

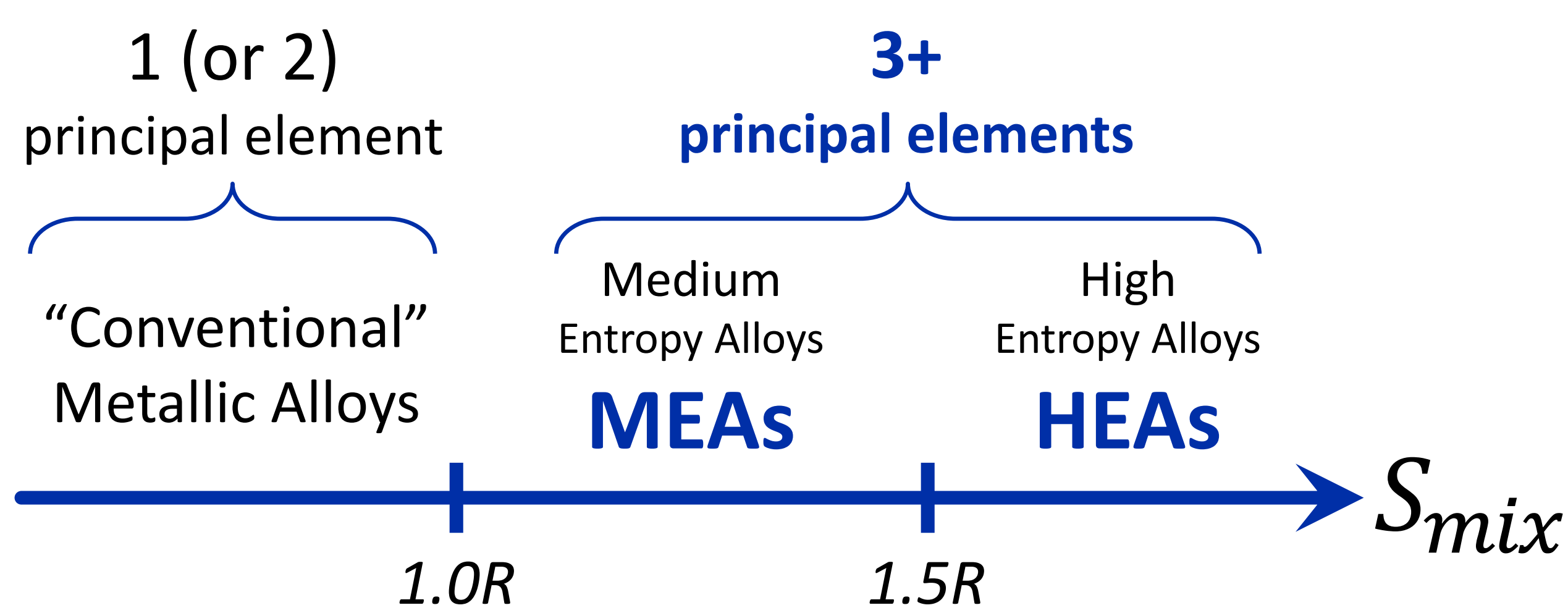
Design of high-C Cr-Co-Ni medium entropy alloy for tribological applications

G. Bertoli^a, G.Y. Koga^b, F.C. Puosso^b, A.J. Clarke^c, C.S. Kiminami^b, F.G. Coury^b

^a Federal University of São Carlos, Graduate Program in Materials Science and Engineering, Rod. Washington Luis, km 235, CEP 13565-905 São Carlos, SP, Brazil
^b Federal University of São Carlos, Department of Materials Science and Engineering, Rod. Washington Luis, km 235, CEP 13565-905 São Carlos, SP, Brazil
^c Colorado School of Mines, George S. Ansell Department of Metallurgical and Materials Engineering, 1500 Illinois St., Golden, CO, 80401, USA

Corresponding author:
gustavo.bertoli@estudante.ufscar.br

Background / Motivation



$$\uparrow S \xrightarrow{G = H - TS} \downarrow G \Rightarrow \uparrow \text{Phase Stability}$$

Multiple principal elements [1]:

- ❖ S and H contribute to the phase competition. Depending on the alloys system and processing a single-phase or multi-phase microstructure may form;
- ❖ There are an almost infinite number of possible combinations;
- ❖ It is possible to achieve different combination of properties.

