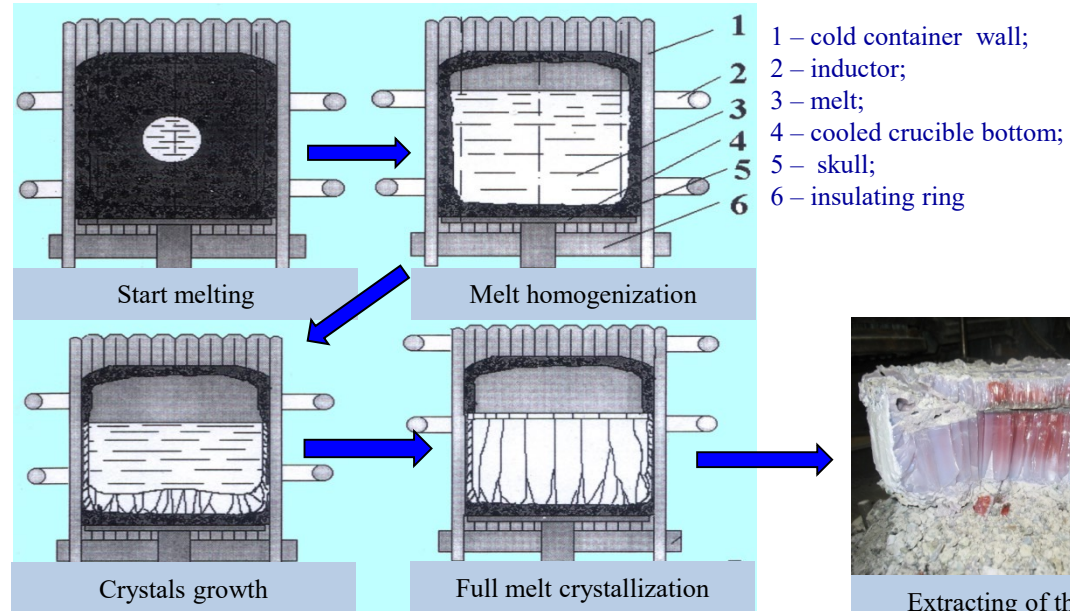
The main advantages of this method of synthesis :

No temperature limits (up to 3000 °C)

No contact with crucible material

No requirements for the particle size distribution of raw materials

The possibility of re-melting crystalline waste



Extracting of the bulk

The "Kristall-407" installation

Electric power	60 kW
Electromagnetic field frequency	5.28 MHz
Container diameter	130 mm
Mass of melt	4 -6 kg
Working atmosphere	air



The separation of the bulk into individual crystals

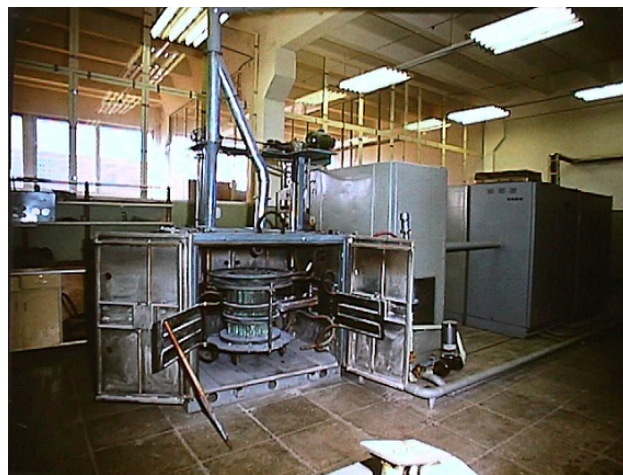
Growth installations for the synthesis of the crystals



