

# Electronic, Magnetic, and Optical Properties of Double Perovskites $\text{Pb}_2\text{XOsO}_6$ ( $\text{X}=\text{Co}, \text{Ni}$ )

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# Outline

- Introduction
- Methods and Computational Details
- Crystal Structure
- Results and Discussion
- Conclusions

# Introduction

## Materials Type

- Metals
- Half-metals
- Semiconductors
- Insulators

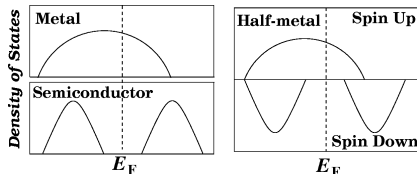


Fig: Classification of materials according to their energy gap [I. Galanakis *et al.*, *arXiv preprint cond-mat/0510276* (2005)]

## Double Perovskites (DPs)

- Most common DPs are  $A_2BB'X_6$
- DPs covers all types of materials
- Shows remarkable properties like structural stability, superconductivity, half-metallicity etc

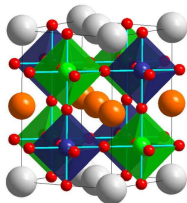


Fig: Ideal DPs structure [K. Momma *et al.*, *J. Appl. Crystallogr.* **44**, 1272 (2011)]

# Introduction

## Applications

- Spintronics devices
- Photovoltaic application
- Laser light and light-emitting diode
- Microelectronics and telecommunication

## Osmium based DPs

- $\text{Sr}_2\text{CrOsO}_6$  → Half metallic antiferromagnet <sup>1</sup>
- $\text{Ba}_2\text{NiOsO}_6$  → Dirac-Mott insulating ferromagnet <sup>2</sup>
- $\text{Ca}_2\text{CrOsO}_6$  → Ferrimagnetic insulator <sup>3</sup>
- $\text{SrLaFeOsO}_6$  → Metallic antiferromagnet <sup>4</sup>

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<sup>1</sup>K. W. Lee *et al.*, *Phys. Rev. B* **77**, 115101 (2008)

<sup>2</sup>H. L. Feng *et al.*, *Phys. Rev. B* **94**, 235158 (2016)

<sup>3</sup>S. R. Bhandari *et al.*, *RSC Adv.* **10**, 16179 (2020)

<sup>4</sup>D. K. Yadav *et al.*, *Comput. Mater. Sci.* **170**, 109168 (2019) 

# Methods and Computational Details

## Theoretical approach

- Performed density functional calculation to explore electronic and magnetic properties
- Used GGA functional
- GGA+ $U$  functional considered to account strong correlation effect of Co-3*d*, Ni-3*d* and Os-5*d*
- $U$ -value for  $\text{Pb}_2\text{CoOsO}_6 \rightarrow \text{Co}=4 \text{ eV}$  and  $\text{Os}=2 \text{ eV}$
- $U$ -value for  $\text{Pb}_2\text{NiOsO}_6 \rightarrow \text{Ni}=5 \text{ eV}$  and  $\text{Os}=2 \text{ eV}$

## Computational details

- Full potential local orbital (FPLO) code is used for calculation
- $K$ -mesh used for band structure and density of state  $\rightarrow 12 \times 12 \times 10$
- $K$ -mesh used for optical properties  $\rightarrow 20 \times 20 \times 20$
- Convergence criteria: Energy  $\rightarrow 10^{-8}$  Hartree and charge convergence  $\rightarrow 10^{-6}$  electronic charge

# Crystal Structure

$\text{Pb}_2\text{CoOsO}_6$

## Crystal Information:

- Space group- 14 ( $P121/c1$   
→ monoclinic)
- Lattice parameters:  
 $a = 5.6365 \text{ \AA}$ ;  $b = 5.5836 \text{ \AA}$ ;  
 $c = 7.82833 \text{ \AA}$
- Angles:  
 $\beta = 89.815^\circ$ ;  $\alpha = \gamma = 90^\circ$
- By replacing Co by Ni from  $\text{Pb}_2\text{CoOsO}_6$ ,  $\text{Pb}_2\text{NiOsO}_6$  is formed.