Methods / Results

- ❖ Computational thermodynamic calculations (CALPHAD method) (Fig. 1) was used to predict the effect of C additions in Cr-Co-Ni MEAs (some of the toughest materials ever made [2], promising corrosion resistance [3]);
- ❖ C can be incorporated into the alloy by melting in a graphite crucible, enabling C saturation in the melt (controlled by the casting T, see Fig. 1);
- ❖ In the present work, 24 at% C was incorporated by melting at 1500 °C.

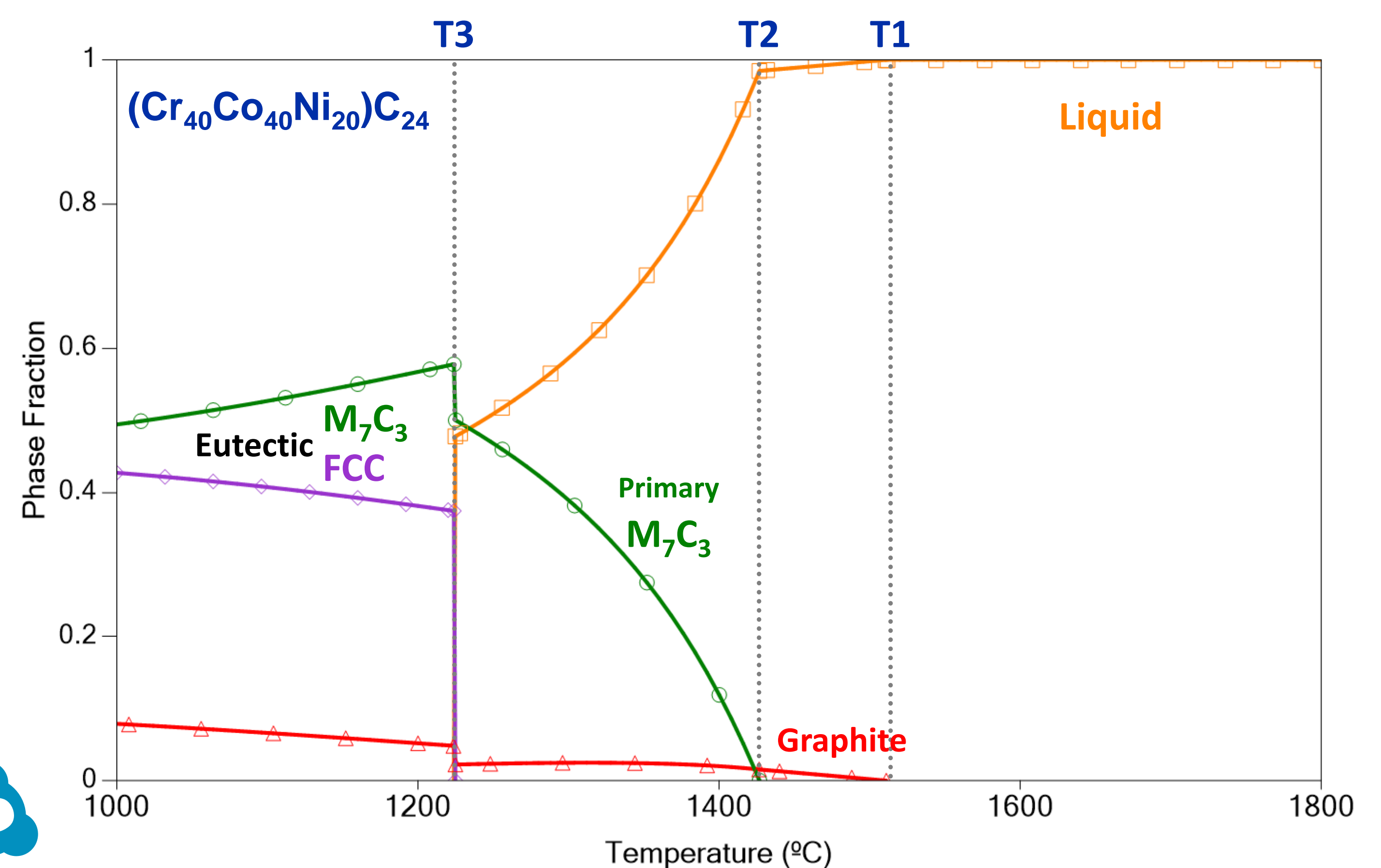
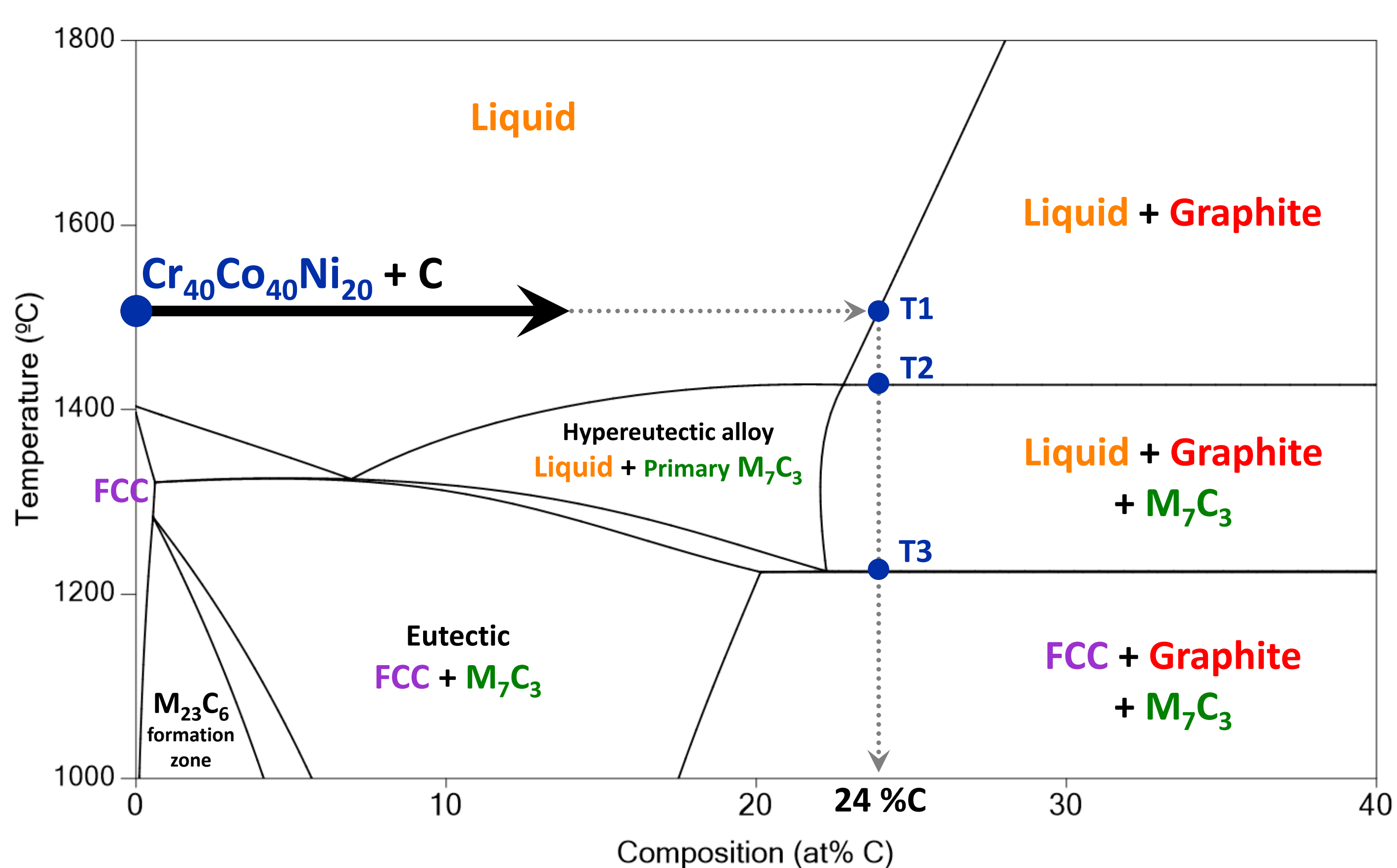


