



**UMS**  
UNIVERSITI MALAYSIA SABAH

# HIGHLY ACTIVE PANDANUS NANOCELLULOSE-SUPPORTED POLY(AMIDOXIME) COPPER (II) COMPLEX FOR ULLMANN CROSS- COUPLING REACTION

**BY:**

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**SUPERVISED BY:**

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CO-SUPERVISOR	DR. MOHD SANI BIN SARJADI

# BACKGROUND

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- ❑ Synthesis process is necessary due to the **molecular complexity** and various type of **bond formation** through the organic transformations is being grown in parallel fashion.
- ❑ C-C or C-N bonds formation (cross-coupling) reactions are **important** for the synthesis of essential chemicals such as:
  1. Fine chemicals,
  2. Drug and intermediate products,
  3. Natural products etc.
- ❑ Transition metal catalysts (Pd, Cu, Ni) are normally used for cross-coupling reactions (suzuki, heck, sonogashira, click etc.).

# Catalyst

Homogeneous

Heterogeneous

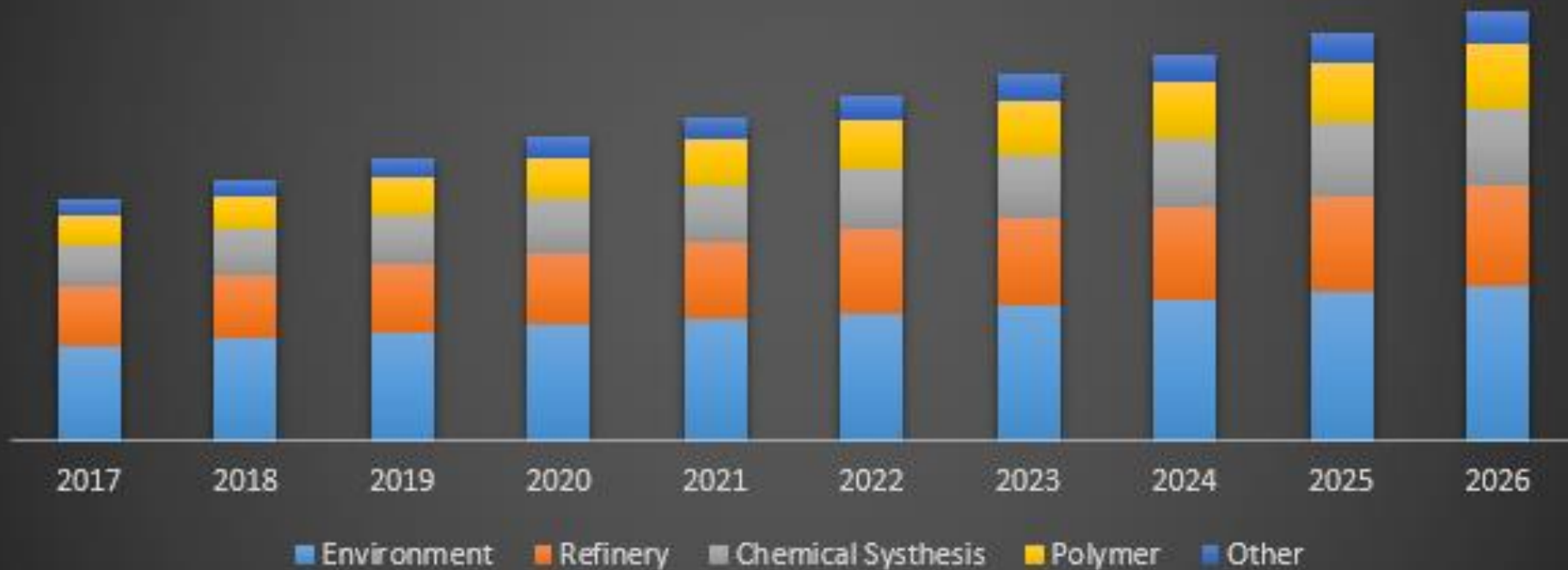
## Homogeneous catalyst

- Expensive
- Purification of product is difficult
- Can not be reused
- Environmental pollution
- Not stable in the reaction media

## Heterogeneous catalyst

- Less expensive
- Easy to purify the product
- Reusable
- Environmental-friendly
- Stable under harsh conditions

## Global Catalyst Market Size By Application



Price of  
Therefore  
chemist  
research

sustainable  
of the



# OBJECTIVES

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- To extract the cellulose from the agro-waste (pandanus fruit fibre) and utilize to synthesize a poly(amidoxime) ligand,
- To prepare heterogeneous copper catalysts from the cellulose-supported amidoxime ligands,
- To evaluate the catalytic ability and reusability of synthesized catalyst in Ullmann cross-coupling reaction.



# METHODOLOGY

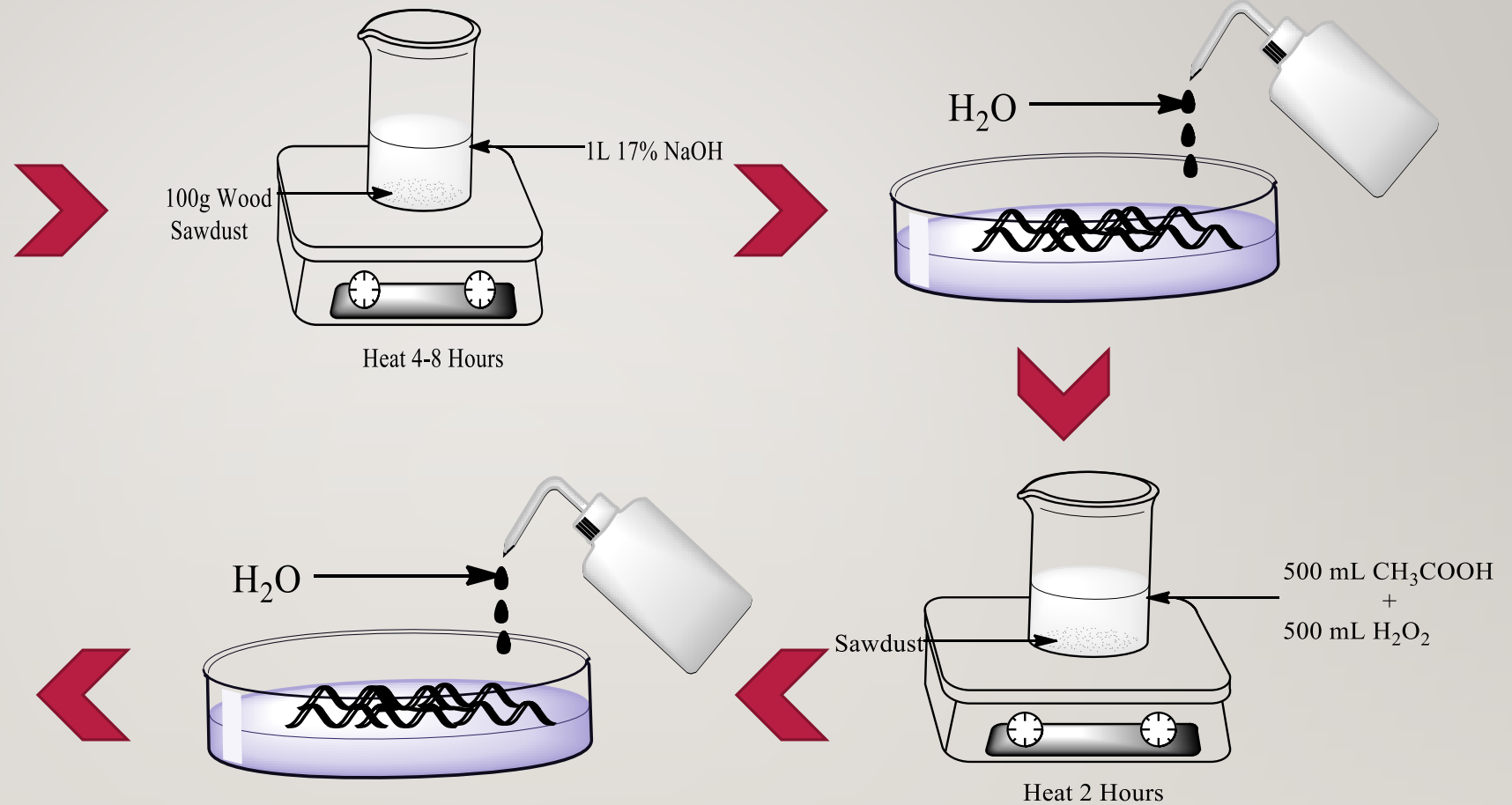
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# Cellulose Extraction

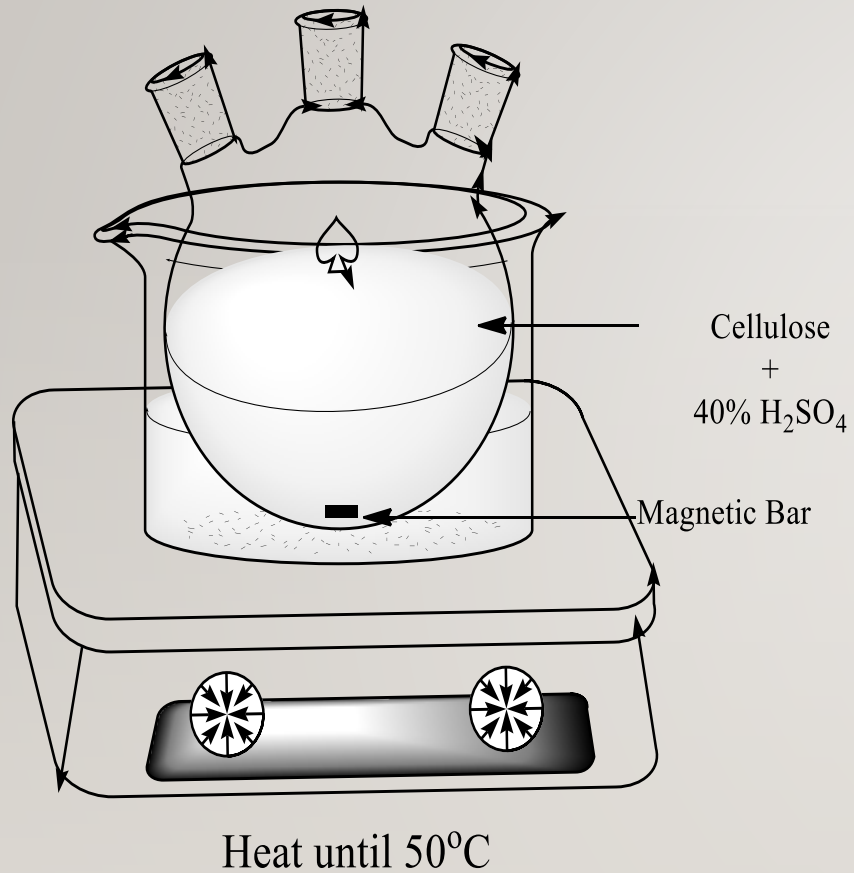
Filter the wood sawdust and weight 100g of fiber.



Dry the cellulose with oven at 50°C for one day and weight the mass of cellulose obtain

(Sources: Rahman *et al.*, 2016)

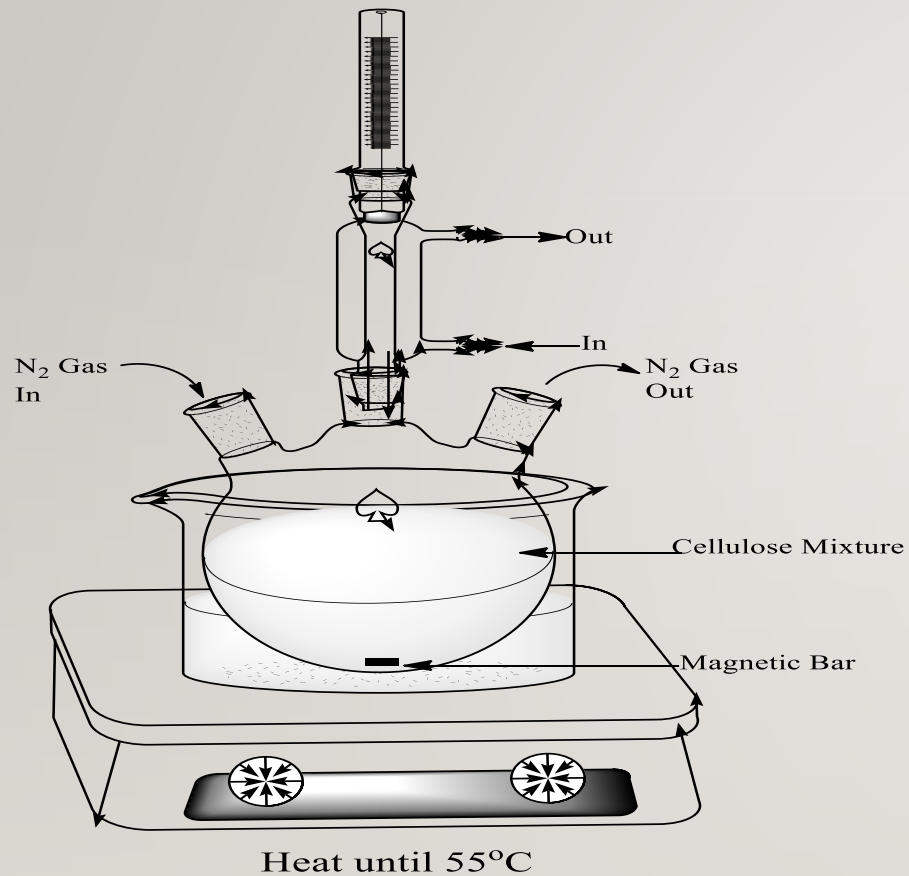
# Synthesize nanocellulose



Procedure:

1. Boil the cellulose with 40% of H<sub>2</sub>SO<sub>4</sub> for 1 hours.
2. Pour the mixture into the cool water after reaction done.
3. Neutralize the solution using NaOH .
4. Wash and dry.