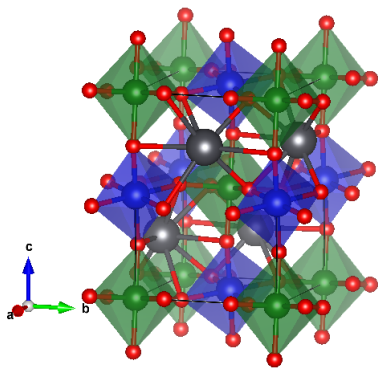


Fig: Crystal structure of  $\text{Pb}_2\text{XOsO}_6$  ( $X = \text{Co}, \text{Ni}$ ). Here green octahedra are due to  $\text{XO}_6$  and blue octahedra are due to  $\text{OsO}_6$ .

# Results and Discussion

## Ground State and Magnetic Moment

- Performed energy calculation for different magnetic configurations: one ferromagnetic state (FM- $\uparrow\uparrow\uparrow\uparrow$ ) two antiferromagnetic states (AF1- $\uparrow\downarrow\uparrow\downarrow$ , AF2- $\uparrow\downarrow\downarrow\uparrow$ ), two ferrimagnetic states (FI1- $\uparrow\uparrow\downarrow\downarrow$ , FI2- $\uparrow\uparrow\uparrow\downarrow$ ), and non-magnetic state.
- $\text{Pb}_2\text{CoOsO}_6$  has antiferromagnetic (AF2- $\uparrow\downarrow\downarrow\uparrow$ ) ground state.
- $\text{Pb}_2\text{NiOsO}_6$  has ferrimagnetic (FI1- $\uparrow\uparrow\downarrow\downarrow$ ) ground state.

Table: Magnetic moment of  $\text{Pb}_2\text{CoOsO}_6$

Atoms	GGA ( $\mu_B$ )	GGA+U ( $\mu_B$ )
Co	2.68	3.00
Os	0.84	1.63
Total	0.00	0.00

Table: Magnetic moment of  $\text{Pb}_2\text{NiOsO}_6$

Atoms	GGA ( $\mu_B$ )	GGA+U ( $\mu_B$ )
Ni	1.47	1.96
Os	1.32	1.69
Total	0.01	0.00

# Results and Discussion

## Density of State

- $\text{Pb}_2\text{CoOsO}_6$  is metal and  $\text{Pb}_2\text{NiOsO}_6$  is half metal.
- No contribution of Co and Ni around Fermi level.
- Strong hybridization between Os-5d and O-2p near the Fermi level.

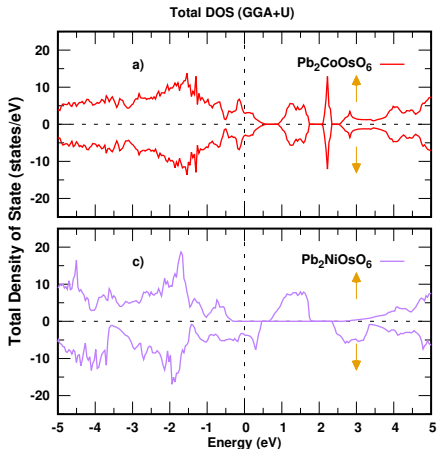


Fig: Total density of state of (a)  $\text{Pb}_2\text{CoOsO}_6$  (b)  $\text{Pb}_2\text{NiOsO}_6$  within GGA+U.



# Results and Discussion

## Partial density of State

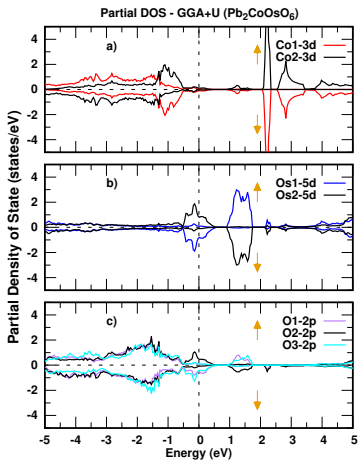


Fig: Partial density of state of  $\text{Pb}_2\text{CoOsO}_6$  within GGA+U.

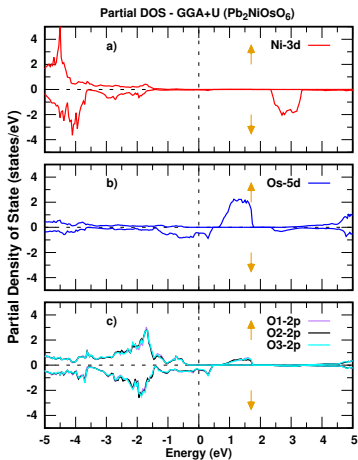


Fig: Partial density of state of  $\text{Pb}_2\text{NiOsO}_6$  within GGA+U.