Fig. 1: Computational thermodynamic calculations (CALPHAD method), Pandat® software, PanHEA2020 database. On the left: isopleth Cr₄₀Co₄₀Ni₂₀ – C (at%), the arrows highlight that the casting temperature control the C saturation in the melt. On the right: equilibrium phases of (Cr₄₀Co₄₀Ni₂₀)C₂₄ at different temperatures. T1 = graphite formation; T2 = primary M₇C₃ carbide formation; and T3 = the remaining liquid solidifies into a eutectic (FCC + M₇C₃). In lower temperatures, M₇C₃ carbide may be completely or partially converted to M₃C₂ carbide, if kinetic conditions allow it.

- ❖ A (Cr₄₀Co₄₀Ni₂₀)C₂₄ MEA was produced, as described above, and characterized by XRD and SEM-EDS (Fig. 2).

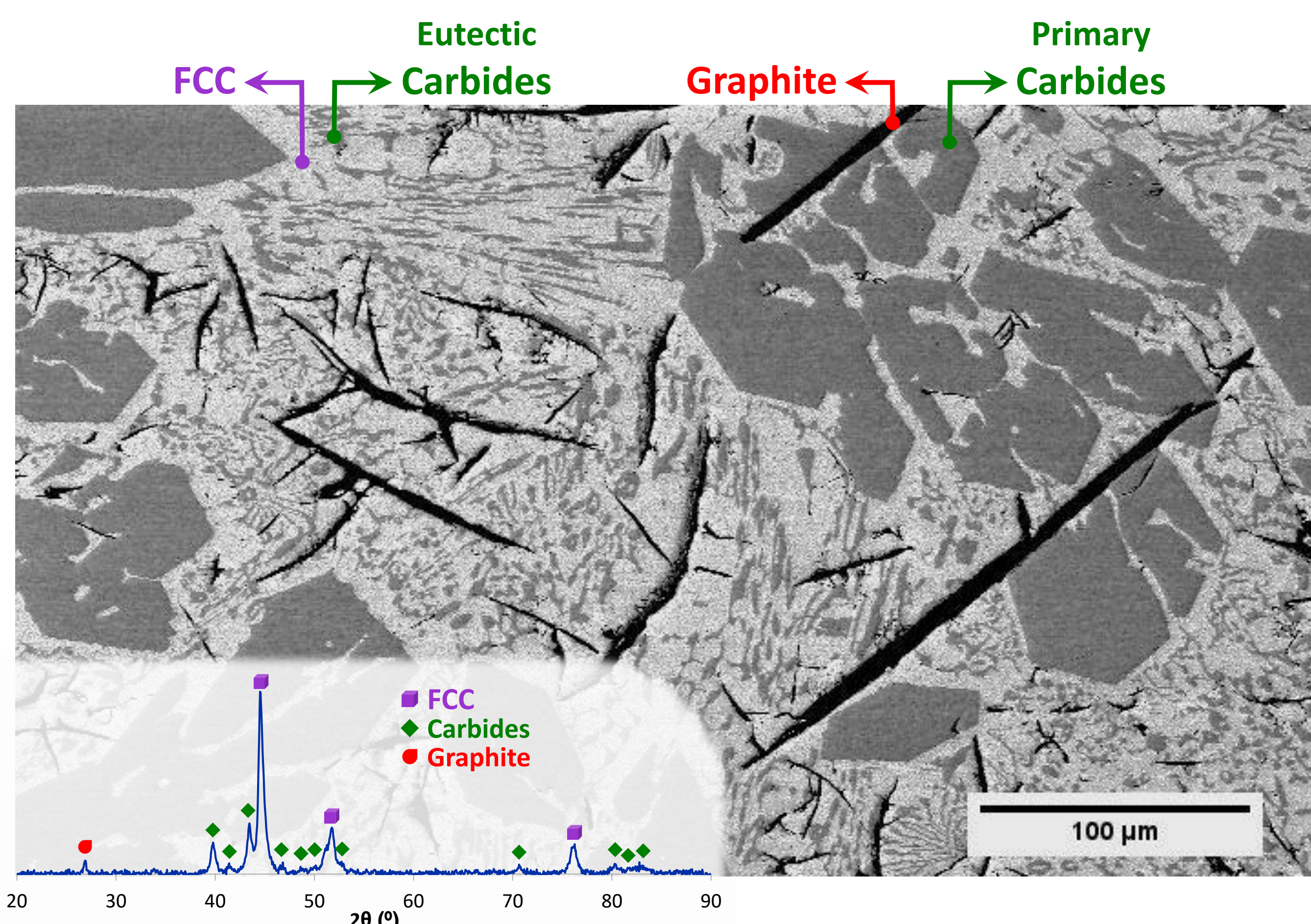


Fig. 2: SEM-BSE image and X-ray diffraction pattern of as-cast (Cr₄₀Co₄₀Ni₂₀)C₂₄ MEA.

Conclusions

- ❖ A MEA with promising microstructure for resisting wear was developed;
- ❖ In good agreement with CALPHAD, (Cr₄₀Co₄₀Ni₂₀)C₂₄ displayed:
 - ❖ Graphite (lubricating component)
 - ❖ Primary carbides (hard component)
 - ❖ Eutectic matrix of carbides and FCC phase (tough component)

Future Work

- ❖ Refine size, morphology and distribution of carbides ⇒ treatment by electric current [4] and/or rapid cooling;
- ❖ Increase graphite fraction ⇒ solid-state electropulsing [5];
- ❖ Wear testing to confirm expected good behavior.

References

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