# Graft Copolymerization (Poly(acrylonitrile))



Procedure:

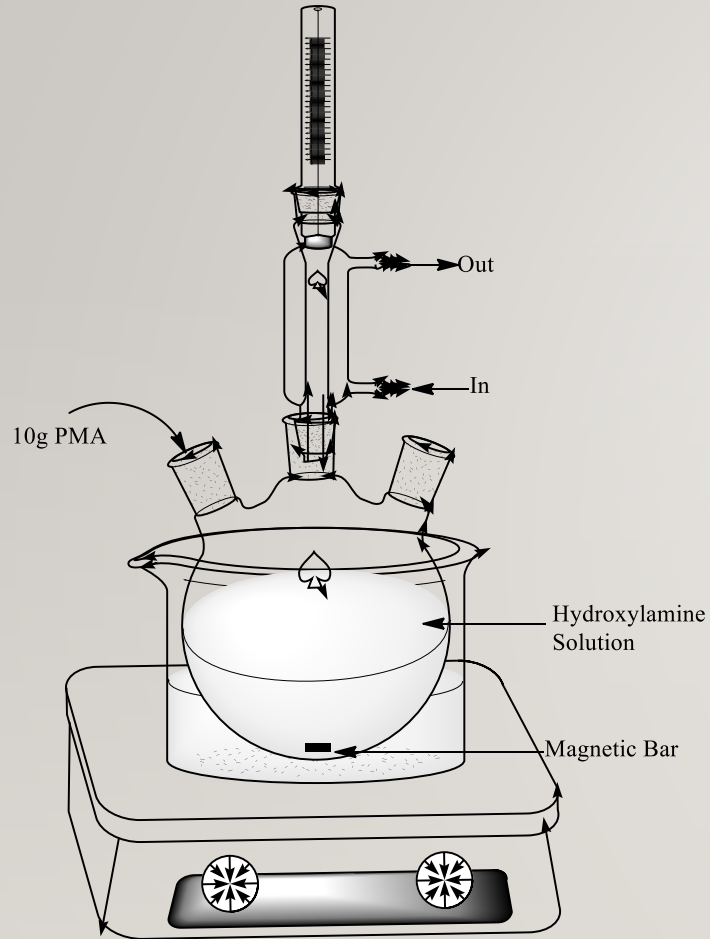
1. Hydrolyze cellulose react with ceric initiator in inert condition for 15 min
2. Purified monomer (methyl acrylate) is added.
3. Heat for 4 hours at 55°C.
4. Wash and dry

(Sources: Mandal *et al.*, 2016)

# Synthesis of Poly(amidoxime) Ligand

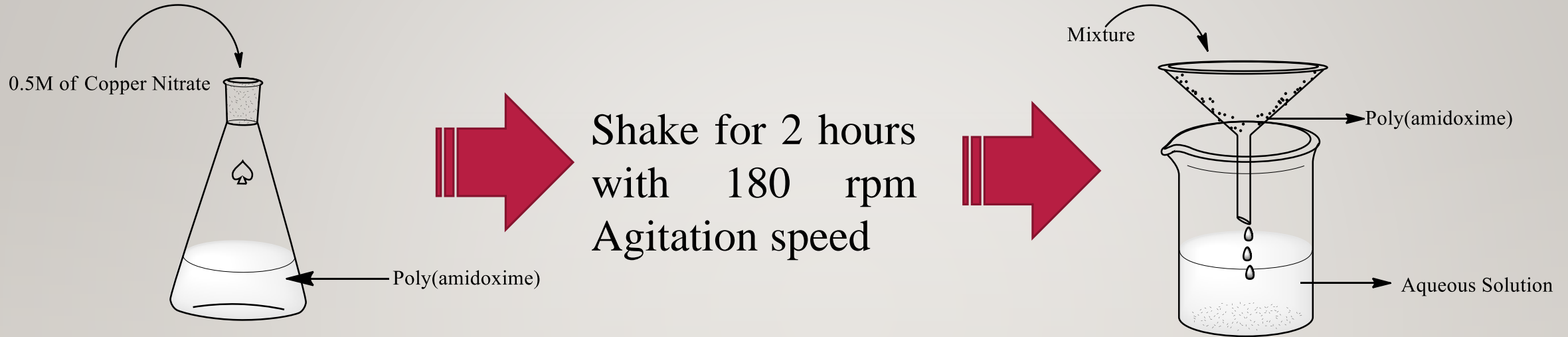
## Procedure

1. Hydroxylamine hydrochloride is dissolved into 4:1 methanolic solution.
2. PMA is added into the hydroxylamine solution and heat for 6 hours at 70°C.
3. PHA ligand is washed by methanolic solution.
4. In order to cover chelating polymeric ligand into H-form ligand, the ligand was treating with 100 mL of 0.1 M of hydrochloric acid (HCl) in methanolic for five minutes.
5. Wash and dry.



(Sources: Shaheen *et al.*, 2016)

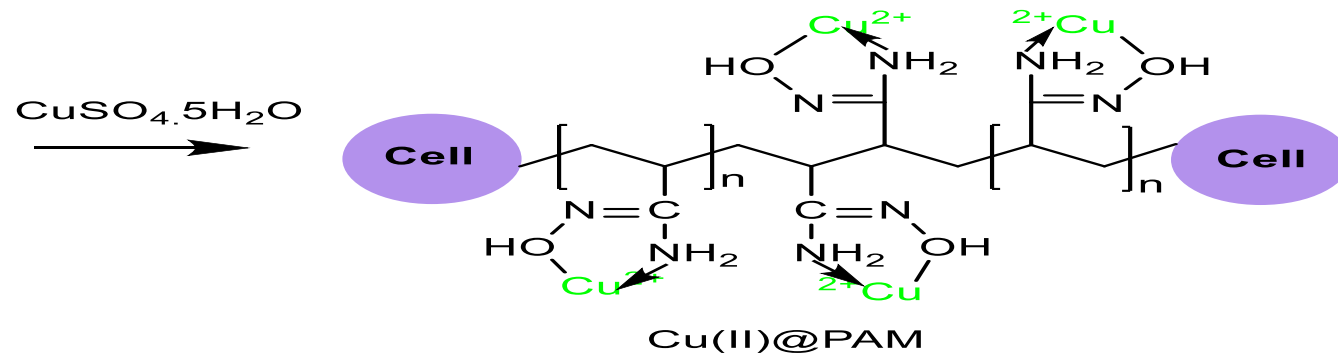
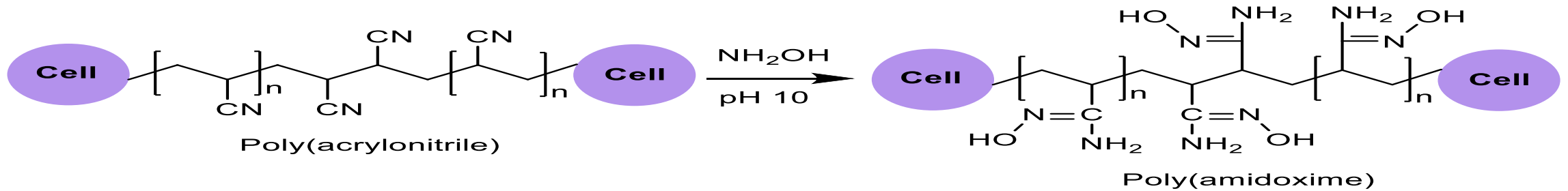
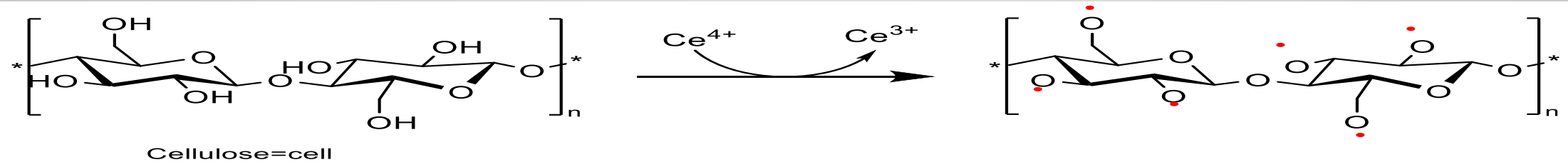
# Preparation Of Metal Catalyst ( $\text{Cu}^{2+}$ )



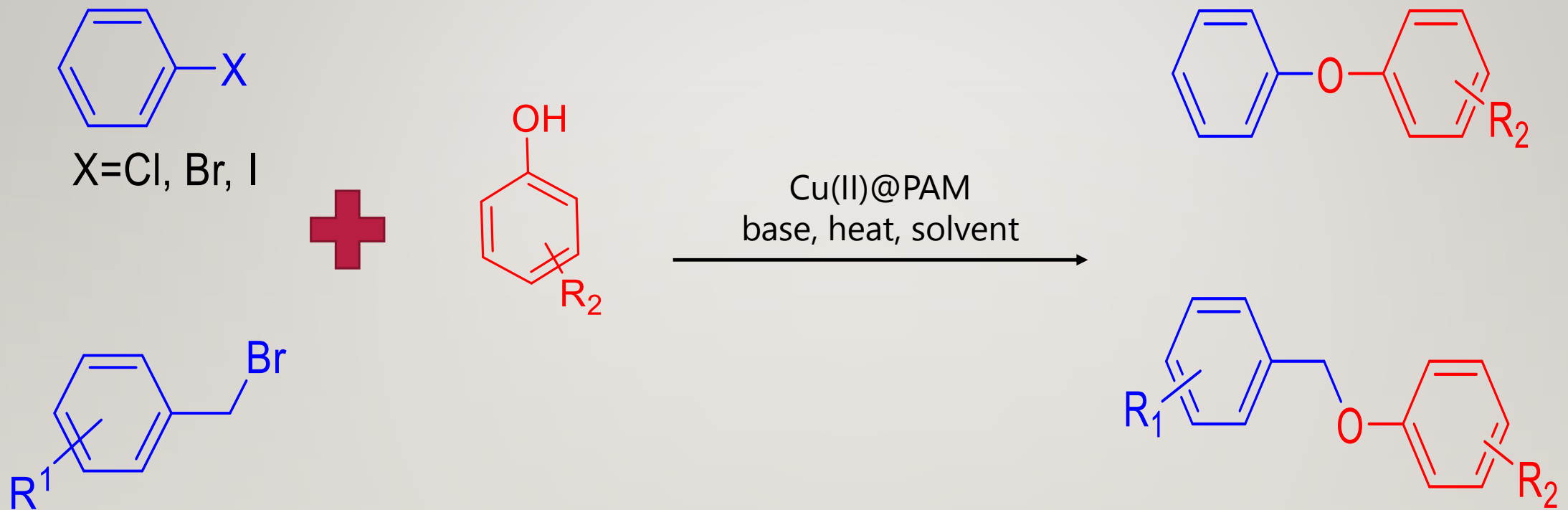
Wash the ligand and dry it.  
The ICP-OES analysis should be used to estimate the copper adsorbed

(Sources: Rahman *et al.*, 2016)