*A view of cold container (CC)
of 130 mm diameter*



*The "Kristall-401" installation
(CC of 180 mm diameter)*



*The "Kristall-403" installation
(CC of 300 mm diameter)*

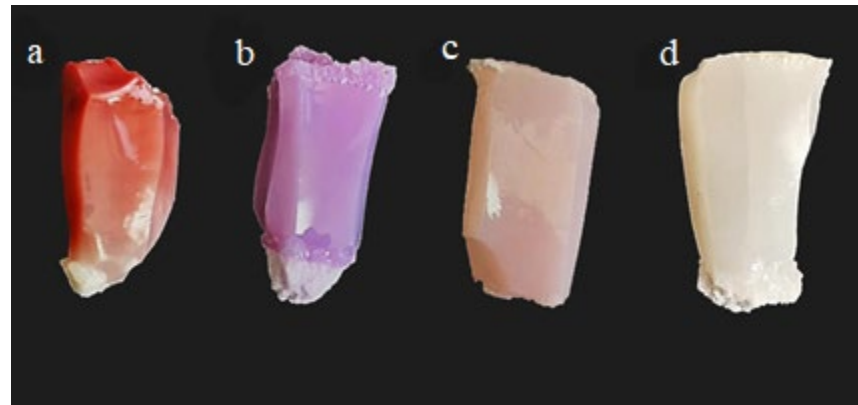


*The "Kristall-403M" installation
(CC of 700 mm diameter)*

Growth of the crystals

The compositions of the grown crystals, their symbols and density

Compositions of the crystals	Symbol	Density, g/sm ³
$(\text{ZrO}_2)_{0.97}(\text{Y}_2\text{O}_3)_{0.025}(\text{Ce}_2\text{O}_3)_{0.005}$	2.5Y0.5CeSZ	6.097±0.002
$(\text{ZrO}_2)_{0.97}(\text{Y}_2\text{O}_3)_{0.025}(\text{Nd}_2\text{O}_3)_{0.005}$	2.5Y0.5NdSZ	6.104±0.002
$(\text{ZrO}_2)_{0.97}(\text{Y}_2\text{O}_3)_{0.025}(\text{Er}_2\text{O}_3)_{0.005}$	2.5Y0.5ErSZ	6.117±0.001
$(\text{ZrO}_2)_{0.97}(\text{Y}_2\text{O}_3)_{0.025}(\text{Yb}_2\text{O}_3)_{0.005}$	2.5Y0.5YbSZ	6.119±0.003



Appearance of the crystals

2.5Y0.5CeSZ (a), 2.5Y0.5NdSZ (b), 2.5Y0.5ErSZ (c), 2.5Y0.5YbSZ (d)

Phase composition of the crystals

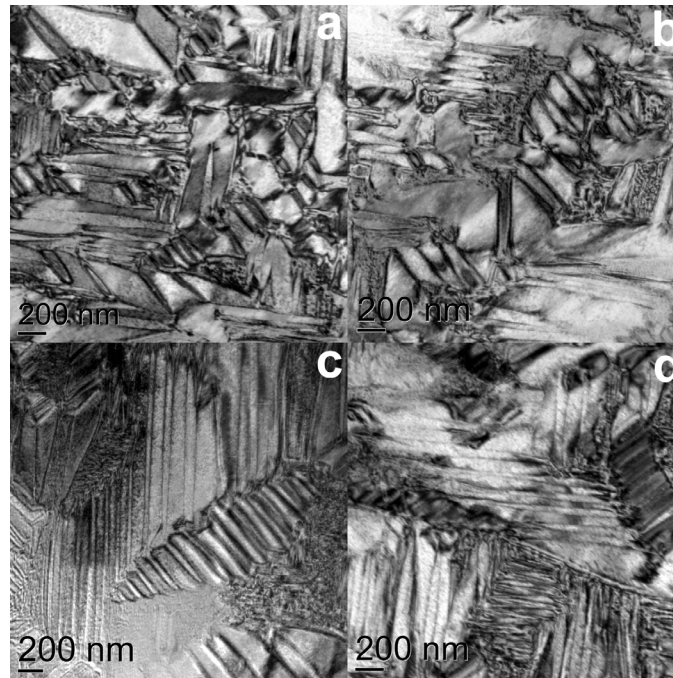
The phase analysis of crystals was studied by X-ray diffractometry on a Bruker D8 installation on plates cut from different parts of the crystal perpendicular to the $\langle 100 \rangle$ direction.

