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In vitro antioxidant activity, phytochemical screening and total phenolic and flavonoid contents from the leaves extracts of Stachys germanica subs cordigera briq

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In vitro antioxidant activity, phytochemical screening and total phenolic and flavonoid contents from the leaves extracts of

Stachys germanica subs cordigera briq



Abstract

This study is planned to perform phytochemical screening, evaluate antioxidant activity and assess total phenolics and flavonoids content of methanolic and ethyl acetate extracts from the leaves of Stachys germanica subs cordigera briq. The dried powdered leaves of Stachys germanica subs cordigera briq (80 g) were extracted exhaustively by Soxhlet apparatus with increasing polarity of solvents (hexane, ethyl acetate and methanol). The total phenolic and flavonoid contents in the methanolic and ethyl acetate extracts were determined by using the Folin-Ciocalteu reagent and aluminum chloride method, respectively. The antioxidant activities were examined by three different methods, namely 2,2-diphenyl-1-picrylhydrazyl (DPPH) free radical scavenging activity, reducing power scavenging activity (FRAP) and total antioxidant capacity. Phytochemical analysis of all extracts showed the presence of major classes of phytochemicals such as, flavonoids, tannins and polyphenols. Total phenolic and flavonoid contents results are showed in a large dominance in ethyl acetate extract. In vitro antioxidant activities of both extracts (ethyl acetate and methanol) were significant and ethyl acetate extract showed a higher potency than reference antioxidant Butylated Hydroxy Toluene (BHT) in total antioxidant capacity essay. It can be concluded that the crude extracts from the leaves of *Stachys germanica subs cordigera brig*. are a potential source of natural antioxidants which can be used in preventing the progression of many diseases.

<u>Keywords</u>: *Stachys germanica subsp cordigera briq*, phytochemical screening, total phenolics and flavonoids content, antioxidant activity

OUTLINE Introduction **Objectives Materials and Methods** Plan **Results** Conclusion





humans, along with many other creatures, need oxygen in the air we breathe to stay alive



highly reactive atom that is capable of becoming part of potentially damaging molecule commonly called "free radical"



The excessive production of these free radicals creates what is known as oxidative stress.





Oxidative stress





Free radicals

What are Free Radicals?





Free radicals

Free radicals are capable of attaching cells of body, causing them to lose their structure and function.

Although the initial attack cause the free radical to become neutralized, another free radical is formed in this process, causing a chain reaction is occur.



And until subsequent free radicals are deactivated, thousands of free radical reaction can occur within second of initial reaction.

Free radicals have been implicated in the pathogenesis of at least 50 diseases cardiovascular diseases, cancers, neurodegenerative diseases, Alzheimer's disease inflammatory diseases and other related diseases







Antioxidants

Antioxidants are chemical substances that donate an electron to the free radical and convert it to harmless molecule.





Selection cretiria of antioxidants

Following criteria should be considered while selecting an antioxidant

- It should be able to produce desire redox reaction
- It should be physiologically and chemically compatible
- It should be physiologically inert
- It should be non-toxic both in the reduced and oxidized forms
- It should be effective in low concentration
 - It should be provide prolonged stability to the formulation





Natural compounds derived from medicinal plants play a central part in the health care and drug development in classical as well as advanced systems of medicine.



Introduction Objectives Materials Methods Results Conclusion



Stachys germanica subs cordegira briq







Many *Stachys* species are used in the preparation of food such as yoghurt or jelly to improve the taste and as flavors and seasoning



Aims and Objectives

The present investigation is undertaken by utilizing the plant *Stachys* germanica subs cordegira briq with following objectives:

Extraction of the leaves with different solvents



Preliminary phytochemical analysis



Determination of total phenolic and flavonoid contents at the different extracts



Evaluation the antioxidant activities



Material and methods

1-COLLECTION OF MATERIAL AND EXTRACTION PROCEDURE





2- PHYTOCHEMICAL ANALYSIS



Sterols and terpenes: Burchard's test



Polyphenols: Fecl₃ solution



Flavonoids: cyanidine reaction



Tannins: Stiasny's test



Alkaloids: Dragendorff's test



3-TOTAL FLAVONOIDS CONTENT

Total flavonoid content was determined spectrophotometrically by using the AlCl₃ reagent as described by Dewanto et al 2002

250µl of crude extract + 75µl of 5% NaNO₂





TOTAL PHENOLIC CONTENT

Total phenolic content was determined spectrophotometrically by using Folin-Ciocalteu method as described by El Hajaji et al 2012

100µl Folin-ciocalteu reagent + 1.58mL distilled water + 20µl crude extract





ANTIOXIDANT ACTIVITIES

Antioxidant activities were examined by three different methods namely,

***** 2,2-diphenyl-1-picrylhydrazyl (DPPH) free radical scavenging activity

Reducing power scavenging activity (FRAP)

♦ Total antioxidant capacity.