# Reaction Mechanism



# Ullmann Reactions



# RESULT

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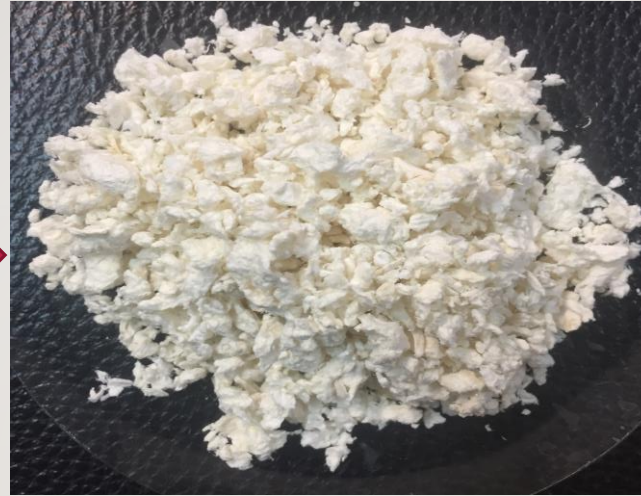




# Product



Pandanus fruit and its fibre



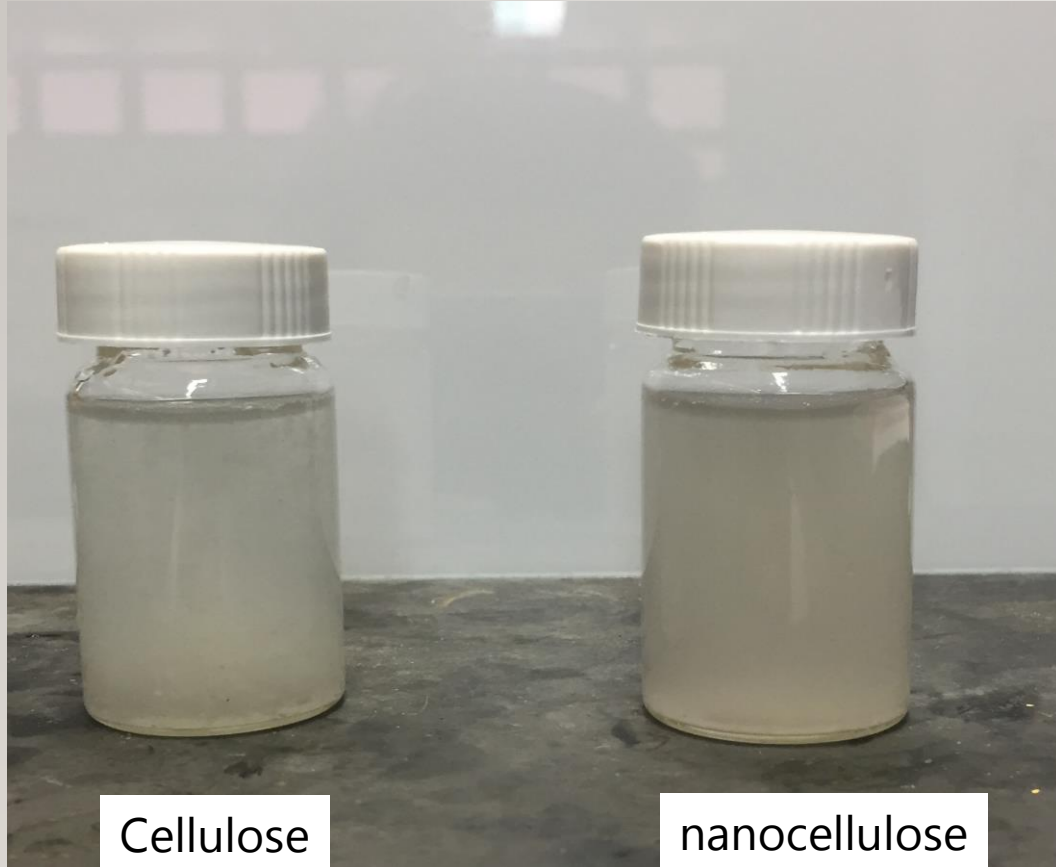
Cellulose from Pandanus  
fruit fibre

Extraction:  $49.5 \pm 1.0\text{g}$  from  
100g dried fruit fibre  
Yield: ( $\approx 50\%$ )

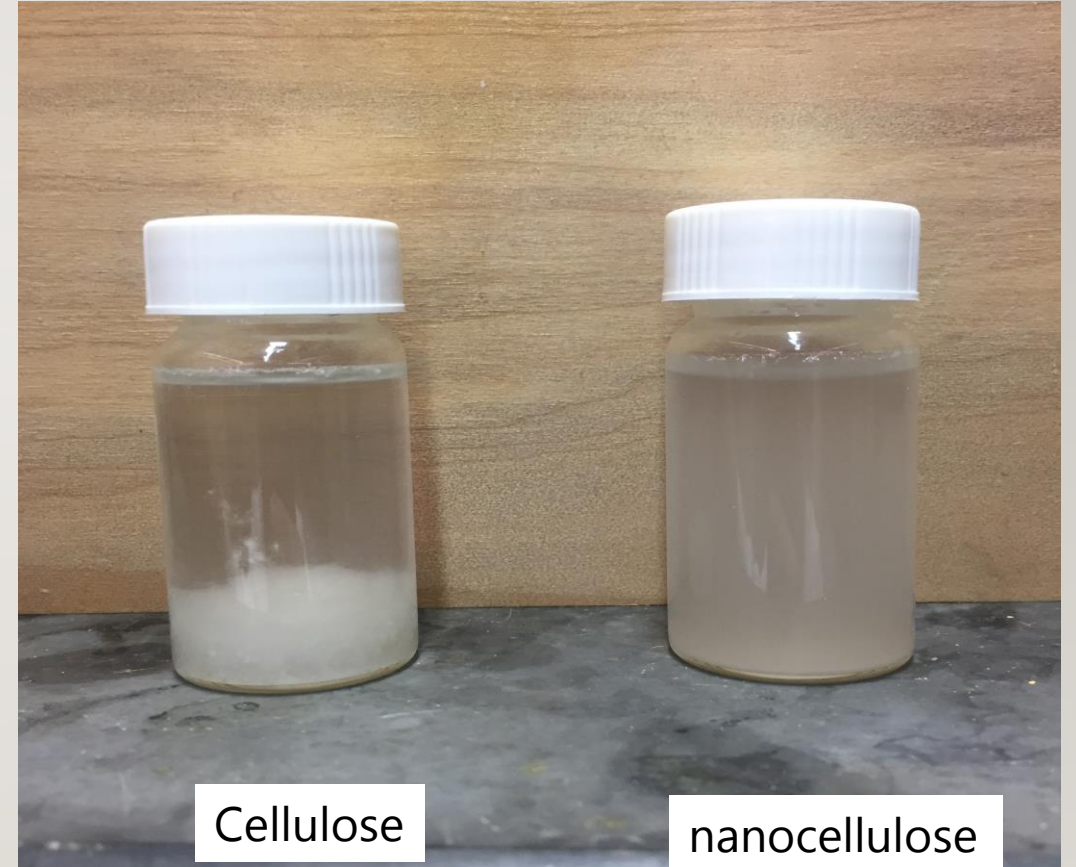


(nanocellulose)

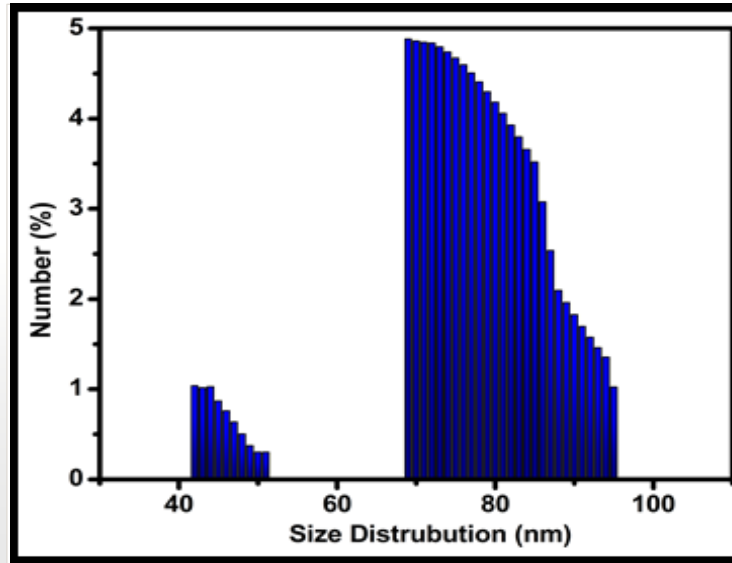
# Physical stability of cellulose and nanocellulose



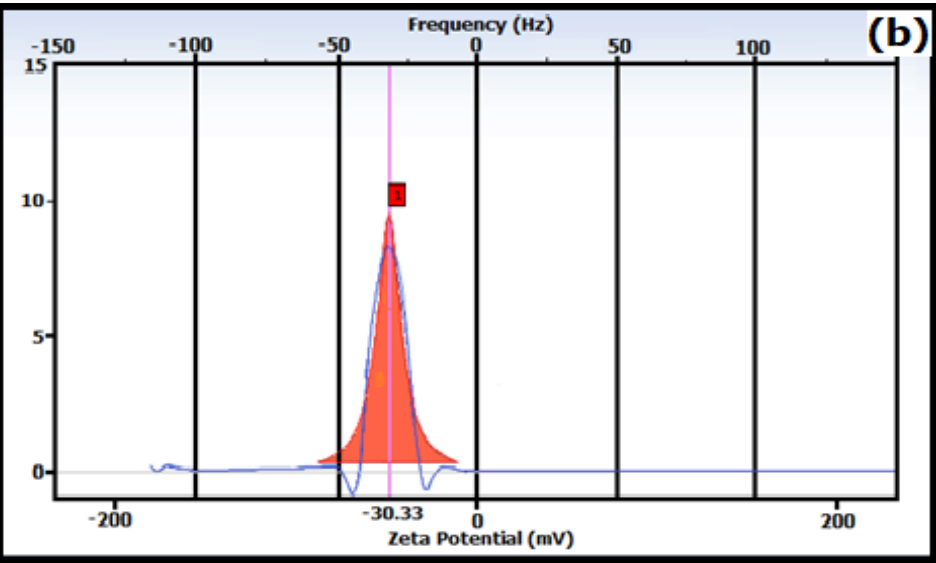
**15 min String**



**Leave For 3 hour**



(a)



(b)

# Particle size and zeta potential analysis

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# Product



Poly(acrylonitrile)