Sample	Phase composition	Weight Fraction, %	Lattice parameters, Å	$c/\sqrt{2a}$
2.5Y0.5CeSZ	t	90(5)	$a = 3.605(1), c = 5.177(1)$	1.015
	t'	10(5)	$a = 3.621(1), c = 5.155(2)$	1.0065
2.5Y0.5NdSZ	t	90(5)	$a = 3.606(1), c = 5.177(1)$	1.015
	t'	10(5)	$a = 3.623(1), c = 5.155(2)$	1.006
2.5Y0.5ErSZ	t	85(5)	$a = 3.606(1), c = 5.175(1)$	1.015
	t'	15(5)	$a = 3.620(1), c = 5.152(2)$	1.006
2.5Y0.5YbSZ	t	85(5)	$a = 3.606(1), c = 5.174(1)$	1.0145
	t'	15(5)	$a = 3.618(1), c = 5.151(2)$	1.007

PSZ crystals are constituted of two tetragonal phases: untransformable t' -phase and transformable t-phase which is transformed into a monoclinic phase when a mechanical load is applied.

Structure of the crystal

Study of the crystal structure by transmission electron microscopy showed that all crystals contained twins. For all doped crystals, a mixture of large and finely dispersed twins was observed, similar in character to the twin structure of YSZ crystals [1].

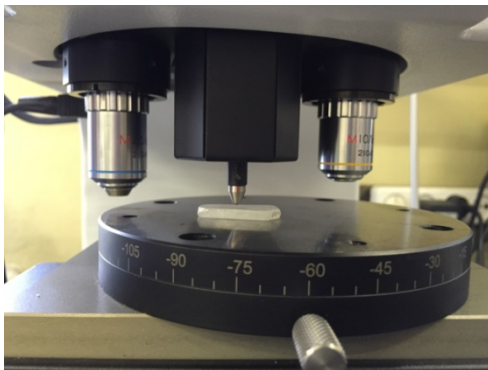


Images of the twin domain structure in the crystals obtained using TEM:
2.5Y0.5CeSZ (a), 2.5Y0.5NdSZ (b), 2.5Y0.5ErSZ (c), 2.5Y0.5YbSZ (d)

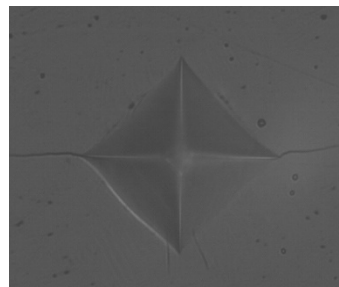
Properties of the crystal

Microhardness and fracture toughness obtained on plates cut from crystals perpendicular to different crystallographic directions for crystals. These values were measured by the Vickers method

Sample	Microhardness H, GPa	K_{1c} , $\text{MPa}\cdot\text{m}^{1/2}$
2.5Y0.5CeSZ	13.93± 0.50	10.1±0.7
2.5Y0.5NdSZ	12.94 ± 0.50	9.12± 0.6
2.5Y0.5ErSZ	12.92 ± 0.50	7.90±0.6
2.5Y0.5YbSZ	13.20 ± 0.50	7.45±0.6



DM 8 B AUTO microhardness tester



Indentation with cracks

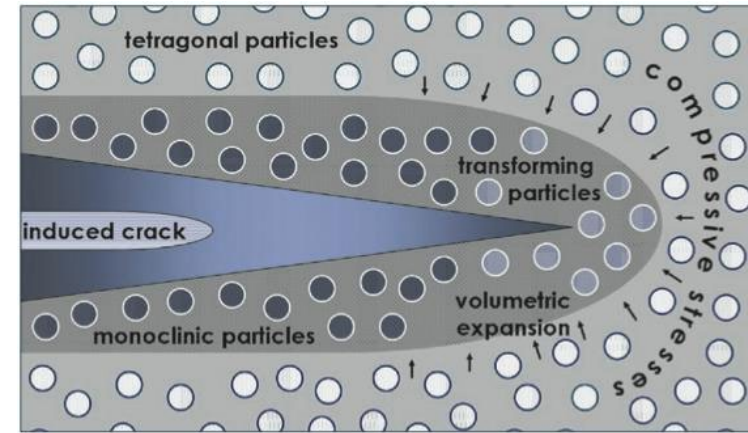
$$K_{1c} = 0.035(L/a)^{-1/2}(CE/H)^{2/5}Ha^{1/2}C^{-1}$$

K_{1c} — fracture toughness [$\text{MPa} \cdot \text{m}^{1/2}$],
 L — radial crack length [m],
 a — indentation half-width [m],
 C — Poisson's ratio,
 E — Jung modulea [Pa],
 H — Microhardness [Pa].