* 2,2-diphenyl-1-picrylhydrazyl (DPPH) free radical scavenging activity

The leaves extracts of *S.germanica subs cordigera briq* were tested in their free radical scavenging activity using (DPPH), according to the protocol described by El Hajaji et al 2012.

100µl of extract + 10mL of methanolic solution of DPPH



Inhibition ratio : % inhibition = $[(A_0 - A_1)/A_0] \times 100$

 $\mathbb{A}_{\mathbb{Q}_{2}}$ absorbance of control reaction ; \mathbb{A}_{1} absorbance of test compounds

BHT : positive control. The test was carried out in triplicate



Reducing power scavenging activity (FRAP)

The reducing power was determined according to the protocol described by El Hajaji et al 2012.

Extract at various conc + 0.2M phosphate buffer $(pH 6.6) + (K_3Fe(CN)_6) (1\%)$



Ascorbic acid : standard



***** Total antioxidant capacity.

The total antioxidant capacity of the extracts was evaluated by using the phosphomolybdenum method as montioned by El Hajaji et al 2012

- The test is based on the reduction of Mo (VI) to Mo(V) in the extract and subsequent formation of a green phosphate/Mo(V) complex at acid pH.
- The absorbance of the solution was measured at 695 nm using a U.V.visible spectrophotometer
- The antioxidant capacity of each sample was served as Ascorbic Acid equivalent using the following linear equation in using ascorbic acid as standard: [A= 0.0037C + 0.0343; R²=0.991]

A: absorbance at 695 nm ; **C:** concentration as ascorbic acid equivalent (µg/ml). The values represent the triplicate analysis.





Figure 1. Yields of hexane, ethyl acetate and methanol extracts of Stachys germanica subs cordigera briq.



Preliminary phytochemical analysis from the leaves extracts of *Stachys germanica subsp cordigera briq* showed the presence of major classes of secondary metabolites.

Extracts Metabolites	Hexane	Ethyl acetate	Methanol
Polyphynols	-	+	+
Sterols/steroids	+	-	-
Terpenes/Terpenoids	+	-	-
Tannins	-	+	-
Alkaloids	-	-	-
Flavonoids	-	+	+

 Table. 1 : Results of preliminary phytochemical screening of S.germanica subsp cordigera

 briq. leaves extracts.



TOTAL PHENOLIC CONTENTS

Total phenolic content was expressed as GAE using the following linear equation as standard: y= 0.0171x+0.2091, R²=0.975.

Ethyl acetate extract = $23.2 \mu g/mL \text{ GAE}$; Methanol extract = $15.7 \mu g/mL \text{ GAE}$



Fig 2. Total phenolics contents of ethyl acetate and methanol extracts of *S. germanica subs cordigera briq*.



TOTAL FLAVONOIDS CONTENTS

Total flavonoids contents, they are made as quercetin equivalent using also the following linear equation with quercetin as standard: y=1.657x+0.0317; R²=0.994.

Ethyl acetate extract = $82 \mu g/ml QE$; Methanol extract = $42\mu g/ml QE$.



Fig 3. Total Flavonoid contents of ethyl acetate and methanol extracts of *S. germanica subs cordigera briq*.

TFCs



DPPH ANTIOXIDANT ACTIVITY



Fig 4. Antioxidant activity of *Stachys germanica subs cordigera briq*. leaves extracts against DPPH



FERRIC REDUCING ANTIOXIDANT POWER



Fig 5. Reducing power of of *Stachys germanica subs cordigera briq*. leaves extracts



TOTAL ANTIOXIDANT CAPACITY



Fig 6. Antioxidant capacity of *Stachys germanica subs cordigera* briq. leaves extracts



Conclusion

Phytochemical screening showed the presence of various classes of bioactive chemical constituents in all extracts of *Stachys germanica subs cordigera briq* including sterols/steroids, terpenes /terpenoids, polyphenols, tannins and flavonoids.

Total phenolic and flavonoid contents results showed a large dominance in ethyl acetate extract.

The leaves extracts from *Stachys germanica subs cordigera briq* showed a significant antioxidant activity as well as, ethyl acetate extract exhibited greater antioxidant than BHT as measured by total antioxidant capacity.

These results suggested that the leaves extracts from *Stachys germanica subs cordigera briq.* can be used as possible in natural antioxidant source. It is then necessary to identify and isolate the compounds that are responsible to these antioxidant activities.