Yield: 18.15 g from 10g dried nanocellulose



Poly(amidoxime) Ligand

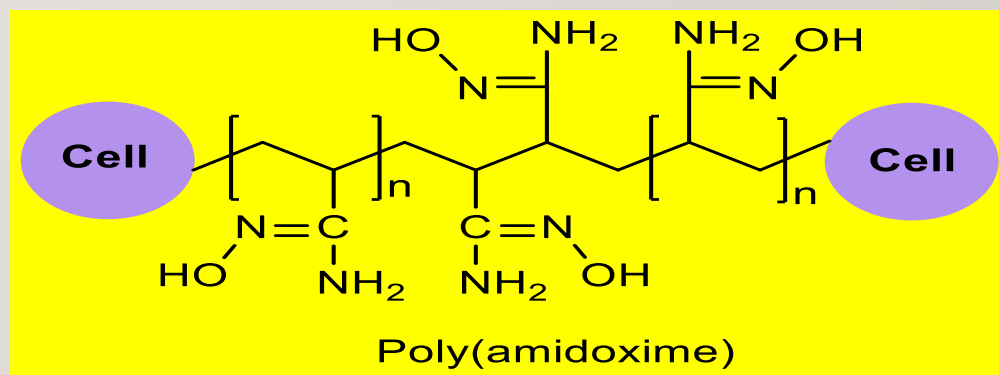
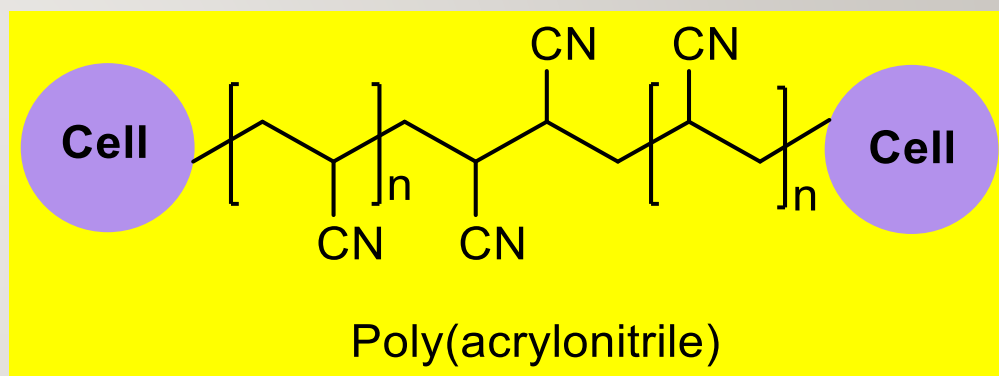
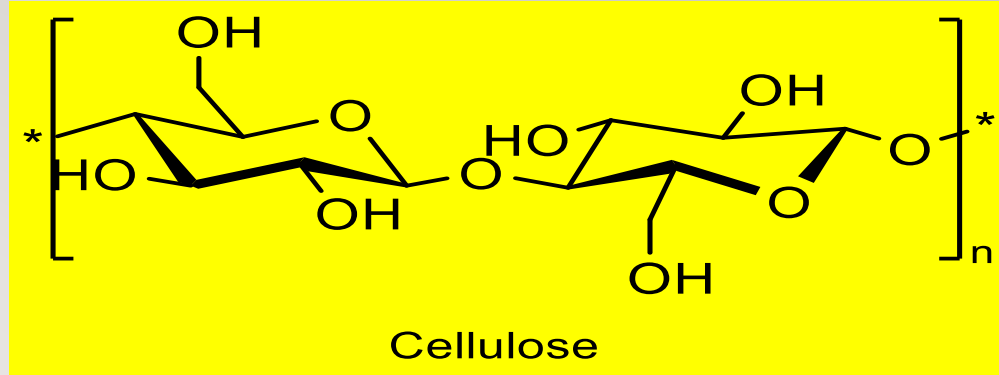
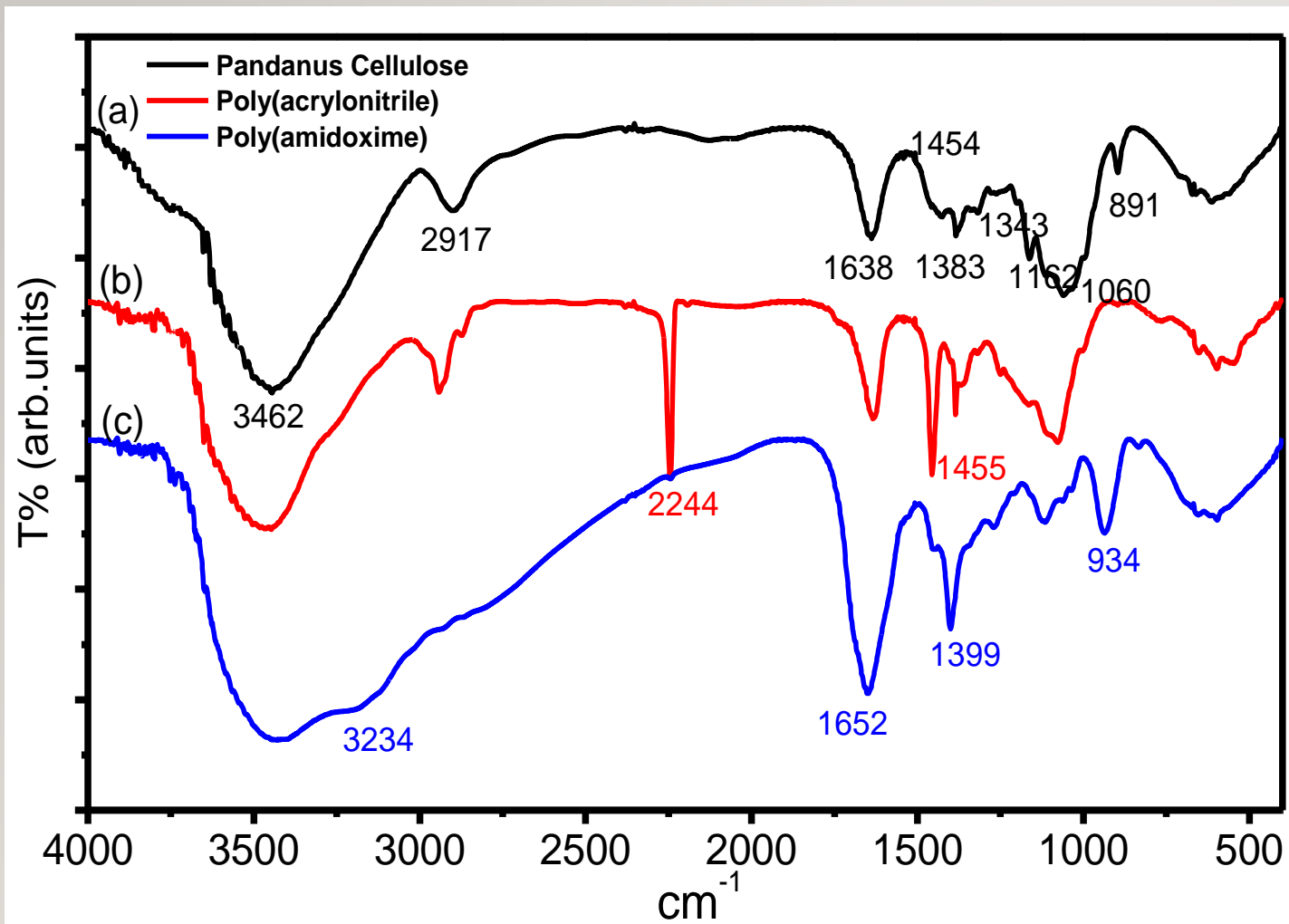
Yield: 28.50g from 10g dried poly(acrylonitrile)