Transformational hardening mechanism in PSZ crystals

Sample	Phase composition	Weight Fraction, %	K_{1c} , $\text{MPa}\cdot\text{m}^{1/2}$
2.5Y0.5CeSZ	t	90(5)	10.1±0.7
	t'	10(5)	
2.5Y0.5NdSZ	t	90(5)	9.12± 0.6
	t'	10(5)	

$t \rightarrow m$ phase transformation!



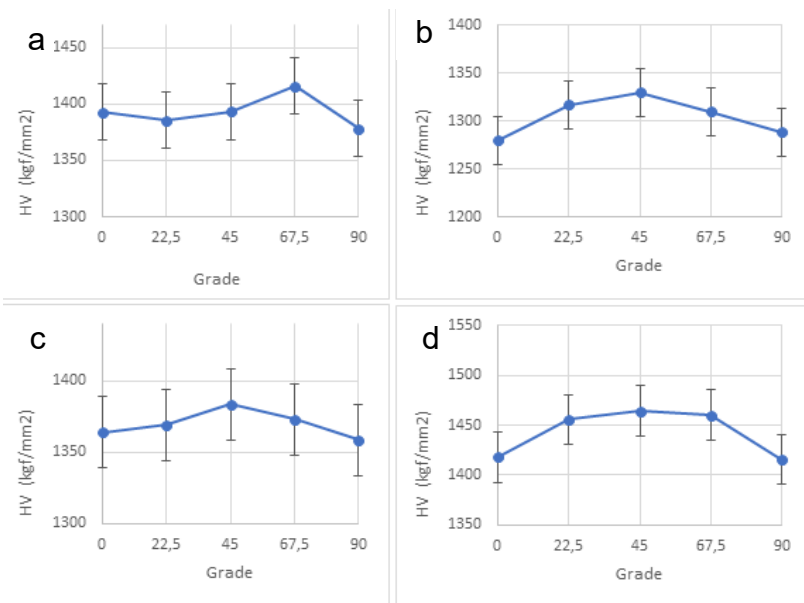
- The tetragonal - monoclinic phase transformation involves a 4.7% volume increase.
- This volume increase is the basis for transformation toughening.

The fracture toughness of zirconia base materials is commonly associated with the transformation hardening the main mechanism of which at room temperature is the phase transition from the **metastable tetragonal phase to the stable monoclinic one induced by the mechanical stresses** around the propagating cracks.

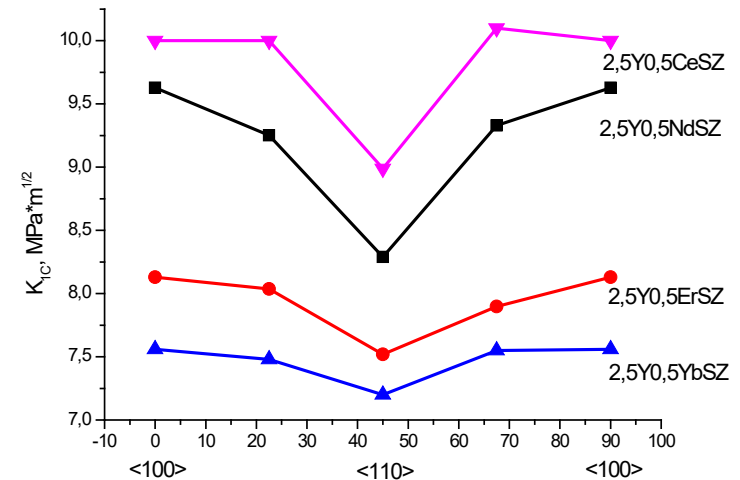
The fracture toughness is then proportional to the transformable **t-phase content**.

Anisotropy of crystal properties

- No anisotropy of microhardness was found. The microhardness values for different orientations of the indenter diagonals in the $\{100\}$ plane of the sample varied within the measurement error.
- The magnitude of the fracture toughness when the indenter diagonals are oriented in the $\langle 100 \rangle$ crystallographic direction is higher than when the indenter diagonals are oriented in the $\langle 110 \rangle$ crystallographic direction.