# Results and Discussion

## Band Structures

- Metallic nature in  $\text{Pb}_2\text{CoOsO}_6$  and half-metallic nature in  $\text{Pb}_2\text{NiOsO}_6$ .

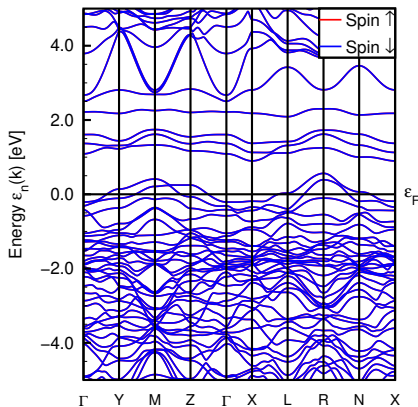


Fig: Band structures of  $\text{Pb}_2\text{CoOsO}_6$  within GGA+U.

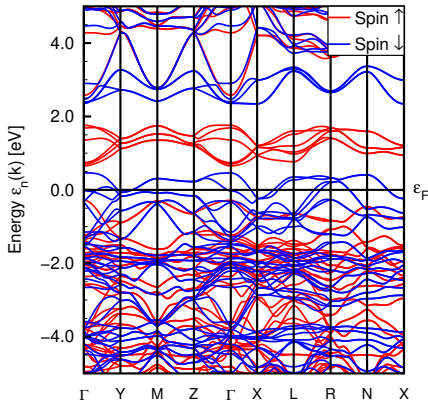


Fig: Band structures of  $\text{Pb}_2\text{NiOsO}_6$  within GGA+U.

# Results and Discussion

## Optical Properties

### $\text{Pb}_2\text{CoOsO}_6$

- Optically active in the energy range  $\rightarrow$  0 - 6.4 eV
- Significant intraband transition

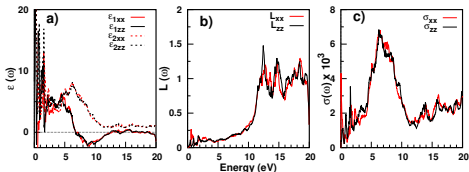


Fig: a) real ( $\epsilon_1$ ) and imaginary ( $\epsilon_2$ ) part of dielectric function, b) Loss function c) optical conductivity ( $\text{ohm}^{-1}\text{cm}^{-1}$ ) of  $\text{Pb}_2\text{CoOsO}_6$  within GGA+U.

### $\text{Pb}_2\text{NiOsO}_6$

- Optically active in the energy range  $\rightarrow$  1.4 eV - 6.8 eV.

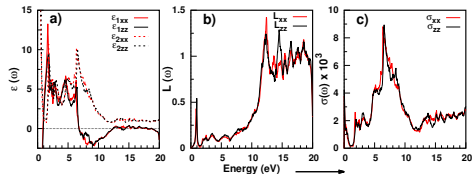


Fig: a) real ( $\epsilon_1$ ) and imaginary ( $\epsilon_2$ ) part of dielectric function, b) Loss function c) optical conductivity ( $\text{ohm}^{-1}\text{cm}^{-1}$ ) of  $\text{Pb}_2\text{NiOsO}_6$  within GGA+U.

# Conclusions

- $\text{Pb}_2\text{CoOsO}_6$  is antiferromagnetic metal whereas  $\text{Pb}_2\text{NiOsO}_6$  is ferrimagnetic half-metal.
- With electron doping in  $\text{Pb}_2\text{CoOsO}_6$ , We found electronic and magnetic phase transition.
- We observed high intraband contribution in  $\text{Pb}_2\text{CoOsO}_6$  as compare to  $\text{Pb}_2\text{NiOsO}_6$
- $\text{Pb}_2\text{CoOsO}_6$  is optically active in infrared, visible, and lower energy range of the ultraviolet region.
- $\text{Pb}_2\text{NiOsO}_6$  is found to be more active in the visible region and lower energy range of the ultraviolet region.

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Thank YOU