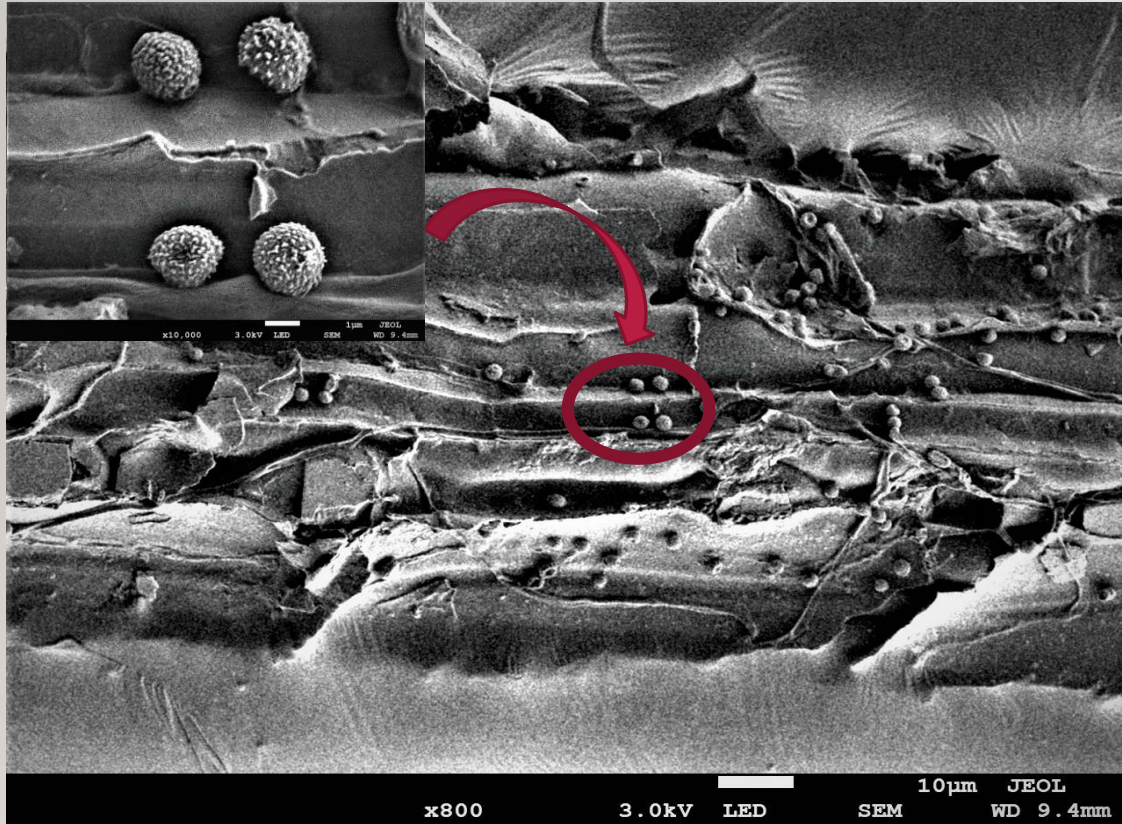
Cu(II)@PAM

ICP-OES= 0.50mmol/g of copper

# FT-IR Analysis



# FE-SEM Analysis



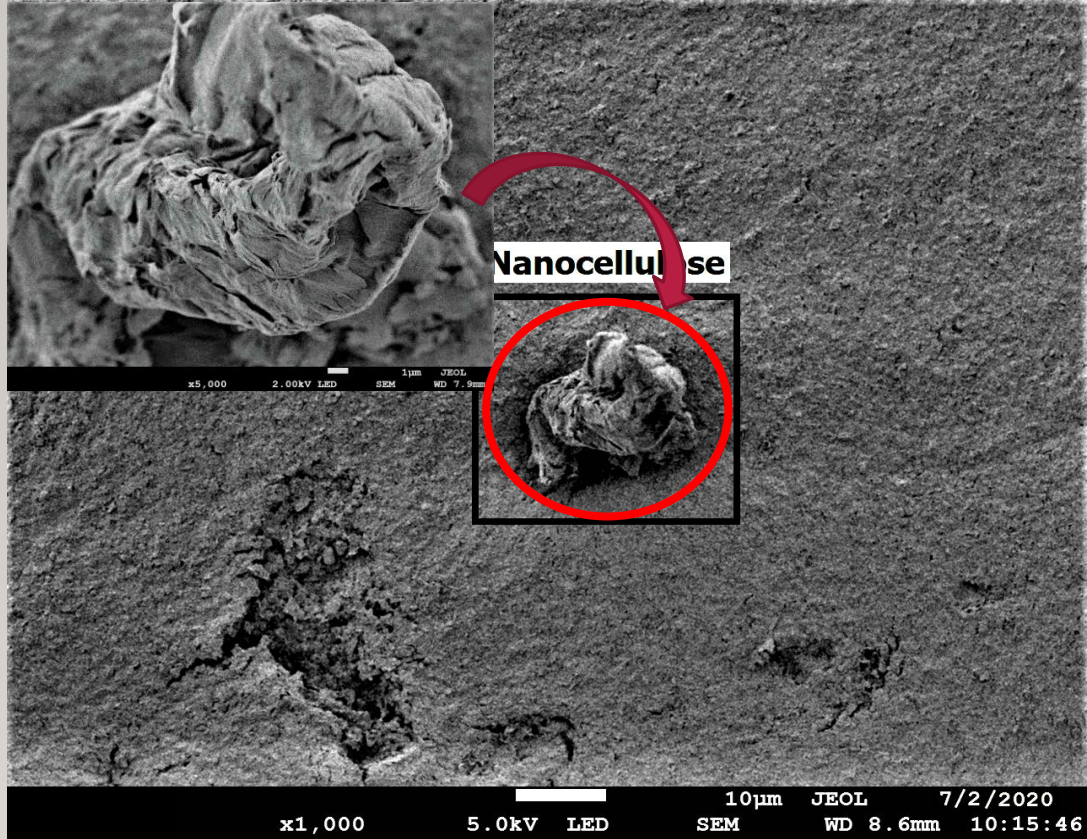
Pandanus fruit fiber



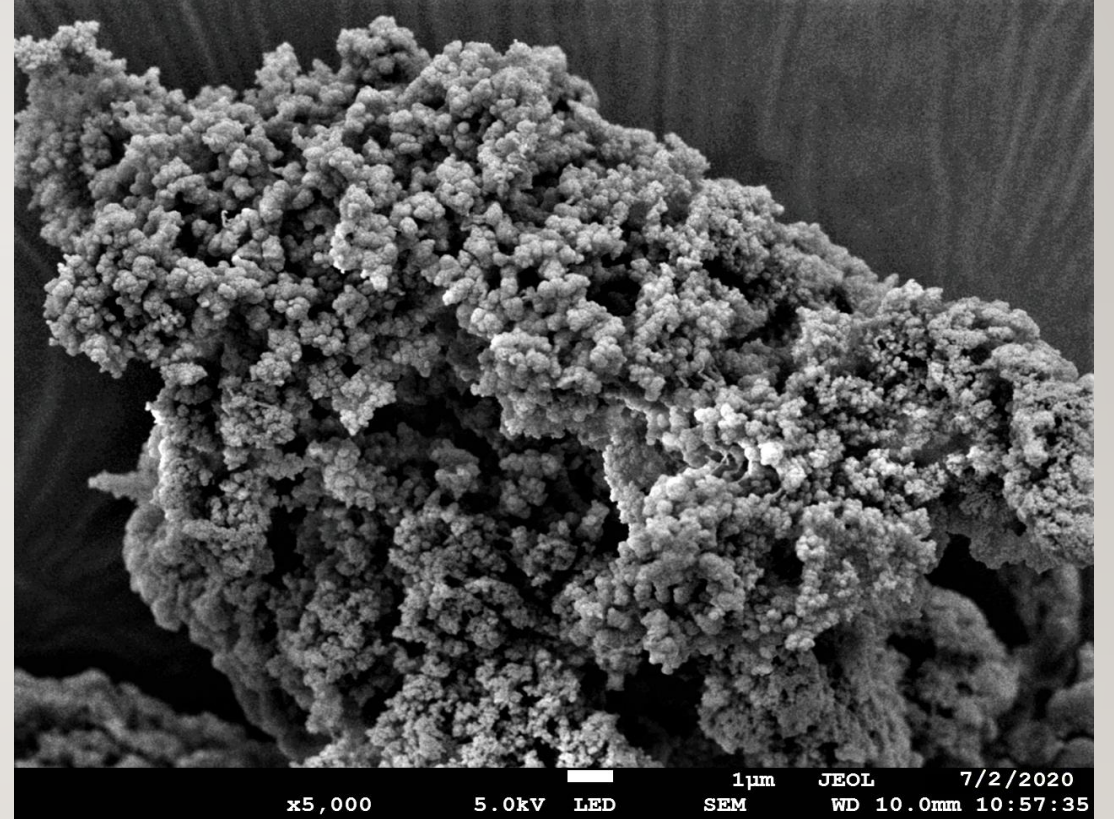
Pandanus-cellulose

Magnification: 1000X

# FE-SEM ANALYSIS

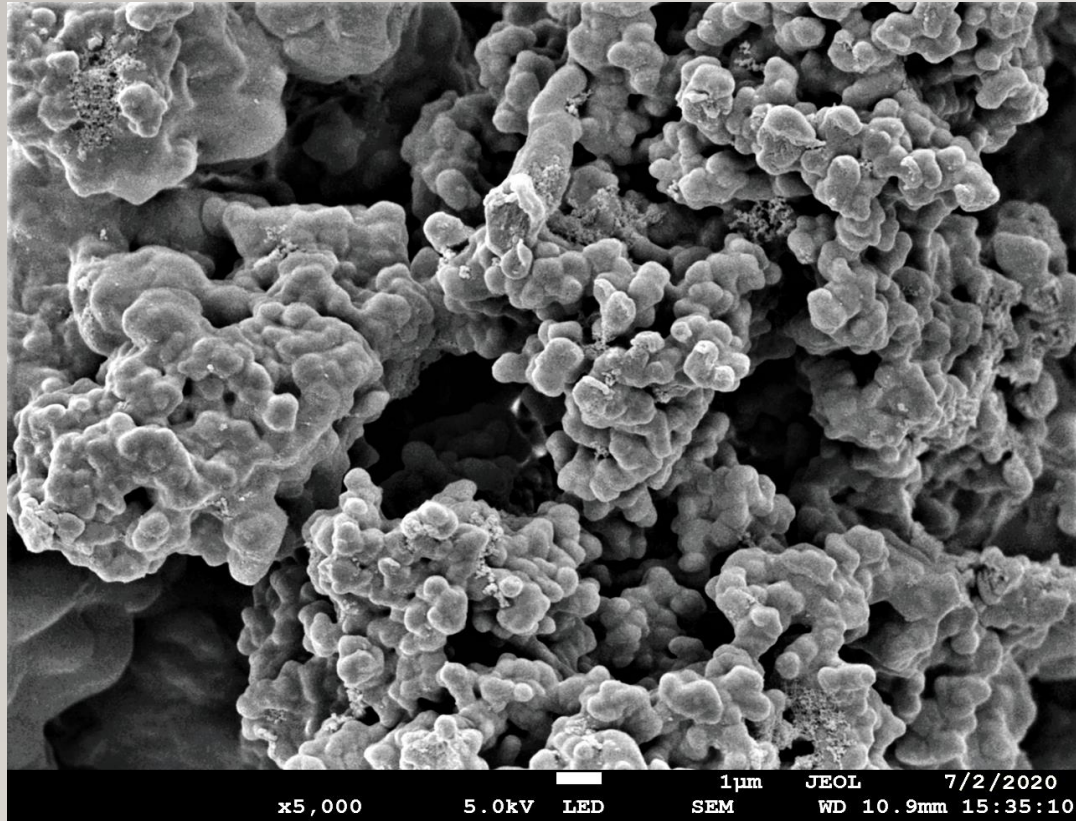


nanocellulose  
Magnification: 1000X

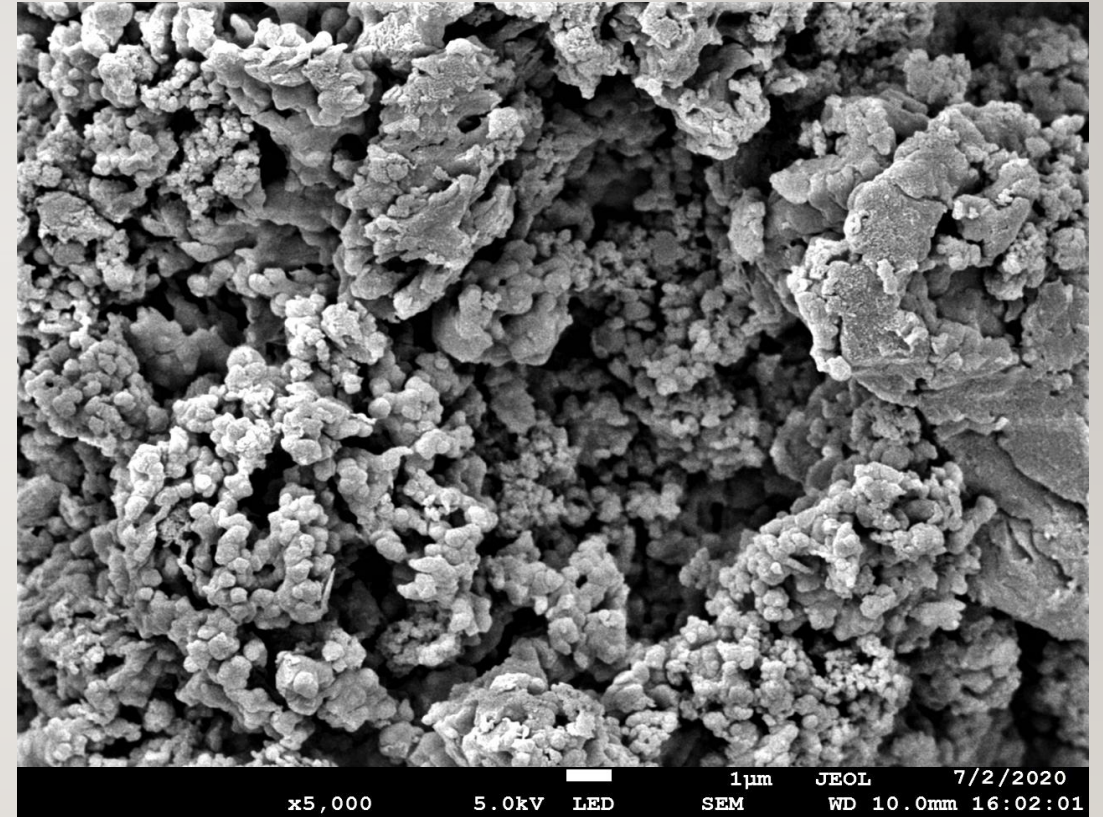


Poly(acrylonitrile)  
Magnification: 5000X

# FE-SEM Analysis



Poly(amidoxime)