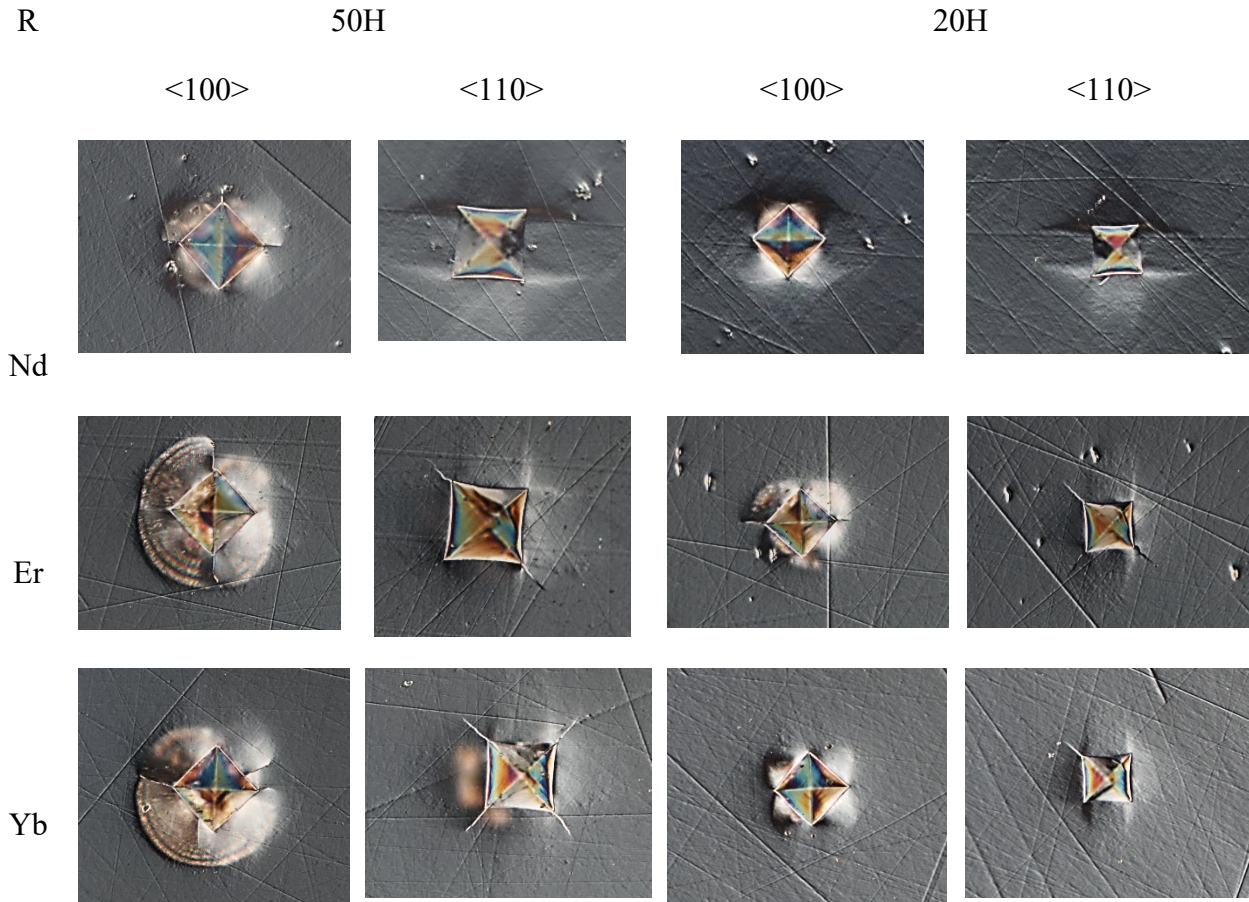


Anisotropy diagrams of microhardness values for crystals 2.5Y0.5CeSZ (a), 2.5Y0.5NdSZ (b), 2.5Y0.5ErSZ (c) and 2.5Y0.5YbSZ (d)



Anisotropy diagrams of fracture toughness of crystal samples depending on the direction of the indenter diagonals relative to the crystallographic orientation in the plane of the sample $\{100\}$