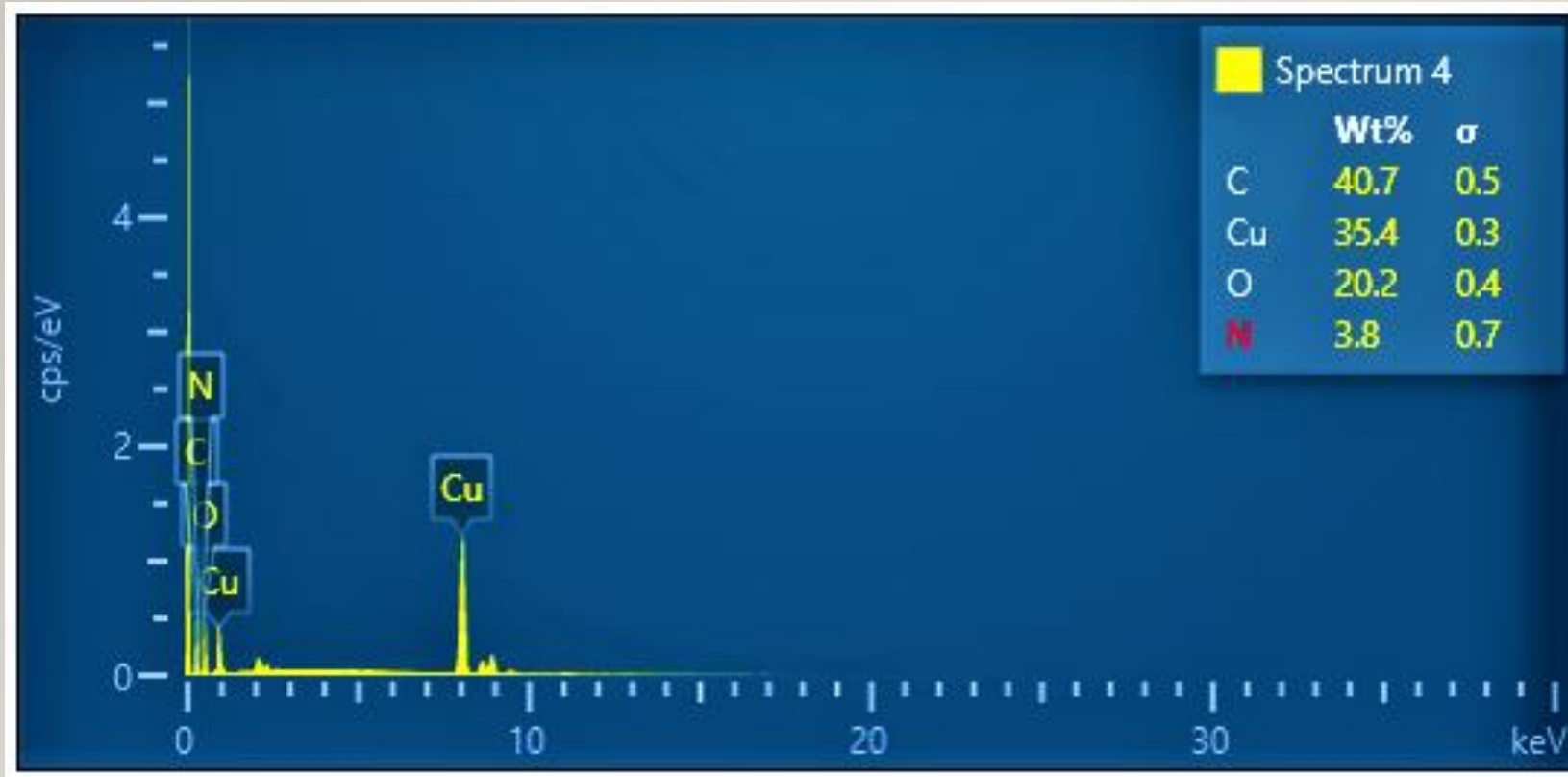


Cu(II)@PAM

Magnification: 5000X

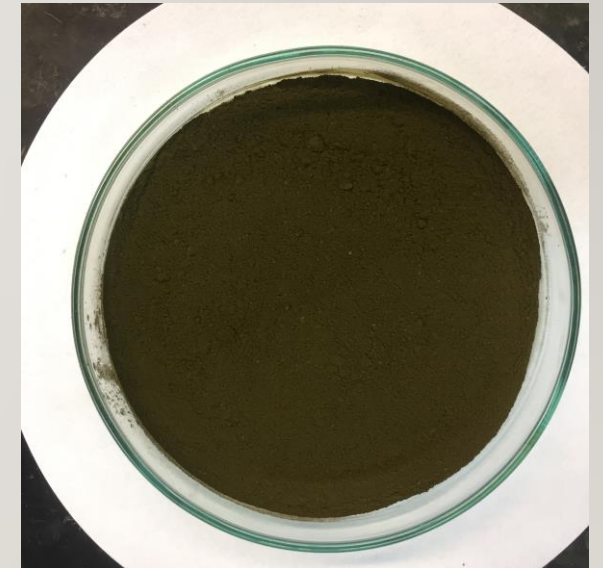


# Energy Dispersive X-ray Diffraction (EDX Analysis)

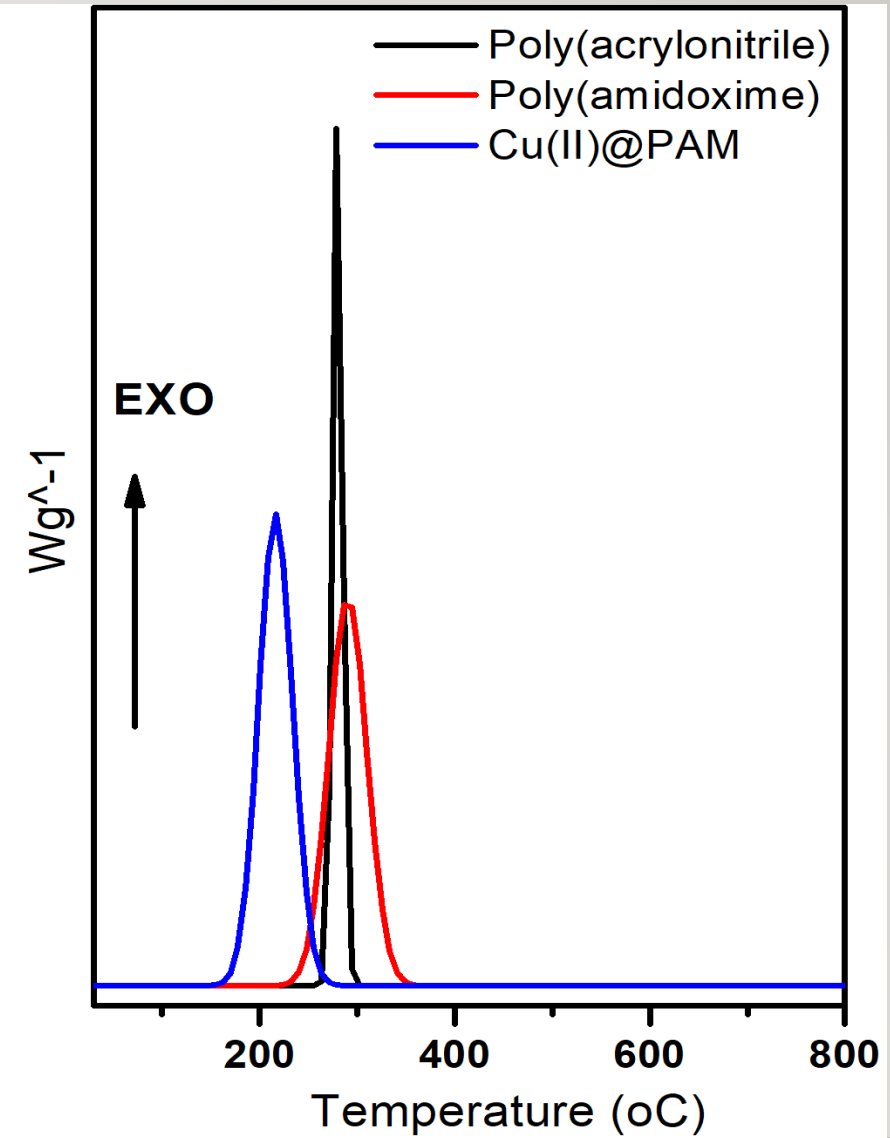
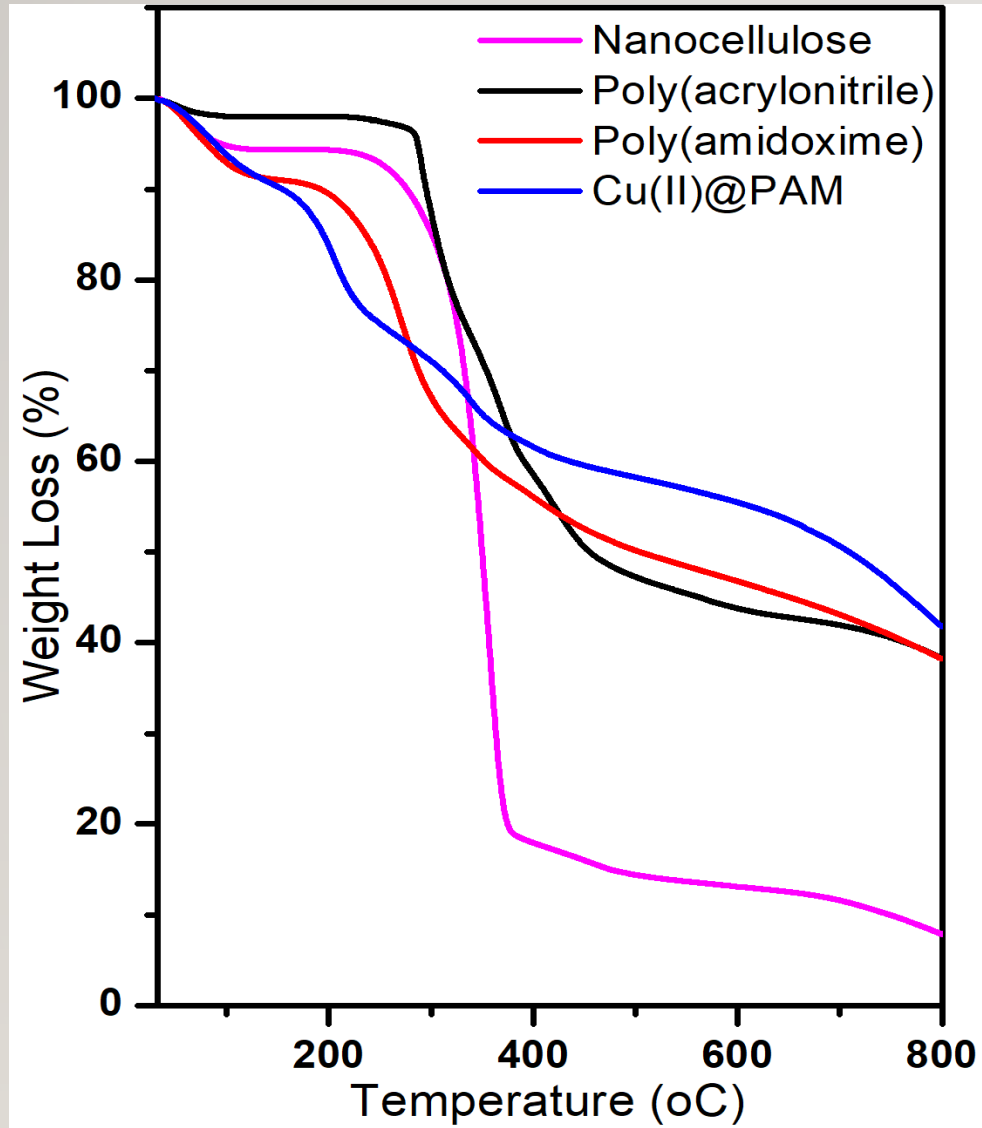


Cu(II)@PAM

Copper = 35.4%  
Carbon = 40.7%  
Oxygen = 20.2%  
Nitrogen = 3.8%

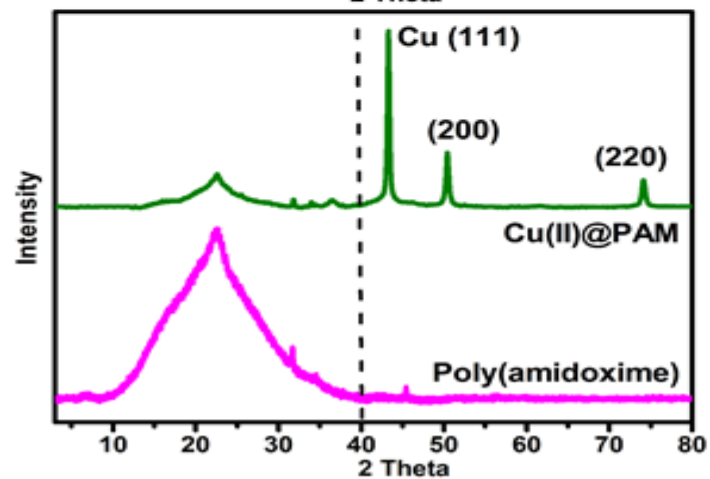
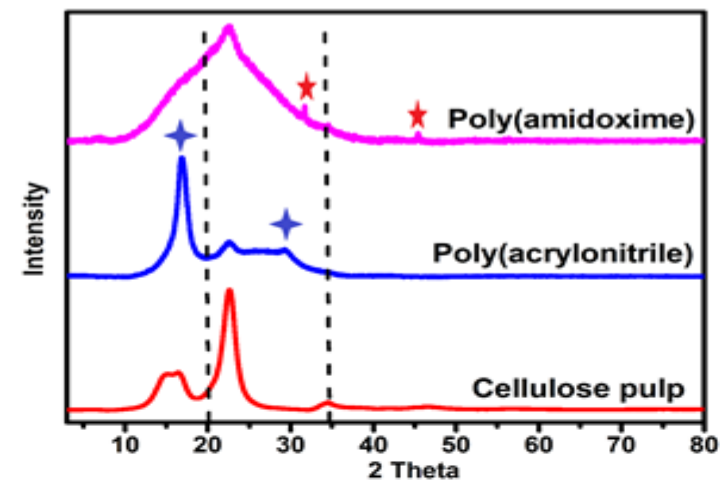
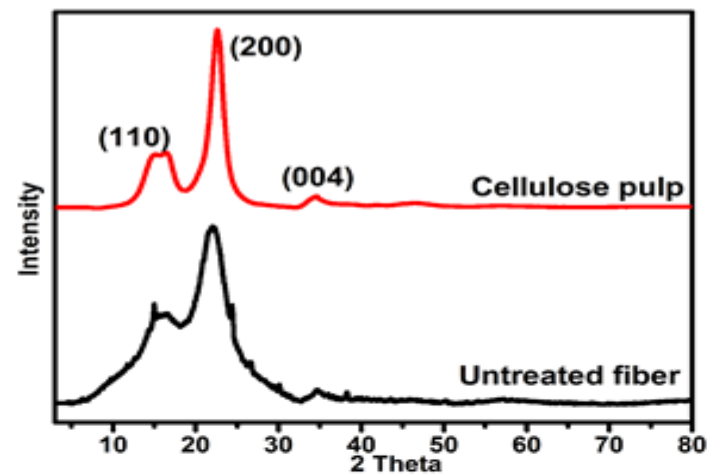


# TGA & DSC

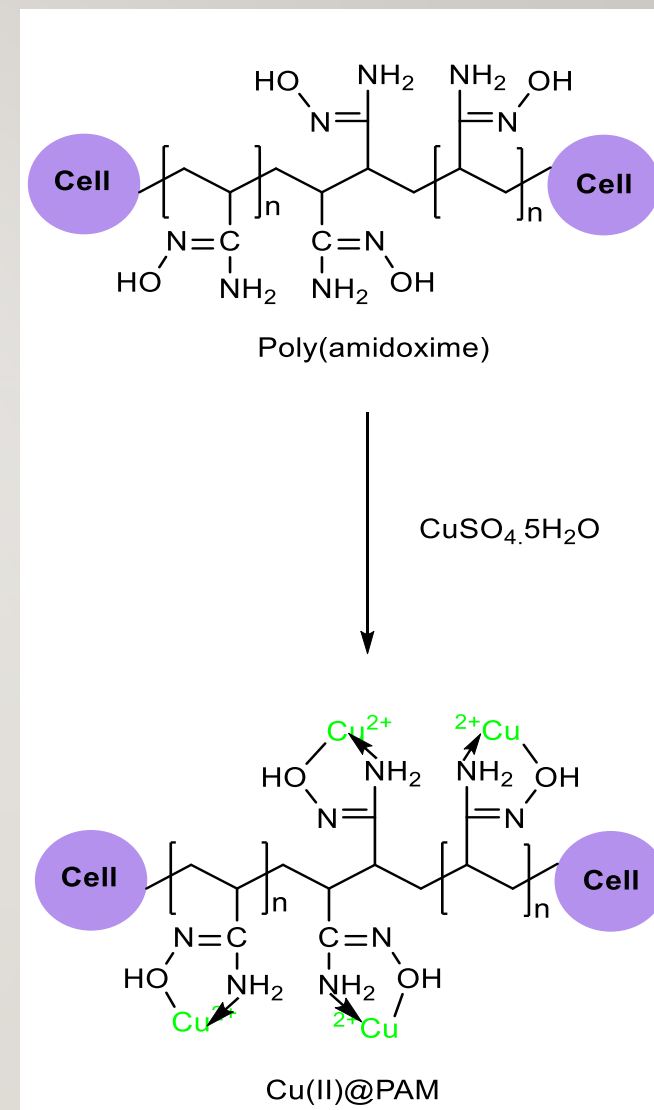
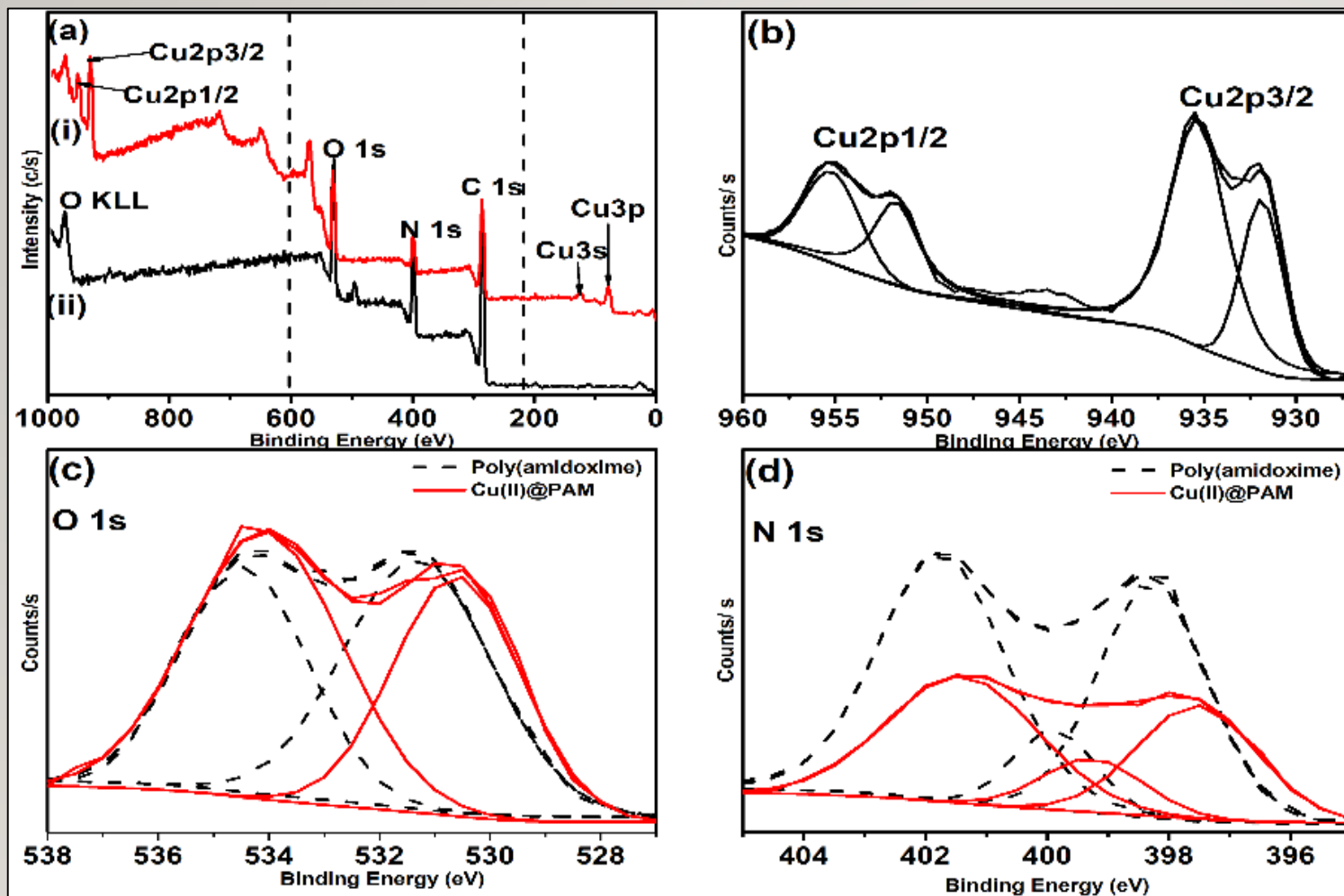


# XRD RESULT

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# XPS ANALYSIS

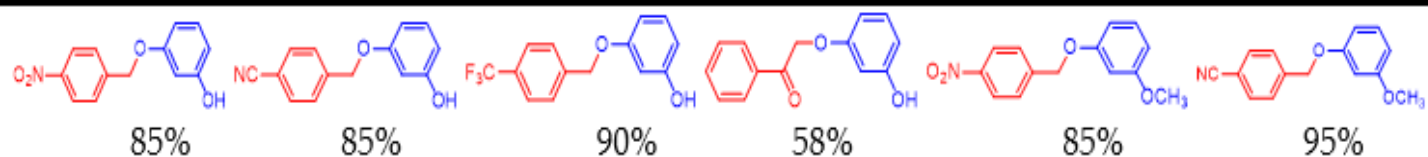
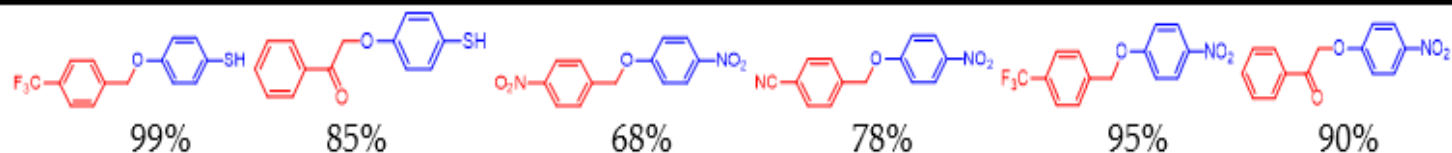
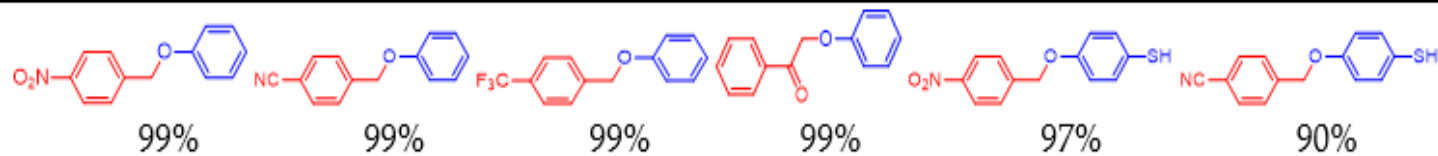
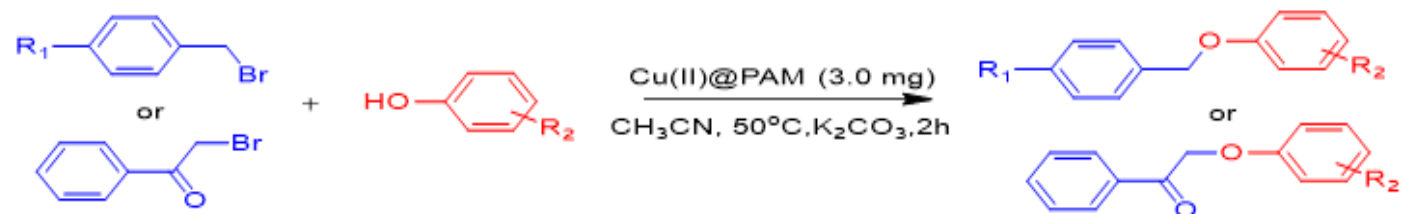


# Optimization Of Ullmann Reaction

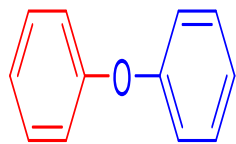
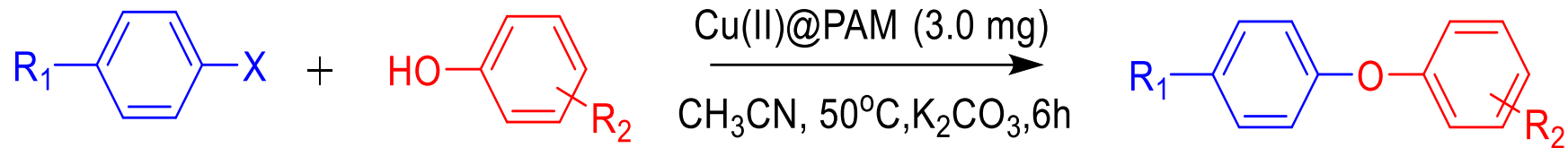
Entry	Cu(II)@PAM (mg)	Temperature (°C)	Time (h)	Yield (%)
1	15	80	8	98
2	5	50	8	99
3	3	50	2	99
4	1.5	50	2	89

- Conditions: 4-nitrobenzyl bromide (1 mmol), phenol (1.2 mmol), a catalytic amount of complex copper and 3 mol equiv. of  $K_2CO_3$  in 5 mL of acetonitrile.

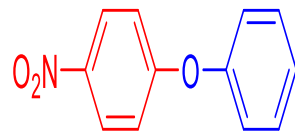
# REACTION



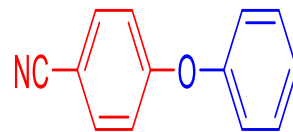
- Conditions: Benzyl halide (1 mmol), phenol (1.2 mmol),  $\pm 3.0$  mg of Cu(II)@PAM and 3 mol equiv. of  $K_2CO_3$  in 5 mL of acetonitrile. All the compound was determined by GC-MS and NMR.



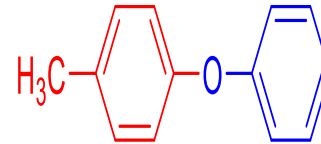
I= 60%; Br= 55%;  
Cl= 20%



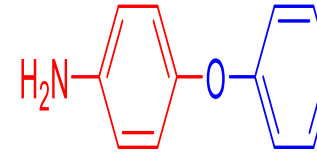
I= 77%; Br= 69%;  
Cl= 51%



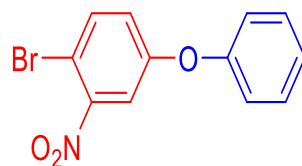
I= 90%; Br=82%;  
Cl= 60%



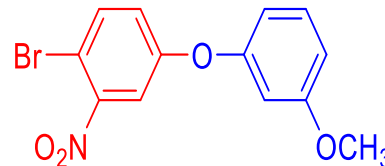
I= 65%; Br= 50%;  
Cl= 20%



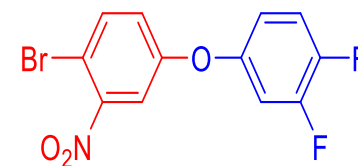
I= 50%; Br= 45%;  
Cl= 15



Br= 67%

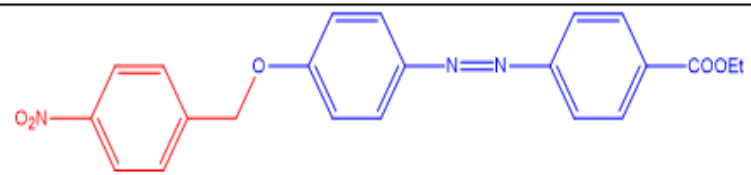
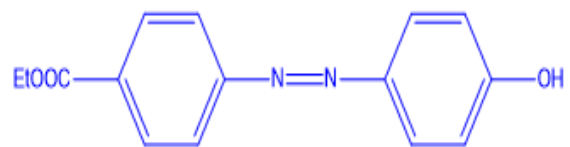


Br= 75%

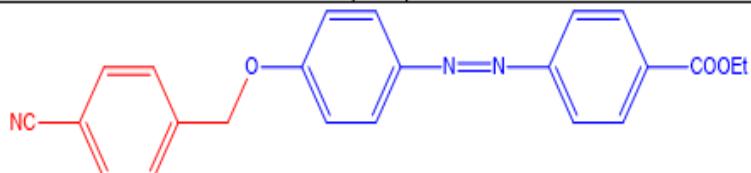


Br= 65%

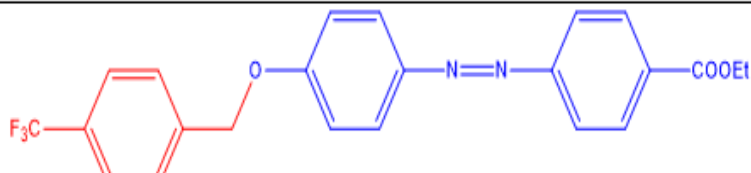
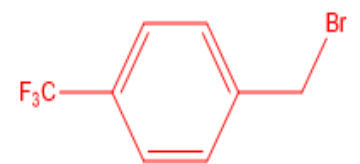
- Conditions: Benzyl halide (1 mmol), phenol (1.2 mmol),  $\pm$ 3.0 mg of Cu(II)@PAM and 3 mol equiv. of  $K_2CO_3$  in 5 mL of acetonitrile. All the compound was determined by using GC-MS and NMR.