Anisotropy of crystal properties

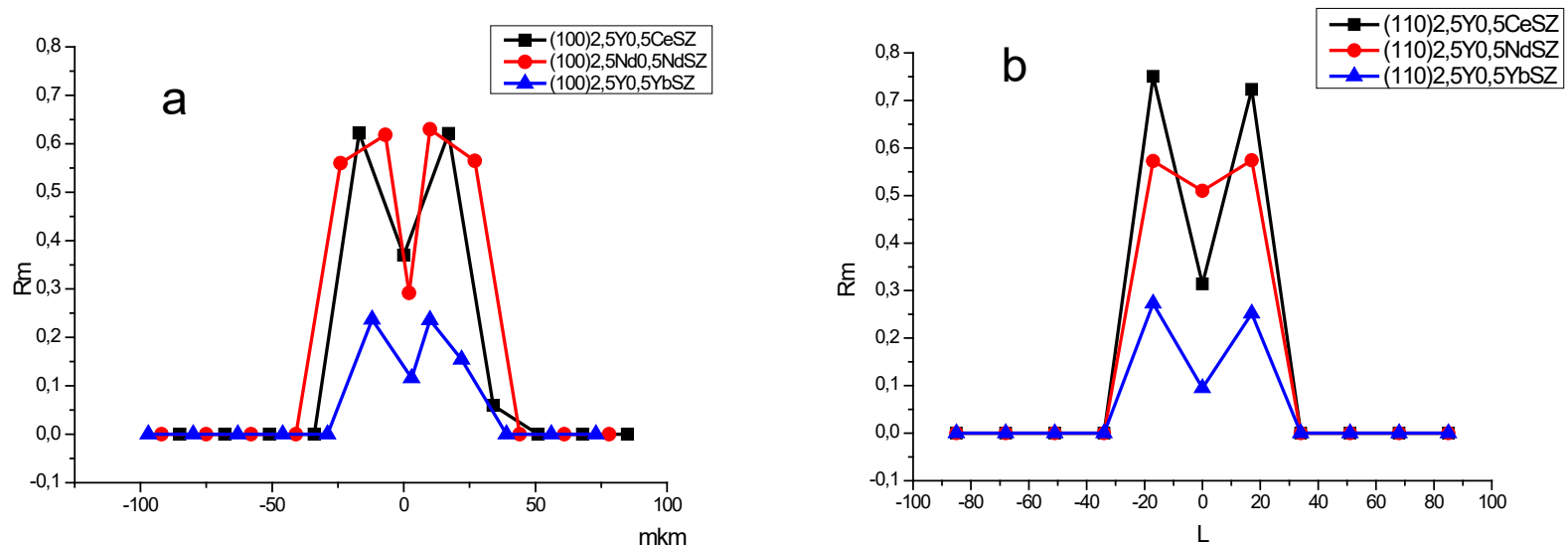


Images of the surface of 2.5Y0.5RSZ crystals (R-Nd, Er, Yb) with indentation at a load of 20 and 50 N for the orientation of the indenter diagonals along the directions $\langle 100 \rangle$ and $\langle 110 \rangle$.

Properties of the crystal

The distribution of the monoclinic phase around the indentation was studied. A comparative analysis of the intensity of the tetragonal-monoclinic transformation was carried out using the method of Raman spectroscopy. The estimation of the degree of intensity of the tetragonal-monoclinic transformation depending on the ratio of the intensities of the lines of the tetragonal and monoclinic phases of the Raman spectrum inside and around the indentations was carried out using the relation

$$R_m = \frac{I_{178}^m + I_{190}^m}{I_{146}^t + I_{178}^m + I_{190}^m}$$



Dependence of the value of the degree of transformation R_m in the region of the indentation for 2.5Y0.5RSZ (R-Ce, Nd, Yb) crystals on the {100} plane with the orientation of the indenter diagonals in the <100> direction (a) and in the <110> direction (b).

Summary

- PSZ crystals are constituted of two tetragonal phases (t and t') differing by its tetragonality (c/a).
- PSZ crystals have developed twin domain structure.
- The study of the mechanical characteristics of the crystals showed that co-doping has an insignificant effect on the change in microhardness values.
- The value of the fracture toughness of crystals increases with an increase in the radius of the rare earth element of the co-doped oxide and depends on crystallographic orientation.
- Comparison of the obtained data on the dependence of the tetragonal - monoclinic phase transformations intensity with the data on the fracture toughness of crystals shows a general tendency towards a decrease in the values of fracture toughness with a decrease in the intensity of the tetragonal - monoclinic phase transformations.

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