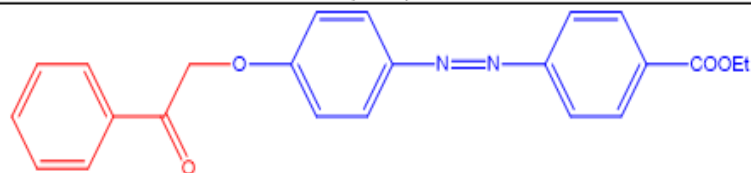
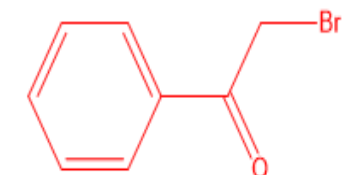
(75%)



(80%)



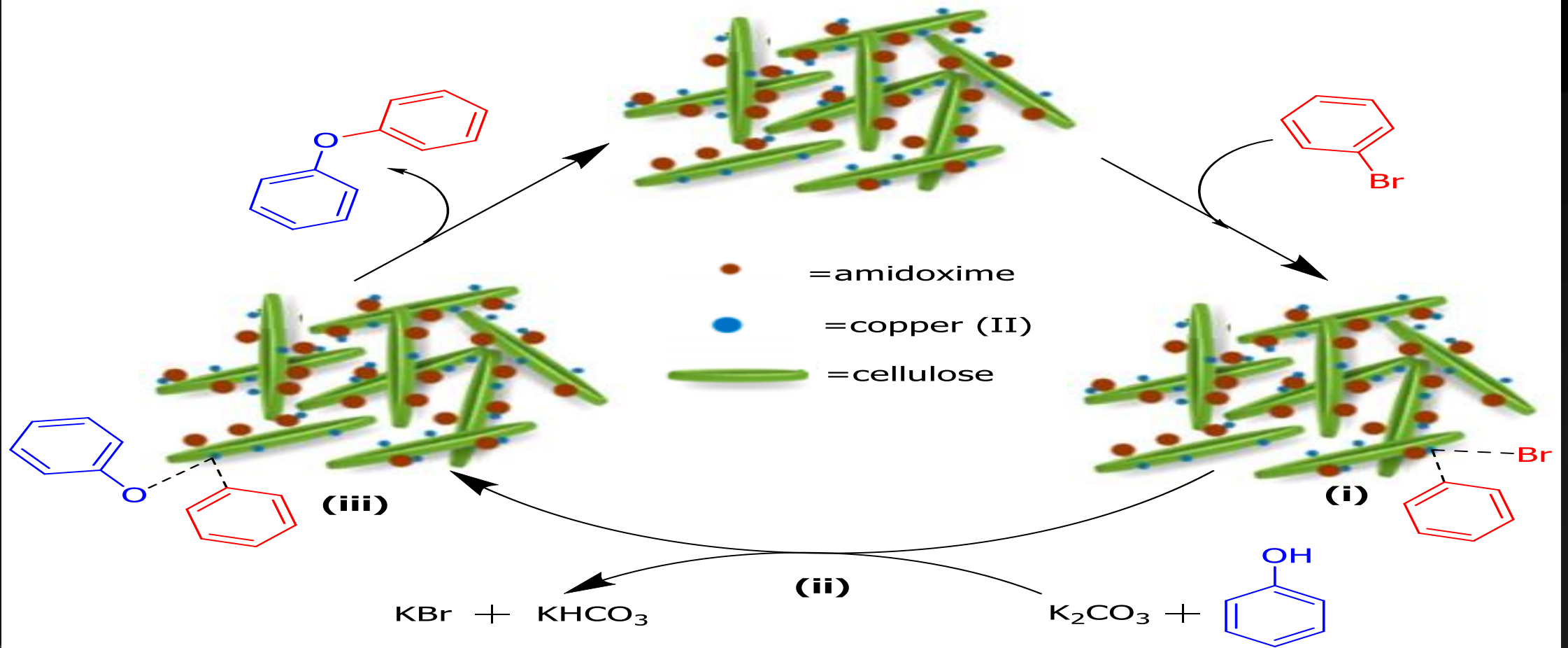
(88%)



(85%)

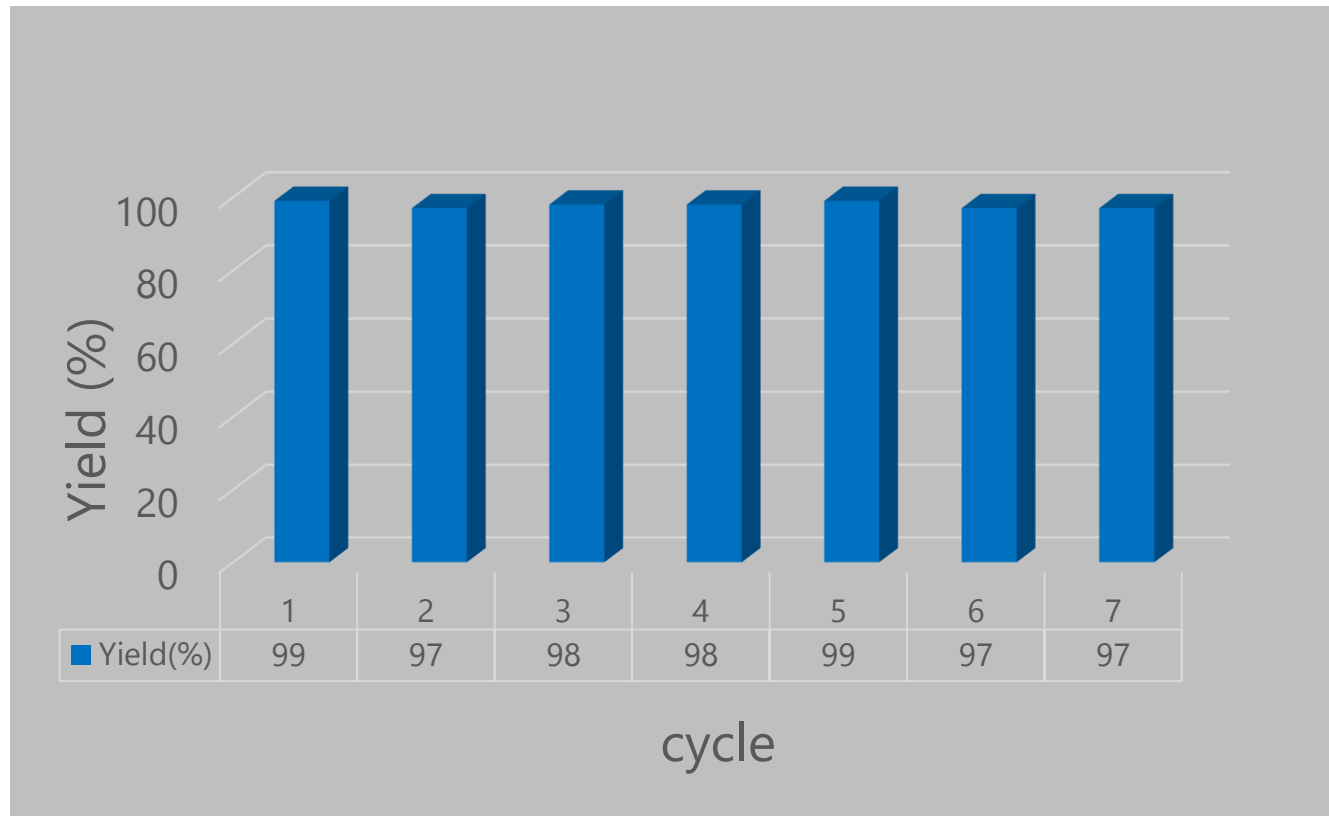


# Mechanism for Ullmann Reaction



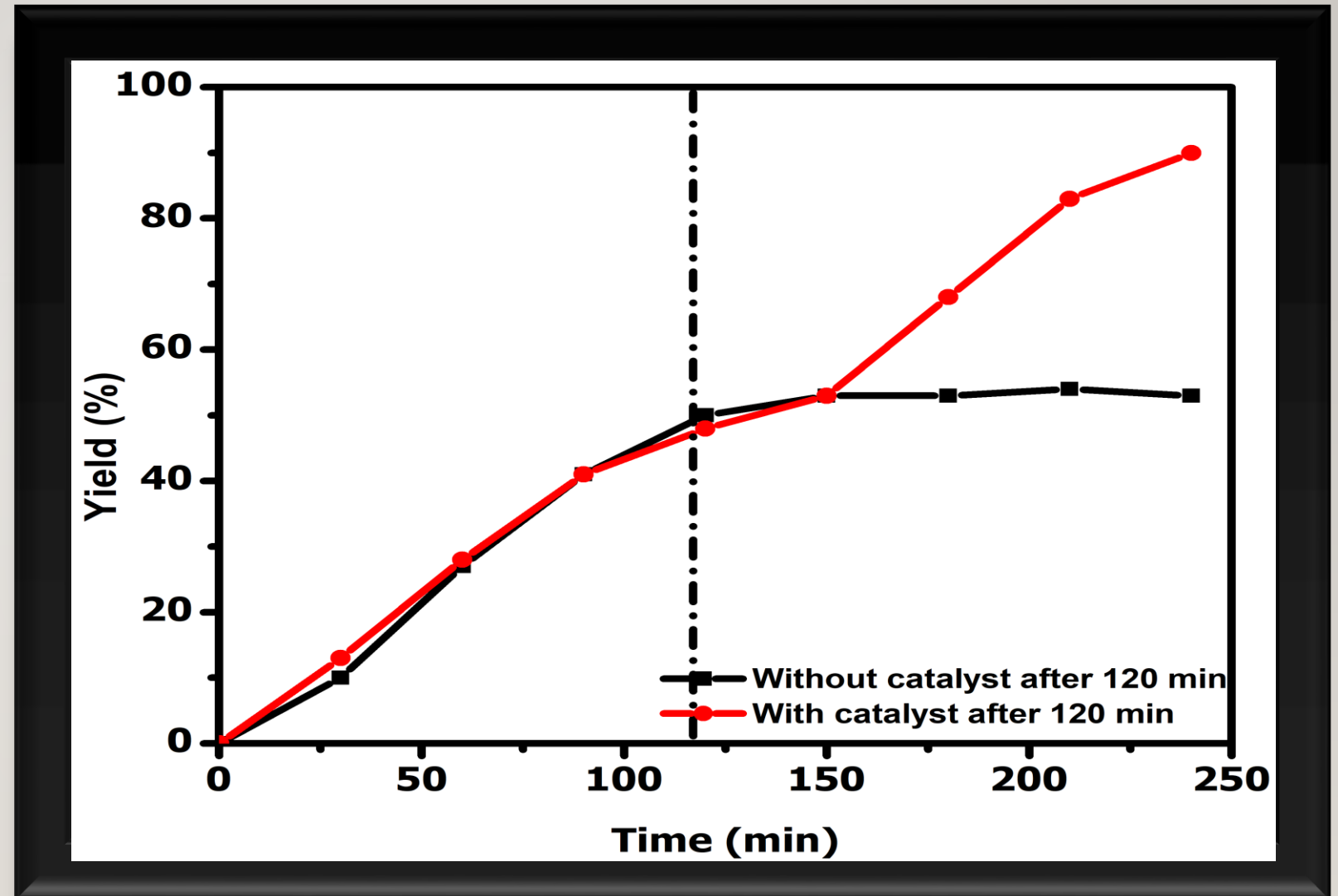
# REUSABILITY

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# LEACHING STUDY

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# CONCLUSION

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Successful synthesis  
high active, stable and  
safe copper catalyst for  
Ullmann etherification  
reaction.

The synthesized copper  
catalyst can afford the  
Ullmann etherification  
in good to high yield of  
product.

# FUTURE WORK

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- Test the synthesized catalyst in other cross-coupling reaction (C-C, C-N, C-S, etc.)
- Utilize the Cu(II)NPs@PAM in total synthesis of natural product, medicine compound.

# REFERENCES

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Thank

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